# M0011244: Table of Atomic weights determined by Berzelius, 19th century

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### BERZELIUS'S ATOMIC WEIGHT TABLES

1	2	3	4	.5	6	7	8
Ť1	1814		1818		1826		
Element						-6	
O S P	cos cos	16	CO2 CO3	16	CO2 CO3	16.03	60 60
5	SO <sup>2</sup> , SO <sup>3</sup>	32.16	SO <sup>2</sup> , SO <sup>3</sup>	32.19	SO <sup>2</sup> , SO <sup>3</sup>	32.24	SO <sub>2</sub> , SO <sub>3</sub>
P	$P^2O^3$ , $P^2O^5$	26.80	PO³, PO⁵	62.77	P <sup>2</sup> O <sup>3</sup> , P <sup>2</sup> O <sup>5</sup>	31.43	$P_2O_3$ , $P_2O_5$
Cl C N	00 000	(35.16)	CO CO2	(35.41)	Cl <sup>2</sup> O <sup>5</sup>	35.47	60.60
C	CO, CO <sup>2</sup>	11.986	CO, CO <sup>2</sup>	12.05	CO, CO <sup>2</sup>	12.25	CO, CO,
N	****	(14.36)	7700	(14.05)	N <sup>2</sup> O, NO	14.19	N <sub>2</sub> O, NO
H	H <sup>2</sup> O	1.062	H <sup>2</sup> O	0.9948	H <sup>2</sup> O	I	$H_2O$
As	AsO <sup>3</sup> , AsO <sup>5</sup>	134.38	AsO3, AsO5	150.52	$As^2O^3$ , $As^2O^5$	75.33	$As_2O_3$ , $As_2O_5$
Cr	CrO3, CrO6	113.29	CrO3, CrO6	112.58	Cr <sup>2</sup> O <sup>3</sup> , CrO <sup>3</sup>	56.38	$Cr_2O_3$ , $CrO_3$
Si	SiO <sup>3</sup>	48.696	SiO <sup>3</sup>	47.43	SiO <sup>3</sup>	44.44	SiO <sub>2</sub>
$_{\mathrm{Hg}}$	HgO, HgO <sup>2</sup>	405.06	HgO, HgO <sup>2</sup>	405.06	Hg²O, HgO	202.86	Hg <sub>2</sub> O, HgO
Ag	AgO <sup>2</sup>	430.107	$AgO^2$	432.51	AgO	216.6	Ag <sub>2</sub> O
Cu	CuO, CuO <sup>2</sup>	129.03	CuO, CuO <sup>2</sup>	126.62	Cu <sup>2</sup> O, CuO	63.42	Cu <sub>2</sub> O, CuO
Bi	BiO <sup>2</sup>	283.84	$BiO^2$	283.81	Bi <sup>2</sup> O <sup>3</sup>	213.22	$Bi_2O_3$
Pb	PbO <sup>2</sup> , PbO <sup>3</sup>	415.58	PbO <sup>2</sup> , PbO <sup>3</sup>	414.24	PbO, Pb <sup>2</sup> O <sup>3</sup>	207.46	PbO, Pb <sub>2</sub> O <sub>3</sub>
Sn	SnO2, SnO4	235.29	SnO2, SnO4	235.3	SnO, SnO <sup>2</sup>	117.84	SnO, SnO,
Fe	FeO <sup>2</sup> , FeO <sup>3</sup>	110.98	FeO <sup>2</sup> , FeO <sup>3</sup>	108.55	FeO, Fe <sup>2</sup> O <sup>3</sup>	54.36	FeO, Fe <sub>2</sub> O <sub>3</sub>
Zn	ZnO <sup>2</sup>	129.03	ZnO <sup>2</sup>	129.03	ZnO	64.62	ZnO
Mn	MnO <sup>2</sup> , MnO <sup>3</sup>	113.85	MnO <sup>2</sup> , MnO <sup>3</sup>	113.85	MnO, Mn2O3	55.43	MnO, Mn <sub>2</sub> O <sub>3</sub>
Al	AlO <sup>3</sup>	54.88	AlO <sup>3</sup>	54.77	Al <sup>2</sup> O <sup>3</sup>	27.43	$Al_2O_3$
Mg	MgO <sup>2</sup>	50.47	MgO <sup>2</sup>	50.68	MgO	25.38	MgO
Ca	CaO <sup>2</sup>	81.63	CaO <sup>2</sup>	81.93	CaO	41.03	CaO
Na	NaO <sup>2</sup>	92.69	NaO <sup>2</sup>	93.09	NaO	46.62	Na <sub>2</sub> O
K	KO <sup>2</sup>	156.48	KO <sup>2</sup>	156.77.	KO	78.51	$K_2O$

In this table, columns 2, 4 and 6 give the formulae of the oxides assumed by Berzelius, column 8 the modern formulae of the oxides; columns 3 and 5 give the atomic weights recalculated from Berzelius's values, referred to oxygen=100, to oxygen=16; column 7 gives Berzelius's values on his alternative scale of hydrogen=1.