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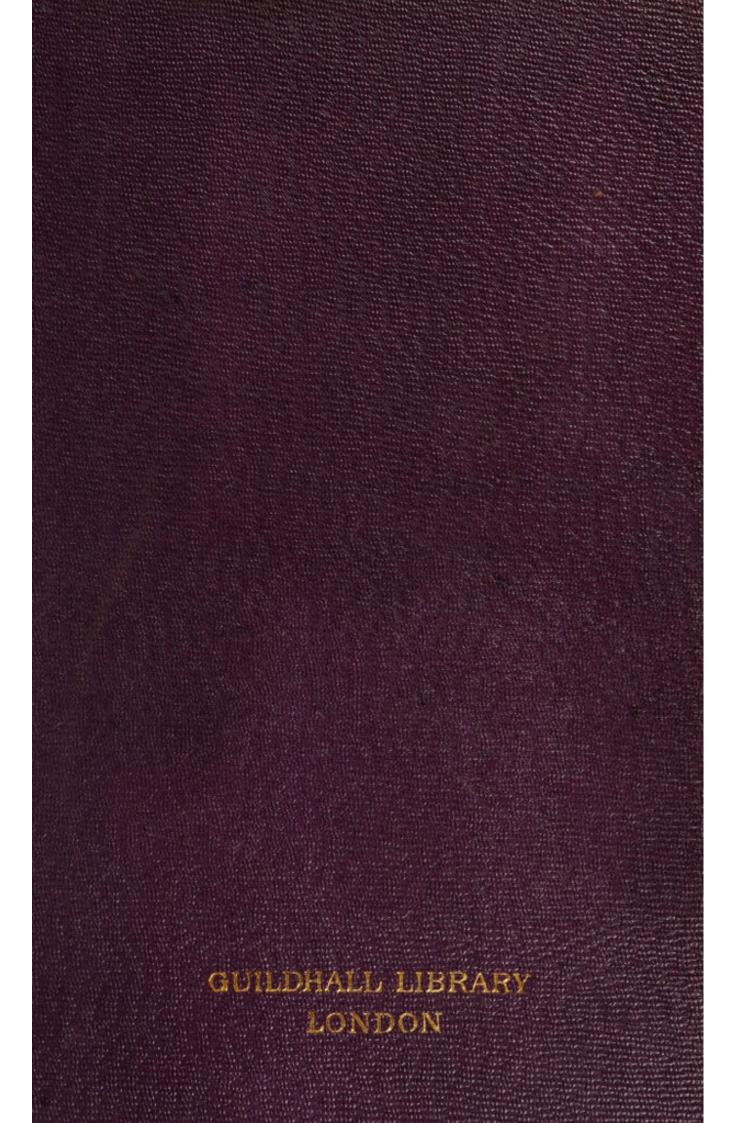
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AN

ESSAY

ON THE

MEDICINAL PROPERTIES

OF

FACTITIOUS AIRS.

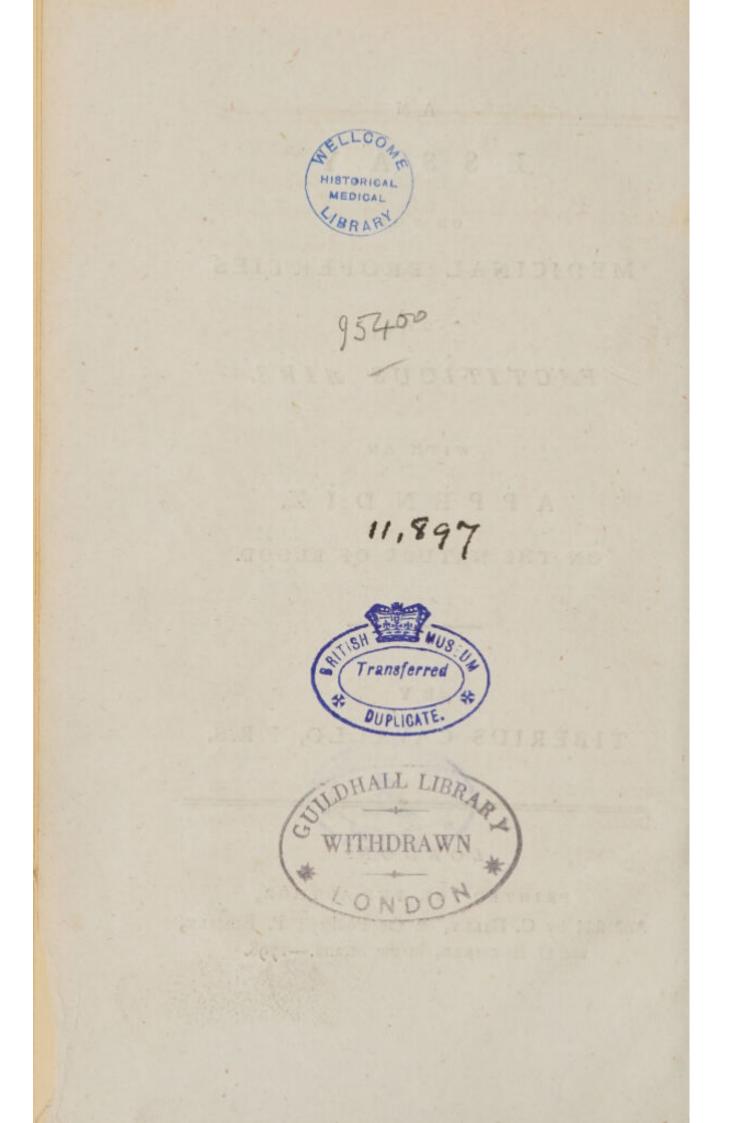
WITH AN

APPENDIX,

BY

TIBERIUS CAVALLO, F.R.S.

PRINTED FOR THE AUTHOR, And fold by C. DILLY, in the Poultry; P. ELMSLY, and D. BREMNER, in the Strand.-1798.



LONDON LONDON

PREFACE.

T is not quite forty years fince the artificial aerial fluids began to be adminiftered as remedies to the human body. The uncertainty, and the errors of the early applications, rendered the progress of the practice flow and doubtful; nor has the experience, or the fuccess of recent and more numerous practitioners, been fufficient to determine the precise power of the aerial fluids, or to diffipate the doubts which are ftill entertained concerning their use.

The defire of extricating the fubject from the conflict of contrary opinions, eftablifhed prejudices, and opposite interests, has induced the Author, perhaps too hastily, to publish the present work, which, in every sense of the word, deserves the epithet of impersect. But he hopes that the importance of an object so highly in-A 2 teresting iv

terefting to the human fpecies, may palliate, if not juftify, the imperfections of the performance, which might, perhaps, have been lefs excufable in other fubjects.

To exhibit a concife view of afcertained facts, to feparate them from fuppofitions and hypothefes, and to point out the ways of inveftigating the farther use of factitious airs, has been the Author's principal aim in the compilation of the prefent Essay.

In the courfe of his inquiries, he has frequently found caufe to admire the ingenuity, the caution, and the perfeverance of feveral gentlemen, who either have adminiftered the aerial fluids, or have otherwife exerted themfelves in the promotion of their ufe. Yet he has taken particular care to avoid paying them any compliments, or even making frequent ufe of their names, left his defire of promoting the fubject fhould be apparently converted into an endeavour of promoting the intereft of certain practitioners.

The first four chapters contain fuch facts as may be of theoretical use in the applications tions of aeriform fluids, and in the inveftigation of their action, independent on medical cafes. The fifth chapter exhibits a concife view of the modern theory of aerial fluids, and of the proceffes that are principally depending thereon, fuch as refpiration, combustion, &c. The fixth and feventh chapters flew the practical application of those fluids by way of remedies to the human body; and this practice is exemplified in the eighth chapter, in which a felect number of authentic cafes is related. The ninth, or last chapter, contains feveral practical remarks, hints, &c. which could not be conveniently inferted in the preceding part of the work.

Laftly, a differtation on the nature and properties of blood has been added by way of Appendix, that fluid being evidently and principally concerned in refpiration, and in the general dependance of the animal exiftence on the aerial fluids.

By the mixt use of the old and the new chemical names in various parts of the work, the author imagines that his meaning may

PREFACE.

may be rendered lefs equivocal, and more generally intelligible; for at a time when the old names are not quite difufed, and the new chemical nomenclature not univerfally underftood, it is difficult to determine whether the greatest number of readers may remain fatisfied with the exclusive use of either.

Wells Street, January the 8th, 1798.

CONTENTS.

CHAP. I. THE principal Properties of those Airs, or permanently elastic Fluids, which have been applied as Remedies to the Human Body page 1 Facts concerning the Respiration II. of Common, and of Oxygen, Airs 22 Phænomena arifing from breath-III. ing other Aerial Fluids, befides the Common and the Oxygen Airs 40 Phænomena arifing from the Ap-IV. plication of the above-mentioned elastic Fluids to other Parts of the Animal Body befides the Lungs 5I V. Theory of the Nature of Aerial Fluids, and of Respiration 58 CHAP.

CONTENTS.

viii

Снар. VI. A general Idea of the Application of Aerial Fluids for the Cure of Diforders incident to the Human Body - page 88 VII. Of the particular Administration of Aerial Fluids in different Diforders - 114 VIII. Medical Cafes in which Aerial Fluids were administered 149 IX. Practical Remarks, Hints, &c. 202

APPENDIX.

On the Nature of Blood - - - 217



AN

ESSAY

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MEDICINAL PROPERTIES

OF

FACTITIOUS AIRS.

CHAPTER I.

The principal Properties of those AIRS, or permanently elastic Fluids, which have been applied as Remedies to the Human Body.

THE philosophical investigations of the two last centuries, and particularly of the prefent age, have ascertained the existence of various elastic fluids, analogous to common air, with respect to elasticity and invisibility; but otherwise effentially different from it, as also different from each other; such are the depblogisticated air, or vital air, or oxygen air; the pblogisticated air, or gas azote; the fixed air, or carbonic acid

gas:

2

gas; the inflammable air, or hydrogen gas; the nitrous gas, &c. But as of all the different airs five only appear to be applicable to the human body, viz. the common, the oxygen, the azotic, the carbonic acid, and the hydrogen airs, we shall not therefore extend our notice to any other fort of elastic fluid; nor shall we deferibe more than the principal properties of those five; viz. such properties only as may be useful to elucidate their action on the human body.

Of the Common, or Atmospherical Air.

THAT invisible elastic fluid, which furrounds the earth, and in which we live, is indispensably necessary to animal life, to combustion, and to other process. No animal can live, nor can any combustible body burn, without air. For either purpose the atmospherical air is more or less useful in proportion to its purity.

WHEN common air is mixed with another particular fort of air, called *nitrous gas*, a diminution of bulk takes place, which is proportionate to the purity of the air; the *



pureft air being diminished most, and vice verfa; so that very impure air fuffers no diminution. Hence the quality or goodnefs of common air may be ascertained by mixing a certain quantity of it with a determinate quantity of nitrous air, and then measuring the diminution of bulk that ensues. The instrument in which this operation for ascertaining the purity of the air is made, has been called an *eudiometer*.

THE purity of common air is not the fame in all places, nor is it conftant in the fame place at all times. The variation in the latter cafe is much more confiderable than in the former; yet, upon the whole, it is not very great. If in the usual state of the atmosphere, and in places that are reckoned healthy, 100 parts or measures of common air be mixed with an equal quantity of nitrous air, their bulk, after the mixture, will be found, instead of 200 parts, to be between 100 and 120, more or lefs, according to the time of the year, fituation of the place, state of the atmosphere, &cc. But B 2

But in caves, mines, crowded rooms, hofpitals, work-fhops, and the like, the air is lefs pure; yet even in this cafe the difference, as indicated by the teft of nitrous air, is but trifling; excepting indeed thofe places in which the communication with the external air is abfolutely or almost entirely interrupted *.

NOTWITHSTANDING the fmall difference which is manifested by this method of trying the purity of common air, it is however evident, from the oppression which is felt in certain instances, and the reviving effect which is experienced in other cases,

• Dr. Prieftley having dined one day in company with eight or ten perfons, in a large and very lofty room, and happening to go out of the room for a fhort time, was, on his return, ftruck with the offenfivenefs of the air, and his curiofity prompted him to afcertain the degree in which the air was injured. On trial he found that 100 parts of that air, with 100 parts of nitrous air, were reduced to 131 parts; whereas the like experiment being performed with the air of a well-ventilated room of the fame houfe, the 200 parts of mixt aerial fluid were reduced to 125 parts.

that

that the human lungs are fenfibly affected by the finalleft differences in the purity of the air. But it is neceffary to remark, that noxious particles are frequently fufpended in common air, which do not alter the effect of nitrous gas upon it, though, at the fame time, they render it very offenfive to animals.

CONSIDERING the variety of vapours, minute bodies, &c. that are continually fcattered through, and float in, the air, the atmosphere must be looked upon as being always contaminated by the prefence of minute animal, vegetable, and even mineral, particles ;—of bodies, in short, that are foreign to, or unconnected with, the nature of air.

THE quality of common air is not altered by merely heating or cooling *, or by

* Every degree of Fahrenheit's thermometer rarifies or increases the bulk of common air, by about $\frac{1}{473}$ part of the whole.

B 3

keeping,

keeping, or by being for a time loaded with the vapour of water, nor by rarefaction or condenfation; but it is contaminated principally by refpiration, by combustion, by the fermentation and putrefaction of animal and vegetable bodies, by the calcination of metallic substances, by the prefence of vegetables when they are not under the influence of the fun's rays, and by the admixture of every other gas, or permanently elastic fluid, except the oxygen.

WHEN the common air is completely contaminated, or rendered unfit for combuftion and refpiration, it is (according to the prefent nomenclature) called gas azote, whereas it was formerly called *phlogiflicated* air.

VITIATED air is capable of being meliorated various ways, and the methods of effecting it may be diftinguished into natural and artificial. The natural means are far from being known to their full extent; but the vegetation of plants, in certain circumftances,

cumftances, and the contact of water, as in rains, dews, &c. are two very powerful correctors of contaminated air. Whether those and other natural means, are fufficient to preferve the atmospherical air nearly in the fame degree of purity, or whether that degree be constantly undergoing a gradual change, so as to render the air either continually better or worse, is a very interesting question, but it can only be answered by the philosophers of future generations. For my part, I am led to suffect that the purity of the air is subject to a periodical fluctuation, or to an alternate increase and decrease for an uncertain number of years.

VENTILATION, and whatever promotes ventilation, does nothing more than remove vitiated air from those places in which it is generated, and disperse it through the atmosphere.

THE artificial methods of correcting vitiated air are few and imperfect. Ventilation, by means of bellows and other ma-B 4 chines,

cells, then reft alone, or at molt a flight

chines, is the most efficacious, and at the fame time the most practicable way of improving the air of hospitals, fick rooms, prifons, &c. viz. by removing the vitiated, and introducing a fresh current of purer air. A fire purifies the air of certain places, only by promoting the ventilation or circulation, and by drying the moisture; but the air which has passed through the fire must not remain in those places, otherwise the injury will be infinitely greater than the advantage. It has been confidently afferted, and denied, but it is now with limitation believed, that the vapours of nitrous, or of marine acid, will diveft common air of the poifonous effluvia of contagious diforders; hence the vapours of those acids are now frequently dispersed through the air of hospitals, crowded ships, &c. When noxious vapours are merely fuspended in the air, as it often takes place in feveral natural and artificial proceffes, then reft alone, or at most a flight agitation in water will be fufficient to purify the air. By the admixture of oxygen gas, a quantity of common air may be improved

proved to almost any degree; but the method is difficult and expensive; hence it can only be used with limitation in certain cases, which will be specified in the sequel.

Of the Dephlogisticated, or Oxygen Air.

THE oxygen is a fort of aerial fluid, that poffeffes the ufeful properties of common air in a much more eminent degree; viz. it affifts combustion and animal respiration for a much longer time, and with superior energy. When a lighted candle is introduced into a vessel full of oxygen air, its flame becomes larger, and surprisingly brighter than in common air. Its heat is likewise increased to a very great degree.

THIS air is not found pure or unmixed in nature, but it may be extracted from various fubftances by means of artificial proceffes. The leaves of plants, indeed, yield a confiderable quantity of it whilft they are exposed to the light of the fun; but the oxygen air which is thus produced, mixes with,

with, and is difperfed through the circumambient air as foon as it is generated; fo that the air contiguous to the plants is feldom fenfibly better than that of the neighbouring country.

By the addition of nitrous air the oxygen is diminished much more than common air. When 100 parts of good oxygen air are mixed with an equal quantity, viz. 100 parts, of nitrous air, their joined bulk will not exceed 50 parts, the other 150 parts having loft the aerial form. Nor is this the utmost degree of diminution that can be produced; for if 100 parts of the pureft oxygen that can be procured, be mixed with twice its quantity of nitrous gas, almost the whole bulk of elaftic fluid will difappear; at most, the refiduum will not exceed five or fix parts. By putting a lighted candle into a vefiel full of any species of respirable air, and observing the effect of that air on the flame, one may estimate the degree of its purity near enough for feveral purpoies.

THE

THE following are the principal methods of procuring this air. The green leaves of vegetables, when placed in a glafs receiver full of, and inverted in fpring water, and thus exposed to the direct rays of the fun, vield a confiderable quantity of oxygen air, which afcends to the upper part of the receiver, and may be eafily removed from it for use. One hundred leaves of Indian crefs, nasturtium Indicum, in a gallon of fpring water, will, in about three hours exposure to the fun, yield about ten cubic inches of oxygen air, not indeed quite pure, but yet vaftly better than common air. I do not know of any plant whole leaves produce this fort of air in greater abundance.

THERE are feveral fubftances from which oxygen air may be extracted by the action of heat or of acids; but those which upon the whole yield it in greatest plenty, and are fit to be used, are faltpetre or nitre, and the metallic calces.

ONE ONE

ONE ounce of nitre, by remaining exposed to a full red, or rather a white, heat in an earthen retort for about four or five hours, will give between 700 and 800 cubic inches of oxygen air, which is not equally good in every period of the process, but at a medium it is fuch that if Ico parts of it be mixed with 150 parts of nitrous air, the whole will be reduced to about 100 parts. This oxygen gas contains a quantity of nitrous acid in the form of vapour, and therefore, when it is to be used for respiration, the acid vapour must be previously separated from it, which may be done by agitating the air in an alkaline lixivium, or at leaft in lime water.

IF an ounce of *mercurius precipitatus per* f fe be exposed to a barely red heat in a glass vessel, it will yield at least 66 cubic inches of very good oxygen air.

RED precipitate of mercury, when treated in the like manner, does also yield a confiderable quantity of this fort of air.

THE

THE action of a red heat alone, or of vitriolic acid and a moderate degree of heat, expels from minium, or red lead, about ten or twelve times its bulk of oxygen, mixed with about one third of carbonic acid, air; the latter of which may be feparated from the former by wafhing in lime water. If the red lead be previoufly moiftened with nitrous acid, and then ftrong vitriolic acid be poured upon it, a greater quantity of oxygen gas will be obtained in a fhorter time, and even without the application of heat.

THIS fort of elaftic fluid may be alfo obtained in fmall quantities from feveral other metallic calces; but the mineral called manganefe, gives a great quantity of it in an eafy manner; it is at the fame time a very cheap article, fo that, upon the whole, manganefe is at prefent the most eligible fubftance for the purpose of procuring oxygen air.

MANGANESE

MANGANESE is not always of the fame quality, and of course the elastic fluid, which is extracted from a given quantity of it, is variable both in quantity and quality. One ounce of good manganese, free from large calcareous particles, will, in a red heat, yield more than two pints and a half wine measure, or about eighty cubic inches of elastic fluid, about one tenth of which is carbonic acid, and the reft is oxygen gas. By means of vitriolic acid and a gentle heat, about an equal quantity of elastic fluid, nearly of the fame quality, may be extracted from manganese; but in this case fome acid vapours come over with it, which must be carefully washed off in order to render the oxygen air fit for refpiration.

THE oxygen air is diminished to a much greater degree than common air, not only by the admixture of nitrous gas, but alfo by all the proceffes which are known to diminish atmospherical air; and indeed fometimes the whole quantity of oxygen air is abforbed or deprived of its aerial form.

form. Thus, by refpiration, this air will be entirely abforbed, excepting indeed that part which is converted into fixed air.

Of Fixed Air, or the Carbonic Acid Gas.

THIS gas, which is the heavieft of the aerial fluids, is of an acid nature, but it reddens only light blue vegetable colours; it cryftallizes with fixed alkali, and is poffeffed of a confiderable antifeptic power. It is abfolutely incapable of affifting refpiration and combuftion *; nor is it diminifhed by nitrous air. It combines with various fubftances, and is readily abforbed by water, to which it communicates an acidolous tafte and fparkling property. It is alfo abforbed by, and precipitates the calcareous earth in lime water, but when in greater quantity, it again diffolves the cal-

* Even a mixture of one part of fixed, and eight parts of common, air will extinguish the flame of a candle. See Cavendish's paper, in the Phil. Trans. for 1766.

careous

careous earth in the water. It also diffolves iron in water, and keeps it diffolved therein.

THIS elastic fluid is produced in a great many natural as well as artificial proceffes. It is frequently found in fubterranean places, especially in the vicinity of volcanos, and hot fprings, where, on account of its great fpecific gravity, it remains for a confiderable time, unlefs it be removed by means of ventilation, &c. It is contained more or lefs in almost all the mineral waters; it is abundantly produced in vinous fermentation. Respiration, combustion, and some other proceffes, do likewife produce a certain quantity of carbonic acid gas. It is contained in a variety of mineral fubstances, and particularly in calcareous earth, as chalk, marble, &c. from which fubftances a great quantity of that gas may be extracted by means of heat or of acids *; the calcareous

* The fixed air which is contained in white marble amounts to about one third part of its weight.

bodies

bodies remaining, after the loss of that gas, in a cauftic or acrid ftate; fo that the calcareous earth, by being in a mild ftate whilft it contains that elastic fluid, may be juftly confidered as a neutral falt, confisting of an earthy basis and an aerial acid.

Of the Inflammable Air, or Hydrogen Gas.

INFLAM MABLE Air is the lighteft of the elaftic fluids. It is, as its name imports, a combuftible fluid, which, like other combuftible fubftances, may be inflamed by the contact of an ignited body, and will burn only when in contact with common, or oxygen, air.

THOUGH this fort of elaftic fluid be abfolutely unfit for refpiration, it is not, however, fo noxious as the carbonic acid. It fuffers no diminution when mixed with nitrous air. Its bulk is increafed of $\frac{1}{200}$ part of the whole by each degree of Fahrenheit's thermometer.

C

HYDROGEN

HYDROGEN gas is abundantly produced during the diffolution of animal and vegetable bodies ; hence it is often found to come out of ponds, burying grounds, and other places that contain animal and vegetable matter in a state of decay. This gas does alfo frequently come out of the earth, where inflammable minerals are contained, as in coal mines, and mines of fulphureous metallic ores. But in all those cafes the inflammable gas, by being much lighter than common air, afcends to the upper regions of the atmosphere as foon as it is produced, and leaves the air, adjacent to the ground, very little, if at all, infected, excepting in vaulted fubterranean places, where, indeed, befides its infecting the common air, it fometimes takes fire and explodes, to the great danger of the miners.

By means of heat, or of acids, this gas may be obtained from almost all forts of bodies, whether they be vegetable, animal, or mineral. But the greatest quantity of it may be extracted from iron, or from zinc, 3 by

by means of diluted vitriolic acid; and likewife from iron, by paffing the fteam of boiling water over its furface, the iron being red hot. When charcoal is treated in the laft-mentioned manner, it likewife yields abundance of a peculiar fort of inflammable gas, called *bydrocarbonate*, which however is mixed with a confiderable proportion of carbonic acid gas.

HYDROGEN gas has the property of diffolving and holding in fufpenfion, for a longer or fhorter time, a variety of fubftances, fuch as iron, charcoal, fulphur, phofphorus, &cc. from which circumftance it acquires a variety of particular names as well as properties. Hence we hear of the *phofphoric hydrogen gas*, or *phofphuret* of *hydrogen*; of the *fulphuric hydrogen gas*, or *fulphuret* of *hydrogen*, &c.; hence alfo we find that the hydrogen gas is not always of the fame fpecific gravity, nor has it always the fame fmell.

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IT

IT has been obferved, that the hydrogen gas fometimes lofes its inflammability, and degenerates into azotic air. This change happens more frequently when the hydrogen gas is mixed with common air. The caufe of this phenomenon has not yet been fully afcertained.

For the fake both of brevity and of perfpicuity I have omitted to mention the fpecific gravities of the abovementioned elaftic fluids in the preceding pages, and fhall add them all together in the following table, which contains their fpecific gravities as well as the abfolute weight of a cubic inch of each elaftic fluid.

THE gravity of common air is confiderably affected by the variations of heat, wind, purity, &cc. fo that its fpecific gravity, compared with that of water, has fometimes been known to be as one to fix hundred and fix, and at other times as one to nine hundred and thirty-one*. The gra-

* Mufichenbrock, tom. II. §. 2059.

vities

vities of other elastic fluids are likewise fubject to the fame variations. But the following table has been calculated for a mean and temperate state of the air, viz. when its gravity is to that of water, as one to eight hundred, when the height of the barometer is 29,85 inches, and when Fahrenheit's thermometer is at 55°.

	· Specifi wities.			4	Absolute Weight of a Cubic Inch of each in Troy Grains.
Common air I		-	-	-	0,31648
Azotic gas, or com- mon air completely diminished by ni- trous gas	948		-	2	0,3
Oxygen air 1,0	0427		-	-	0,33
Carbonic acid gas 1,		-	-	-	0,475
The lightest hydro- }0,0	833		-	-	0,02637

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CHAPTER II.

Facts concerning the Respiration of Common, and of Oxygen, Air.

HAT the whole mass of air which furrounds the earth is called the atmofphere, that this atmosphere extends to a confiderable but unknown distance above the furface of the earth, that it decreafes in denfity as it recedes from the earth, that its motion is called wind, that it acts upon all other bodies by its temperature, its weight, and other qualities, that it abforbs vapours, or keeps them fufpended, and fuch other like properties of the atmospherical fluid, have been rendered fo common by the prefent flate of knowledge and of polite education, as not to demand any particular elucidation in this work; we shall, therefore, proceed immediately to enumerate the phenomena which have been afcertained relatively to the respiration of common air, upon

upon which, as upon a folid bafis, we may afterwards eftablish the theory and the practice of applying the factitious airs to the human lungs.

A CERTAIN quantity of air will fupport animal life, or combustion, but for a limited time. If a lighted middle-fized tallow candle be confined in a vessel that holds one gallon of common air, the flame will, in a few seconds of time, begin to grow dim, and it will be extinguished at the end of about one minute; after this, if another lighted candle be introduced into the same vessel, its flame will be extinguished immediately.

IF a man be confined in a veffel that holds ten gallons of common air, he will begin to feel an opprefiion, and a difficulty of refpiration, at the end of eight or ten minutes; this difficulty will gradually increafe, and at the end of about half an hour, reckoning from the beginning of his confinement, he will lofe his fenfation, C_4 his

his motion, and, prefently after, his life. The fame effect will take place with other animals, in a longer or fhorter time, proportionably to their fize, nature, and difpofition of body.

In the ufual way of breathing, when refpiration is performed in a natural and eafy manner, a full grown perfon confumes about five cubic feet, or thirty gallons and a half, beer meafure, of common air per hour.

A MAN generally performs one infpiration and one expiration for every feven or eight pulfations of his arteries; therefore reckoning, at a mean, eighty pulfations per minute, a perfon may be faid to perform eleven or twelve infpirations, and as many expirations, in a minute. But refpiration is quickened by various caufes; viz. by the quickening of the pulfe, by agitation of the body, by heat, by furprife, by difeafes of the lungs, by a rarefied atmosphere, and by impure air. Thus when a man is confined

in

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in a certain quantity of air, his refpiration is quickened in proportion as that quantity of air becomes contaminated; he alfo takes in and expels a greater quantity of air at a time, in order to compenfate for the want of purity. The fame quickening of refpiration takes place on high mountains, where the air is more rare than on the level of the fea.

AT a medium, about 30 cubic inches of air are taken in at one infpiration, and a quantity, nearly equal to it, is thrown out at every expiration; but a great deal of air remains in the lungs, wind-pipe, and mouth; fo that by a violent expiration after a natural infpiration, a double quantity, viz. fixty cubic inches of air, may be expelled, and even then fome air neceffarily remains in the lungs, wind-pipe, and mouth.

THE air which has ferved for one infpiration is not thereby completely contaminated, but it may be refpired again and again. 350 cubic inches of common air were

were confined in a bladder that was furnifhed with a wooden tube; this tube was applied to the mouth of a healthy middleaged man, who, ftopping his noftrils, endeavoured to breathe that quantity of air as long as he poffibly could. After having performed forty infpirations, his ftrength began to fail, and he was obliged to defift.

OLD perfons, people of a bad habit of body, or labouring under difeafes, and fuch as eat and drink immoderately, will contaminate the air much fafter than the healthy, the moderate, and the young.

It has been afferted, that fome human beings can live with a much fimaller quantity of air than has been mentioned above, and that divers have fometimes been known to remain under water ten or fifteen minutes, and even a longer time *. It has

* See Beckman's Hiftory of Inventions, article Diving-Bell; and Gmelin's Reife Durch Rufsland, II, p. 199.

been

been likewife discussed, whether fuch divers were enabled to remain fo long under water, and without air, by any particular conformation of the internal parts of their bodies, or from long practice and particular artifices. But there are ftrong reafons for difcrediting the above-mentioned affertions. The inaccurate way of reckoning the time in fuch cafes, and the common fondnefs for the marvellous, are in general the foundation of fuch extraordinary reports. Upon the whole, it will be found, that the most experienced diver can hardly remain without air longer than a minute and a half; but most perfons will begin to feel a degree of uneafinefs in about half a minute's time.

THE air, which has been completely contaminated by refpiration, is deleterious to other animals, though fmall and young animals will live a fhort time in it : it extinguishes flame, is diminished very little by nitrous air, contains about one-thirtieth of carbonic acid gas, and is contracted in bulk,

bulk, the diminution being various, but hardly ever exceeding one-fifth part of the original quantity.

THE deleterious quality of the air that has been contaminated by refpiration is in great measure owing to the carbonic acid gas, which is formed in the process of refpiration; and it is for this reafon that, when an animal is confined in a veffel full of respirable air, he will be able to live longer in it when fome lime-water is placed in the veffel, than otherwife; becaufe the lime-water abforbs the carbonic acid gas as foon as it is generated. An animal will likewife live longer in a veffel full of air, when he is placed at the upper than at the lower part of the veffel; because in the former cafe the carbonic acid gas will, on account of its great specific gravity, fall towards the lower part of the veffel, and will, of course, be at a distance from the body of the animal.

THE

THE refpiration of oxygen air is attended with peculiar phenomena. The oxygen, like the common, air, is diminished by refpiration; but the diminution proceeds to a much greater degree, for almost the whole quantity of elastic fluid will be reduced to a fmall proportion of carbonic acid gas; and if the experiment be performed on limewater, the whole quantity of oxygen air will difappear. By repeatedly performing the experiment in this manner, it has been found that a healthy middle-aged man will entirely confume two gallons of pure oxygen air in about five minutes time *. But in this cafe the oxygen air is confumed fafter than is neceffary for the ufual fupport of life; and, in fact, if the fame quantity of it be mixed with an equal quantity of azotic

* Amongft the various ways of producing oxygen air, it frequently happens, as we have already hinted, that acid vapours, or other volatile fubftances, are mixed with it; and in that cafe the animal which is confined in it may feel an oppreffion on his lungs, or he may even be fuffocated, when, by the teft of nitrous gas, that air will actually appear to be much better than common air.

gas, it will then last as long again, viz. about ten minutes. It is therefore evident, that as the azotic gas is absolutely incapable of affisting respiration, the mixing of it with the oxygen air produces no other effect than that of presenting a smaller quantity of oxygen to the surface of the lungs in each inspiration. It is for the same reason that oxygen air is confumed faster, and that common air is vitiated soner, when respired under an increased, and flower when respired under a diminished atmospherical pressure.

THE air which is expelled from the lungs after every infpiration, whether it be oxygen or atmospherical air, contains, besides the portion of carbonic acid gas, a confiderable quantity of aqueous vapour, which, in cold weather, is manifested by its condensation as soon as it comes out of the mouth; for air can hold in solution a much greater quantity of water when hot than when cold.

THE

THE breathing of pure oxygen air is generally, if not always, attended with an increafe of heat, efpecially about the lungs, and a quickening of the pulfe; but on fome individuals those effects are increased to fuch a degree as to produce fevers, inflammation of the lungs, and even confumptions, whilst with other individuals they are moderate, temporary, and even falutary. But I shall endeavour to impress the reader's mind with a clearer idea of those phenomena, by subjoining a short account of the principal experiments that have been performed relatively to this interesting part of our subject.

Dr. PRIESTLEY is, as far as I know, the first perfon who had the curiofity of breathing oxygen air. "I have," *fays he*, " gra-" tified that curiofity, by breathing it, " drawing it through a glass-fyphon, and " by this means I reduced a large jar full " of it to the standard of common air. The " feeling of it to my lungs, was not fen-" fibly different from that of common air, " but

" But I fancied that my breaft felt peculi-" arly light and eafy for fome time after-" wards *."

THE following experiment was performed, with great accuracy, before a philosophical fociety of gentlemen, at Dr. Higgin's house, in the year 1794 .- Nineteen pints of pure oxygen gas were put into a receiver which flood inverted in limewater. A tube proceeded from the upper part of the receiver to the mouth of the experimenter, a healthy man of about twenty-two years of age, who, after having accurately stopped his nostrils, and having expired as much air from his lungs as he poffibly could in a bent pofture of the body, took the end of the tube in his mouth, and began to breathe the oxygen air in a natural and flow manner, during which the receiver was permitted to play freely up and down in the lime-water, in order to prevent any increase or decrease of

> * Experiments on Air, &c: vol. ii. p. 102. pressure

pressure on the lungs. An affistant was employed to keep the lime-water in continual agitation, in order to promote the abforption of the carbonic acid air that was formed in the course of the experiment. The bulk of oxygen air was vifibly diminished at every inspiration, and the limewater became turbid. The whole of the oxygen air was confumed in fix minutes time, and the experimenter ftopped only when the lime-water came to his mouth. " During the refpiration his pulfe (which, " previous to the experiment, was only " fixty-four) quickened to ninety beats in a " minute, and was confiderably increafed " in fulnels and ftrength; but he felt no " inconvenience whatever.

" THE veffel being immediately charged again with nineteen pints of gas, he refpired thefe alfo, and confumed them entirely in fix minutes. His pulfe was increafed to 120 beats in a minute, and was vigorous withal. He felt no inconvenience, but had a fenfe of unufual D "warmth

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" warmth in his lungs. In one hour after the experiment his pulfe returned to fixty-four *."

DR. BEDDOES found the breathing of oxygen air extremely hurtful. "To my own "lungs," *fays be*, " it feels like ardent fpi-" rit applied to the palate; and I have " often thought I could not furvive the in-" fpiration of oxygen air, as it is driven " from manganefe by heat, many mi-" nutes +."

A SINGLE infpiration of oxygen air may be kept in the lungs much longer than an infpiration of common air.

WHEN oxygen air is mixed with common air, and is then breathed in that diluted flate, the lungs are lefs affected with

* Minutes of the Society for Philosophical Experiments and Conversations, page 146.

+ Confiderations on the Medicinal Use of Factitious Airs, vol. i. p. 14.

the

the fenfation of heat, nor is the pulfe quickened fo much as when pure oxygen is ufed; yet in this diluted state the oxygen air has been found beneficial in a variety of cafes, which will be mentioned in the fequel. We shall likewife mention the proportion of the two elaftic fluids, which has been found to answer best for each particular cafe; but in the prefent chapter it will be neceffary to ftate the effect which the breathing, or the action, of oxygen air has been observed to have upon particular parts of the animal body, whence proper conjectures may be formed of its general use in the animal economy, and of its application for the cure or alleviation of particular diforders.

THAT oxygen air is a powerful ftimulus to the lungs, has been evinced by various experiments, but by none better than the following, which has been repeated with equal fuccefs by different perfons :-- Some young rabbits were kept under water till every appearance of life, and even a hope of

D 2

of recovery, had vanished; they were then withdrawn, and oxygen air was forced through the mouths of fome of them into their lungs, whilst a fimilar operation with common air was performed on the others : the latter remained dead, whilft the former recovered. Young dogs and kittens were fubjected to the like experiment, the general refult of which was, that the oxygen air brought them to life where common air proved ineffectual. Animals thus apparently deprived of life have frequently revived by only being placed in a vefiel full of oxygen air, without forcing it into their lungs. From this fact we derive a powerful method of reftoring fufpended animation.

RABBITS, dogs, kittens, and birds, have been often confined in veffels full of oxygen air, and have been fuffered to remain in that quantity of air for various lengths of time. It has been conftantly obferved, that they live longer in that, than in an equal quantity of common air. But whenever the experiment

experiment has been protracted to a certain length, it has almost always been attended with illnefs, with a strong inflammation, and even with death. The diffection of the animals that have been thus *oxygenated*, has principally exhibited the following phænomena:

THE lungs appear of a florid red colour, often marked on the edges with figns of mortification; the heart appears of a florid red colour; the pleura is generally inflamed; the colour of the liver, kidneys, and the blood-veffels of the mefentery, is more inclining to red than is otherwife known to be; their blood coagulates fooner; their mufcles are more vigorous, and fhew figns of ftronger irritability.

ANIMALS that have breathed oxygen air, previoufly to their being immerfed in water, will not die fo foon as those which have breathed common air only. The quantity of purer air, which remains in the lungs of the former, is what in great mea-D 3 fure,

fure, if not entirely, contributes to the prefervation of their lives.

AFTER having defcribed, in the preceding paragraphs, the principal phænomena, which are produced by the refpiration of pure, or nearly pure, oxygen air, it will be hardly neceffary to add, that a mixture of common and oxygen airs, or of azotic and cxygen airs, must produce phænomena analogous to those which have been mentioned above, but nearly proportionate to the quantity of oxygen air which is contained in the mixture. There is, however, a remarkable circumstance, which must be carefully attended to, as being of the utmost consequence in the application of oxygen air to medicinal uses. This circumftance is, that whilft the refpiration of pure oxygen air, or of fuch air as contains a great proportion of oxygen, is attended with inflammation and other bad confequences, the refpiration of common air a little improved by the admixture of a moderate proportion (as for inftance, one-15th,

15th, or even one-20th) of oxygen air, is attended with remarkably falutary effects.

THE inhalation of fuch diluted oxygen air, or we may call it improved atmospherical air, for about 10 or 15 minutes a day, has been found to produce a florid colour in the face, to conciliate fleep, to ftrengthen the organs of digestion, to promote circulation, to strengthen the pulse, &c.

HOWEVER ftrange and unaccountable those effects may at first fight appear, especially to those who are not conversant in philosophical investigations, the facts are certainly true, and a simple reflection may contribute to diffipate the wonder; namely, that people of all descriptions, but especially such as are weak and emaciated, derive a confiderable degree of exhilaration and improvement by a short excurfion out of a town, or of a house, when the superior purity of the country air, D 4 above 40 MEDICINAL PROPERTIES of above that of the town, is not equal to that which is produced by mixing common air with even one-twentieth of its bulk of oxygen air. But we shall have occafion to notice this circumstance again in the fequel.

CHAPTER III.

Phænomena arifing from breathing other aerial Fluids, befides the Common and the Oxygen Airs.

I has already been noticed, that of the various forts of elastic fluids, two only, viz. the common and the oxygen airs, are capable of affisting respiration, from which it may be naturally deduced, that by the admixture of any other gas, either of those two will be rendered less respirable in different degrees. But this diminished goodness of the respirable airs, this mixture of respirable

refpirable and unrefpirable aerial fluids, has proved beneficial in a variety of medical cafes; hence many experiments have been made for the purpofe of afcertaining the mixtures that are more applicable to any particular cafe, and likewife the phænomena which arife from the refpiration of those mixed gass.

I WOULD not be understood to affert or think that the action of the unrefpirable gaffes confifts merely in lowering the quality of common air, or of oxygen air; for that purpose could be more commodiously anfwered by breathing a certain quantity of common air longer than in the ufual way. The fact is, that, befides rendering the common or oxygen air lefs refpirable, each particular gas imparts peculiar and remarkable properties to the mixture, which mixtures are of course applicable to particular cafes. With refpect to those mixtures, much has already been afcertained; but a great deal more remains to be examined and tried under a variety of circumstances, to which

42 MEDICINAL PROPERTIES of which object we must look forward with anxious expectation.

It has been repeatedly afferted and denied, that pure and unmixed hydrogen, or inflammable gas, may be refpired with impunity for a confiderable time, and many experiments are related to prove each of those contradictory affertions. The equivocal refults of those experiments arise from two causes, viz. from the variable nature of the gas, and from the different quantity of common air, which remains in the lungs, mouth, &c. of the animals that are subjected to such experiments.

INFLAMMABLE gas, in the common way of producing it, is feldom very pure; but even when that is the cafe, its coming into contact with the lungs is naturally prevented by the common air, which remains in that organ previoufly to the application of the inflammable gas, the latter being much lighter than the former. By a ftrong expiration in a bent pofture of the body, the common

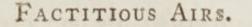
common air may, in great meafure, be expelled; but even in that cafe a certain quantity of it unavoidably remains in the mouth, wind-pipe, &c.

OF the different forts of inflammable gas, that which is obtained by paffing the fteam of water over red hot iron feems to be the least offensive. Next to this is the gas which is obtained from iron and diluted vitriolic acid. The other species are more variable in their quality; but they are all incapable of affifting refpiration; and if a perfon will carefully expel as much air from his lungs as he poffibly can by a forced expiration in a bent pofture, and will then apply his mouth to a vefiel, or to a tube that communicates with a veffel, full of pure inflammable gas, keeping his noftrils stopped at the fame time, he will find, after about three or four infpirations, that the florid colour of his face is vanished, and his ftrength is fo far diminished as to prevent the profecution of the experiment. Having myfelf been more than once witnefs

nefs to this experiment, I have always obferved an evident change of colour in the face of the experimenter after the fecond infpiration.—The gas had been extracted from iron and diluted vitriolic acid.

INFLAMMABLE gas may be rendered lefs noxious by agitation in water.

WHEN this gas (meaning that which is obtained from the vapour of water and red hot iron, or from iron and diluted vitriolic acid) is mixed with about an equal quantity, or even a fmaller proportion of common air, it may then be breathed with fafety for a confiderable time; and it is remarkable, that the lungs are affected by it with a peculiar fenfation of levity. This fingular property has rendered it useful and beneficial in inflammations of the lungs, convulfive coughs, &c. where the object is to diminish the irritability of the parts affected. During this operation the face will be found to grow dark or livid, but the natural colour will be fpeedily recovered by afterwards 3



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afterwards breathing the common air in the ufual way.

THE hydrocarbonate, viz. that fpecies of inflammable gas which is produced by paffing the fteam of water over the furface of red hot charcoal, is much more pernicious to the lungs. Animals will die much fooner in this than in the above-mentioned fpecies of inflammable gas. Sometimes two or three infpirations of pure hydrocarbonate are fufficient to occafion the death of the animal.

THE active quality of this gas is perceivable even when diluted with 20 or 30 times its own bulk of common air. A perfon who breathes it in that diluted ftate for about a quarter of an hour, is generally made fick and vertiginous; feeling at the fame time a fenfation of cold throughout his whole body; his lips become blue, the face livid, and the pulfe feeble, though frequent; but the fenfibility of the lungs is confiderably diminifhed by it, on which account

account it has been administered in various cafes with advantage to the patient. Some patients, after the respiration of this diluted gas, have experienced such levity or infensibility about the region of the lungs, as to remain for a time entirely free from pain.

It is remarkable, that the ficknefs, dizzinefs, or, in fhort, the bad effects of the diluted hydrocarbonate, frequently come on after the operation, and fometimes come on and go off two or three times repeatedly, at the interval of an hour or longer; which fhews that this fort of gas can hardly be administered with too much care and caution.

PURE carbonic acid gas is likewife very pernicious to the lungs. Sometimes one or two infpirations of it have been quite fufficient to kill an animal; and, indeed, animals will die in carbonic acid gas, and likewife in hydrocarbonate, much fooner than if they did not refpire at all, or if they were

were immerfed in water, which proves that fome noxious principle is introduced by those gaffes into the body.

OF the animals, those which have large lungs in proportion to their bulk, and are formed to live in the air, are fooner affected by this gas; thus the birds have in general been found to die foonest in carbonic acid air; the dogs come next, then the cats, then the amphibious animals, and lastly, the infects *. If they are not less too long in this gas, they will, in general, revive, by being removed into the common air. When they die in it they shew no ftruggles. By being frequently exposed to this gas, the animals may be so habituated as not to be killed by it so foon as others that were never exposed to it.

THE following are the appearances which have been more commonly obferved on the diffected bodies of the animals that have been killed by carbonic acid gas.—The lungs are a little collapfed, fhewing a few

> * Bergman de Acido Aereo, fect. 26. inflamed

inflamed places. The right ventricle and right auricle of the heart, the pulmonary artery, the vena cava, the jugulars, and the veffels of the brain, are turgid with blood; but the pulmonary veins, the *aorta*, the left ventricle, and left auricle of the heart, are moftly flaccid. The mufcular fibres of the body are found deprived of irritability, fo that even the heart, extracted whilft the body is ftill warm, fhews no figns of irritability *.

FISHES die in a few minutes time, in water impregnated with carbonic acid gas +.

WHEN this gas is diluted with twice or three times its own bulk of common air, it may then be breathed for a certain time, but not nearly fo long as the mild forts of inflammable air fimilarly diluted.

PURE azotic gas is about as deleterious as the inflammable gas from iron and di-

· Bergman de Acido Aereo, fect. 26.

+ Priestley's Exp. and Observ. vol. ii. sect. 13. Nº 3.

luted

luted vitriolic acid; yet the animals that are confined in it until they appear to be dead, will, on being withdrawn, recover more frequently than those which are confined in the inflammable gas.

THE artificial gaffes have likewife been breathed in combinations of three or four at a time, one of them always being either the common or the oxygen air; but it does not appear that those triple or quadruple mixtures have been tried in a great variety of cases.

In the refpiring of combined gaffes, due regard muft be had to their fpecific gravities, as this circumftance is often the caufe of phænomena that are erroneoufly attributed to other fources. The difference between the fpecific gravities of the common, the oxygen, and the azotic, airs, is indeed trifling; but the inflammable and the carbonic acid gaffes differ confiderably from the reft, and efpecially from each other; the former being a great deal E lighter,

lighter, and the latter much heavier, than common air. If the inflammable, the carbonic acid, and the common or the oxygen, airs, be not well mixed together in a veffel, they will remain feparate for a confiderable time in their refpective places, viz. the carbonic acid air in the loweft part, the common in the middle, and the inflammable in the higheft part of the veffel; but even when they are well mixed together, they always fhew a tendency to feparate, fo that after a fhort interval each of them will be found lefs mixed in its refpective place.

It is hardly neceffary to add, that the fame peculiarity of fituation must also take place within the lungs, and that this is, perhaps, the fole cause which renders the carbonic acid gas more noxious than the inflammable gas, and the heavy fort of inflammable gas, *bydrocarbonate*, more offenfive than the lighter species of it.

CHAPTER IV.

Phænomena arifing from the Application of the abovementioned elastic Fluids to other Parts of the Animal Body besides the Lungs.

T has been found that the pores of the fkin imbibe and expel a finall quantity of air, and it is faid, that in equal times they will abforb a much greater quantity of oxygen, than of common, or of any other, air.

DIFFERENT forts of elaftic fluid were feparately injected into the cellular membrane of animals, through incifions made in the fkin, and the apertures were clofed immediately after. The appearances, as obferved by Dr. Maxwell *, and confirmed by others, were in general as follows:

* See his Thefis, Edinburgh, 1787.

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COMMON

COMMON air fwelled or puffed the animal, rendeted it uneafy for a day or two, after which the fwelling began to decreafe, and vanished entirely at the end of about three weeks.

OXYGEN air fwelled the animal, and rendered it fomewhat uneafy for a fhort time; the uneafinefs, however, foon vanished, the animal became unufually lively, and the fwelling difappeared much fooner than in the cafe where common air had been ufed.

AZOTIC gas fwelled the animal, and rendered it dull, by fuperinducing a fort of ftupor, which, in a few days time, degenerated into convultions, and at laft killed the animal.

CARBONIC acid gas was rapidly abforbed, and feldom produced any flight and temporary uneafinefs.

HYDROGEN gas fwelled the animal, produced heavinefs and fhiverings; but the fwelling

fwelling difappeared fooner than in the cafe of common air.

MR. GIRTANTER is faid to have injected azotic gas into the jugular vein of a dog, in confequence of which the animal died at the end of twenty feconds. On opening its thorax, the pericardium, and the heart, the right auricle and right ventricle were filled with black blood; the left ventricle was of its ordinary dark colour ; the heart and muscles had loft their irritability almost entirely. A fimilar experiment being made with carbonic acid gas, inftead of azotic gas, nearly the fame phænomena took place.

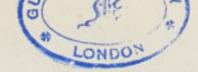
BLOOD recently taken from the veins of an animal, and exposed to the common air, becomes of a bright red colour ; and if expofed to oxygen air, its colour will become still brighter, and the oxygen air will be diminished, and partly converted into carbonic acid air. On the contrary, if the blood thus brightened, or the blood taken from the arteries of an animal, which is well E 3 known

known to be of a florid red, be exposed to any of the unrefpirable gaffes, its colour will be darkened prefently, and a fmall part of the elastic fluid will be abforbed. It is to be remarked, that those effects take place even when an animal membrane, as a piece of bladder, intervenes between the blood and the respirable or unrespirable elastic fluids *. Even the colour of the fleshy parts is made to incline more towards a florid red by the action of oxygen air.

That the oxygen air acts as a ftimulus on other parts of the body, as well as on the lungs, is clearly proved by the following often repeated experiment : A blifter being formed on the hand, or a finger, by the application of the ufual plaifter of cantharides, the fkin was cut off, and the hand was immediately introduced into a veffel full of oxygen air : the confequence was, that the experimenter felt a very acute pain. The hand was then removed into a veffel full of

* Prieftley's Exp. and Obf. vol. III. fect. 5.

carbonic



carbonic acid gas, the action of which removed the pain in a very fhort time. On the hand being exposed to the common air, a degree of pain returned, and on being, as at first, placed in oxygen air, the pain became acute.

THE contact of inflammable gas does neither accelerate nor retard the putrefaction of animal matter.

WHEN the ftream of carbonic acid air is iffuing out of a fmall aperture, as that of the tube of the phial in which this gas is ufually produced from calcareous earth and diluted vitriolic acid, if the mouth or noftrils be prefented to it, they will be affected with a peculiar, and rather pleafing, pungency.

THIS gas is poffeffed of confiderable antifeptic power. And for this property, it is administered to the animal body either internally or externally, and separate parts of animal or vegetable substances may be preferved in it for a confiderable time.

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IT is applied internally to the ftomach, or externally, either in the aerial form, or combined with water and other fubstances. Many fluid or folid bodies derive their antifeptic property from their containing this gas in confiderable quantity; fuch are liquors in a state of vinous fermentation, ripe fruit, certain mineral waters, &c.

FRUIT may be preferved feveral days longer in carbonic acid than in common This is also the cafe with animal air. fluids, or with pieces of meat that are not very large, but they are apt to loofe their flavour. Large pieces of meat are faid to have been preferved for feveral days longer than in the ufual way, by only washing them three or four times a day in water ftrongly impregnated with carbonic acid air.

DISTILLED water, or water that has been deprived of its air by boiling, will, in forty days time, and in a temperate atmofphere, abforb, without needing any agitation,

tion, about $\frac{1}{1+}$ th of its bulk of oxygen air, whereas of common air it will abforb about the half of that quantity, viz. $\frac{1}{2+5}$ th part. It will abforb in a few hours time a quantity of carbonic acid gas little greater than its own bulk; but a cold temperature and an increased atmospherical preffure will enable it to abforb a much greater quantity of that gas. Of inflammable gas it will abforb about as much as it does of common air, viz, $\frac{1}{2+5}$ th part of its bulk.

THIS abforption of elastic fluids by water is much expedited by agitation of the latter in the former.

CHAPTER V.

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Theory of the Nature of Aerial Fluids, and of Respiration.

HAT respiration and life can not be maintained without atmospherical air, is a fact known to the philosophers of the remotest antiquity; but their ideas of the use of air in respiration were vague, and unfupported by experiments. On the revival of learning in Europe, and efpecially after the fixteenth century, the fcientific inquiries of philosophers, physicians, and chemists, ascertained that the air was fubfervient to other natural as well as artificial proceffes, befides refpiration ; and likewife that there actually existed various species of air, fome of which were highly noxious *. The progrefs and diffemination of fcience gradually added new articles to

* See the works of Van Helmont and Dr. Mayow.

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the flock of knowledge relative to the aerial fluids; but the great improvements, the furprifing difcoveries, which have produced a total revolution in this branch of natural philofophy, were referved for the prefent age, and are undoubtedly due to the labours of modern philofophers.

It is entertaining to perufe the works of authors previous to the late difcoveries, and to obferve how near the ideas of fome of them approached the modern theory of refpiration. Hippocrates confidered air as one of the aliments of the body. Dr. Mayow afferts, that fome nitre, or aerial fpirit of nitre, enters the body through the lungs, and furnifhes the animal fpirits at the fame time that it communicates heat to the blood*.

DR. WHITE fuppofed that the ftimulating quality of the air is neceffary to keep the heart in motion. Mr. Hewfon, obferv-

* See his work, printed at Oxford in the year 1674, under the title of Trastatus quinque Medico-Physici.

ing that the blood has a more florid red appearance in the left, than in the right, auricle of the heart, concludes with faying, that as the change of colour in blood out of the body is occafioned by the contact of air, fo it may be prefumed that the fame change within the body is occafioned by air alfo, and that the change takes place in the lungs.

DR. PRIESTLEY formed a very ingenious hypothefis concerning the ufe of air in refpiration, which he eftablifhed by a train of well-conducted experiments on the then prevailing phlogiftic theory. The principal law of this hypothefis is, that the air ferves to abforb the fuperfluous phlogifton from the blood through the lungs, and that the more or lefs florid rednefs of the blood depends on the different quantities of phlogifton in it *. The phlogifton, however, or principle of inflammability, is not

* For a full explanation of this hypothesis fee the Doctor's Exper. and Obs. vol. III. fect. 5; or the Phil. Trans. vol. LXVI.; and likewise Dr. Crawford's work on Animal Heat and Inflammation.

a real,

a real, but a fuppofed, agent in nature, which, for want of better information, was applied to explain most of the phænomena of combustion, decomposition, and (by Dr. Priestley's ingenuity) of respiration. But the present state of knowledge being, in confequence of very recent discoveries, fufficient to account for the abovementioned phænomena in a simpler, and, of course, a more natural way, the supposition of the phlogistic principle is become altogether superfluous.

OF this new or antiphlogiftic theory, which may be feen at large in a variety of recent publications, and of the difcoveries which gave rife to it, I fhall briefly mention fuch particulars only, as may be of ufe in elucidating the action of the aerial fluids on the human body. As for the facts upon which its feveral parts are eftablifhed, and likewife for the objections which have been made to it, I must refer the reader to the works of other authors *.

* See Lavoifier's Elements of Chemistry, Dr. Priestley's pamphlet, entitled, Experiments and Observations relating to the Analysis of Atmospherical Air, &c.; Fourcroy's Chemistry, &c.

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THIS theory is at prefent almost univerfally adopted by perfons of the first rank in philosophy, and daily experience is continually throwing new light upon it; yet it must be confessed that it is by no means free from doubts and difficulties. It is in confequence of those deficiencies, and on account of the uncertainty, which is infeparable from the nature of hypotheses, that I have carefully feparated the knowledge of facts from the fuppofition of their caufes. The former have been arranged in the preceding four chapters, and any perfon may account for them in the manner he likes best; but it was deemed necessary, at the fame time, to add the most fatisfactory explanation which can be fuggefted by the prefent state of knowledge, and this explanation will be found in the prefent chapter.

THE fenfation of heat is fuppofed to be produced by a peculiar fluid called *the caloric*, or elementary heat; a fluid extremely fine, penetrating, and fo light that its weight cannot

not be effimated. All forts of bodies are expanded by the addition, and contracted by the abstraction of caloric. The accession of it to the human body produces the fenfation of heat, and the separation of it produces the fenfation of cold. Thus when we touch a fubftance which is of a lower temperature, viz. colder than our bodies, that substance, by robbing us of a portion of caloric, will excite the fenfation of cold; and on the contrary, if the fubstance be hotter than our bodies, it will excite the fenfation of heat, by adding caloric to our bodies.

WHEN a number of bodies of different temperatures are put together, the fum of their quantities of caloric will be difperfed amongst them in such a manner as to render them all of the fame temperature, fo that a thermometer will be found to indicate the fame precife degree of heat in any one of them. But it must be remarked, that though the temperature be the fame, yet the abovementioned fum of elementary heat will not be divided equally amongst the bodies,

bodies, unless the bodies be of the fame fort, as, for instance, three or four parcels of water, or of mercury, &c.; but fome bodies will imbibe more and others lefs of the caloric, in order to be raifed to the fame temperature, or apparent degree of heat; and this peculiar disposition in any particular body is called its capacity for containing .caloric. This property of bodies may be rendered motor intelligible by an example or two. Suppose that a pint of water, at 100° of heat, be mixed with another pint of water at 200° of heat, the heat of the mixture will be nearly 1 50°, viz. an arithmetical mean between the two temperatures; but if a pint of water at 100° of heat be mixed with a pint of quickfilver at 50° of heat, the heat of the mixture will be found to be 80°, (viz. greater than 75°, which is the arithmetical mean) which fnews that either the quickfilver or the water, has imbibed more than its equal fhare of caloric, in order to have its temperature raifed to the common degree of fenfible heat. On the other hand, if the degrees of heat be reversed,

reversed, viz. the water at 50° be mixed with an equal bulk of quickfilver at 100°, the temperature of the mixture will be 70°, which plainly fhews, that water abforbs more heat than quickfilver; and as the difference between their original temperatures and the temperatures of the mixture in the first and last cafe is as two to three, we therefore fay, that the abfolute heat of mercury is to that of an equal bulk of water as two to three; viz. " that the compa-" rative quantities of their absolute heats " are reciprocally proportionable to the " changes which are produced in their " fenfible heats, when they are mixed to-" gether at different temperatures *."

SIMILAR experiments performed on a variety of bodies shew, that unequal quantities of absolute heat must be communicated to them in order to raise their temperature, or apparent heat to the same degree.

* Dr. Crawford on Animal Heat and Inflammation; in which work a full explanation of the doctrine of heat will be found, together with a table of the comparative heats of different bodies.

65

It is in confequence of their various capacities, that whenever bodies of different fpecies are brought together, a change of temperature is generally produced. Thus, if you mix fpirit of wine and water, the mixture will become hotter than the ingredients were before. A much greater degree of heat will be produced by mixing water with vitriolic acid; and, on the other hand, if fal ammoniac de diffolved in water, a confiderable degree of cold will be produced.

IN moft fubftances a total change in their flate of exiftence is produced by the fuperaddition of caloric; thus water is gradually changed from its folid flate of ice, into a fluid, and then into an elaftic fluid, called vapour, by the addition of different degrees of caloric. And it muft be remarked, that this change of flate in bodies is attended with a change of capacity for containing caloric; the lefs denfe flate containing the greateft quantity of caloric. Thus water in the fluid flate contains lefs caloric

caloric than when it is reduced into vapour, and more than when it exifts in the form of ice.

THE aerial fluids are fuppofed to be combinations of certain fubftances with caloric. Oxygen air confifts of a fubftance, *fui generis*, which is called oxygen, combined with caloric, and, in all probability, with the matter of light alfo.

Azorte gas confifts of a particular fubftance, called *azote*, and caloric. Common air confifts of azotic gas and oxygen air, in the proportion of 73 parts of the former to 27 of the latter. By a mixture of those elastic fluids in the faid proportion, an aerial fluid is formed exactly like the atmospherical air *.

HYDROGEN gas confifts of a particular fubstance, called *bydrogen*, and caloric. As

* In general the atmospherical fluid contains a variety of extraneous particles, but they hardly ever exceed the hundredth part of the whole, and feldom amount to that quantity.

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67

for the particular fubftances which are frequently found in the hydrogen gas, fuch as phofphorus, particles of iron, &c.; they must be confidered as extraneous matters fuspended or diffolved in the gas, but not effential to its constitution.

CARBONIC acid gas confifts of a peculiar fubftance, called *carbon*, or the conflituent part of charcoal, and oxygen air, in the proportion of feven of the former to eighteen of the latter.

WATER, which has long been efteemed an elementary fubftance, incapable of decomposition, has been found to confift of hydrogen gas and oxygen gas, in the proportion of three of the former to feventeen of the latter. By the combustion of those elastic fluids water is actually formed; and, on the contrary, water may be reduced into those aerial fluids, by placing it, under certain circumstances, in contact with bodies that attract one of its components, or by the action of electricity *.

* See Fourcroy's Chemistry, the third volume of my Electricity, and the Phil. Trans. for 1797, P. I.

COMBUSTION

COMBUSTION confifts in the abforption of the bafe of oxygen air, viz. the oxygen, by bodies that are faid to be combuftible, and fetting free both the caloric and the light, which, as has been mentioned above, are the two other components of oxygen air. Agreeably to this definition, we muft confider as combuftions not only the burning of coals and other fuel, as is ufually done in our chimneys, but alfo the calcination of metals, and refpiration itfelf, fince in both those processes an abforption of oxygen, and an evolution of caloric, take place.

IF the calcination of a metal (which is now called *oxygenation* of the metal) is carried on flowly, as by merely expofing certain metals to the atmosphere, then the caloric and the light, which is feparated from the oxygen portion of the atmosphere, is too little to affect our fenses, and we can only observe, after a certain time, that by having absorbed a quantity of oxygen, the metallic fubstance has lost its combustibi-F 3 lity

lity (viz. its attraction for oxygen) and has affumed a different appearance, together with an increase of weight. If the oxygenation be carried on in a quick manner, as when an iron wire is made very hot in oxygen air, then both the caloric and the light become manifest.

WHEN the metal has abforbed as much oxygen as its nature admits of, it is then faid to be incombustible, or completely oxydated. But if by any means the oxygen be feparated from it, then the metallic oxyde will be converted again into a metallic fubstance fusceptible of combustion.

In the combustion of animal and vegetable fubftances, which confist of various component articles, the process is accompanied with peculiar phænomena, which vary with the nature of the combustible, the quickness of the combustion, and other circumstances. Thus in the burning of wood, the oxygen of the atmosphere is abforbed, the caloric and the light are difengaged,

gaged, the carbon of the wood combines with a portion of the oxygen, and forms carbonic acid gas, the evolved caloric converts the aqueous part of the wood into fteam, and fo forth.

A VARIETY of phænomena may be obvioufly explained upon the bafis of this doctrine.

WE may eafily comprehend why no fort of combustion can take place where no oxygen air exists; as also why every fort of combustion will proceed rapidly in pure oxygen air, and much less fo in common air; for the latter contains only a small proportion (viz. about one quarter) of oxygen air. Thus may other processes be easily reconciled to, or explained by, this theory. But it is now time to examine the phænomena of respiration.

THE uses of respiration are various and important. They may be divided into mechanical and chymical. Of the mechanical, F 4 fuch

fuch as the voice, the cough, &c. no notice will be taken in this work. The other uses principally confift in furnishing the body with a constant supply of oxygen, and probably in exonerating the blood of the superfluous carbon and hydrogen.

In the process of refpiration a decompofition of the air takes place in the lungs. The blood, in its paffage through that organ, abforbs the oxygen of the common air, difengages the caloric, and leaves the azotic gas, with a fmall refiduum of oxygen air. The blood, therefore, does not imbibe the oxygen air, but the oxygen alone, viz. the bafis of oxygen air, divefted of that quantity of caloric which was necessary to give it the aerial form. The caloric which is fet free in this process, by dispersing itself through the body, keeps up its temperature, and forms the origin of animal heat. However, this part of the theory which relates to the formation of animal heat, is embarraffed with difficulties, which will be noticed prefently.

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THE carbonic acid gas, which is formed in the process of respiration, is supposed to derive its origin from a quantity of carbon, which, being discharged from the blood, combines with a portion of the oxygen air.

THE watery vapour which is expelled with the air that is expired from the lungs, is fuppofed to be formed in that organ by a combination of oxygen with a quantity of hydrogen, which is likewife difcharged from the blood. But it is not unlikely that both the carbonic acid gas and the water, inftead of being formed in the lungs, may come out of the blood, through the exhaling pores of that organ, ready formed; the blood having originally received it in that ftate from the chyle, &c.

THE air then which is expired from the lungs, contains a fmaller quantity of oxygen air than it did before, but it contains alfo fome carbonic acid gas, and fome water, 74 MEDICINAL PROPERTIES of water, in the form of vapour *.-Let us now examine the different parts of this theory.

IT is evident, from the foregoing facts and explanations, that the oxygen air is the only fluid capable of affifting refpiration and combuftion, and that it is indifpenfably neceffary to animal life, fince the common air is useful only on account of the oxygen it contains.

* There are fome modern philosophers, who explain the phænomena of refpiration without admitting the abforption of oxygen by the blood. The blood, they fay, in paffing through the lungs, acquires a vermilion red colour, becaufe it depofits a portion of its carbonated hydrogen upon the air; and it becomes again dark in the course of circulation, because it combines with a fresh quantity of carbonated hydrogen. At the fame time the oxygen of the common air which enters the lungs, by combining with the carbon and with the hydrogen, forms the carbonic acid gas with the former, and the watery vapour with the latter .- It may be eafily perceived, that by only changing the name of carbon into that of phlogiston, this explanation may, in a great measure, be made to coincide with Dr. Priestley's hypothefis.

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THE mixture of nearly one part of oxygen air and three parts of azotic gas, which forms the atmospherical fluid, is, in all probability, the best proportion of ingredients for the maintenance of life; fince we find that with a fmaller proportion of oxygen, not only the respiration becomes unpleasant and laborious, but debility, convulsions, and other bad effects are produced; and on the other hand, that bad symptoms of another fort are brought on by a greater proportion of it, such as a preternatural heat, feverish pulsation, pains, inflammations, &c.

THE phænomena of refpiration and of combustion are not only analogous, but they illustrate each other in an admirable manner. In atmospherical air a candle gives light fufficient for ordinary purposes. In a lefs pure atmosphere the light becomes too dim; and in pure, or nearly pure, oxygen air, the candle will indeed give a much brighter light; but it will waste fo very fast, as not to last perhaps the twentieth part of the time it will in common air.

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THAT the blood abforbs the oxygen of the atmospherical air in the act of respiration, is a proposition which a variety of experiments and analogies feem to prove beyond all doubt. When blood, recently taken out of the veins of an animal, is enclosed in a piece of bladder, and is thus exposed to common air, or to oxygen air, it acquires a florid red colour, and part of the oxygen air is abforbed. The fame thing takes place within the body, viz. the air which is expired contains a smaller proportion of oxygen than it did before, and the blood which returns from the lungs to the heart, and thence proceeds through the arteries, is found to have acquired a bright redness in its passage through the lungs; it is therefore natural to conclude, that the blood has abforbed the oxygen through the pores of the thin membrane, which separates it from the air in the cells of the lungs *.

* This membrane is certainly much thinner than common bladder. Dr. Hales conjectured the thickness of the former to be the thousandth part of an inch.

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THE probability of this conclusion is corroborated by ftrong collateral proofs; as by obferving that the arterial blood of animals that have been fuffocated, or that have died for want of oxygen air, is far from being of its usual florid red colour; as also by obferving, that when a quantity of blood is confined in a veffel full of air, the air is not fo quickly contaminated or deprived of its oxygen by the prefence of arterial, as by that of venous blood. And it is even afferted, that a quantity of blood taken out of the carotid artery of a fheep, being confined in a veffel full of azotic gas, improved the gas fo as to render it, in fome meafure, fit for refpiration, fo that fome oxygen must have been imparted to it by the blood *. This experiment deferves to be repeated with particular care.

THE decomposition of air, and the abforption of its oxygen in combustion and oxygenation of metallic bodies, are also ana-

* Medical Extracts, vol. I. p. 70.

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77

logous to the phænomena of refpiration, and confirm the abforption of oxygen by the blood in that process.

It is true that a finall quantity of carbonic acid gas is found in the air in which blood has been confined; but the formation of this gas does not entirely account for the diminution of the oxygen. Befides, it is not improbable, as we faid above, that the carbonic acid gas comes out of the blood ready formed, at the fame time that the blood abforbs the oxygen.

By examining the course and state of the blood, we find that it preferves its brilliant redness through all the channels which convey it from the lungs through the heart, and to the extremities of the body. 'But in the other vessels, which receive it at the extremities of the former, and convey it through the heart as far as the lungs, the blood is of a dark purple colour. The former course is performed through the pulmonary veins, the left auricle and left ventricle

ventricle of the heart, the aorta and its branches. The latter is performed through the branches and trunks of the afcending and defcending cava, the right auricle and right ventricle of the heart, and laftly, through the pulmonary arteries, which convey it to the fpungy cells of the lungs, where its colour is changed, &c.

THE blood, therefore, having acquired the oxygen in the lungs, conveys it as far as the extremities of the branches of the aorta, where the oxygen is deposited, and the blood returns without it through the veins.

It is difficult to fay under what form is the oxygen combined with the blood, and what becomes of it at the extremities of the arteries where it is left by the blood. For want of direct experimental information concerning this interesting point, we have only the light of analogy and conjecture to lead us in the investigation of truth.

79

IN the combinations of the bale of oxygen air with different bodies, fuch as take place in combustions of every fort and degree, three different effects must be particularly remarked. The first is, that the oxygenation is generally accompanied with colours of different intenfity; the red being produced more frequently than any other colour, as is the cafe with mercurius calcinatus per se, red lead, crocus martis, &c. The fecond is, that by the acceffion of oxygen a body is always rendered firmer or more compact. Thus, by the combustion of hydrogene and oxygen, water is produced, which is a much heavier and more compact fubstance than either of its two components; thus also by oxygenation oils are thickened, and metallic bodies are converted into a fubstance powdery indeed, but whole particles are firmer and harder than the fame bodies in their metallic ftate *. The third is, that a body lofes, in great

* It is in confequence of the fuperior hardness of its particles, that crocus martis (which is oxygenated iron) and

great measure, its capacity for containing caloric, and of course gives out heat whenever it passes from a rare into a more compact state of existence, and vice versa. Thus water contains a great deal more of caloric than ice, but much less than steam; hence when steam is converted into water, it deposits part of its caloric, viz. it communicates sensible heat to the surrounding bodies, &c.

By an eafy application of those facts to the phænomena of respiration, we are led to conclude, first, that the redness which the blood acquires in the lungs, indicates a real oxygenation of that fluid; fecondly, that the oxygen is flightly attached to the blood, for the blood easily parts with it at the extremities of the arteries; thirdly, that the oxygen, which is deposited by the blood at the extremities of the arteries, enters into combination with, and gives firm-

and the oxyde of tin (commonly called *putty*) are employed for polifhing the hardeft fteel, glafs, and even agates.

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nefs and folidity to, those particles of matter which give increment and stability to the animal frame; fourthly, and laftly, that as the bond of union between the blood and the oxygen is not very ftrong, and as the union of the oxygen with other fubstances at the extremities of the body is much stronger, therefore it feems evident that the caloric of the oxygen air is not entirely evolved from it in the lungs; but that the greater portion of caloric is evolved at the extremities of the arteries, where the oxygen is more powerfully attracted by other fubstances than it is by the blood in the lungs. Hence it follows, that the origin of animal heat does not exift in the lungs only, but that it takes place, more or lefs, in every part of the body. And this fhews why the whole body is nearly of the fame temperature; whereas, if the caloric were evolved in the lungs only, that part of the body would be much warmer' than any other, which is not the cafe *.

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* I am happy to find that this explanation coincides with the opinion of a very diffinguished and recent anatomical

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82

WHAT difpofes the blood to abforb the oxygen in the lungs, and what forces it to depofit that principle at the extremities of the arteries, are queftions which the prefent flate of knowledge does not enable us to anfwer fatisfactorily. It has been fuppofed that the oxygen is attracted by the ferruginous particles of the blood, and that the rednefs of the blood is to be attributed to the red colour of the oxyde of iron. But fince

tomical writer, who expresses himfelf in the following words :

"But in reflecting upon this moft difficult of all fubjects, the generation of heat in the living body, many things are to be taken in the calculation, which feem, on the flighteft glance, to be far more important than this deposition of oxygene from the blood. It is a law of nature, to which, as far as we know, no exception is found, that a body, while it paffes from an aerial to a fluid, or from a fluid to a folid form, gives out heat. Now, what is the whole bufinefs of the living fyftem but a continual affimilation of new parts, making them continually pafs from a fluid into a folid form ? The whole nourifhment of the body goes on in the extreme veffels, and is a continual affumption of new parts. The extreme veffels are continually employed in forming fome acids, which appear naked in the fecretions; in forming

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oxyds,

fince it has been proved by a variety of experiments, that the oxygen is attracted by, and combines with, a variety of other fubftances independant of iron or metals, I do not fee the neceffity of attributing the attraction of the oxygen to the ferruginous, more than to other, ingredients of the blood. Nor do I fee the abfolute neceffity of attributing the red colour to the particles of iron, fince other fubftances, in which iron is not concerned, fuch as the oxyde of mer-

oxyds, as the fat and the jellies of the membranous and white parts; in the various depositions of muscle, bone, tendon, &c. for thefe are all continually abforbed, thrown off by the urine, and inceffantly renewed. They are continually employed in filling all the interffices of the body with a bland fluid or halitus; they are continually employed in forming fecretions of various kinds, In performing all this the power of the veffels may do much ; but the ultimate effect in each procefs must be a chemical change, and perpetual changes will produce a conftant heat. Place the organ and focus of this animal heat in the centre of the body, and you are embarraffed in a thousand difficulties; allow this heat to arife in each part according to its degree of action, and each part provides for itfelf." Bell's Anatomy, vol. II. p. 125.

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cury, red lead, &c. owe their rednefs merely to the oxygen which they have imbibed.

IT is difficult to account for the formation of the carbonic acid gas, and of the watery vapour in the lungs; for if those fluids be really formed in that organ by the combination of the carbone, and of the hydrogen, with the oxygene of the infpired air; the whole, or nearly the whole, of the oxygen air would be fo expended, and little or none of it would remain to be imbibed by the blood. The caloric likewife would be employed in the formation of those fluids, instead of being dispersed through the body. Is it not therefore more natural and more fatisfactory to suppose, that both the carbonic acid gas, and the water, are separated from the blood in the lungs, but not formed in that organ? It is certain that carbonic acid gas is introduced into the ftomach by the aliments; and it is certain that the chyle conveys it to the blood, why then should we suppose that there is another formation of this gas in the lungs? As for the G 3 watery

watery vapour, we may account for it in the fame manner; and indeed the exudation of water through the internal membranes of the human body, is fo generally practifed by nature for the purpofe of keeping those membranes, &c. foft and pliant, that it would be irregular not to admit the fame exudation of water in the lungs alfo.

THE expulsion of putrid effluvia from the body is confidered as another office of refpiration. This is shewn by the offensive smell of the breath of certain persons, who have no bad teeth to account for it. But it is difficult to ascertain in what cases this may take place, and how far it may extend.

It is with the appearance of probability fuppofed that the oxygen, which the blood depofits on the various parts of the body, is partly expended in the exercife of mufcular motion; fince we find, that after unufual exertions of the body, a man breathes fafter, and likewife takes in much more air at a time, as if nature endeavoured by that means

means to recruit what has been lately expended in greater quantity than ufual.

THE azotic gas, which is the greatest ingredient of common air, is confidered as only a diluent of the oxygen air, and as being otherwife paffive in the process of respiration. Yet this diluent answers a variety of purpofes; the principal of which is, that it exposes a proper quantity of oxygen air to a great quantity of blood, which could not have been the cafe if the atmofpherical fluid had confifted entirely of oxygen. This object is accomplished by the very extensive furface which the lungs present to the air in its numerous cells; for the more numerous the cavities are, the greater is the furface; and, in fact, we find that in those animals that are not much in want of air, and that must frequently fufpend their respiration for a confiderable time, fuch as the fea-turtle and the frog, the lungs confift of very few and very large cells.

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87

THE great proportion of azotic gas in common air, does also adapt that fluid to the purposes of vegetation, and other natural processes, the enumeration of which is incompatible with the limits of this Essay.

CHAPTER VI.

A general Idea of the Application of aerial Fluids for the Cure of Diforders incident to the human Body.

CONSTANT observation has informed mankind, from time immemorial, that the air of certain places is more or less falubrious than that of other places; and that the various qualities of the air in different fituations, are peculiarly favourable to certain conftitutions. Physicians, availing themselves of this natural variety, have long

long been in the habit of fending their patients to fuch places as experience and analogy indicated to be more favourable to their respiration. The sharp air of one place was reckoned good for one diforder, the damp air of a fecond place was efteemed ufeful in other cafes, the pure air of a third was recommended in particular difeases, and so on. Howfoever defective and erroneous their knowledge of the real conftitution of the atmosphere may have been, how foever they may have abused the application, yet certain it is, that the variety of effects, fuitably to the different qualities of the atmospherical fluid in different fituations, is attefted by innumerable facts and universal observation. Previoufly to the late difcoveries, the ideas of phyficians refpecting the different qualities and effects of the atmospherical fluid, were vague, and generally erroneous. Experience, which shewed them the advantages that had been obtained in a number of fimilar cafes, was their best guide, and all befides was doubt and obscurity. The present state of knowledge has, in great measure,

measure, diffipated the clouds; fince it has not only shewn the reasons upon which certain qualities of the air depend, but has likewise furnished us with the means of procuring airs of opposite qualities, and of any degree of purity, at all times and places, as also of applying them in all the extensive variety of quality, degree of purity, and length of time.

THE apparatus necessary for producing the various factitious airs, may be eafily derived from the particulars that have been mentioned towards the beginning of this book ; but for a general apparatus, that admits of compactness, cheapness, and a fufficiently extensive application, I cannot recommend any better than, or nearly fo good as, that which was contrived by Mr. James Watt, engineer, of Birmingham; by means of which the artificial airs, of fufficient purity, may be produced at a very moderate expense, and easier than by any other general method. Those apparatuses are now made for fale, and a printed defcription, with

with necessary practical directions for the use of its various parts, is given with each apparatus, which fuperfedes the neceffity of adding the fame to the prefent work. I shall, nevertheless, referve, for the end of the book, a lift of the principal precautions which should be attended to in the management of Mr. Watt's, or of any other apparatus of this fort, to which the practitioner may recur for extempore information.

THE artificial elaftic fluids are applied to the lungs by the way of refpiration, to the ftomach and inteftines, by means of injections, or in combination with fluids, and to the external parts of the body, merely by contact.

VARIOUS apparatufes have been used for the refpiration of factitious airs. The leaft exceptionable air-holder for this purpose, confifts of a large glafs receiver filled with the required fort of elastic fluid, inverted, and fwimming in water; out of which the air

air is refpired by means of a bent glafs tube, which, paffing with its bent part through the water, projects one aperture above the water within the receiver, whilft its other extremity is applied to the mouth of the experimenter.

INSTRAD of the above-mentioned bent tube, the receiver may have an aperture at its upper end, to which a tube is adapted, air tight, in an horizontal direction. But as this apparatus requires a large tub full of water for the receiver to fwim in, which renders it rather cumberfome, and not very portable, therefore other contrivances have been substituted to the receivers. The machine more in use for this purpose, confifts of an oil-filk bag, furnished with a short wooden tube or faucet, which, when the bag is full of the required aerial fluid, is applied to the mouth of the patient. Those bags are filled with the proper aerial fluid by means of a glafs receiver, which, befides its large aperture, has a fmall aperture with a ftopple at the oppofite end. The receiver being

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being filled with the required air, and inverted in water, the ftopple is removed from its fmall aperture, and the wooden tube of the bag is applied quickly to it; then by preffing the receivers down into the water of the tub, the air will be forced into the bag. But with the air-holders, which form part of Mr. Watt's apparatus, the operation is rather eafier, for which, fee the defcription of the faid apparatus. The principal imperfection of thefe bags confifts in the fmell of the oil-filk, which proves naufeous, and almoft intolerable to delicate perfons; yet this fmell may, in fome meafure, be removed *.

INSTEAD

* For this purpole Mr. Watt gives the following directions :—" To free oiled filk from its difagreeable " fmell, cut it into pieces of the fize wanted for the " bags, and provide a fmooth table formewhat larger " than the pieces of filk, and a flat board of the fame " fize as the table. Take charcoal frefh burnt in an " open fire, until it is free from fmoke, extinguish it by " fhutting it up in a clean close vessel, and reduce it to " powder. Sift this powder over the table to the " thickness of a quarter of an inch, or more, fpread a " piece

93

INSTEAD of oil-filk bags large bladders may be used; but as a bladder is not capable of holding a quantity of elastic fluid large enough for medicinal use, several of them should be had in readiness, each fur-

" piece of your filk upon it, and fift upon that again ano-" ther layer of your charcoal duft, and thus proceed al-" ternating the layers of filk, and charcoal, until the " whole of your filk is deposited ; then lay your move-" able board upon the top of all, and leave the whole " undifturbed for four or five days. If, upon remov-" ing the charcoal duft, the filk has not loft its fmell en-" tirely, repeat the process. The charcoal dust is to be " fwept off the filk, and the filk to be washed upon a " table with a wet fponge until it is clean. The bags " must then be carefully fowed up, and the feams " anointed with japanner's gold fize, taking care to use " that kind which does not become brittle when dry. "Green oiled filk fhould be avoided, as it is flained by " means of verdigris, which rots it; the yellowish filk " is the beft.

" It is neceffary to obferve here, that although oiled filk be the beft fubftance known for making the bags of, it is very imperfectly air-tight; and although charcoal-duft deprives it of fmell for the time, yet as it can only attract the odoriferous particles from the furface, it re-acquires fome fmell by keeping, but by no means equal to what it had at firft."

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nifhed with a wooden or glafs tube, like the oil-filk bags, through which they may be filled, &cc.; fo that when the air of one bladder is exhaufted, a fecond bladder may be fubftituted, and fo on. Several bladders might be eafily made to communicate with each other, fo that through one tube or faucet they might be filled all at once: four or five large bladders thus joined together, would contain about as much air as an ordinary oil-filk bag, which is a quantity, in moft cafes, fufficient for one application, and it would laft about fix minutes. The bladders have likewife an unpleafant fmell, which may alfo, in great meafure, be removed *.

WHETHER the glafs receiver, or the oil-filk bags, or the bladders be used, the patient must always take care to keep his nostrils accurately stopped whils he draws

* For this purpose turn the bladder infide out, wash it well with a weak folution of falt of tartar, then wash it feveral times over with fair water, so as to remove every particle of alkali; lastly, wash it with spirit of wine.

95

the air into his lungs at every infpiration, and to open them immediately after, fo as to expel the air from his lungs through the noftrils into the atmosphere at every expiration, and not to return it into the bag or receiver. This operation is not easily performed by most perfors, and some there are who cannot perform it at all; in which case they breathe the same air backwards and forwards to and from the bag. But by this means the air of the bag, even when limewater is contained in it, is contaminated fo quickly as to do more harm than good.

THIS inconvenience, however, is completely obviated by the use of a little machine, which is to be interposed between the mouth of the patient and the faucet of the bag. It confifts of a small box of wood, having three apertures, to the two opposite of which two short tubes are fastened; to the third, which is a lateral one, there is an external valve which will only permit the air's going out of the box into the atmosphere. One of those tubes is applied to the

the faucet of the bag, and it contains a valve which prevents the return of the air into the bag; the other tube is applied to the mouth of the patient, who has nothing more to do than to hold his noftrils conftantly ftopped, and to breathe in a natural way as long as there is any air in the bag or receiver *; it being eafy to underftand that whenever he infpires, the air will pafs from the bag into his lungs; but that at every expiration, the air will be forced through the lateral valve of the machine into the ambient air.

OF the various forts of elastic fluids, the carbonic acid gas is the only one that has been fuccessfully applied to the stomach or intestines, and for this purpose it may be administered two ways, viz. either in the aerial form in clysters, or combined with different fluids and given through the

* There are feveral perfons who, with very little attention, can breathe through the mouth only; when this is the cafe, the keeping of the noftrils ftopped is fuperfluous.

mouth.

mouth. For the former of those purposes the gas must be first introduced into a bladder by the method already defcribed. For the latter, the gas is either naturally contained in liquors, as in newly fermented liquors, yeaft, certain ripe fruits, and mineral waters; or is to be first combined with the required liquors, in which cafe water is the fluid which is more generally used. This impregnation of water and other liquors with carbonic acid air, may be accomplished by various methods, fuch as by pouring the liquor backwards and forwards from one veffel to another, over the furface of vegetable fubftances that are in a ftrong state of fermentation; or by filling a veffel partly with earbonic acid air, and partly with the required liquor, and then fhaking it for a minute or two, &c. But the best way of performing this impregnation, is by means of a well-contrived machine, which has been long in use, and is generally known under the name of Dr. Nooth's glafs apparatus, for making artificial mineral waters. There is, however, a contrivance for impregnating

pregnating water with an incomparably greater quantity of carbonic acid gas, than that which can be accomplifhed in Dr. Nooth's apparatus. But this contrivance is kept a fecret by the inventor, though the water, fo highly impregnated by him, may be had in London at a moderate price.

THE application of factitious airs to the external parts of the body, may be performed with the utmost facility. The aperture of a tube, which proceeds from the veffel in which the gas is generated, may be directed towards the part which is affected; a bladder full of the required gas may be gradually preffed, fo as to throw a ftream of the gas upon it; the part itfelf, as far as it is practicable, may be introduced into a vessel full of the required air; or, lastly, a fmall glafs funnel, with a bladder fastened to its fmall end, and filled with the required elastic fluid, may be applied over the part, with the edge of its large aperture clofe to the fkin, fo as to prevent the efcape of the gas into the circumambient air.

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THE medical application of factitious airs, and the effects which have thereby been produced, are as yet labouring under all the viciffitudes of truth and exaggeration, of accuracy and misapplication, of short experience and uncertainty. The anxiety of fome perfons, the ignorance of others, the defire of fame, the love of interest, and the fear of dangerous innovations, have alternately operated in favour and against the administration of the elastic fluids for the alleviation of diforders incident to the human body. In the conflict of fuch oppofite powers, it is difficult to feparate truth from exaggeration and error; it is impoffible to afcertain the precife limits of their use and efficacy.

NOTWITHSTANDING those weighty objections, I have endeavoured to collect, to examine, and to methodize all the useful information which I could procure relatively to the subject, in hopes that a comprehensive view of it might promote the use, and in great measure prevent the abuse,

abufe, of a new class of remedies, which have all the appearance of proving very advantageous to mankind.

In the use of oxygen air we have a fingular stimulus, which admits of its being rendered more or less active by dilution with various proportions of common air. In its pure, or nearly pure, state, it is a powerful exciter of suspended animation; and when diluted with a confiderable quantity of common air, it is a gentle stimulus, which, by invigorating the various parts of the animal body, by communicating firmness to the folids, and energy to the sluids, does frequently obliterate the causes of morbid habits.

THE use of azotic gas, and of the various species of hydrogen gas, produces a diminution of the irritability of the animal fibre to any degree, and hence it becomes useful in a variety of those disorders, which depend on an increased irritability, such as inflammations, coughs, spass, &c.

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In the use of the carbonic acid gas we have a powerful antiseptic, and in certain cases a folvent of confiderable efficacy.

THE use of pure oxygen air is confined to the purpose of exciting the dormant powers of suspended animation, and it is, therefore, to be administered to children born apparently dead, or overlaid; to perfons suffocated by drowning, by steam of charcoal, by foul air, &c. whenever the circumstances of the case may indicate a possibility of recovery.

THOSE cafes excepted, the refpiration of pure, or nearly pure, oxygen air, is almoft always attended with unfavourable fymptoms, fuch as a preternatural heat, efpecially about the region of the lungs; a quickened and feverifh pulfation; inflammation, &c. And those fymptoms come on after a fhorter or longer use of the oxygen air, according to the particular constitution of the experimenter, and the purity of the gas.

BUT

BUT when the oxygen is diluted with much common air, viz. in the proportion of one to eight, and even as far as one to twenty, it then is a fafe and very ufeful remedy, whofe principal action confifts in giving tone, elafticity, and confiftence to the fluid as well as to the folid parts of the body, and of courfe it promotes all the natural confequences of those effects, viz. it quickens languid circulation, it ftrengthens the organs of digestion, promotes fecretions, invigorates debilitated habits, and it affifts nature in throwing off bad humours, and other lurking causes of difeases.

It has been obferved, that fome individuals can bear a much greater proportion of oxygen than others, which is analogous to the various difpofitions for all other applications. Thus a certain quantity of any remedy will act powerfully on fome perfons, whilft it will not be even felt by others. Thus alfo a certain quantity of food produces ftrength and cheerfulnefs in fome individuals, whilft it produces ficknefs and H 4 indigeftion

indigeftion in others. It therefore becomes neceffary, in the application of this remedy, to regulate the proportion of the two elastic fluids agreeably to the constitution of the patient, which may be easily accomplished by means of a very few trials.

In the diluted ftate, the oxygen air is adminiftered by letting the patient breathe it for five or ten minutes once or twice a day. It might probably prove more efficacious, if it were breathed in a more diluted ftate for a longer time; but the preceding mode has undoubtedly been attended with falutary effects.

HOWEVER flight this application may appear, however fmall the unufual quantity of oxygen which is thus introduced may be, the effects have been proved and confirmed by a variety of experiments and medical cafes. But independent of the experimental proof, the improbability of the effect will difappear if it be confidered, that the lungs of most perfons, and effectially of those who labour under certain difeases,

are

are almost immediately relieved or affected by the transition from the air of one place to that of another; as by their going out of town, or even out of the houfe; and yet, as has been already observed, the difference between the air of a town and the air of the country, or of that of a houfe and of the external air, is fo very trifling as hardly ever to be diffinguished by the eudiometer. Extremely minute, and almost inconceivably small, quantities of matter can act with wonderful efficacy, when they are introduced into the circulation of the blood. The inoculation of the fmall-pox, and the experiments with poifons, furnish fufficient confirmation of this obfervation.

By breathing a mixture of common and oxygen air, even when the latter does not exceed one-eighth part of the former, for about ten or fifteen minutes, the pulfe is generally quickened of a few ftrokes, but it is almost always made stronger. The lungs, during the operation, are feldom fensibly affected; but on leaving off the mixed airs, and

106 MEDICINAL PROPERTIES of and returning to the atmospherical air, a degree of tightness is frequently felt on the cheft, which, however, gradually goes off and vanishes after a few minutes time.

WHEN debilitated habits breathe the diluted oxygen air for about a quarter of an hour once or twice a day, the improvement of their health is hardly ever confpicuous in lefs time than a week or a fortnight; but after that period, they will find their ftrength, their appetite, their digeftion, their circulation, and other functions, fenfibly improved; and this improvement goes on progreffively to a greater or lefs degree, according as age, local indifpofitions, times of the year, and other circumftances may allow.

THE mixtures of common air with azotic gas, or of common air and any species of hydrogen gas, are commonly denominated *reduced atmospheres*; for, in fact, they contain a smaller quantity of respirable fluid, than is contained in an equal quantity of

of common air. The principal effect of those reduced atmospheres, is to diminish the irritability of the parts subfervient to respiration, and indeed of the whole body; for which reason they are successfully administered in inflammations of the lungs, in spasimodic coughs, and in all the diforders that are nearly allied to those.

MUCH caution must be used in the administration of those reduced atmospheres, as fome of them are productive of alarming fymptoms. The mixture of azotic and common air, in which the former should never be more than a quarter of the latter, is the leaft dangerous, and at the fame time the leaft efficacious. The fame thing may be faid of the mixture of common air with the mild fort of hydrogen gas, viz. that which is produced from iron and diluted vitriolic acid, or by paffing the fteam of water over red hot iron, excepting that it is rather more efficacious than the preceding. But the hydrocarbonate is much more powerful and more dangerous than any

any of the abovementioned gaffes, especially when fresh made. It should, in general, be mixed with about twenty or thirty times its bulk of common air, unless fome particular case may seem to require a greater proportion of the dangerous gas. For most purposes it will suffice to breath it for about five minutes a day.

THE breathing of the diluted hydrocarbonate is attended with a diminution of fenfibility, efpecially about the cheft, and this effect is frequently fo great, that fome perfons have expressed it by faying, that they felt as if they had no lungs at all, even when they had been a few minutes before in excruciating pains. But this diminution of fenfibility is almost always accompanied with vertigo or giddinefs, with a lowering of the pulse, and with faintness. It must be particularly remarked, that though those fymptoms in general come on immediately after the operation, yet fometimes they return once, and even twice, more in the course of the day. When the breathing of reduced

reduced atmospheres proves very troublefome, it may be interrupted for a few minutes.

Its great power in checking irritability and fenfibility, feems to render the diluted hydrocarbonate applicable to fome diforders that have hitherto eluded all medical application; and as one of the moft likely to be relieved by this treatment, I fhall mention the hydrophobia, or madnefs which is occafioned by the bite of mad dogs, or other mad animals *.

A REDUCED atmosphere, capable of diminishing in some degree the irritation of the lungs in inflammations, coughs, and

• I have been told, and have read, though I cannot at prefent recollect where, that the ufe of opium, and likewife that the fufpenfion of animation for a time by accidental drowning, have actually cured the hydrophobia in two or three cafes. If this be true, the probability of the hydrocarbonate proving beneficial, is thereby much increased.

certain

certain species of afthma, has been expeditioufly formed by mixing the vapour of vitriolic ether with common air. For this purpose the patient needs only hold a small phial of ether open near his mouth, for about an hour at a time or longer, by which means the vapour of the ether mixes with the air that enters the lungs in the ufual courfe of refpiration, and converts it into an inflammable, or rather an explosive, aerial fluid *. For this purpose it has been found useful to mix some powdered leaves of hemloc (cicuta) with the ether. The ether (viz. about a quarter of an ounce of it) may also be put in a common tea-pot, and the mouth may be applied to the fpout of it, fo as to draw the air through it, and through the vapour of the ether.

THE carbonic acid gas has been longer in use as a medicine than any other factl-

* If three or four drops of ether be fhook in a phial full of common air, and if afterwards the aperture of the phial be prefented to the flame of a candle, the air in it will explode like a mixture of common air and hydrogen gas.

tious

tious aerial fluid. Much has been done, and much has been written, relatively to it. But the useful refult of those experiments and investigations will be found condensed in the following few paragraphs.

In putrid fevers the free use of carbonic acid gas has been of confiderable use, whenever the urgency of the case has not been very great, viz. when time was allowed for the gas to operate upon the morbid matter; and when the differition of the bowels was not fo great as to prevent the free use of the gas.

In the fcurvy this gas has been of confiderable ufe in the beginning of the diforder, rather more than in an advanced ftate of it. But the ufe of vegetables, of fugar, and of other fubftances that contain it in great abundance, are acknowledged to be ufeful in all ftates of that diforder. Experience likewife inform us, that in the ufe of carbonic acid gas we are not to expect an unlimited antifeptic, nor a perfect folvent of the

the ftone in the urinary bladder; yet its ufe in putrid cafes, and in fome difeafes of the bladder or kidnies, is attended with confiderable benefit.

THE external application of carbonic acid gas to fores and ulcers of every fort, is unqueftionably very ufeful.

AFTER a careful confideration of the preceding general and comprehensive prospect of the medicinal use and efficacy of the aerial fluids, we may eafily regulate the measure of our hopes by the standard of reafon and experience. The idea of finding in them a remedy, capable of curing confumptions in all their stages, must be laid afide; and the hope of healing all forts of internal ulcers will naturally vanish. The use of reduced atmospheres does undoubtedly diminish the irritability of the fibre, and a diminution of irritability favours the healing of certain ulcers, but by no means of them all; nay, in fome cafes it will even produce the contrary effect. The use of oxygen

oxygen air has been found advantageous in many of those disorders that are called nervous, and it has undoubtedly ftrengthened and invigorated several debilitated or emaciated habits; but it would be absurd to expect that it should prove beneficial in all cases of emaciation and debility, since those visible effects are often produced by causes that may be rather fomented than checked by the use of oxygen air.

IN most of the diforders incident to the human body, the various concurring circumstances are far from being known to their full extent; hence theory may fuggest, but experience must prove the use of certain practices. Improvements and discoveries may be generally urged and expected; but where theory and experience are filent, we have no warrantable guide to affist us in the investigation of new properties and new applications.

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CHAPTER VII.

Of the particular Administration of aerial Fluids in different Disorders.

FTER a general idea of the applica-L tion of factitious airs by way of remedies to the human body, it will be neceffary to flate those modes of treatment, which experience or analogy fhew to be the most efficacious in particular difeases." But this statement cannot be attended either with great accuracy of defeription, or with extensive information concerning the phænomena, that are really produced by the factitious airs in all cafes. The various nature of individuals, the imperfect accounts of feveral cafes, and the frequent administration of other medicines in conjunction with the aerial fluids, limit for the prefent the attainment of the abovementioned objects.

ALL

ALL that the practitioner may expect to derive from the prefent chapter is, a guide or indication for the commencement of the application, a general view of the principal effects that are produced by the particular administrations, and a warning against miftakes. But with respect to the continuation, or suspension, or alteration, of the treatment, he can only be instructed by a careful observation of the phænomena which take place in the course of the application.

I SHALL forbear mentioning other medicines that may be proper to be adminiftered at the fame time with the gaffes, as these must be left to the judgment of profeffional gentlemen. But I would strongly recommend to administer them as sparingly as the nature of the case can possibly admit of; being persuaded, that the good effects of the aerial fluids is frequently counteracted by the action of other medicines.

N. B. The difeases in the following pages are arranged in alphabetical order.

I 2

Animation

Animation Suspended.

IN cafes of this fort, whether they be occafioned by drowning, by noxious vapours, or by any other caufe of the like nature, the oxygen air should be administered pure, or nearly fo. The wooden pipe of a large bladder full of it must be introduced into the mouth of the fubject, the lips must be prefied upon the faid pipe, and the noftrils must be stopped by the hands of an affiftant. Then by preffing the bladder, the oxygen air must be forced into the lungs, as much as poffible, for about eight or ten feconds, after which the mouth and noftrils being unftopped, without removing the pipe of the bladder, the cheft about the region of the lungs must be presed gently; then the bladder being applied as before, the oxygen air is forced again into the lungs, and fo on; continuing a fort of forced and artificial refpiration for about a quarter of an hour at least, if no figns of . life

life appear before that time *. But as foon as any natural or fpontaneous movements are perceived, the preffing of the region of the lungs may be difcontinued, and the bladder, &c. must be removed; for in that state a free ventilation of the ambient air will be found fufficient to restore life.

THIS treatment fhould be accompanied with the communication of a gentle warmth, and perhaps with friction to the hands and feet. But care muft be taken to do what is just necessary, and not too much; for in the attempts to restore animation, the stimuli and other applications are frequently carried fo far, as to destroy that last spark of life, which they were intended to revive.

In cafes of children born apparently dead, or ftrangled in laborious parturition, &c. the use of oxygen air cannot be too forcibly recommended. The application is eafy and highly promising. Independent of

* Several bladders full of oxygen air fhould be kept in readinefs, for a fingle bladder will be foon exhaufted.

I 3

the experiments that have been made on brutes, I know of a cafe, in which a child born apparently dead, was brought to life merely by forcing oxygen air into his lungs, whilft he was held before the fire,

Afibma.

I FIND many creditable accounts of this diforder having been relieved, and fometimes perfectly cured, by the use of diluted oxygen air in some cases, and by the use of reduced atmospheres and the vapour of ether in other cases.

It would be abfurd to imagine, that either of those treatments may be indifcriminately applied to the very fame species of assessment after a set of the distinctions are not clearly stated in all the accounts of the cases. It appears, however, that in a plethoric assessment and when the disorder is attended with confiderable pain, hard cough, and inflammatory symptoms, the reduced atmospheres must be administered.

119

In those cases the patient may be directed to breathe daily fixteen quarts of common air, with four quarts of hydrogen, obtained from iron and diluted vitriolic acid, or, which is better, from the vapour of water and red-hot iron. But should this mixture of elastic fluids prove ineffectual in a day or two, then a mixture of one pint of hydrocarbonate and thirty pints of common air, may be used instead of it; and the ftrength of this mixture may be increafed according to circumstances. If in breathing the diluted hydrocarbonate, giddinefs should come on, the patient must be defired to intermit the operation; refting, that is breathing the ambient air, for a few minutes, and then to refume the inhalation of the diluted hydrocarbonate. Thus the operation may be intermitted three or four times.

THE breathing of the vapour of ether, after the manner defcribed in the preceding chapter, has been found to afford confiderable alleviation of the pain and oppreffion in those cafes.

I.4

IN nervous afthma, and efpecially in debilitated habits, the oxygen air may be adminiftered; and it will be proper to begin by inhaling daily eight quarts of common, with two quarts of oxygen, air, extracted from manganefe by means of heat. The quantity of oxygen may be increased, in case the abovementioned proportion should be found ineffectual; and it is remarkable, that in this species of disease the patients can sometimes bear a great quantity of oxygen without any material effect.

IN all cafes of afthma, the effects of the application of factitious airs may be perceived in the courfe of four or five days; but the accomplishment of the cure will frequently require fix weeks time, or longer.

Cancer.

THE flubborn nature of a cancer, and its dreadful confequences, render every hint, that promifes an alleviation of its effects, extremely interefling.

THE

THE elaftic fluids have been repeatedly applied to cafes of this fort, and fuch applications have been attended with confiderable advantage. 1 do not find any authentic account of a cancer having been completely cured by the ufe of factitious But certain it is, that in a variety of airs. cafes the pain has been confiderably diminished, the fætor as well as the bad aspect of the ulcer, have been almost entirely removed, and the whole habit of body has been confiderably improved, fo that the patients have thereby been enabled to have comfortable nights, more cheerful countenances, &c.

THOSE good effects have been produced by the external application of carbonic acid gas to the ulcer, and the inhalation of diluted oxygen air. Both those elastic fluids must be administered daily for weeks, or as long as the indications of the case may afford a hope of melioration. The manner of applying the carbonic acid gas has been already

already defcribed *; as for the continuance of the application, an hour a day is by no means too much, and it would be better if fuch an application were repeated two or three times in the courfe of each day. With refpect to the oxygen, two or at most three quarts of it, with about fourteen or fixteen quarts of common air, may be fufficient for each daily inhalation.

Catarrh.

In colds and defluxions, efpecially when accompanied with tightnefs about the region of the lungs, and a hard cough, much and almost instantaneous relief has been frequently obtained by breathing a mixture of about four quarts of hydrogen and twenty quarts of common air. There is no need of breathing this quantity at once. It will hardly ever be neceffary to repeat this application longer than the third day. The breathing of the vapour of ether in the

* See chap. VI. p. 99,

manner

manner already deferibed, will anfwer nearly as well as the above-mentioned mixture of elaftic fluids, and it has the advantage of being a much eafier application, fince it requires no particular apparatus *.

Chlorofis.

THE administration of diluted oxygen air has proved beneficial in difeases of this kind, perhaps more often than in any other diforder, as is proved beyond a doubt by feveral authentic cases. The paleness, the debility, the palpitation, the fever, the depraved appetite, and the other bad symptoms which accompany this diforder, generally begin to diminish in about four or five days, and a complete cure is often accomplished in about fix weeks time.

THE daily inhalation of one quart of oxygen, and ten or twelve quarts of common, air, may fuffice for the beginning. But it

* See chap. VI. p. 110.

is to be remarked, that chlorotic patients are fenfible of the leaft excefs in the proportion of oxygen, fo that fometimes they are more hurt than benefited by it, unlefs fuch a quantity of it be administered as may be just neceffary; and this quantity can be shewn only by a careful observation of the effects which take place. The lungs will be enabled to bear the stimulus of oxygen air every day better and better.

Confumption.

THE various states of confumption, or phthifis pulmonaris, its different causes, and the difficulty of discerning a real phthifis from 'certain other disorders, render the treatment of this difease frequently doubtful and perplexing. But its stubborn nature, and the frequency of the disease, demand the utmost attention, and all the affistance which philosophy can suggest, and the medical art can apply. We shall therefore endeavour to state, how far the use of factitious airs has been found useful or promising in cases of this nature.

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It has been faid on one fide, that the factitious airs have the power of arrefting the progrefs of confumption, and often of accomplifting a perfect cure; but on the other hand it has been afferted, that they have never afforded any permanent benefit, and that they have often produced manifest harm. It appears, however, from a difinteress examination of the cases, and from the testimony of patients as well as of practitioners, that both those affertions imply a considerable degree of exaggeration. The result of this examination will be found condensed in the following paragraphs.

THE diluted hydrocarbonate is the only one, or at leaft the principal aerial fluid that has been fuccefsfully administered in cafes of phthifis; and it has generally afforded a fensible and almost immediate relief, by abating the hectic fever, by diminishing fensibility, by promoting sleep, and by reducing the quantity of expectoration.

BUT the use of hydrocarbonate is always attended with a diminution of strength. Hence,

Hence, when the patients are very feeble, which is generally the cafe in an advanced ftate of the diforder, the difadvantage which arifes from the diminution of ftrength, is greater than the advantage which arifes from the other good effects of the hydrocarbonate. When the patients, therefore, are too far gone, the ufe of the hydrocarbonate produces an apparent but not a real melioration.

It is on the fame account that this elaftic fluid cannot be administered to patients, that labour under great weakness of the digestive organs. In such cases the vapour of ether is, perhaps, the only elastic fluid that may be tried with safety; and the use of it is attended with at least a temporary relief.

THERE are two or three cafes of real phthifis creditably related, where a perfect cure feems to have been performed; though in a great many others the application of aerial fluids proved evidently ufelefs. But though

though from those few successful cases no great expectations can be derived, yet in a difease where no remedy has ever been found efficacious, surely it is not improper to try an application which at least affords a ray of hope.

THE quantity of diluted hydrocarbonate, which may be administered daily, is various, according to the conftitution of the patient. It is proper, however, to begin by adminiltering one pint of hydrocarbonate with between twenty and thirty pints of common air; and the quantity of the former may, in procefs of time, be increased conformably to the effects. In breathing this quantity of elastic fluid, it will be proper to let the patient reft four or five times, or in fhort whenever any giddinefs happens to come on; for this giddinefs or vertigo generally goes off in two or three minutes, after which the patient may again apply his mouth to the bag or veffel which contains the diluted hydrocarbonate.

It will be found, that cuftom habituates the lungs to bear the hydrocarbonate, in an increafed proportion, as far as a certain limit. Thus the fame patient who at firft was made vertiginous by a quart of hydrocarbonate, diluted with twenty quarts of common air, will, in process of time, be hardly affected by the double of that quantity.

THE inhalation of the vapour of ether, as alfo of other forts of reduced atmospheres, fuch as a mixture of azote and common air, of carbonic acid gas and common air, of hydrogen and common air, have been of partial use; however, the mixture last-mentioned feems to have proved more beneficial than any of the rest. This fort of reduced atmosphere must be administered more freely than the diluted hydrocarbonate. The vapour of ether may be inhaled with the utmost facility, as no apparatus is required for it, and it will be found at least of temporary use for allaying the cough, the pain, &c.

THOUGH

120

THOUGH the use of reduced atmospheres be more promising in cases of incipient phthis, yet that application should not be neglected in any state of the disorder; fince the elastic fluids are the only remedies which can be applied immediately to the part affected.

OF the various fpecies of phthifis pulmonaris, two only, viz. the chlorotic and the fyphilitic, feem to require a different treatment, and I find a few cafes in which fyphilitic ulcers in the lungs are faid to have been cured by the ufe of diluted oxygen, which was breathed once a day; but this treatment was accompanied with mercurial and other medicines; which, however, when adminiftered by themfelves, had produced no good effect.

THE inhalation of carbonic acid gas is faid to have proved beneficial in hectic diforders, but I do not know how far this practice may be fafe or useful, as I do not find any very particular information concerning it.

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Coughs.

Coughs.

OF the various fpecies of cough, those which originate from catarrh and from phthis have been already mentioned under those articles, to which the reader is referred. But with respect to the application of factitious airs to other species of cough, I do not find much authentic information, and of course must leave it for future investigation.

Debility.

AN univerfal debility is not unfrequently met with amongft perfons of all ages, and efpecially among women. It is fometimes the unconquerable effect of former diforders that are fubdued, or of lurking and invifible caufes. Whatever its origin may be, the fymptoms it produces are numerous and often of the utmost confequence. It produces palenefs, emaciation, difficulty of breathing, palpitation, indigestion, loss of streathing, palpitation, fwellings of the extremities,

ONDON

tremities, weaknefs of fight, lofs of voice, fuppreffion of the ufual evacuations, &c. Thofe fymptoms, of which a greater or lefs number is to be obferved in the fame individual, are, at first, the confequence of the debility, but they foon become the fomenters of that very languor, and confequently of each other.

WHEN the difeafe, which produces the languor, is prefent and known, I need hardly mention that the removal of that caufe should be the first object of the practitioner. But when that is not the cafe, diluted oxygen air may be administered with great hopes of fuccefs; for fuch treatment has been found beneficial in a great many cafes of this fort, and wonderful cures have been performed where no other remedy was found efficacious. The improvement is perceived fooner or later, according to the nature of the cafes; but, generally fpeaking, it becomes manifest in about a fortnight or three weeks time. It operates by ftrengthening and improving the whole habit. The pulse K 2

132 MEDICINAL PROPERTIES of pulfe becomes ftronger, the afpect acquires colour, the laffitude after exercise goes off gradually, the appetite is improved, and the reft of the fymptoms disappear gradually.

IT has been repeatedly observed, that in cafes of this fort the quantity of oxygen must be nicely regulated by the strength of the patient. If too fmall a quantity be administered, little or no improvement will be obtained; but if the quantity be too great, the effects will be hurtful, and fome of the bad fymptoms are thereby increafed. A few days experience will foon indicate the proper dole of oxygen. But I would recommend to begin by giving one quart of oxygen, with twelve or fifteen quarts of common, air, per day, and to increase or diminish the quantity according to circumstances. It must, however, be remembered, that when a fenfible improvement becomes manifest, more advantage is to be derived from a moderate dofe regularly administered, than from an increased proportion of oxygen.

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I DO not find that the leaving off of this application has ever been productive of any harm, but at all events it may be not improper to relinquish it by degrees, viz. by diminishing the quantity of oxygen, and intermitting the application by the interval of a day or two.

Digestion impaired, or Dyspepsia.

WE have not a clear account of the various fpecies of dyfpepfia to which the aerial fluids have been applied, nor indeed have they been tried in a great variety of cafes. But upon the whole it appears, that when debility is the caufe, and efpecially when it is accompanied with what is called a nervous head-ach, the inhalation of diluted oxygen air has been of fingular ufe, and the diforder has been frequently removed in a fhort time.

IF the impaired digeftion be accompanied with other fymptoms of debility befides the head-ach, the administration of K 3 oxygen

oxygen may be regulated agreeably to what has been mentioned in the preceding article, otherwife a greater proportion of it may be administered, as about four or five, or fix pints of it, with between twenty and thirty of common air. In cases of this fort, the good effects of the oxygen may be perceived in the course of a few days.

THIS treatment has proved peculiarly beneficial to fuch perfons as have contracted a weaknefs of digeftion, from having been confined in the foul air of workshops, counting-houses, &c.

I FIND, likewife, the cafe of a man who had been afflicted for upwards of five years with heart-burn, flatulence, lownefs of fpirits, and coldnefs of the extremities, which feemed to indicate a bad digeftion, and who was perfectly cured by the inhalation of diluted oxygen, and by drinking water impregnated with carbonic acid gas, together with fome falt of fteel.

Dropfy.

Dropfy.

IN a variety of dropfical cafes the inhalation of diluted oxygen air has been attended with fuccefs, and this fuccefs has feveral times amounted to a complete cure. This treatment feems to be more efficacious in an incipient dropfy, and when the diforder is confined to the extremities, than in other states of it. Yet I find a remarkable cafe of hydrothorax which was effectually cured, though a fimilar one was not attended with the fame effect; and likewife a cafe of water in the head of a boy of thirteen years, which is faid to have been partially removed by the inhalation of diluted oxygen air.

ONE quart of oxygen, and about fifteen of common air, per day, may be fufficient for the beginning; but the proportion of oxygen must be increased in the course of three or four days (provided no bad effects enfue) to two quarts; and foon after it will be K 4

136 MEDICINAL PROPERTIES of be proper to double the quantity both of oxygen and of common air.

THIS regimen must be perfisted in for weeks and months, according to the nature of the cafe; and should any inflammatory symptoms appear in the course of this application, the inhalation of the diluted oxygen may, in that case, be suspended or moderated for two or three days.

Eruptions.

I FIND a few cafes of fcorbutic eruptions on the face, as alfo on other parts of the body, in which a complete cure was accomplifhed by the daily inhalation of diluted oxygen air.

In cases of tumors and eruptions, which derive their origin from debility and a poor or thin state of the blood, the like treatment has been found beneficial.

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Two or three pints of oxygen, with about ten times that quantity of common air, per day, is fufficient for the beginning; but in cafes of this fort, the proportion of oxygen should not be much increased.

I NEED hardly add, that in fuch cafes it is proper to continue the ufual dreffings of the parts affected, the means of keeping the body gently open, &c.

Fevers.

I DO not know whether the factitious airs have been tried with fuccefs in other forts of fever, befides the putrid and the hectic. With refpect to the latter, the reader may confult what has already been faid under the articles of *chlorofis* and *confumption*. But in putrid fevers the carbonic acid gas is generally allowed to be an ufeful remedy; yet the ufe of it has not proved fo generally beneficial as it was at first believed.

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137

THIS gas is applied internally, not to the lungs but to the ftomach and inteffines, in three different manners, viz. by way of clyfters, either in the aerial form or in combination with water; by way of drink when combined with water; and, laftly, by giving through the mouth fuch fubftances as contain carbonic acid gas in abundance, that is liquors in a ftate of fermentation, certain fruits, &c.

WHEN a large quantity of it is given either in the aerial form or in combination with water, the abdomen is frequently diftended by it; for though this gas is pretty eafily imbibed by animal fluids, the fluids which it ufually meets with in the flomach, &c. are feldom capable of abforbing more than a moderate quantity of it. However, the differition of the abdomen is not fo very detrimental, but it may be fupported to a certain degree.

WHAT feems to render the carbonic acid gas not fo efficacious in cafes of putrid fevers,

fevers, as from its ufual properties one might be led to expect, is the difficulty of its infinuation into the vafcular fystem of the whole body. The lacteals imbibe it in fmall quantity, and the difficulty becomes greater in certain states of the diforder; whenever, therefore, the difeafe is not in a very alarming state, viz. fo as to give time for the infinuation of the carbonic acid gas into the fluids of the body, then more benefit is to be expected from it. There are, however, fome cafes in record, where the free use of carbonic acid gas proved efficacious in the worft state of putrid difeafes; and I do not find that it was ever attended with noxious effects.

OF the various fubftances which are administered in putrid diseases, on account of the carbonic acid gas which they contain, the following are the principal ones, viz. effervescing alkaline and acid mixtures, confisting of a folution of falt of tartar, to which lemon juice, or diluted vitriolic acid, or diluted nitrous acid, is added the moment

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ment before it is to be drank; fweetwort, or an infufion of malt, yeaft, and certain acidulous fruits, fuch as oranges, lemons, &c.

I SHALL not attempt to define the circumftances in which one or other of those articles may be preferable, nor is it neceffary to limit the dofes. The circumstances of fuch cafes being very numerous and diverfified, must be left to the skill of the attending practitioners. If the carbonic acid gas be administered in the aerial form, the quantity of it can hardly be too great, provided it does not diftend the abdomen too much ; but if the gas be given in combination with other fubftances, the quantity of those other substances must be limited, not by the quantity of carbonic acid gas that may be contained in them, but by their other properties, which must be proportionate to the state of the patient.

IT has been proposed (not without expectations of fuccess, though with difficulty of

of execution) to introduce, in certain cafes, the whole body of the patient, the mouth excepted, into a veffel full of carbonic acid gas; for as this gas is abforbed by the pores of the fkin, a greater quantity of it might thereby be imbibed.

Head-Ach.

THE various origin of this diforder, and the fmall number of cafes that are circumftantially related, prevent our forming a comprehensive idea of the use of factitious airs in cafes of this fort. The inhalation of diluted oxygen air, has fometimes been of use in what is commonly called nervous head-ach; and it appears that in fuch cafes, a very great proportion of oxygen air has been administered, even as much as five or fix gallons per day. I would not, however, recommend fo free a ufe of it.

In head-achs that arife from a weak digeftion, the inhalation of diluted oxygen air is an useful remedy. See the article Digestion. Hæmoptyfis,

Hæmoptysis, or Spitting of Blood.

I FIND a few cafes of this diforder, in which the administration of hydrocarbonate gas was administered with good effect. The account of the most remarkable one will be found in the next chapter.

Ophthalmia.

IN cafes of ophthalmia, and weaknefs of fight, when accompanied with a relaxed habit of body, the inhalation of diluted oxygen air has been of fingular ufe. About two quarts (when a finaller quantity has proved ineffectual) of oxygen air, with about fifteen of common air, is a dofe fufficient for each day.

Phthifis Pulmonalis.

See Confumption.

Paralyfis.

Paralyfis.

I FIND a few cafes of that fpecies of paralyfis which is occafioned by prepararations of lead, the *colica pictonum*, where the inhalation of diluted oxygen air proved beneficial. Three or four pints of it, with about thirty pints of common air, is a dofe fufficient for each day.

Scurvy.

THE use of carbonic acid gas has long been confidered as a powerful remedy in fcorbutic diforders; and certain it is, that when the diforder is not too far advanced, a perfect cure may be generally expected from it; and even in cases of the worft fort, the free use of this gas has frequently accomplished a perfect recovery.

ALL the various ways of administering this gas, which have been mentioned for the cure of putrid fevers (See the article Fevers) 144 MEDICINAL PROPERTIES of Fevers) are applicable to this fort of diforder.

SINCE much has been written concerning the fcurvy, and fince the methods of administering such substances as contain abundance of carbonic acid gas, are generally known and fuccefsfully administered, I shall not detain the reader with long extracts from more able writers; but shall only add, that, whilft the carbonic acid gas is applied to the ftomach and inteffines, a moderate dofe of diluted oxygen air should be applied to the lungs by the way of refpiration; for whilft the former acts as an antifeptic, and corrects the putrid tendency, the latter gives energy and vigor to the fibre, and enables the body to throw off the morbid humours with greater quicknefs.

Stone in the Bladder, &c.

WATER impregnated with carbonic acid gas, has been long known to afford relief in calculous complaints of the bladder and

and urinary paffage. But by the addition of a fixed alkali, the remedy has of late been rendered much more efficacious in cafes of the above-mentioned fort, and even when a large stone has actually existed in the bladder. I do not know how far this acidulous foda water, as it is commonly called, may operate by way of a folvent of a large ftone; but certain it is, that even in those cases it affords confiderable relief, and it feems effectually to prevent the farther accumulation of the ftony matter, by diffolving the mucus as well as the finall ftony concretions, and washing them off from the kidnies, ureters, bladder, &c. It is, therefore, given in all complaints that originate from a thickening of, or deposition of gross matter by, the urine in the above-mentioned parts, fuch as strangury, pain in voiding the urine, ulceration of the parts, &c.

For this purpose one ounce of soda is diffolved in four or five pints of rain, or of boiled soft, water; and the solution is then impregnated, as much as possible, with L carbonic

carbonic acid gas *. Of this water, a pint a day is the quantity ufually given for the above-mentioned diforders, and it is to be drank not all at once, but at three different times, viz. morning, noon, and night.

INDEPENDENT of those difeases, the acidulous foda water is fuccessfully adminiftered in fcorbutic cases, bilious complaints, weakness of the digestive organs, fome nervous affections, &c. but in those the proportion of the alkali, as well as the daily allowance, must be diminissed according to the circumstances of the case.

Swellings.

I FIND one cafe only of a white fwelling of the knee recorded, in which a per-

* In Dr. Nooth's glass apparatus for impregnating water with carbonic acid gas, the quantity of gas that can be thrown into it is very moderate, yet efficacious; but the foda water which is now prepared and fold in London by a Mr. Schweppe, contains an incomparably greater proportion of carbonic acid gas, and accordingly is much more efficacious.

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fect cure is faid to have been accomplished by the daily inhalation of diluted oxygen air. As those fwellings owe their origin in great meafure to weaknefs of body, it is likely that the use of oxygen, which invigorates the animal fibre, may prove an ufeful remedy.

THE like treatment is faid to have been found fometimes useful in fcrophulous tumours.

Ulcers.

THE factitious airs have been frequently administered in cases of ulcers on different parts of the body, and especially on the legs; but the indifcriminate and injudicious application, which feems evident in many cafes, has been productive of equivocal effects. However, a careful examination of the particular circumstances shews, in agreement with the theory, that when the ulcers originate from a poor state of the blood, and a debilitated habit, the daily inhalation

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inhalation of three or four pints of oxygent air, with about ten times that quantity of common air, is of fingular ufe; and by this means fome ulcers of the worft kind, viz. painful, fætid, ftubborn, &c. and when they were accompanied with fcurfy eruptions over great part of the body, with want of appetite, &c. have been completely cured in about fix weeks time.

In ulcers of other fort the flate of the patient, as alfo the origin of the diforder, must be carefully attended to, and the elaftic fluids, when they may be thought ufeful, must be administered accordingly, otherwife they will produce more harm than good. In fact, I find a cafe of a scrophulous ulcer, where the oxygen air proved detrimental; but a mixture of oxygen, hydrocarbonate, and common air, accomplished the cure. This cafe will be found in the next chapter.

IN all cafes of ulcers, the external application of carbonic acid gas generally affords

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an alleviation of pain, as well as of the fœtor, and a better difcharge.

CHAPTER VIII.

Medical Cafes in which Aerial Fluids were administered.

COLLECTION of medical cafes, in , which the factitious airs have been administered with great fuccess, forms the content of the present chapter. Those cafes have been either extracted from other publications, or they have been communicated by intelligent friends; and they have been felected out of a great number, merely for the purpose of shewing the practical methods of administering the artificial elastic fluids. Such cafes, therefore, have been preferred, as by the variety of circumstances feemed more likely to manifest the modes of applying, proportioning, varying, and fufpending L 3

150 MEDICINAL PROPERTIES of fufpending the administration of the factitious airs.

CONVINCED that the unfkilful application of this new fet of remedies has produced confiderable harm, and has thrown a degree of difcredit on the practice, I take the liberty of warning the practitioners againft drawing hafty conclusions from a few crude, and, in all probability, ill-managed cafes. For however fkilled those gentlemen may be in other branches of physic, it is at least likely that in this new application their management of patients may not be generally correct; and of course the failure is not always to be attributed to the want of power in the aerial fluids.

CASE I.

Communicated by Dr. J. Lind, of Windfor.

THE first time I applied the modified atmosphere as a remedy, was in the case of an officer of the Excise, who, during the fevere

fevere weather of January, 1797, being much exposed to the cold in the exercise of his duty, had got a violent cough, which caufed the rupture of a confiderable blood veffel in his lungs, and this was foon followed by the fymptoms of a rapid confumption. On the 25th of January he first applied to me, when I ordered him an infusion of rofes, acidulated with vitriolic acid, and fmall doses of ipecacuanha, to stop the hæmoptoe; and for the cough and hectic fever I recommended him to breathe, feveral times in the course of the day, the vapour of vitriolic ether, in which the powdered leaves of cicuta were infused, after the manner recommended by Dr. Pearfon, of Birmingham. The benefit which he received from this application was really remarkable, for after not more than four or five days, almost all the bad fymptoms were wonderfully diminished; yet, finding that he got hardly any fleep at night, and that he had been a bad fleeper for above a year, I made him inhale about a quart of hydro-. carbonate gas, diluted with about fourteen quarts

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quarts of common air, at bed-time, which procured him an uninterrupted night's reft, fuch as he had not experienced for many months before. He continued to follow the fame courfe till the 20th of February, when his health being perfectly reftored, he returned to his duty.—N. B. When he inhaled the diluted hydrocarbonate, he drew it in over the fleam of hot water, with the inhaler which I have conftructed on the plan of Mr. Watt's refrigeratory *.

CASE II.

Communicated by the fame.

I TRIED the diluted hydrocarbonate likewife with fuccefs, in an inflammation of the lungs. The patient was a man of fixty-fix years of age, labouring under an inflammation of the lungs, but attended

* This inhaler is made of japanned tin, and being filled with hot water, is interposed between the bag or receiver of air and the mouth of the patient, fo that the air is heated by the vapour of hot water in passing through it.

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with fo fmall a pulfe that bleeding did not appear advifeable. I therefore directed him to breathe the hydrocarbonate gas, diluted nearly in the proportion mentioned in the preceding cafe, which he did every night, and occafionally whenever the pain returned. The effect of the modified air was immediate, and very remarkable, for not only the pain was removed, but he ufed to fay that the hydrocarbonate had deprived him of his body, and had left him only his head; fuch was the diminution of irritability which this gas is capable of producing.

THE diforder vanished in a very short time: for in seven days from the commencement of the application, his health was perfectly restored.

THIS cafe shews that in inflammations of the lungs, when the pulse is weak, which is fometimes the cafe, the reduced atmospheres are, perhaps, the only application practicable.

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CASE III. Related by Dr. Carmichael.

Birmingham, March, 1795.

I.B. æt. 45, was attacked about four months fince with difficulty of breathing, attended at times with pain under the sternum, and commonly with a fense of tightness of the thorax, frequent cough, with copious expectoration of a tough whitish fluid, pulse 96, body regular, appetite variable. He has feldom paffed four and twenty hours without a material aggravation of all his fymptoms. Was first attacked with this diforder fix years ago, and has regularly fuffered very feverely from it every winter fince that period; it has always left him about the beginning of May, and he has kept free from complaint during the fummer and autumn months. He has tried many remedies, but never with more than very transitory relief.

FEBRUARY 14th, 1795, I directed him to inhale daily a mixture of hydrocarbonate and

and atmospheric air, in the proportion of 1 to 19.—15th, No fensible effects from the use of the hydrocarbonate; the strength of the mixture was therefore increafed in the proportion of two to 18 .- 16th, No vertigo, nor any other fensible effect, produced by the use of the modified air. The proportion still farther increased to 4 to 18 .- 17th, Confiderable vertigo produced by yesterday's defe, which returned at intervals, attended by head-ach during the day. Breathing much relieved, even during the act of inhaling the modified air, and has fince continued tolerably eafy. Slept better last night than he has been accustomed to do for fome months. - 22d, Hydrocarbonate continues to produce confiderable giddiness; breathing, except some short intervals of flight return, continues much easier. Cough less frequent, expectoration much diminished. Continues to enjoy comfortable fleep .- 27th, Had a confiderable return of difficulty of breathing on the afternoon of the 25th, which, however, abated fo much before his usual bed-time.

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as not to prevent him from paffing the night comfortably. Cough infrequent, and rarely attended with expectoration. Has for fome time paft had no pain under his sternum, and rarely any fense of tightness of his thorax. -March 4th, He is in every respect fo much better, that he intends to return to his usual occupation (making moulds in a caft-iron foundry) on the 9th inft. Modified air continues to produce vertigo .---March 9th, He continued without any return of his complaint, and returned to his employment as he intended; but after working for a few hours only, he was obliged to defift, by a return of the fenfe of tightnefs on his thorax, and confiderable difficulty of breathing. Breathing increafed in difficulty towards evening, and still continues, attended by frequent dry cough .---13th, Continues to breathe with confiderable difficulty, pulse 100; fleepless nights; cough more frequent, but now attended with confiderable expectoration. -17th, Difficulty of breathing continued until yefterday; has paffed a better night than ufual;

157

ufual; and this morning finds himfelf much better.—20th, Breathing continues eaffer; cough much lefs frequent, and quantity of expectoration diminifhed. Has flept for fome nights paft comfortably; pulfe 86. Modified air continues to produce confiderable vertigo.—29th, Continues uniformly to recover; his cough is very trifling, and he expectorates better; his ftrength is fo much improved, that he can ufe confiderable exercife without inconvenience. Sleeps uniformly well.—He returns to his work to-morrow, but for the prefent is to work within doors. He is of opinion that he is in every refpect equal to the undertaking.

CASE IV.

Related by the Same.

J.T. æt. 40. has for two years paft been affected, during the winter and fpring months, with cough and expectoration, and at times with pains in his breaft, accompanied with flight dyfpnœa. Thefe fymptoms,

toms, in general, left him during the fummer months, and never at any time arole to, fuch a degree as to prevent him from following his ufual occupation. In the beginning of October laft he was feized with pain on his fide, cough, dyipnœa, and after fome time with copious expectoration. He applied for my advice in the beginning of November. At that time he had an almost inceffant cough, attended with copious expectoration; he complained of a fenfe of tightnefs acrofs his thorax, and much dyfpnœa on the flightest exertion; his pulse was in general from 110 to 120, his nights were reftlefs, and attended with profuse perfpirations, his body was irregular, his appetite much impaired, his frame much emaciated. I ordered for him, at different times, emetics, squills, ammoniacum, blifters, &c. but from none of them did he derive more than a very temporary relief .---November 27th, he began the use of the hydrocarbonate; I directed him at first to inhale a mixture containing a quart and a half of this species of factitious air, and nineteen

nineteen of atmospheric air. This quantity he used in about twenty minutes, breathing it for twenty feconds together, and then refting for one, two, or three minutes, according to the degree of vertigo produced. -28th, The vertigo produced by yesterday's inhalation was very fevere, and returned at intervals during the evening. He has paffed a much better night than usual, and fays that the dyfpnœa and fense of stricture on the thorax are much relieved. The quantity of hydrocarbonate diminished to one quart, diluted as above .-- 30th, Cough much relieved, sense of stricture gone, dyspnæa less troublesome on motion, has had better nights, and his perspirations are less profuse; pulse 106, appetite rather better .--December 7th, Cough evidently better, expectoration confiderably diminished, pulse 95, body for fome days past regular; breathing fo much improved that he can with eafe walk up flairs to his chamber and undrefs himfelf, without return of dispnœa, which he could not before accomplish without the greatest difficulty; fleeps better than he

159

he has done for months paft, perspirations entirely left him, appetite mended .--- 15th, Continues to recover in every refpect, has at times some return of tightness on his breaft, but which is uniformly relieved or completely carried off by the hydrocarbo= nate. His countenance is evidently altered for the better, and he is of opinion that his strength returns. Notwithstanding that the modified air still continues to produce confiderable vertigo, I increased the quantity to two quarts, diluted as before .- 27th, Cough very much relieved, expectorated matter reduced to one-third of its former quantity, pulse from 84 to 90. He has evidently acquired flesh, and he is of opinion that his firength continues to improve .----January 6th, 1795, Cough rather more frequent, and attended with fome degree of difpnœa. On account of the feverity of the weather, which evidently affects him, I ordered him not to ftir from home. At this time he began to breathe the modified air, of the ftrength directed above, twice a day. -16th, Cough relieved, quantity of expectorated

torated matter much the fame as reported on the 27th ult.; in other respects the fame .- February 1st, On account of the unufual feverity of the weather, no advance has been made fince last report. Cough more variable, and at times attended with fome degree of difpnœa, expectorated matter increased, he does not, however, emaciate. --- 12th, Cough much abated, quantity of expectoration reduced to one-fifth of its former quantity, his strength is fo much recruited that it is with difficulty I can reftrain him from returning to his occupation. In every refpect he is much better .- March 1st, continues to gain strength, cough less frequent, and expectoration still diminishing in quantity, appetite good, fleeps well. As I could not prevail with him to remain longer at home, I advifed him, before he returned to his usual occupation, to walk out a little daily.

HE complied with my advice, and continued to gain ground till the 9th of that month, when in the evening he was feized M with

with the usual fymptoms of the influenza, an epidemic catarrhal infection, which at that time prevailed much in this place. The febrile fymptoms ran high, and were attended by frequent cough and confiderable pain on his fide; he complained alfo of fevere head-ach, and unufual langour; he was thirfty; his tongue was white, and his pulse 110.-March 15th, febrile symptoms continue; cough frequent, but now attended with increased expectoration; pain of his fide less fevere; confiderable dispnæa on the flightest motion; pulse 115, fmall and weak. Until this attack he inhaled twice daily a gallon of hydrocarbonate, diluted with four times the quantity of atmofpheric air, but, as his ftrength wafted, it was found neceffary to leffen the quantity to one quart diluted as above .- March 20th, pain of his fide fomething eafier, but his cough is increafed in frequency, and his expectoration more copious. Reftless nights; no appetite; ftrength fo much impaired that, for the laft four days, he has not been able to inhale the modified air; pulse 120. I directed

I directed a warm ftimulating plaister to be applied to his fide, and five drops of tinctura opii to be given every four hours .- March 28th, pain of his fide gone, but his other fymptoms continue; bowels regular; has, had better nights, but his fleep has been attended with profuse perspirations; the tinctura opii was omitted, and he was directed to take at nearly the fame intervals afmall glafs full of port wine .- April 15th, complaints continue without material alteration. From this date he re-commenced the use of the hydrocarbonate, beginning with it of the ftrength of one pint to fixteen quarts of common air .- April 25th, at first the modified air occafioned confiderable vertigo, but he foon became fo much habituated to its operation that the quantity was increased to one, and afterwards to two, quarts. His perspirations have abated, his cough has been lefs urgent, the quantity of his expectoration has diminished, and the difpnœa, with which he has for fome time past been troubled on the flightest motion, is greatly alleviated.

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163

MARCH 3, Since the laft report he has experienced confiderable amendment, pulfe 98. The quantity of hydrocarbonate was further increafed to a gallon, diluted with four times that quantity of atmospheric air. May 15th, he has continued to recover fo much in every respect, that yesterday he was able to walk fourteen miles into the country.

FROM this time I did not fee him till the middle of June, when he returned to this place with an intention to follow his usual occupation. He was in every particular fo much better, that he feemed to have recovered his health completely. I advifed him, however, to the contrary, to which he confented, and he has fince been occupied in hay-making, and more lately in reaping. I faw him a few days ago; he cannot be faid either to cough or expectorate, except in the morning, and then in the most trifling degree, and his ftrength is fo completely restored, that he has been earning wages equal to those of the stoutest of his fellowlabourers,

labourers, with both ease to himself and fatisfaction to his employer.

CASE V.

Related by Dr. W. Pearfon.

ELIZABETH VYSE, aged 27, having been feized at the end of autumn with cough, fever, and fpitting of blood, applied at the hofpital for relief, and came under my care last October. She informed me she had been subject to a cough for three winters. She had a quick and finall pulfe, flushed cheeks, dyspnœa, pain on the fide, conftant cough attended with copious expectoration, and night-fweats. She was very feeble and much emaciated. The hæmoptoe was foon removed by the medicines commonly prefcribed in fuch cafes; but the fymptoms continued. I therefore ordered her, on the 12th of November, to breathe the vapour of vitriolic æther, impregnated with extract of cicuta two or three times a day. On the 19th, when I faw her again, fhe informed me that fhe had obtained great relief M 3

relief from the æther vapour, having much lefs tightnefs acrofs the cheft, and lefs pain of the fide. She faid fhe was fomewhat giddy after every inhalation.—December the 3d, lefs fever, lefs cough, and confiderably better in every refpect. Has found more benefit (to ufe her own words) from the æther application than from any thing elfe.—December 10th, cough and other complaints fo flight, that fhe fays fhe does not require any more medicines.

N.B. During the use of æther vapour, she took a decoction of Peruvian bark and farsaparilla, and pills composed of extract of cicuta and rhubarb.

CASE VI.

Related by Mr. Barr. Birmingham, October 9, 1795.

MR. BARROR, of Barton-under-Needwood, being in this town on a vifit to a friend in the fpring of 1793, was feized with an highly inflammatory fever, attended with

with a violent pain of the fide. This fever was followed by a dry tickling cough, a fense of tightness in breathing, much languor, and a great degree of reftleffnefs and anxiety. His bowels felt full, tenfe, and uneafy; his pulfe intermitted; and he complained that his urine, though nearly in the usual quantity, did not flow freely, and that he had always the fenfation of not having evacuated the whole. Blifters, boluffes of triturated mercury, and a decoction first of Peruvian, and afterwards of Angustra, bark were prefcribed. He was relieved by thefe medicines, but he neither recovered his ftrength nor his fpirits. In this fituation nearly he paffed the remainder of the year in the country; in the fpring of 1794 he came to Birmingham again, with all the fymptoms of his diforder increased, particularly the oppreffion in breathing. He could neither lie down in bed with comfort, nor afcend the fmallest aclivity without the greatest uneafiness. His urine was diminished in quantity, and voided with difficulty. A decoction of Seneka root, and finall M 4

finall dofes of Digitalis were directed and continued for two or three weeks; but they rather feemed to amufe than relieve him. He called on me again laft April, and told me that all medicines had loft the power of relieving him, that his breathing was now more generally difficult, that his urine was very fcanty, and that his appetite was entirely gone. I prescribed the Digitalis with a itter infusion. He went into the country and continued thefe medicines for fome time. Towards the end of July he called upon me again-but, alas! how changed ! His face was now become pale and emaciated, his eyes flared as if taking a laft confcious view of their object; which laft circumstance much alarmed his friends .--His legs were fwelled to fuch a degree that the fkin was become much inflamed, and in danger of burfting. He had a continual tenefinus, and made very little urine; he could not endure an horizontal posture for a moment, but was under the neceffity of being bolftered upright in bed through the night; even then he flept little, and that little

169

it

little was difturbed and unrefreshing, for he frequently started from his sleep under an impression of instant fuffocation.

HAVING feen an account of the happy relief Sir William Chambers had experienced from oxygen in a fimilar fituation, I wrote to my patient, and advifed the adoption of the pneumatic plan. I did this, I confefs, in the prefent inftance, with little hope of advantage; but as the most powerful medicines had produced no falutary effect, I felt it my duty to him, as well as to the cause of humanity, to urge this compliance. I procured him a reading of the cafe, and the fimilarity of the circumstances was fo ftriking, that he agreed to place himfelf immediately under my care.

HE arrived here on the 12th of August, and began to inspire the factitious air on the 13th. I directed one quart of oxygen mixed with nineteen of atmospheric air, to be inhaled every day; but as the symptoms were become extremely urgent, I thought

4

it right to join the use of those active medicines that I had prefcribed for him before. Accordingly I directed him to take half a grain of Digitalis in fubftance, every evening, and four ounces of a decoction of Angustra bark in the course of each day. On the third night after infpiring he found himself more composed, he could remain longer in one pofture, and the ftartings, during fleep, feemed both lefs frequent and lefs violent. Every night he was fenfible of amendment; in ten days he could bear the removal of feveral of the pillows that bolftered him up in bed; and he could fleep for three or four hours without one flarting fit. The fwelling of his legs too began now to fubfide; the tenefmus was entirely removed; the quantity of urine was much increafed, and he could walk up ftairs with much eafe; his appetite and cheerfulnefs began to return, and the pale face of difease to give place to the florid countenance of health. In the courfe of the fecond week I had gradually increafed the quantity of oxygen to two quarts a day, diluted as before.

fore. In four weeks from his beginning to infpire the vital air, not a veftige of the diforder remained, except weaknefs; he could lay his head as low in bed as when in perfect health, and fleep the whole night; no fwelling of the legs remained ; no difficulty of breathing upon ordinary exertion; and every function was performed with regularity and eafe. He then went home provided with a pneumatic apparatus, and directions how to use it, and laid afide the use of all medicines except a laxative pill occafionally. He paffed through this town yesterday in perfect health. His strength, agility, and vivacity, are greater than in most men of his age (60).

CASE VII.

Related by Dr. Alderson.--Hull, June 5th, 1795.

MISS ——, aged 16, had all the fymptoms of approaching phthifis, cold tremors about twelve o'clock; fever, heat, and flufhing

ing every afternoon, pulse 120, countenance uncommonly florid, breathing rather difficult, cough fevere, accompanied with expectoration; as feveral of her family had died of confumption, there could be little doubt of the tendency of these fymptoms; and after finding nitre, spermaceti, vomits, &cc. to have no good effect, I advised the inhalation of hydrogen air. She therefore daily inhaled a quart of pure hydrogen from water, by every now and then taking an infpiration at the mouth-piece of the tunnel. It frequently occafioned naufea and even vomiting. The pulse fell, the fluthings and fever fubfided, and the whole train of phthifical fymptoms left her, but at the expence of her fine florid colour, her countenance having ever fince been of a darker tint than before fhe was ill.

CASE

CASE VIII.

Related by Mr. Barr.—Birmingham, 14th March 1795.

ABOUT four months ago, a gentleman of this neighbourhood applied to me for advice in the management of a fcrophulous ulcer of confiderable extent. He had tried various remedies, but had derived no lafting advantage from any of them. When I first visited him he was worn down by a long courfe of night-watching. The deep-feated pain of the arm was fo conftant and fevere, that it had in great measure deprived him of fleep. His countenance was pale and fickly; his limbs were continually afflicted with aching pains; every exertion, even the most gentle, feemed beyond the measure of his strength, for his body had loft much of its active power, and his mind much of its wonted energy. The difcharge from the ulcer was copious, thin, bloody, and corrofive; and befides, the whole

173

whole furface of the fore was fo exceedingly irritable, that the mildeft dreffings, applied in the gentleft manner, produced very fevere and lafting pain. During the first fix weeks of my attendance, he regularly took as much Peruvian bark in fubstance as his ftomach and bowels could bear; and the ulcer was dreffed with various emollient, fedative, and aftringent applications, but without any permanent advantage. I then recommended a trial of oxygen air, which was readily complied with. He began by infpiring four ale quarts diluted with fixteen of atmospheric air twice a day, and gradually increased the quantity of oxygen to a cubic foot and a half in the day; by purfuing this plan for about a month, his health was wonderfully improved, but the ulcer shewed no disposition to heal. The deep-feated pain was now entirely removed, but in the fpace of a few days more he complained of a burning fenfation over the whole furface of the fore, fimilar to the pain arifing from erifepelatous inflammation. This unpleafant fenfation first commenced

menced after infpiring the whole quantity of oxygen in the fpace of two hours, which before had been taken in equally divided portions morning and evening. We ftill purfued our plan, thinking that this new pain might be owing to fome accidental circumftance, "and that it would foon pafs away. But it every day continued to increafe, and the ulcer began to fpread wider and wider. The edges became thick, and were turned outwards, and the difcharge became more thin and acrid.

In this fituation a local application feemed proper. I wifhed to have applied hydrocarbonate externally to the ulcer, but this, from fome circumftance of the cafe, was not practicable. I then thought to moderate the ftimulus of the oxygen by a mixture of hydrocarbonate, which Mr. Watt told me would occafion no chemical change in the two airs. Accordingly a mixture of three parts of oxygen, and one of hydrocarbonate, was preferibed. Four quarts of this mixed air were added to about fixteen of atmospheric,

ric, and this quantity infpired morning and evening. In lefs than a week the burning fenfation was much diminished, and the ulcer put on a more healing appearance. The mixed air was then increased to five quarts, and used as before, which produced an increase of all the pleasant symptoms. After a few days trial of this proportion of the mixed air fix quarts were prescribed. This is the quantity now inspired morning and evening.

My friend, at prefent, enjoys good health and a good appetite, and feels himfelf as ftrong as at any former period of his life. The ulcer is now reduced to lefs than half its original fize, and healing rapidly. There is neither fuperficial nor deep-feated pain remaining, and the action of the contiguous mufcles is free and eafy.

CASE

CASE IX.

Related by Dr. Redfearn, Lynn, Norfolk, June 26, 1795.

MR. B. F- æt. 23, of a florid complexion, narrow cheft, prominent shoulders, finooth skin, and of a delicate slender form, has been afflicted with hæmoptyfis about two years and a half, attended with dyfpnœa, cough, a difagreeable fense of burning in the cheft, and expectoration of a purulent nature. Pulse about 100, and invariably accelerated by the hydrocarbonate air. The hectic fever was not completely formed, but he had at times a fenfe of chillinefs in the day-time, with heat towards the evening. He began by taking one quart of hydrocarbonate, diluted with twenty-one quarts of atmospheric air, once a day. From this mixture he experienced much vertigo during its inhalation, and two hours after dinner he fuddenly became vertiginous, from which, however, he foon recovered, although N

178 MEDICINAL PROPERTIES of although a violent head-ach remained during the reft of the evening.

THE following days he only inhaled one pint of hydrocarbonate mixed with twenty quarts of common air, once a day, which generally affected him with fome flight vertigo and tightnefs over his forehead; the hydrocarbonate was increafed gradually to two quarts or more at one dofe, but I find it always neceffary to begin with the original dofe, where the air has been recently generated.

My patient has been perfevering in this plan about three months, and has had no return of the hæmorrhage; his cough and expectoration are very much diminifhed; fometimes he does not expectorate more than one table-fpoonful in the fpace of three days; he has alfo never experienced any of the diftreffing heats in his cheft, which baraffed him before the adminiftration of the air; his dyfpnæa is perfectly removed; he can ride upon horfeback twelve miles

iniles without feeling much fatigue; his appetite is very good, and he fleeps well; pulfe 80; he fays he thinks his health is perfectly eftablished.

CASE X.

IN the month of June 1797, a lady began the use of vital air, for an entire loss of voice, which misfortune she had fustained about three years. Her constitution was extremely nervous; had been the greatest part of her life subject to deplorable spafmodic affections, particularly in the organs of respiration, on any trifling exertions of exercise, or in a confined atmosphere; she had been long habituated to an uncommon quantity of opium, to suspend the frequency and violence of her attacks, and many means had been tried in vain to diminish materially the quantity she found necessary for her support.

SHE commenced the use of the vital air under the difadvantage of remaining in N 2 London

179

London during the fummer, a circumftance which was likely to be attended with great aggravation of her complaints, being contrary to her ufual practice, and, in fact, fhe had already begun to experience extraordinary fymptoms of debility from the attempt.

ABOUT three quarts of oxygen, with twelve quarts of common air, were adminiftered daily for about a fortnight, the effect of which was, that a flight degree of tightnefs about her cheft came on foon after the inhalation of the air, and generally went off in five or eight minutes time; the pulse was likewife rendered fuller though not more frequent, and the nights were often attended with a fort of reftlefinefs. In confequence of this last effect it was judged proper to diminish the quantity of oxygen air, and accordingly it was found, after repeated augmentations and diminutions of the dose, that about one quart or three pints of oxygen air, with about twelve quarts of common air, was the proportion of the acrial

aerial fluids, which feemed to agree beft with her conftitution. This application was perfifted in for upwards of five months, excepting fome flight intermiffions of a day or two occafionally, and it produced the most falutary effects. The whole habit of body began to be improved in about a month after the commencement of the application. The fhortness of breath vanished gradually, as also the symptoms of debility. Her afpect became healthy, and the voice improved by flow degrees, fo that by the end of October its tone was become fully equal to what it used to be previoufly to the illnefs. In fhort, this lady now enjoys a better state of health than she has experienced for many years.

A REMARKABLE circumftance was obferved in this cafe with refpect to the effect of opium, which is, that from long habit fhe had accuftomed herfelf to take an extraordinary quantity of opium daily, in order to fuftain her ufual exertions of the day; for, in fact, the opium produced in her rather a N 3 ferenity

ferenity of fpirits than drowfinefs; but after having inhaled the oxygen air for a few days, fhe found that fhe could do with lefs opium, and in procefs of time fhe further obferved, that the opium, initead of fupporting, difcompofed her fo much as to oblige her to diminifh the quantity of it to a very fmall part of what fhe had been accuftomed to take before the ufe of the oxygen air.

CASE XI.

Related by Mr. Hey, in a Letter to Dr. Priestley.

January 8th, MR. LIGHTBOWNE, a young gentleman who lives with me, was feized with a fever, which, after continuing about ten days, began to be attended with those fymptoms that indicate a putrefcent flate of the fluids.

18th. His tongue was black in the morning when I first visited him, but the blackness

blacknefs went off in the day-time upon drinking; he had begun to doze much the preceding day, and now he took little notice of thofe that were about him; his belly was loofe, and had been fo for fome days; his pulfe beat 110 ftrokes in a minute, and was rather low; he was ordered to take twenty-five grains of Peruvian bark with five of tormentill-root in powder every four hours, and to ufe red wine and water, cold, as his common drink.

19th. I was called to vifit him early in the morning, on account of a bleeding at the nofe which had come on; he loft about eight ounces of blood, which was of a loofe texture; the hæmorrhage was fuppreffed, though not without fome difficulty, by means of tents made of foft lint dipped in cold water ftrongly impregnated with tincture of iron, which were introduced within the noftrils quite through to their pofterior apertures, a method which has never yet failed me in like cafes. His tongue was now covered with a thick black pellicle, N 4 which

which was not diminifhed by drinking; his teeth were furred with the fame kind of fordid matter, and even the roof of his mouth and fauces were not free from it; his loofenefs and ftupor continued, and he was almoft inceffantly muttering to himfelf; he took this day a fcruple of the Peruvian bark with ten grains of tormentill every two or three hours; a ftarch clyfter, containing a drachm of the compound powder of bole, without opium, was given morning and evening; a window was fet open in his room, though it was a fevere froft, and the floor was frequently fprinkled with vinegar.

20th, HE continued nearly in the fame ftate; when roufed from his dozing, he generally gave a fenfible anfwer to the queftions afked him, but he immediately relapfed, and repeated his muttering. His fkin was dry and harfh, but without petechiæ. He fometimes voided his urine and fæces into the bed, but generally had fenfe enough to afk for the bed-pan. As he now naufeated

nauseated the bark in substance, it was exchanged for Huxham's tincture, of which he took a table-spoonful every two hours in a cup full of cold water; he drank fometimes a little of the tincture of rofes, but his common liquors were red wine and water, or rice-water and brandy acidulated with elixir of vitriol; before drinking he was commonly requefted to rinfe his mouth with water, to which a little honey and vinegar had been added. His loofenels rather increased, and the stools were watery, black, and fætid; it was judged neceffary to moderate this discharge, which seemed to fink him, by mixing a drachm of the theriaca andromachi, with each clyfter.

21ft, THE fame putrid fymptoms remained, and a *fubfultus tendinum* came on; his ftools were more foetid, and remarkably hot; the medicine and clyfters were repeated.

REFLECTING upon the difagreeable neceffity we feemed to lie under of confining this

this putrid matter in the intestines, left the evacuation should destroy the vis vita before there was time to correct its bad quality, and overcome its bad effects, by the means we were using, I confidered that, if this putrid ferment could be more immediately corrected, a ftop would probably be put to the flux, which feemed to arife from, or at least to be increased by it, and the fomes of the difease would likewise be in a great measure removed : I thought nothing was fo likely to effect this as the introduction of fixed air into the alimentary canal, which, from the experiments of Dr. Macbride, and those you have made fince his publication, appears to be the most powerful corrector of putrefaction hitherto known. I recollected what you had recommended to me as deferving to be tried in putrid difeases; I mean the injection of this kind of air by way of clyfter, and judged that in the prefent cafe fuch a method was clearly indicated.

THE next morning I mentioned my reflections to Dr. Hird and Dr. Crowther, who

who kindly attended this young gentleman at my requeft, and proposed the following method of treatment, which, with their approbation, was immediately entered upon. We first gave him five grains of ipecacuanha, to evacuate, in the most easy manner, part of the putrid colluvies; he was then . allowed to drink freely of brifk orangewine, which contained a good deal of fixed air, yet had not loft its fweetnefs. The tincture of bark was continued as before, and the water, which he drank along with it, was impregnated with fixed air from the atmosphere of a large vat of fermenting wort, in the manner I had learned from Inftead of the aftringent clyfter, air you. alone was injected, collected from a fermenting mixture of chalk and oil of vitriol : he drank a bottle of orange-wine in the courfe of this day, but refufed any other liquor, except water and his medicine ; two bladders full of fixed air were thrown up in the afternoon.

187

23d, H1s stools were less frequent ; their heat likewise, and peculiar factor, were confiderably diminished; his muttering was much abated, and the *fubfultus tendinum* had left him. Finding that part of the air was rejected when given with a bladder in the ufual way, I contrived a method of injecting it which was not fo liable to this inconvenience. I took the flexible tube of that inftrument which is used for throwing up the fume of tobacco, and tied a small bladder to the end of it that is connected with the box made for receiving the tobacco, which I had previoufly taken off from the tube; I then put some bits of chalk into a fixounce phial until it was half filled; upon thefe I poured fuch a quantity of oil of vitriol as I though capable of faturating the chalk, and immediately tied the bladder, which I had fixed to the tube, round the neck of the phial; the clyster-pipe, which was fastened to the other end of the tube, was introduced into the anus before the oil of vitriol was poured upon the chalk. By this method the air paffed gradually into . the

the inteffines as it was generated, the rejection of it was in a great meafure prevented, and the inconvenience of keeping the patient uncovered during the operation was avoided.

24th, HE was fo much better that there feemed to be no neceffity for repeating the clyfters; the other means were continued. The window of his room was now kept fhut.

25th, ALL the fymptoms of putrefcency had left him; his tongue and teeth were clean; there remained no unnatural blacknefs or *factor* in the ftools, which had now regained their proper confiftence; his dozing and muttering were gone off, and the difagreeable odour of his breath and perfpiration was no longer perceived. He took nourifhment to-day with pleafure, and, in the afternoon, fat up an hour in his chair.

His fever, however, did not immediately leave him; but this we attributed to his having caught cold from being incautioufly uncovered when the window was open, and the weather extremely fevere; for a cough, which had troubled him in fome degree, from the beginning, increafed, and he became likewife very hoarfe for feveral days, his pulfe at the fame time growing quicker; but thefe complaints alfo went off, and he recovered, without any return of the bad fymptoms above-mentioned.

CASE XII.

Related by Dr. Thomas Percival.

ELIZABETH GRUNDY, aged feventeen, was attacked, on the 10th of December, with the ufual fymptoms of a continued fever. The common method of cure was purfued, but the difeafe increased, and foon affumed a putrid type.

ON the 23d, I found her in a conftant delirium, with a *fubfultus tendinum*. Her *

fkin was hot and dry, her tongue black, her thirft immoderate, and ftools frequent, extremely offenfive, and for the most part involuntary. Her pulse beat 130 strokes in a minute; fhe dofed much, and was very deaf. I directed wine to be administered freely; a blifter to be applied to her back; the pediluvium to be used feveral times in the day, and fixed air to be injected under the form of a clyfter every two hours. The next day her ftools were lefs frequent, had loft their fætor, and were no longer difcharged involuntarily; her pulfe was reduced to 110 ftrokes in the minute, and her delirium was much abated. Directions were given to repeat the clyfters, and to fupply the patient liberally with wine. These means were affiduously purfued feveral days, and the young woman was fo recruited by the 28th, that the injections were discontinued. She was now quite rational, and not averfe to medicine. A decoction of Peruvian bark was therefore preferibed, by the use of which she speedily recovered her health.

CASE

191

CASE XIII.

Communicated by an intelligent Gentleman in the West of this Island.

A YOUNG lady of 18, the daughter of a neighbour of mine, had been long indifpofed with a fort of diforder which the medical gentlemen of the neighbourhood could neither properly define, nor in the leaft relieve. The origin of this indifpofition is attributed to a violent cold, which this young lady caught about two years ago at a ball; for fince that time fhe had never enjoyed her health, and, in fpite of all medicines, fhe rather loft ground by flow degrees than fhewed any appearance of amendment. The fymptoms, as nearly as I can defcribe them, were as follows:

SHE had loft all colour from her face and hands, had a remarkably keen appetite, and eat much more than other perfons of her age are wont to do; but this food gave her neither firength nor increment, and fhe conftantly

conftantly complained of wearinefs, refufing to take any fort of exercife. At night fhe frequently had a flight fever, which terminated by the morning with an head-ach; but this fever did not come on every night, nor did it feem to follow any determinate period. She perfpired profufely every night, and even in the day-time the leaft exertion would throw her into a profufe perfpiration. She had tried bark, fteel medicines, mineral waters, flight emetics, rhubarb, &c. but all in vain.

In this flate of things I firft took the liberty of recommending the ufe of the vital air, or oxygen air, concerning which much has been faid of late, and to the phyfical properties of which I had two years ago been witnefs in a courfe of chemical lectures, which I attended in London. After feveral conversations with the father of the young lady on the fubject, it was at laft agreed to try the oxygen air, and I undertook to perform the experiment with a few chemical veffels which I happened to have by me.

ON

193

ON the tenth of April, 1797, I put eight ounces of nitre in a fmall and luted green glafs retort, and by expofing the retort to a barely red heat, I obtained nearly two quarts meafure of oxygen air, which, being mixed with about eight quarts of common air, was given to the young lady in an awkward manner; for it was introduced into an old glafs receiver of an air pump, to the upper aperture of which a leathern tube was adapted, with a glafs tube at the extremity of the leathern pipe, to which the patient applied her mouth, &cc.

THE effect of this application was rather difcouraging. The young lady felt a tightnefs, as the expressed it, about her head, for at leaft three hours after the inhalation of the vital air, and was very reftlefs at night, in confequence of which the could not be prevailed on to repeat the inhalation for feveral days. At laft, finding that no other bad confequence had been produced by it, the confented to make another trial, which was managed nearly in the fame manner; but it I

was attended with much lefs tightnefs about her head, though with an equal degree of reftlefsnefs at night; notwithftanding which a third attempt was made on the following day, and the operation was again repeated after an interval of one day.

ALL those trials were more or less attended with the like effects as the first; yet our patient thought that, notwithstanding the restless nights which she had passed, her strength seemed to be in some measure improved, which encouraged us all to follow the application; and in order to avoid both trouble and expence as much as possible, we procured one of Mr. Watt's apparatuses from Birmingham, and some good manganese from Devonshire, with which we began to work in a large and more expeditious way.

WE found that Mr. Watt's apparatus requires a nicety of management, without which one may do more harm than good. In the first trial of this apparatus we got, O 2 instead

inftead of pure air, an exploiive elaftic fluid; for on lowering a lighted match into a bottle full of it, the air took fire and exploded. It was foon found that this inflammability was occafioned by the moifture which was contained in the manganefe, in confequence of which the manganefe was made very dry for the fubfequent trials, and thus we obtained abundance of oxygen air, which was freed from the carbonic acid air by wafhing in lime water.

BEING now in pofferion of the proper materials, and having fome expectation of fuccefs, we began, on the 22d of May, to work affiduoufly and regularly; and I took care to note, every three or four days, all the circumftances that feemed at all remarkable.

THREE pints of oxygen air and eight quarts of common air were administered daily, which constantly produced the tightness of the head and the restless at night.

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MAY the 28th, the young lady feemed to have gained ftrength; but complaining much of the tightnefs of her head, the quantity of oxygen was diminifhed to one quart, with eight quarts of common air, per day, which was continued until the 10th of June, by which time her ftrength was unqueftionably improved, and the perfpiration at night was confiderably diminifhed; but as a cough happened to come on, we intermitted the application of the vital air for a whole week, after which, the cough having difappeared, the inhalation of the air was recommenced, and continued as before.

By the beginning of July the good effects of our application were very confiderable. The ftrength of the young lady was fuch as might be expected in a perfon of her time of life; the healthy colour was in great meafure returned to her face and arms; the perfpiration at night was trifling, and fhe feemed to acquire flefh.

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ON the 15th of August, the inhalation of the diluted oxygen air was difcontinued, finding that the young lady's health was completely reftored.

CASE XIV.

A GENTLEMAN, 35 years of age, of a fcorbutic habit, and fubject to violent head-achs, was induced to try the artificial airs in December 1796, every other medical application having proved ineffectual, and his health gradually declining. He was at first advifed to try the diluted oxygen air, which he accordingly did, but after three days inhalation of this air, a confiderable degree of inflammation on his lungs obliged him to defift.

THE inflammation being fubdued, he again inhaled the oxygen air, and a fimilar effect took place, though this fecond time the inflammation was not fo confiderable.

FINDING, therefore, that the oxygen air was not fit for him, he was recommended to drink

drink the water impregnated with carbonic acid gas, and to take fome other medicines of a demulcent kind. By following this plan for about fix weeks, and by breathing the falubrious air of Devonshire, his health improved to a certain degree; the fcorbutic fymptoms were reduced, and the head-achs were not quite fo frequent as they used to be; but after this improvement, the continuance of the above-mentioned medicines for full three months produced no other effect.

CONSIDERING that in this improved state his constitution might, perhaps, bear the ftimulus of the vital air better than it had done before, he was recommended to try that air again, but to take it in fmaller Accordingly he inhaled not quantities. more than one pint of it with about fixteen pints of common air every day, which, not producing any inflammation upon his lungs, he continued for upwards of two months, at the end of which time his head-achs were quite vanished, his digestion, which had 04

199

had always been defective, was confiderably improved, and he reckoned himfelf quite well.

IN the account of the preceding cafes, the reader may obvioufly remark, that not one unfuccefsful cafe has been introduced; on which it will be proper to mention, that in fo doing I did not mean to imprefs the reader's mind with an exaggerated idea of the power of factitious airs; but that my only meaning was, to render him better acquainted with the practical administration of the aerial fluids, which feemed more likely to be accomplished by adducing examples, in which the practice was in fome measure fanctioned by fucces, than otherwife.

WITH respect to the estimate of the efficacy of factitious airs in different diforders, my reader must confult the preceding chapter; for I have expressed in it the result, or what seemed to be the fair result of all

all the cafes that have come to my notice; and of fuch cafes the few that are contained in the preceding pages form a very fmall part.

I WOULD not be underftood to mean, that the application of the aerial fluids, in the cafes of the prefent chapter, is to be confidered as the model of practical perfection, for in fome of them the administration is evidently incorrect; but they certainly give a great infight into the practice, and hope that, with the affistance of the cautions and remarks of the following chapter, they may in great measure prevent the abuse of a new set of remedies, which have all the appearance of becoming very useful tools in the hands of skilful practitioners.

CHAPTER IX.

PRACTICAL REMARKS, HINTS, &c.

Concerning the Production of Factitious Airs.

IN particular fituations the difficulty of procuring proper materials and proper tools may prevent the poffibility of adopting the most expeditious, or, upon the whole, the most advantageous, methods of procuring the aerial fluids; and when that is the case the practitioner must confult the first chapter of this effay, for the method which may be more fuitable to the circumstances of his fituation. But when there is the opportunity of procuring both materials and instruments, it is then proper to follow the plan which may appear less exceptionable.

THE cheapest article for the production of oxygen air is the mineral called manganese,

nefe, which is found plentifully in many parts of this ifland, and elfewhere. A very good fort of it is found near Exeter. It ought to be free from extraneous, and particularly noxious, minerals; but it frequently contains a confiderable proportion of calcareous matter, which may be detected by powdering a little of the mineral, and pouring fome nitrous acid upon the powder; for this will produce an effervefcence proportionate to the quantity of calcareous matter. It must not, however, be expected to find manganefe perfectly free from it; for though this may be the cafe with fmall pieces of that mineral, yet in confiderably large quantities of it, fuch as are required for the production of oxygen air, fome calcareous earth is almost always contained; but the only effect which arifes from it, is the production of carbonic acid gas, together with the oxygen air, the former of which is eafily feparated from the latter by the well-known method of wafhing in lime-water.

203

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THE greatest quantity of oxygen air is extricated from manganese merely by the action of a full red heat; it is, therefore, necefiary to put that mineral in a veffel capable of refifting the action of fuch a degree of heat. Earthen-ware and certain metals are the materials fit for the conftruction of fuch veffels. The former is certainly unexceptionable in point of purity; but it is not managed very eafily for this purpose, and befides, the use of it is attended with confiderable expence, for a vefiel of that fort will hardly ever ferve more than once, as on cooling after the first experiment it generally breaks; and indeed it frequently breaks in the courfe of the experiment. Of the metals, gold or platina veffels would be the fitteft for the purpose, did not their value offer a material objection. Those metals excepted, iron is the beft; for though the use of a vessel of this metal be attended with evident objections, yet, when managed with care and attention, the oxygen air may be produced of fuch a degree of purity as to be more

205

more than fufficiently useful for medicinal purposes.

It is neceffary to remark, that in all cafes, but efpecially when an iron veffel is ufed, the manganefe, as well as the veffel in which it is contained, and the pipe or tube which conveys the air from it to the receiver, must be quite free from animal or vegetable matter, and perfectly dry, otherwise the elastic fluid, which is produced, may be injured in point of purity, and it may even degenerate into a noxious fluid.

WHEN those particulars are attended to, the oxygen air will principally contain a certain proportion of carbonic acid gas, and fome light powder of manganese, the former of which is to be separated by means of lime, and the latter will be deposited by standing, in about ten or sourceen hours time.

THE species of inflammable gas mostly in use are extracted by means of diluted vitriolic

vitriolic acid from zinc or iron, and by paffing the fleam of water over the furface of red hot zinc, or iron, or charcoal.

THE gas, extracted by means of diluted acid, holds in fufpenfion fmall particles of the metals concerned, viz. of the zinc or the iron, the latter of which in particular may be rendered manifeft by burning the gas in a bottle full of it, in which cafe fome fmall particles of a dark red light will be difcerned within the pale flame of the gas, which are the ferrugineous particles; for those minute red sparks are not to be feen in the inflammable gas which is obtained from pond water, or putrid matter, or, in short, from such such as do not contain any metallic fubftance.

THE gas obtained by paffing the fteam of water over red hot zinc, holds in fufpenfion a confiderable quantity of the flowers of zinc, which it depofits in about a day's time.

THE

207

THE gas obtained in a fimilar manner from iron is the most abundant, and of course the cheapest.

· For the production of the heavy inflammable gas, or hydrocarbonate, Mr. Watt recommends to use " charcoal made of the "twigs of fofter woods, fuch as willow, " poplar, hazle, birch, or fycamore, avoid-" ing fuch as have refinous or aftringent " juices. Prepare the charcoal by heating it " to full ignition in an open fire, and " quenching it in clean water, or by filling " a crucible with it, covering it with clean " fand, and exposing it to a ftrong heat in " an air furnace, and then fuffering it to " cool. In either of these cases it will be " found free from any bituminous matter, " which might contaminate the air, as ge-" nerally happens with common charcoal."

MR. WATT likewife mentions, amongst other forts of inflammable gas, that which is extracted from a mixture of charcoal powder and flaked lime, which, on account

count of its peculiar properties feems likely to prove very ufeful : "In refpect, *fays be*, " to the medicinal properties, all I know is, " that the inflammable air from charcoal " and lime contained no fixed air feparable " by wafhing with quick lime and water, " and that it did not caufe vertigo when in-" haled pure."

ONE or other of those species of inflammable gas may be preferred in particular cafes, and it is not only likely, but in great measure proved by actual experiment, that the particles of iron, or other matter, which are suspended in a particular fort of gas, may be peculiarly useful in certain difeases.

In the production of inflammable gas, the introduction of any extraneous matter, and efpecially of vegetable or animal fubftances, and of minerals that contain acids, fhould be carefully guarded againft. It is likewife advifable, for a very obvious reafon, not to conduct this procefs by candle light.

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THE carbonic acid gas may be extracted from chalk in Mr. Watt's apparatus, according to the directions given with the faid apparatus; but when no extraordinary large quantity of it is required, it is far more commodious to extract it from chalk or marble powder, and diluted vitriolic acid, in a glafs veffel. The difference between chalk and marble in this refpect is, that the former gives out the gas quicker, but is foon exhausted; whereas the latter gives it out more gradually, and for a greater length of time; hence, in fome cafes, the former, and in others the latter, may be preferred.

Concerning the Prefervation of Aerial Fluids.

OXYGEN Air is not contaminated by keeping in glafs receivers, or in fuch veffels as do not communicate any thing to it, nor does the contact of pure water injure it; but in wooden veffels, or veffels painted with oil paint, and when a confiderable quantity of common river water is in con-

210 MEDICINAL PROPERTIES of tact with it, the oxygen air will be contaminated more or lefs.

THE various fpecies of inflammable gas are apt to degenerate in process of time, especially if they be kept mixed with common or with oxygen air. The hydrocarbonate, in particular, is vastly more powerful when fresh made, than two or three days after. Due allowance, therefore, must be made for the loss of power in the administration of those airs.

WHEN oxygen air, or inflammable gas, is to be taken out of an air-holder or bottle, &c. by putting water in the veffel after the ufual manner, it is advifable to ufe limewater; for the lime will not only abford any carbonic acid gas that may be mixed with those airs, but will also prevent the putrefaction of the water.

FOR this purpose there is no occasion to filtrate the lime-water, as is practised in the usual manner of preparing it; but it will be

be fufficient to mix the quick lime with the water, and after leaving it at reft for an hour or two, to feparate the fluid and ufeful part from the fediment, by decanting it gently.

THE carbonic acid gas is not contaminated by keeping; but as it is abforbed by moft fluids, it fhould not be kept in contact with much water. In moft cafes it will be better to produce it afresh every time it is wanted.

In order to manage the aerial fluids with the greateft propriety, the practitioner fhould make himfelf acquainted with the modes of afcertaining their purity. This may, in great meafure, be derived from what has been mentioned towards the beginning of this effay; but if a more particular defcription of those methods be required, and especially concerning the use of the nitrous-gas eudiometer, or phosphoric eudiometer, the reader must confult those books which have been written expressly

P 2

011

on the fubject of aerial fluids, as the addition of those methods would increase the bulk of this effay beyond its prescribed limits.

Concerning the Administration of Factitious Airs.

WHEN oil-filk bags or bladders are ufed, the air or mixture of airs fhould be introduced immediately before it is to be inhaled, in order to avoid the airs acquiring an unpleafant flavour.

THE oil-filk bags, when not actually in ufe, fhould be hung up by means of a ftring, which may be fastened to the pipe, or they may be put over the back of a chair; but they must not be folded or preffed.

In the ufual way of making the mixture of airs, the factitious gas, in any required quantity, is first introduced in the bag, after which the common air is forced in by means of a common pair of bellows, until the bag 6 is

213

is quite inflated; for when the capacity of the bag is once known, one may eafily determine the quantity of oxygen or other factitious air, which may be required in order to form a mixture in any given proportion.

WHEN common bellows are used for this purpose, care should be had that they be made free from dust and ashes, which are generally contained in such bellows as are used for common fire-places.

WHEN a perfon is inhaling any fpecies of inflammable gas, or the vapour of ether, the operation fhould be conducted at a diftance from a candle, left the gas fhould catch fire, and at leaft occasion a furprife.

THE queftion, which is frequently afked, whether a patient muft be confined to his houfe, or to any particular diet, whilft he is under a courfe of aerial application, fuggefts the propriety of obferving, that there is no particular confinement or diet required merely on that account.

P 3

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WHEN aerial fluids are administered, it is proper to feel the patient's pulfe both before and after the inhalation, at least for the three or four first inhalations, as thereby one may, in great measure, be informed of the effect which the aerial fluid is likely to produce, and may regulate the subsequent applications accordingly.

THE patient should be enjoined to breathe the mixture of aerial fluids in an easy and natural way, and not in a forced manner, as some are apt to do.

WITH fome perfons the fenfibility of the lungs is fo very great, that they are affected with the fenfation of preternatural heat, and even of inflammation, by a remarkable fmall quantity of oxygen; half a pint of it, mixed with about twenty times that quantity of common air, has been known to produce fuch an effect; and this obfervation has been made where there was not the leaft appearance of miftake, or of any equivocal circumftance. This is particularly the cafe with

with perfons that have recently recovered from an inflammation of the lungs. In fuch cafes, therefore, the practitioner ought to be particularly careful, and he ought to begin by administering very finall quantities of oxygen.

THE above-mentioned fenfation of heat generally comes on immediately after the inhalation, but fometimes it comes on fome hours after, and efpecially in bed. It is, therefore, neceffary to enquire whether any particular effect has been obferved at any time between one inhalation and the next, in order to form a proper effimate of the effect of the application.

WHEN this fenfation of heat or reftlefsnefs is in a triffing degree, the daily inhalation may be continued; but it muft he fufpended, or at leaft moderated, whenever it be found to increase by daily repetition.

WHAT has been observed with respect to the effect of oxygen air, may, with pro-Р 4 per

per and obvious changes, be alfo applied to the inhalation of other aerial fluids, and particularly of the hydrocarbonate.

HOWEVER ftrange and unaccountable fome of these effects may appear, as that produced by a very small quantity of an aerial fluid in certain circumstances; or that of the preternatural heat coming on so long after the inhalation, &c. my reader may reft affured that the facts are true; and though we cannot reconcile the phænomena with the theory, yet as long as absurdity does not intervene, we must not deviate from the path which is pointed out by experience, because we are unable to understand the real causes of the effects.

APPEN-

APPENDIX.

On the Nature of Blood.

THE intimate connexion between refpiration and the ftate of the blood, the neceffary dependance of animal life on the oxygen part of the atmosphere through the intermediation of that fluid, and the various discordant opinions which have been entertained concerning the nature of blood; will eafily excuse the introduction of this concise account of the nature of that fluid in the present work, whose principal object is the investigation of the action of aerial fluids on the human body.

THE name of blood has been used in a more or less extended sense by different writers.

writers. Some confine it to the red fluid, which circulates through the veins and the arteries of the animal body; others have, with propriety, extended it to that fluid, which, whether coloured or colourlefs, is the most abundant in the animal body, and upon the circulation of which the life of the animal principally depends; hence the red colour is not an abfolute characteristic of blood; and, in fact, the blood of certain animals has not the least tint of red. Lastly, the name of blood has been bestowed even upon the fluid which circulates through the vessels of plants.

IN the following pages we fhall extend our obfervations not farther than the red blood, and hardly beyond that of the human fpecies; confining ourfelves principally to the account of facts that are independent on particular opinions.

THIS fluid, fo effentially neceffary to animal life, has been examined with all the ingenuity

ingenuity of man in a mechanical and phyfiological fenfe, as it circulates through the veffels of the body; it has been carefully viewed, under a variety of circumftances, through the moft powerful microfcopes, and it has been analyzed by the moft ingenious chemifts. By this means many difcoveries have been made, and many doubts have been cleared relatively to it; but after all we can form a very inadequate idea of its extensive use and properties. We must, however, remain fatisfied with the present knowledge of facts, and must leave the farther investigation of the subject to the labours and fortune of future observers.

BLOOD is a fluid confifting of a great variety of ingredients, fome of which are always to be found in it; whilft others are adventitious, or are to be obferved in it in particular circumftances; but the proportion of them all is not only various in different ages, and fexes, and ftates of the body, but is not the fame even in the different parts

parts of the fame body *. A difference not fo great with refpect to the number, as with refpect to the proportion, of the ingredients, has been obferved between the blood of man, and that of other animals, fuch as the ox, the horfe, the fheep, the hog, &c. +.

LEAVING it to the phyfiologists to explain how the blood circulates through the fanguiserous vessels, how the chyle is mixed with it, how a variety of fluids are fecreted from it, &cc. we shall examine its nature as a fluid out of the body.

BLOOD is of a uniform rich red colour, which inclines towards the florid vermilion

* Fourcroy found the blood of a human fœtus to differ in three remarkable particulars from that of an adult; viz. it contains no fibrous fubftance, ftrictly fpeaking, but a fort of gelatinous matter; it does not take a bright colour from the contact of air; and it does not afford any marks of its containing phofphoric acid. The colour of blood is paler and thinner in infants, in women, and in phlegmatic perfons, than in men of a healthy and robuft conffitution.

+ Rouelle's Analyfis.

in blood that comes out of the arteries, and to the dark purple in blood that comes out of the veins, but the latter, as has been obferved in the preceding pages, becomes brighter by exposure to refpirable air. It is not fo fluid as water, it feels unctuous or faponacious to the touch, and has a little fweetifh or faline tafte.

Soon after its extraction from the body, as the blood cools and remains at reft, a fpontaneous decomposition, or feparation, of parts takes place. A thick lump of coagulated red matter is formed in the middle, called the *craffamentum*, or *clot of blood*, and a fluid of a flight greenish yellow colour furrounds it, which is called the *ferum*. The quantity of ferum thus formed is lefs at first, than a few days after; for as the coagulable part contracts and grows harder, fo more and more ferum is forced out of it.

By washing the lump of coagulated matter, the colouring substance is entirely separated from it, and the remainder is an insipid,

fipid, tenacious, white, and fibrous, fubftance. The latter is called the *coagulable lympb*, or *fibrous matter*, of the blood. The former, or coloured portion, when viewed through the microfcope, is found to confift entirely of feparate particles, circular and pretty uniform in their fhape; whereas the ferum and the coagulable lymph, when examined with the beft microfcopes, do not appear to contain any diftinct particles in their composition.

THE blood, therefore, confifts of, or is first of all refolvable into, three distinct parts; namely, the *ferum*, the *coagulable lympb*, and the *red particles*; each of which is likewise a compound substance, but whose components cannot be so easily separated from each other.

THE fpecific gravity of human blood is variable, but it always exceeds that of water; the latter being to the former, at the leaft as one to 1,04, and at the most as one to 1,063. Each of its three principal components

components is likewife heavier than water; but with refpect to each other, the red particles are the heavieft, and the ferum is the lighteft.

THE ferum remains fluid in the ufual temperature of the atmosphere, as far down as a few degrees below the freezing point. But it coagulates in a degree of heat about equal to 160° of Fahrenheit's thermometer. The coagulation of ferum by heat is attended with two peculiar circumstances, *viz.* 1st, a confiderable quantity of air is extricated from it in the act of congelation; and, 2dly, a small part of it does not coagulate, but remains fluid.

THE coagulable lymph has been juftly confidered as the most important part of the blood, and as being the substance, from which all the other parts of the animal body derive their increment and their support. The fibrous and tenacious nature of this part, which the blood seems to derive from the gluten of our food, is so remarkable that

224 MEDICINAL PROPERTIES of that it may be ftretched out to a confiderable length, and by the continuance of a moderate degree of heat, it may be rendered gradually more and more confiftent; fo that at laft it may be brought to equal the confiftency of horn and even of bone.

THE red particles, from which the whole mass of blood derives its colour, feem to have no particular attraction for each other, nor for the other two components, fo that in the coagulum they are only entangled and detained by the vifcid part. Their peculiar and uniform shape has attracted the attention of philosophers fince the latter end of the last century, about which time they were first discovered. They have been attentively examined with the best microfcopes, and the appearances which have been partly observed and partly supposed, have given origin to a variety of conjectures and hypothefes, generally fanciful, and often abfurd.

WHEN any thin and femitransparent part of a living animal, such as the tale of a small

a fmall fish, the membrane which is between the toes of a frog, &c. is viewed through a good microfcope; the circulation of the blood through its fanguiferous veffels, is rendered manifest only by the motion of the red particles, which follow each other at a greater or lefs diftance; though in general each particle feems to touch, or, at leaft, almost to touch the following particle. They never run into each other and incorporate; and though not very hard, they are however poffeffed of a certain degree of confiftency and elafticity; for in paffing through fmall veffels they are frequently feen to affume an elliptical shape, and from other finaller veffels they are abfolutely excluded.

THOSE particles lose their shape, and are diffolved in certain fluids. They are not diffolved in the serous part of the blood, nor in urine, except when they are left in those fluids for some days, or when those fluids are diluted with water. But water is a powerful, and almost an instantaneous, fol-O vent

225

vent of those particles; yet water may be deprived of this property by the addition of common falt, or nitre, or of almost any other neutral falt, as also by the admixture of a very small proportion of vitriolic acid.

MARINE acid much diluted with water, does not diffolve those particles, but it deprives them of their colour.

VINEGAR is likewife a folvent of the red particles, though not fo powerful as water.

WHEN those particles have been once dried or diffolved in water, they cannot, by any known method, be made to reaffume their former fhape; and indeed even their formation in the animal body feems to be difficultly accomplished, at least much less expeditionally than that of the other components of blood; for in perfons that have lost much blood, the fanguiferous veffels are indeed fpeedily filled with new blood; but this blood continues thin and pale for a con-

227

a confiderable time, and if examined through the microfcope, few red particles will be found in it.

UNWILLING to interrupt the account of the chemical properties of blood, I shall referve the farther examination of the shape and size of its red particles for the latter part of this appendix, and shall now subjoin the farther analysis of this shuid, which is principally extracted from Fourcroy's late chemical works.

BLOOD, exposed to a gentle and continued heat, passes into the state of putrid fermentation. When distilled on a water bath, it affords a phlegm of a faint smell, which is neither acid nor alkaline, but easily putrifies, in confequence of its containing an animal substance dissolved through it. Exposed to a more intense heat, blood gradually coagulates and becomes dry; it then loses seven-eighths of its weight, and becomes capable of effervescing with acids. Deficcated blood, exposed to the open Q = 2

air, attracts from it some degree of moisture, and, in the course of a few months, there is formed on it a faline efflorescence, which Rouelle has determined to be carbonate of foda. When diftilled by naked fire, it affords a faline phlegm; that is, a phlegm holding in folution an ammoniacal falt, fuperfaturated with ammoniac. After this phlegm, a light oil paffes, then a ponderous coloured oil, and ammoniacal carbonate contaminated with a thick oil. There remains in the retort a fpungy coal, very difficult to be incinerated, which is found to contain muriate of foda, carbonate of foda, oxyde of iron, and a matter apparently earthy, which feems to be calcareous phofphate.

BLOOD, when burnt in a crucible, affords feveral products, in the following order: 1. water, and a little ammoniac; 2. oil, and carbonate of ammoniac, which forms a yellowifh vapour, thicker than the former; 3. Pruffic acid, which is eafily diftinguifhed by its fætid fmell of peach-flowers; 4. phofphoric

229

phoric acid, which is formed by the combuftion of phofphorus, and is not difengaged till the blood be reduced to a coal; 5. carbonate of foda, which is volatilized at an intenfe heat; 6. after this there remains in the crucible only a blackifh, granulated, cryftallized oxyde of iron, mixed with calcareous phofphate. The ferrugineous particles of this laft product may be feparated by the magnet, efpecially when the refiduum has been previoufly heated together with charcoal-powder in a covered crucible.

BLOOD combined with alkalis, without previous decomposition, becomes more fluid by ftanding. Acids inftantaneously coagulate it, and alter its colour. By filtrating this fubstance, evaporating the liquor passed through the filter, drying it before a moderate fire, and lixiviating the matter that has been dried, neutral falts are obtained, confisting of foda, with the acid that was mixed with the blood.

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IF entire blood, mixed with a fourth part of its weight of water, be coagulated by heat, and if a part of the fluid that fwims above the coagulated portion be evaporated, a fubftance of a brown yellow is obtained, which is cafily diffinguished to be true bile.

THE ferum, which has been lately called the albuminous fluid, communicates a green tinge to fyrup of violets. By distillation on a water-bath, it affords a phlegm of a mild infipid tafte, which is neither acid nor alkaline, but speedily putrifies. After losing this phlegm, it is dry, hard, and transparent like horn : it is no longer foluble in water : by diffillation in a retort, it affords an alkaline phlegm, a confiderable quantity of ammoniacal carbonate, and a very fœtid thick oil. All thefe products, in general, have a peculiar fœtid fmell. The coal of the ferum, when diffilled by naked fire, almost entirely fills the retort. It is fo difficult to incinerate, that it must be kept burning for feveral hours, and exposed to a great

great deal of fresh air, before it can be reduced to ashes. The ashes are of a blackish grey colour, and contain muriate and carbonate of soda, with calcareous phosphate.

THE ferum, if exposed for some time to an hot temperature in an open vessel, passes readily into a state of putrefaction, and then affords a considerable quantity of ammoniacal carbonate, with an oil, the smell of which is insufferably nauseous.

THIS liquor combines with water in any proportion, and then it lofes its confiftency, its tafte, its greenifh colour. When poured into boiling water, coagulates, almost wholly, inftantaneoufly. A portion of this fluid forms, with the water, a fort of opaque and milky white liquor; which, according to Bucquet, posseffes all the characteristic properties of milk, viz. it is rarified, and caufed to mount up, by heat, and is coagulated by the fame agents, viz. by acids, and by alcohol.

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THE ferum poffeffes the property of fixing and rendering folid by heat, two or three times its weight of water. But when the water exceeds feven times the quantity of ferum, then no coagulation takes place.

ALKALIES render the ferum more fluid, and acids coagulate it. This laft mixture, filtrated and evaporated after filtration, affords a neutral falt formed of foda and the acid employed; which fhews that foda exifts in the ferum in a naked state, in full possession of all its properties. The coagulum formed in this liquor by the addition of an acid, is very fpeedily diffolved in ammoniac, which is the general folvent of the albuminous part of the blood; but it is not diffoluble at all in pure water. Acids precipitate this matter in union with ammoniac. The coagulum affords, by distillation, the fame products as the ferum deficcated, and its carbonaceous refidue contains a good deal of carbonate of foda.

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THE ferum, infpiffated, affords azotic gas by the action of the nitric acid, with the help of a moderate heat. On increasing the fire, there is a quantity of nitrous gas difengaged from the mixture.

THE ferum does not decompose calcareous or aluminous neutral falts; but it acts with fufficient energy in decomposing metallic falts.

THIS fluid is liable to be congealed by alcohol; but this coagulum differs from that formed by means of acids, chiefly for its folubility in water.

THE ferum, therefore, appears to be an animal mucilage, confifting of water, acidifiable oily bafes, muriate and carbonate of foda, with calcareous phofphate.

THE clot of the blood affords, by expofure to the heat of a water-bath, an infipid water; it becomes, at the fame time, dry and brittle. It affords, in the retort, an alkaline

kaline phlegm, a thick oil, of a fœtid, empyreumatic fmell, and a good deal of ammoniacal carbonate. The refiduum which it leaves, is a fpongy coal, of a fparkling metallic afpect, difficult to incinerate, and affording, when treated with fulphuric acid, fulphate of foda and fulphate of iron; there remains, after thefe operations, a mixture of calcareous phofphate with carbonaceous matter. When expofed to a hot atmofphere, the clot of blood readily putrifies.

WHEN the clot is divided, by wafhing, into its two principal components, viz. the red part which is diffolved in the water, and the coagulable lymph; if the former be treated with different menftrua it will be found poffeffed of the fame characteriftics with the ferum; excepting that it contains a greater proportion of iron. The latter, after being well wafhed, will remain white, colourlefs, and infipid. It affords, by diftillation on a water-bath, an infipid phlegm, without fmell, and liable to putrefaction. Even the gentleft heat hardens this fibrous matter

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matter in a fingular manner. When expofed fuddenly to a ftrong fire it fhrinks like parchment. By diftillation, in a retort, it affords an ammoniacal phlegm, a ponderous oil, which is thick and very fœtid, and a good deal of ammoniacal carbonate, contaminated with a portion of oil. The refidual coal is not very bulky, but compact, ponderous, and eafier incinerated than that of the ferum. Its afhes are very white; it contains no faline matter, as it must have loft, by the washing, whatever is contained of that kind; and no iron; it is a fort of refidue of an earthy appearance, and feemingly calcareous phosphate.

THE fibrous part of the blood putrifies very quickly and eafily. When exposed to a hot moift atmosphere, it swells, and affords a good deal of ammoniac. It is not foluble in water; when boiled in that fluid, it becomes hard, and acquires a grey colour. Alkalies do not diffolve it, but even the weakest acids combine with it. The nitric acid difengages from it a considerable quantity

235

quantity of azotic gas, and of Pruffic acid, which comes out in vapour, and at length diffolves it with effervescence, and the difengagement of nitrous gas. When it ceafes to emit nitrous gas, the refidue is observed to contain oily and faline flakes, fwimming in a yellowish liquor : this liquor affords, by evaporation, oxalic acid in crystals; and at the fame time, deposits no inconfiderable quantity of flakes, composed of a peculiar oil, and calcareous phofphate. It appears, that hydrogene, carbone, and azote, which constitute the fibrous substance, are separated in different proportions, to combine with the oxygen of the nitric acid, and thus form the Pruffic and carbonic acids that are disengaged in gas, and the oxalic and malic acids, that remain in folution, and are feparated only by crystallization.

THE fibrous matter diffolves also in the muriatic acid, which converts it into a fort of green jelly. The acid of vinegar diffolves it with the help of heat: water, and especially alkalies, precipitate this fibreus matter

matter from acids. This animal matter is decomposed in these combinations; and when separated, by whatever means, from acids, no longer exhibits the same properties.

THUS much may fuffice with refpect to the chemical properties of blood. I shall now return to the examination of the configuration of its red particles, with which I shall conclude this essay.

THE red particles, which form a very finall part of the human blood, were difcovered by means of the microfcope, towards the end of the laft century. They were found to be circular and uniform; a tranfparent flat furface appearing to be furrounded by a dark circumference. This peculiar fhape feemed to indicate their being of fingular ufe to the animal æconomy, and excited the induftry of philofophers to the further inveftigations of their ftructure. As this could only be obtained without ufing more perfect microfcopes, and as the perfection of microfcopes depended on

on the conftruction of fmall lenfes, various methods were contrived for the attainment of this object, and microfcopical lenfes of very fhort focufes, and of courfe of great magnifying power, were foon produced; but the utmost power of those lenses could only difcover that when the particles of blood were magnified beyond a certain number of times, they exhibited a dark speck in their middle, as a center to their circumference.

THIS is all that could be clearly difcerned in those particles by means of ground lenses; but a vast deal more was suggested by the imagination; and it is curious to observe how much the eye and the understanding were deceived by the natural imperfection of the instruments, and by the influence of premature theories.

FINDING that the improvement of ground lenfes, beyond the abovementioned power, was obstructed by weighty practical difficulties, the deficiency was attempted to be fupplied

supplied by the use of globules of glass made by melting; for in the ftate of fusion, the natural attraction between its particles, will eafily form the glass into a spherical body. Several methods were accordingly devifed for conftructing those globules, as may be feen in Dr. Smith's Optics, and other publications; but those methods are either defective or absolutely impracticable. And, in fact, I do not find that any globules of very great magnifying power, were ufed before the time of Father della Torre, who, about the middle of the prefent century, constructed globules of wonderful minutenefs, and at the fame time clear and diftinct.

THIS Neapolitan Friar, who, without much fcientific knowledge, poffeffed a confiderable fhare of ingenuity, made many obfervations with those magnifiers, which he published, together with a minute and faithful account of his method of constructing the glass globules, in a pamphlet, about 30 years ago. But both the construction 8 of

239

of those globules, and their use as magnifiers, are very difficult; fo that few perfons have attempted to repeat Torre's experiments, and amongst those, fewer still have been fuccefsful. This want of fuccefs has thrown a confiderable degree of fuspicion on Torre's obfervations; and as few people are liberal enough to acknowledge their want of fufficient patience and addrefs, the failure of the attempts has generally induced people to confider Torre's affertions in the light of miftakes or exaggerations. " The " Abbé Torre," fays a recent writer, " ex-" amined the red particles of blood with " fimple lenfes too; but they magnified fo " highly, that from this caufe all his noify " mistake has arifen; for he used not ground " lenfes, but fmall fphericles of glafs, form-" ed by dropping melted glass into water: " they magnified fo much that to him the " central fpot appeared much darker; he " faid that these were not globules, but " rings. He fent his fphericles of glass, " and his observations from Italy, his own " country, to the Royal Society; and for a " long

" long while, though nobody could fee " them, ftill the public were annoyed by " Abbé Torre's rings *."

for this purpote 1 obtained,

Some years ago, when Torre's publication first became known to me, I endeavoured to construct microscopical globules after his method, and to repeat his observations. The undertaking, which at first fight appeared clear and eafy, proved on trial very difficult and laborious; however, after perfevering for a confiderable time, I at last procured three or four useful globules out of a vast number of imperfect ones. With these globules, and an apparatus made exprefsly for fuch delicate experiments, I repeated feveral of Torre's obfervations, and (as far as I now recollect, for both the globules and the journal of observations have been long fince loft) I found that his description of appearances is very accurate, though his conjectures may fometimes be crude or mistaken.

* Bell's Anatomy, vol. ii. p. 89.

BEING

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BEING lately intent on the fubject of the prefent work, I was defirous to repeat the above-mentioned microfcopical obfervations, and for this purpose I obtained, after a confiderable expenditure of time and labour, a few glass globules, fufficiently useful, and with them I made the observations which I shall now lay before the public. But it will be proper to premise a concise account of the principal opinions that have been entertained by various ingenious perfons, concerning the construction of the red particles, as the origin of fome of those opinions will be evidently pointed out by the observations that follow.

LEEUWENHOECK thought that each red particle of the blood confifted of, and was refolvible into, fix fmaller globules, and that every one of thefe fecondary globules confifted of other fmaller particles. Hewfon took them for bladders which contained a nucleus or central body that feemed to roll from one fide of the bag to the other. .Torre

Torre faw them like rings; viz. confifting of an internal and an external circle, and this ring appeared to be divided, or to confift of parts joined together like the rim of a common coach wheel. Falconer confidered them as flat or fpheroidical bodies; for he thought he fometimes faw them fideways. " The red globules," fays the late Mr. J. Hunter, " are always nearly of the " fame fize in the fame animal, and when " in the ferum do not run into one another " as oil does when divided into fmall glo-" bules in water. This form, therefore, " does not arife fimply from their not unit-" ing with the ferum, but they have really " a determined shape and fize. This is " fimilar to what is observed of the globules " in milk; for milk being oily, its globules " are not foluble in water; neither do they " confift of fuch pure oil as to run into " each other; nor will they diffolve in oil. " I fuspect, therefore, that they are regular " bodies, fo that two of them could not " unite and form one *." Dr. Wells is of

* Treatife on the Blood, p. 41.

R 2

opinion,

243

opinion, that the red globules confift of two parts, one within the other, and that the outer, being infoluble in ferum or dilute folutions of neutral falts, defends the inner from the action of those fluids *.

MUCH having been faid against the use of microscopical glass globules, especially by perfons who had never seen them, I thought it necessary to ascertain the limits of the supposed differsion of the image, and other imperfections that had been attributed to them, and for this purpose I viewed certain objects of simple or determinate figures through lenses and globules of different powers, increasing gradually from a magnifying power of about eight or ten as far as that of about 400 times in lineal extension.

could of fuch pure oil as

A DELICATE straight line made by means of a diamond on a piece of glass, and which was quite invisible to the naked

* Phil. Tranf. P. II. for 1797.

eye,

eye, when thus gradually magnified, appeared always ftraight, provided it was made to pass through the axis of the lens or globule. The feathers of a butterfly, or rather any particular part of one of those feathers, never changed its figure though magnified upwards of 400 times.

THERE is an evident diffortion of the image when the object is viewed through the edge of the lens, and efpecially of a globule; but no perfon verfed in fuch experiments will ever obferve through the edge of lenfes, though the lenfes be ever fo perfect.

WHEN the object is not very flat, it is then evident that a perfect view of it can not be had at once; for if one part of it be in the focus, the reft of the object muft of courfe be out of it; yet by alternately bringing one part of the object and then another to the focus, one may, in moft cafes, acquire a fufficiently accurate idea of its fhape.

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V

VARIOUS

245

VARIOUS observations of this fort gave me reason to conclude that the glass globules are by no means so imperfect as they have been represented. Their distortion of the image is trifling and limited; the transparency of some of them (and such only should be used) is equal to that of the best polished lenses; but the use of those globules is very difficult, and it is on account of this difficulty that they have been neglected and misrepresented.

For the fake of those who may be willing to repeat fuch experiments, I shall barely mention the principal difficulties which attend the use of the globules.

THEIR focus is confiderably nearer the furface than that of a lens of the fame magnifying power; and as a globule, in order to magnify more than the ufual microfcopical lenfes, must be lefs than the 30th of an inch in diameter, and its focus shorter than the hundredth part of an inch, it follows that the common microfcopical apparatufes are

are in general inapplicable to fuch globules, fince the deviation of one or two thousandth parts of an inch in the adjustment of the focus will occasion a confiderable degree of indistinction.

It is for the fame reafon, that the globule must be set for as to have part of its furface actually out of the brass cell, and yet it must be secured for as not to drop out.

THE brafs cell must admit of the globule being easily taken out and replaced; for when they are obscured by dust, &c. to which they are very subject, they can feldom be cleaned without removing them from the cell.—Let us now return to the particles of blood.

I HAVE repeatedly meafured the diameters of the red particles, both by means of my mother-of-pearl micrometer in a compound microfcope, and likewife by looking at them with one eye through a fingle R 4 lens,

lens, and referring their image to a fcale properly divided, and viewed with the other eye out of the microfcope.

IN perfons of nearly the fame age the mean fize of the particles differs very little indeed. In the fame perfon they differ a little, and their figure is not very circular. This deviation from the circle is not fuch as a flat circular furface would affume in its different inclinations to the axis of vifion; for, according to the rules of orthographic projection, the flat circular furface must appear either circular, or elliptical, or as a straight line; whereas I never faw the particles of blood as straight lines, viz. edgewife, and the elliptical figure, which they fometimes affume, is by no means regular.

IN an adult of the human fpecies, the diameters of the red particles run from about 0,0003 to about 0,0004 parts of an inch, and I very feldom faw one fmaller or larger than those limits. If, therefore, we take

take the fmalleft particles and fet them in a row, we fhall find that about 3334 of them will equal one inch, and if we take the largeft, about 2500 of them will meafure one inch.

WHEN the particles are magnified more than 40 or 50 times, and lefs than 80 (meaning always in diameter), they appear like colourlefs transparent spots inclosed within dark circles.

WHEN magnified more than 80 times, and lefs than about 160, a dark fpot, like a dot made with ink on paper, appears in the middle of each particle.

IF the reflector which illumines the particles, inftead of being fituated ftraight before the object, be fet on one fide of the axis of vifion, fo as to throw the light obliquely on the object, then the half of the dark circle of each particle difappears, viz. that half which is on the fide oppofite to the reflector.

250 MEDICINAL PROPERTIES of reflector. The central fpot does at the fame time appear to change its place.

WHEN the particles are magnified above 200 times, the central fpot appears converted into a circle inclosing a transparent fpace. The diameter of this inner circle is about the half of that of the external one; but the proportion of these diameters, or the fize of the internal circle, may be caufed to increase or decrease by the least alteration of the distance between the object and the microfcopical lens; and by the fame means the space within the inner circle may be rendered clearer or darker than that between the two circles. The polition of the inner circle is changed by the direction of the light; for if the particle of blood be viewed through a microfcopical globule, directly facing the flame of a candle, without the intermediation of any lens or reflector, the inner circle will appear concentric with the outer one; but if the candle be moved a little to one fide, fo that the light may fall obliquely on the particle of

of blood, then the inner circle will be obferved to move towards the oppofite fide, and to acquire an elliptical shape.

WHEN the particles of blood are magnified above 400 times, an imperfect image of the candle, which is placed before the microscope, may be seen within the inner circle of each particle.

THROUGH a glass globule of 0,018 of an inch in diameter, I have seen the red particles of blood magnified about 900 times, in which case the image of the flame of the candle could be seen within the inner circle of each particle very clearly, at least so so thew to which side the motion of the air in the room inclined it.

NOTWITHSTANDING this great magnifying power, the annulus or fpace between the two circles did not appear to be divided, excepting fome accidental fractures, which now and then could be feen in a few of the particles.

THESE

251

- THESE observations seem to prove, that the red particles of blood are not perforated, but that they are globular, and of fome uniform substance much less transparent than glafs. They likewife fhew that Mr. Hewfon's idea of their containing a central body or nucleus, moveable within the external fhell, arole from the apparent change of place which the various direction of the light produces on the central fpot or inner circle of each particle. Warned, however, by the example of other obfervers, I shall not attempt to offer any farther conjectures concerning the nature and construction of those particles. My reader may draw what conclusion he thinks proper from the abovementioned facts, and he may alfo, with little trouble, fatisfy his curiofity concerning those appearances, as I find that microfcopical glafs globules may be had at Mr. Shutleworth's philosophical inftrument shop on Ludgate Hill. I shall therefore conclude with the account of a few experiments which I have made, with a view of imitating the phenomena that are exhibited

by

253

by the particles of blood, the refult of which feems to corroborate what has been already obferved.

On the fupposition of the red particles being globular, I expected that globules of other transparent matter would exhibit the fame appearances as the particles of blood, and my expectations were in great measure verified by actual experiments.

A GLASS globule was placed as an object upon the ftage of the microfcope, and was fucceffively viewed through lenfes of various, but not great, magnifying powers. As every part of the globule could not be at once in the focus, the whole of it was not of courfe equally diffinct. This indiffunction, however, being not very great, I fhall proceed without taking any farther notice of it.

THE globule appeared like a dark circular furface, with a transparent circular spot in the middle, and in this spot there appeared

peared a diftinct image of the candle or the window, or, in fhort, of any other object that was placed directly before it.

In this experiment three points of difference between the glass globule and the particles of blood were remarked, viz. 1ft, that the globule shewed a diftinct, whereas the particle shewed an indistinct image of the candle; 2dly, that the inner circle of the globule is much smaller in comparison with its external boundary, than the inner circle of the particle is in comparison with its external one; and, 3dly, that the annulus or space between the two circles is uniformly dark in the glass globule, whereas in the particle it is about as clear as the internal furface, or rather clearer.

THE first and the last of these points of difference seem to depend on the imperfect transparency of the particles of blood; for in semitransparent bodies, whatever light falls upon any part of them is scattered through the whole body.

THE

THE fecond point of difference I attributed to the particles of blood being furrounded by a coagulated fluid of nearly an equal refracting power with themfelves, whereas the glafs globule was furrounded by air only. In order to verify this fuppofition, I placed the glafs globule in water, and viewing it in that flate through the fame magnifiers that had been ufed before, I found that the transparent part or circle appeared much larger than in the former cafe *.

In the globule of glass, as well as in the particle of blood, the inner circle may be made to appear larger or smaller, by altering

* These appearances are perfectly reconcileable to the doctrine of optics. The light, which falls from a luminous object upon the glass globule, illuminates at most one half of its furface, and in entering the furface of the glass, it is refracted towards the axis of the globule; hence the whole cone of light being contracted, must pass through a small part only of the opposite furface, and must leave the rest destitute of light. Now this contraction of the light must vary according to the difference between the refractive power of the globule and that of the furrounding medium.

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the

the distance between the object and the microscopical lens.

In the glafs globule the inner circle may be feen to move from the middle of the dark furface, according as the candle is moved from the direct line between the object and the microfcopical lens.

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INDEX.

INDE Х.

A:

AERIAL fluids, different species of, page 1-breathed in combinations of three or four at a time, 49-injected into the cellular membrane, 51-absorbed by water, 56-theory of, 58-their constituent principles, 67 -how applied to the human body, 91, &c. 114-remarks concerning the production of, 202-prefervation of, 209 -remarks concerning the administration of, 212.

Acidulous soda water, 145.

Air, common or atmospherical, its properties, 2—its purity afcertained by means of nitrous gas, 3—contains a variety of fubstances, 5—vitiated various ways, 6—methods of meliorating it, 6—respiration of, 23—imbibed through the pores of the skin, 51—injected into the cellular membrane, 52—its constituent principles, 67 decomposed in the lungs, 72—its various qualities, 88,

Animal heat, origin of, 82.

Animation, fulpended, treatment of, 116.

Apparatus,

Apparatus, neceffary for the production of factitious airs, 90-for administering the factitious airs, 91.

Afthma, treatment of, 118.

Azotic gas, what, 6-deleterious to animals, 48-injected into the cellular membrane, 52-injected into the jugular vein, 53-its conftituent principles, 67-a diluent of oxygen air, 87.

Β.

- Bag, oil filk, for containing air, &c. 92-must not be folded, &c. 212-how to correct its bad fmell, 93.
- Bladders, for containing air, 94 how to correct their bad fmell, 95.
- Blood, its colour altered by the prefence of different aerial fluids, 53, 77—imbibes the oxygen in the lungs, 72, 76, 81—its colour in the courfe of circulation, 78, 83, 220 —name of, 217—confifts of various ingredients, 219 decomposition of, 221—its specific gravity, 222.

C.

Cancer, treatment of, 120.

Calculous complaints, treatment of, 144.

- Caloric, what, 62—laws relative to the communication of, 63—capacity of bodies for containing it, 64—produces a total change of properties in bodies, 66.
- Capacity, for containing caloric, different in different bodies, 64-proportion of, in different bodies, 65.

Carbonic acid gas, its properties, 15-whence obtained, 16 —refpiration of, 46-appearances in the bodies of animals mals that have died in it, 47—injected into the cellular membrane, 52—injected into the jugular vein, 53—its pungency, 55—an antifeptic, 55, 102, 111—how adminiftered, 56, 97—its conflituent principles, 68—produced in the process of respiration, 73, 85.

Cases, medical, 149.

Catarrh, treatment of, 123.

Chlorofis, treatment of, 123.

Combustion, theory of, 69-analogous to respiration, 75.

Confumption, treatment of, 124.

Coughs, treatment of, 130.

Craffamentum, or clot of blood, what, 221.

D.

Debility, treatment of, 130. Dephlogifticated air, the fame as oxygen air, 1. Digeftion, impaired, treatment of, 133. Dropfy, treatment of, 135. Dyfpepfia, treatment of, 133.

E.

Eudiometer, what, 3. Eruptions, treatment of, 136. Ether, its use, 110.

F.

S 2

Fevers, treatment of, 137. Fixed air, the fame as carbonic acid gas, 1.

Head-

H.

Head-ach, treatment of, 141.

Heat, its origin, 62, 66-animal, whence derived, 82.

Hæmoptyfis, treatment of, 142.

Hydrocarbonate, what, 19-refpiration of, 45-remarks concerning the production of, 207.

Hydrogen gas, its properties, 17—whence obtained, 18 refpiration of, 42, 44—properties of its various fpecies, 43—injected into the cellular membrane, 52—it does neither accelerate nor retard putrefaction, 55—its confituent principles, 67—remarks concerning the production of, 205, 208—contains extraneous particles, 206—apt to degenerate, 210.

Ι.

Inflammable air, the fame as hydrogen gas, 2. Inflammation, of the lungs, how treated, 107, 109, 118.

L.

Lime-water, use of, 210.

Lungs, human, affected by the smallest differences in the purity of the air, 4, 214.

Lymph of blood, 222, 223.

M.

Manganese yields abundance of oxygen air, 14, 202—it generally contains calcareous earth, 203—should be made quite dry before it is used, 205.

Mouth-piece, used for breathing any particular fort of air, 96.

N. Nitrous

Nitrous gas, diminishes common air, 2-diminishes oxygen air, 10.

Oil-filk bags, ufeful for refpiring particular airs, 92-how to remove their oily fmell, 93.

Ophthalmia, treatment of, 142.

Oxygen air, its properties, 9—obtained from various fubftances, 9, 11—diminifhed in various circumftances, 10, 14—refpiration of, 29—its ftimulant quality, 34, 35, 36, 37, 54, 101—effects produced by it when diluted with common and other airs, 34, 38, 40—imbibed through the pores of the fkin, 51—injected into the cellular membrane, 52—its conflituent principles, 67—indifpenfably neceffary to animal life, 74—when to be adminiftered, 102, 104.

Oxygenation, of metals, 69-of blood, 72.

Ρ.

Paralysis, treatment of, 143. Phlogisticated air, what, 6. Phlogiston, what, 60. Phosphuret of bydrogen, what, 19. Phthis, treatment of, 124.

R.

Red particles of blood, 222, 224—opinions concerning their conftruction, 242—microfcopical appearance of, 249—their diameter, 248.

Respiration of common air, 23-of oxygen air, 29-of other factitious airs, 40-theory of, 58, 71.

S.

Scurvy, treatment of, 143. Serum of blood, 221, 223.

Skin.

INDEX.

Skin, the aerial fluids are imbibed and expelled through its pores, 51.

Specific gravities of the aerial fluids, 20-table of, 21effects arifing from it, 49.

Stone in the bladder, treatment of, 144. Sulphuret of hydrogen, 19.

outpourte or injoing of injoing

Swellings, treatment of, 146.

Т.

Theory, of aerial fluids, 58-of refpiration, 59, 60, 72.

U.

Veffels, proper, for the production of aerial fluids, 204-for the prefervation, &c. 209. Ulcers, treatment of, 147.

w.

Water abforbs aerial fluids, 56—its conflituent principles, 68—expelled from the lungs, 73—origin of, in the lungs, 85—how impregnated with carbonic acid gas, 98
White fwelling, cafe of, 146.

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