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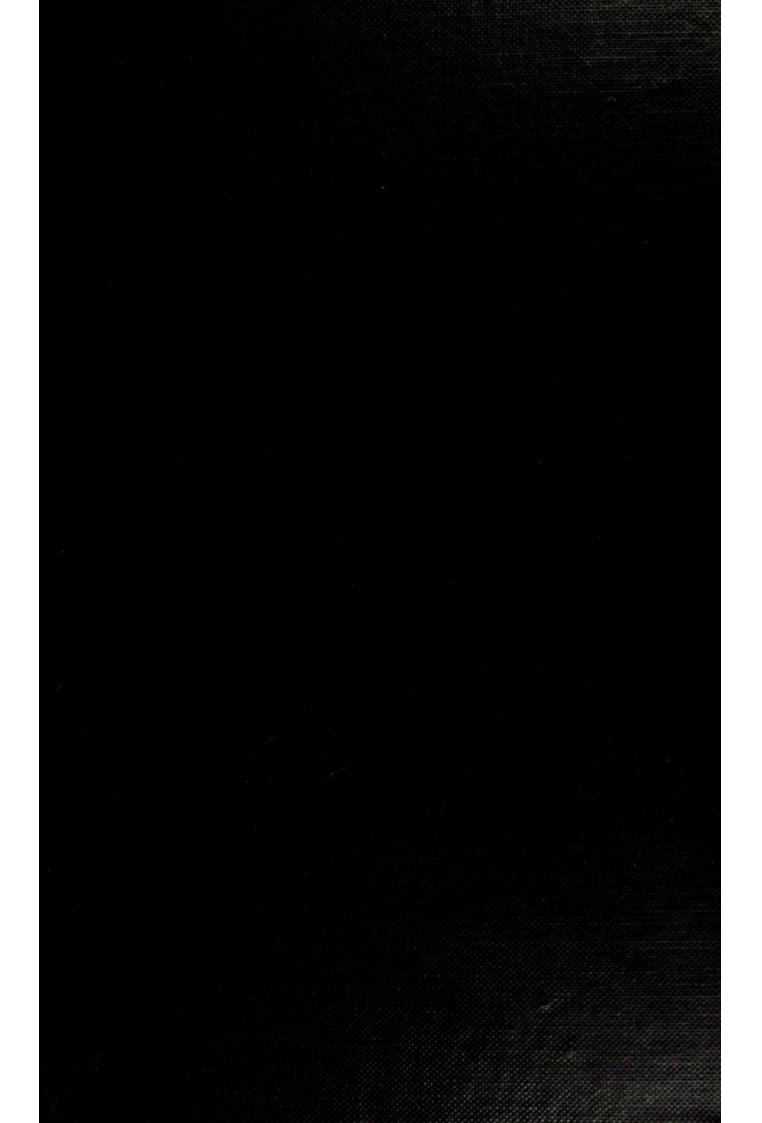
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1314-

CHEMICAL SOCIETIES

OF THE

NINETEENTH CENTURY.

BY

HENRY CARRINGTON BOLTON.



CITY OF WASHINGTON: PUBLISHED BY THE SMITHSONIAN INSTITUTION.

1902.

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Wellcome Lion for the History and Understanding of Medicine

CHEMICAL SOCIETIES OF THE XIX CENTURY.*

BY HENRY CARRINGTON BOLTON, PH.D.

The beginning of a new century affords an opportune period for chronicling the progress of chemistry as shown by the organizations formed to foster its study and to stimulate its adherents. In the following pages an attempt has been made to place on record the statistics of the Chemical Societies of the World for the year 1900, and to indicate those that ended their careers within the nineteenth century. The data have been obtained chiefly by correspondence, and thanks are due to the officers of societies who have responded to inquiries. I am also under special obligations to Dr. Paul Dorveaux, Librarian of the École Supérieure de Pharmacie, Paris; to Professor Bohuslav Brauner, of the Bohemian University, Prague; to Professor George W. A. Kahlbaum, of the University of Basel; and particularly to the Smithsonian Institution, for aid in securing the information sought.

The fact that chemical societies were organized and in operation in the United States of America long before they existed in Europe has been shown in my paper, "Early American Chemical Societies," read to the Chemical Society of Washington, April 8, 1897. The two pioneers in this field were the "Chemical Society of Philadelphia," founded in 1792, and the "Columbian Chemical Society of Philadelphia," founded in 1811. Of these some particulars will be found in their proper order.

In the following list the societies are placed in chronological order under each country, and the countries are arranged alphabetically. Of each society the following data are given so far as attainable:

> Seat, and date of founding, Name of President, and membership in 1900,† Serial publications,‡ Remarks.

^{*} Read at the 25th Anniversary of the American Chemical Society held in New York City, April 12-13, 1901.

[†] No deductions have been made for duplication.

[‡] For full details consult: A Select Bibliography of Chemistry, by Henry Carrington Bolton, Washington, 1893-1899. 3 vols. 8vo.

The results of this census are given in the following Table:

CHEMICAL SOCIETIES OF THE WORLD.

Membership in 1900.

Country. N	o. of Sou	cieties.	No. of Me	mbers.
Austria	7		3,07	2
Belgium	3		74	0
France			4,06	5
Germany	10		7,55	9
Great Britain			7,55	0
Italy	5		47	9
Japan			I,01	2
Russia			32	7
South Africa	I		(?)
Switzerland	2		9)4
United States America	ı. 5		2,37	9
Victoria	I		10	0
TOTALS	56		27,37	7

March, 1901. WASHINGTON, D. C.

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AUSTRIA.

CENTRALVEREIN FÜR RÜBENZUCKER INDUSTRIE IN DER OESTER-REICHISCH-UNGARISCHEN MONARCHIE.

Founded in 1854 at Vienna. In 1900: President, August Freiherr von Stummer; members, 213; associates, 49.

Publications: Organ des Vereins f
ür R.-I., 1863-1874; Organ des Centralvereins f
ür R.-I., 1875-1887; Oesterreichisch-ungarische Zeitschrift f
ür Zucker-Industrie und Landwirthschaft, 1888-1900.

(Beilagen): Der Marktbericht, 1874–1885; Wochenschrift des Centralvereins für Rübenzucker in der oesterreichisch-ungarischen Monarchie, 1886–1900.

NOTE.—The Society maintains a Chemical Experiment Station under the direction of F. Stohmer.

VEREIN ZUR HEBUNG DER ZUCKERFABRIKATION IM KÖNIGREICH BÖHMEN. [Also with a Bohemian name.]

Founded in 1868 at Prag under the presidency of Ferdinand Urbánek; the languages used were Bohemian and German. It was disbanded in 1874.

Publication: Zeitschrift für Zucker-Industrie. Organ des Vereins. Prag. 3 vols., 1872-74.

NOTE.—This journal is not to be confounded with: Zeitschrift für Zucker-Industrie in Böhmen, established at Prag in 1877 and current.

CHEMICKA SPOLEČNOST: SPOLEK ČESKÝCH CHEMIKŮ. [Chemical Society: Union of Bohemian Chemists.]

Founded in 1872 at Prag. In 1900: President, K. Preis; honorary members, 11; active members, 318; correspondents, 77.

Publications: Zprávy spolku českých chemiků. 2 vols., 1872-76. (Reports.) Listy Chemické, 1877-1900. (Letters.)

NOTE.—The Society has also published a Chemická Knihovna (Chemical Library) in 8 vols.

ZEMSKY SPOLEK PRO PRŮMYSL CUKROVARNICKÝ V ČECHÁCH; VEREIN DER ZUCKERINDUSTRIE IM KÖNIGREICH BÖHMEN.

Founded in 1876 at Prag. In 1900: President, Gustav Hodek; members, 325. Publications: Zeitschrift für Zucker-Industrie in Böhmen, 1877-1900. (Beilage): Prager Zuckermarkt, 1881-1900.

Oesterreichische Gesellschaft zur Förderung der chemischen Industrie.

Founded in 1878 at Prag. In 1900: President, Georg Zetter; honorary members, 3; members, 196.

Publications: Bericht der oesterreichischen Gesellschaft zur Förderung der chemischen Industrie, 1879-98. Since 1899 the organ of the Society is: Oesterreichische Chemiker-Zeitung.

SPOLEČNOST PRO PRŮMYSL CHEMICKÝ. [Society of Chemical Industry.]

Founded in 1892 at Prag. In 1900: President, J. B. Lambl; honorary members, 20; active members, 440; correspondents, 54; founders, 57.

Publications: The organ of the Society since 1892 is: Časopis pro průmysl chemický, of Prag, which had been established in 1891. The Society has also published three volumes of a technological library: Knihovna technologicko chemická.

WIENER VEREIN ZUR FÖRDERUNG DES PHYSIKALISCHEN UND CHE-MISCHEN UNTERRICHTS.

Founded in 1895 at Vienna. In 1900: President, Victor von Lang; members, 317.

Publication: Vierteljahrsberichte der Wiener Verein zur Förderung des physikalischen und chemischen Unterrichts, 1895-1900.

VEREIN OESTERREICHISCHER CHEMIKER IN WIEN.

Founded in 1897 at Vienna. In 1900: President, J. Klaudy; members, 878; founders, 14.

Publication: Oesterreichische Chemiker Zeitung, 1898-1900.

BELGIUM.

Association Belge des Chimistes.,

Founded August 4, 1887, at Brussels. In 1900: President, L. L. de Koninck; honorary members, 4; active members, 482; associates, 21; correspondents, 8.

Publication: Bulletin de l'Association Belge des Chimistes, 1887-1900.

Note.—The Association has 8 sections, viz: Liége, Louvain, Gembloux, Charleroi, Mons, Gans, Antwerp, Brussels.

CHEMICAL SOCIETIES OF THE XIX CENTURY

Société technique et chimique de sucrerie de Belgique.

Founded February 26, 1896, at Brussels. In 1900: President, Eugène Meeus; members, 173; patron, 1.

Publications: La sucrerie Belge, which was established August 31, 1872, has been the organ of the Society since its foundation. The Society has also published several pamphlets on technical topics.

NOTE.—The formation of Sections was under discussion in 1900.

SYNDICAT DES CHIMISTES PUBLICS DE BELGIQUE.

Founded in 1897 at Brussels. In 1900: President, François Sachs; members, 51.

Publication: Bulletin du Syndicat des chimistes publics de Belgique, 1897-1900.

NOTE.—The Society has in preparation: Recueil générale des méthodes d'analyse usitée dans les laboratoires publics.

Société GÉNÉRALE DES FABRICANTS DE SUCRE DE BELGIQUE is not a chemical society; its organ is: La sucrerie Belge, 1872-1900.

FRANCE.

Société industrielle de Mulhouse.

Founded in December, 1825, at Mulhouse (first meeting, May 11, 1826). In 1900: President, Auguste Dollfus; honorary members, 9; resident members, 190; non-resident members, 378; correspondents, 54.

Publication: Bulletin de la Société industrielle de Mulhausen (sic), 1827-1900.

NOTE.—This is not purely a chemical society, but it has a Committee on chemistry, and its Bulletin contains many papers on applied chemistry.

Société chimique de Paris.

Founded June 4, 1857, at Paris. In 1900: President, Edouard Grimaux; members, 365; patrons, 121; life members, 91; corresponding members, 449.

Publications: (a) Bulletin des séances de la Société chimique de Paris, 1858-62; (b) Répertoire de chimie pure et appliquée, 1858-63; (c) Bulletin de la Société chimique de Paris, 1864-1900; (d) Conférences et Leçons, 5 vols. Association des élèves de M. Fremy.

Founded in 1878 at Paris. In 1900: President, Louis Barthélemy; members, 200.

Publication: Bulletin trimestriel de l'Association des élèves de M. Fremy, 1878-1900.

NOTE.—A social organization which, however, publishes the work of its members.

Association des chimistes de sucrerie et de distillerie de France et des Colonies.

Founded in 1883 at Paris. In 1900: President, M. Durin; honorary members, 3; resident members, 160; non-resident members, 710; corresponding members, 395.

Publication: Bulletin de l'Association des chimistes de sucrerie et de distillerie de France et des Colonies, 1883-1900.

Association amicale des anciens élèves de l'École de physique et de chimie industrielle de la ville de Paris.

Founded in 1885 at Paris. In 1900: President, Octave Boudouard; honorary members, 41; members, 300.

Publication: Bulletin mensuel de l'Association amicale des anciens élèves de l'École de physique et de chimie industrielle de la ville de Paris, 1885-1900. Annuaire [etc.], 1885-1900.

Association amicale des anciens élèves de l'École de chimie industrielle de l'Lyon.

Founded in 1886 at the Institut chimique de Lyon. In 1900: President, Alphonse Seyewitz; honorary members, 6; members, 104.

Publication: Bulletin des séances de l'Association amicale des anciens élèves de l'École de chimie industrielle de Lyon.

SYNDICAT CENTRAL DES CHIMISTES ET ESSAYEURS DE FRANCE.

Founded in 1890 at Paris. In 1900: President, Ferdinand Jean; members, 125.

Publications: Revue de chimie analytique appliquée à l'industrie, 1893-98. Annales de chimie analytique appliquée à l'industrie became the organ of the Society in 1899; the Annales had been established in 1896, and was united with the Revue (above named) in 1899. Société chimique du Nord de la France.

Founded at Lille in 1891. In 1900: President, A. Pouriez; members, 100.

Publication: Bulletin mensuel de la Société chimique du Nord de la France, 1891-1900

Association amicale des anciens Élèves de l'Institut chimique de Nancy.

Founded November 9, 1892, at Nancy. In 1900: President, M. Noel; honorary members, 8; patrons, 7; members, 52; associates, 75. *Publication:* Bulletin (annuel) de l'Association.

Association amicale des Élèves et anciens Élèves du laboratoire d'Enseignment pratique appliquée de l'uni-

versité de Paris.

Founded in 1897 at Paris. In 1900: President, M. Loyer; honorary members, 12; members, 110.

Publication: Gazette de chimie, Paris, 1900.

GERMANY.

VEREIN FÜR DIE RÜBENZUCKER INDUSTRIE IM ZOLLVEREIN [later, DES DEUTSCHEN REICHS; later, VEREIN DER DEUTSCHEN ZUCKERINDUSTRIE].

Founded in 1850 at Berlin. In 1900: President, De Coste; members, 447.

Publication: Zeitschrift des Vereins [etc.], 1850-1900.

DEUTSCHE CHEMISCHE GESELLSCHAFT ZU BERLIN.

Founded in 1867 at Berlin. In 1900: President, G. Volhard; honorary members, 15; life members, 92; members, 2637; associates, 372.

Publication: Berichte der deutschen chemischen Gesellschaft zu Berlin, 1868-1900. Since 1897 also: Chemisches Centralblatt (established in 1830).

VEREIN ANALYTISCHER CHEMIKER.

Founded in 1878 at Magdeburg, and merged in 1887 with the Deutsche Gesellschaft für angewandte Chemie. See Verein deutscher Chemiker.

Publication: Correspondenzblatt des Vereines analytischer Chemiker, 1878-80.

FREIE VEREINIGUNG BAYERISCHER VERTRETER DER ANGEWANDTEN CHEMIE.

Founded in May, 1883, at Munich. In 1900: President, Albert Hilger; honorary members, 2; members, 124; correspondents, 69.

Publications: Bericht über die 1 [- 18] Versammlung der freien Vereinigung bayerischer Vertreter der angewandten Chemie, 1883-1900. Also reports in: Forschungsberichte über Lebensmittel und ihre Beziehung zur Hygiene, 1894-97: and in: Zeitschrift für Untersuchung der Nahrungs- und Genuss-Mittel, 1898-1900.

VEREIN DEUTSCHER BERUFS-CHEMIKER.

Founded in 1887 at Dresden.

Publication: The "Chemiker und Droguist" (Dresden, 1885) had in 1887 the sub-title: Correspondenzblatt des Vereines deutscher Berufs Chemiker. This title was dropped in 1888.

DEUTSCHE GESELLSCHAFT FÜR ANGEWANDTE CHEMIE.

Founded November, 1887, at Berlin, absorbing the Verein analytischer Chemiker. In 1896 the Society became: Verein deutscher Chemiker, q. v.

Publication: Zeitschrift für angewandte Chemie, 1888-1900. This was begun as: Zeitschrift für die chemische Industrie in 1887.

VEREINIGUNG ÖFFENTLICHER ANALYTISCHER CHEMIKER SACHSENS.

Founded in 1890 at Plauen in Vogtland. In 1900: President, Arthur Forster; members, about 25.

Publication: Zeitschrift für öffentliche Chemie. 1897-1900. Also: Bericht über die Hamptversammlung des Vereines öffentlicher analytischer Chemiker Sachsens.

VEREIN AKADEMISCH-GEBILDETE [later, DEUTSCHER] ZUCKERTECH-NIKER.

Founded in 1891 at Berlin. In 1900: President, H. Claassen; honorary members, 1; members, 406; correspondents, 3.

Publication: Zeitschrift des Vereins akademisch-gebildete Zuckertechniker, 1891-92.

Note.-The organ of publication changed several times.

VERBAND DES LABORATORIUMS-VORSTÄNDE VON DEUTSCHEN HOCH-SCHULEN.

Founded in 1898 [?].

Zweigverein der Zuckertechniker für das Ausland.

Founded at Berlin. In 1901: President, C. Huck.

DEUTSCHE ELEKTROCHEMISCHE GESELLSCHAFT.

Founded in October, 1894, at Berlin. In 1900: President, J. H. van't Hoff; members, about 700.

Publication: Bericht der deutschen elektrochemischen Gesellschaft. 1894-1900.

VEREIN DEUTSCHER CHEMIKER.

Founded in 1896 at Berlin, as successor to Gesellschaft für angewandte Chemie (1887). Its seat is the residence of the President for a given year. In 1900: President, H. Caro; honorary members, 4; members, 2271. Embraced in 1900 the following sections (Bezirk-Vereine): Aachen, Belgien, Berlin, Frankfurt, Hamburg, Hannover, Mittel-Franken, Mittel- und Niederschlesien, Oberrhein, Oberschlesien, Pommern, Rheinland, Rheinland-Westphalen, Saar, Sachsen-Anhalt, Sachsen-Thüringen, Württemberg.

Publication: Zeitschrift für angewandte Chemie, 1887–1900. Cf. Deutsche Gesellschaft für angewandte Chemie.

VERBAND SELBSTÄNDIGER ÖFFENTLICHER CHEMIKER DEUTSCHLANDS.

Founded May 30, 1896, at Nürnberg. In 1900: President, Robert Kayser; members, 161; associates, 102.

Publication: Zeitschrift für öffentliche Chemie (established in 1895), 1897–1900. Also Vol. I. as Vol. III., 1897, of Centralblatt für Nahrungs- und Genussmittel Chemie, sowie Hygiene.

GREAT BRITAIN.

SOCIETY FOR PHILOSOPHICAL EXPERIMENTS.

Founded in 1794 at London.

Publication: Minutes of the Society for Philosophical Experiments, 1794.

NOTE.—A German translation of the *Minutes* was edited by Alex. Nic. Scherer and published at Halle in 1803.

CHEMICAL SECTION OF THE BRITISH ASSOCIATION FOR THE ADVANCE-MENT OF SCIENCE.

Founded in 1831. This is, however, a corporate part of the British Association, and the papers read to the Section are published in the annual Reports of the British Association, 1831-1900. In 1900: President, W. H. Perkin, jun.; number of members not given.

CHEMICAL SOCIETY OF LONDON.

Founded in 1841 at London. In 1900: President, T. E. Thorpe; honorary and foreign members, 33; members, 2300.

Publications: Memoirs and Proceedings of the Chemical Society of London (1841-48); Quarterly Journal, 1849-62; Journal of the Chemical Society, 1863-1900.

SOCIETY OF PUBLIC ANALYSTS.

Founded in 1874 at London. In 1900: President, Walter W. Fisher; honorary members, 9; members, 260.

Publication: Proceedings of the Society of Public Analysts, 1876; The Analyst, 1877-1900.

INSTITUTE OF CHEMISTRY OF GREAT BRITAIN AND IRELAND.

Founded October, 1877, in London; incorporated, 1885. In 1900: President, John Miller Thomson; members, fellows, and associates, 1008 (resident members, 904); students, 118. Total, 1126.

Publications: Proceedings, half yearly, 1878-1900; Register, yearly, 1878-1900; Regulations, yearly, 1878-1900.

SOCIETY OF CHEMICAL INDUSTRY.

Founded in 1881 at London. In 1900: President, Charles F. Chandler; honorary member, 1 (John Glover); members, 3459.

Publication: Journal of the Society of Chemical Industry, 1882-1900.

NOTE.—The Society has eight sections: London, Liverpool, Manchester, Newcastle, New York, Nottingham, Scotland, and Yorkshire.

SOCIETY OF DYERS AND COLOURISTS.

Founded in 1884 at Bradford. In 1900: President, H. Grandage; honorary members, 3; members, 553.

Publication: Journal of the Society of Dyers and Colourists, 1884-1900.

ALEMBIC CLUB.

Founded in 1889 at Edinburgh. This is a private club of only six members and has no president; the Secretary is Leonard Dobbin.

Publishes no journal, but has issued 15 Reprints of Chemical Monographs, etc., 1893-1900, and other works.

CHEMICAL SOCIETIES OF THE XIX CENTURY

INTERNATIONAL ASSOCIATION OF LEATHER-TRADES CHEMISTS.

Founded September, 1897, at London. In 1900: President H. R. Proctor; number of members, —.

Publication: Report of the Proceedings of the Conference of Leather-Trades Chemists, 1897.

ITALY.

ASSOCIAZIONE CHIMICO-FARMACEUTICA FIORENTINA.

Founded in 1877 at Florence. In 1900: honorary members, 20; resident members, 100.

Publication: L'Orosi, Bollettino di chimica, farmacia e scienze affini. Firenze, 1878-1900.

Società chimica di Milano.

Founded in February, 1895, at Milan. In 1900: President, Angelo Menozzi; resident members, 152; correspondents, 133.

Publication: Annuario della Società chimica di Milano, 1896-1900.

ASSOCIAZIONE CHIMICO-INDUSTRIALE DI TORINO,

Founded June 25, 1899, at Turin. In 1900: President, Vittorio Sclopis; honorary members, 4; resident members, 103; correspondents, 87.

Publication: La Chimica industriale, 1899-1900.

SOCIETÀ ITALIANA DEI CHIMICI ANALISTI.

Founded in 1893 at Pavia.

Publication: Atti ufficiale delle Società italiana dei chimici analisti, 1893. This forms a pamphlet of 18 pp. only, and is perhaps a mere prospectus, as the Society ceased to exist before 1900.

JAPAN.

CHEMICAL SOCIETY OF TOKYO.

Founded April, 1878, at Tokyo. In 1900: President, Naokichi Matsui; number of members, 156; associates, 197.

Publication: Tokyo Kagakkai Kaishi, 1880-1000.

SOCIETY OF CHEMICAL INDUSTRY OF JAPAN.

Founded February, 1898, at Tokyo. In 1900: President, Takeaki Enomoto; honorary members, 7; members, 223; associates, 429.

Publication: Kögyö Kagaku Zasshi, 1898-1900.

RUSSIA.

RUSSKAGO KHIMICHESKAGO OBSHTCHESTVA [Russian Chemical Society].

Founded October 26, 1868. The Chairman of the first meeting was D. Mendeléeff. In 1900: President, F. F. Petrushevsky; members, 327.

Publications: Zhurnal Russkago Khimicheskago Obshtchestva. St. Petersburg, 1869-72, 4 vols.

Continued as:

Zhurnal Russkago Khimicheskago Obshtchestva i Fisicheskago Obshtchestva, 1873-78. 6 vols.

Continued as:

Zhurnal Russkago Fisiko-Khimicheskago Obshtchestva, 1879-1900.

SOUTH AFRICA.

CHEMICAL AND METALLURGICAL SOCIETY OF SOUTH AFRICA.

Founded May, 1894, at Johannesburg.

Publication: Proceedings of the Chemical and Metallurgical Society of South Africa, 1894-1897.

SWITZERLAND.

Société chimique de Genève.

Founded February 10, 1878, at Geneva. In 1900: President, F. Kehrmann.

Publication: The Minutes of the monthly meetings are published in: Archives des sciences physiques et naturelles de Genève, and in the Chemiker Zeitung.

VEREIN SCHWEIZERISCHER ANALYTISCHER CHEMIKER. .

Founded March 12, 1887, at Zürich. In 1900: President, A. Bertschinger; number of members, 94.

Publication: The organ of the Society is Schweizerische Wochenschrift für Chemie und Pharmacie, which was established under the title, Schweizerische Zeitschrift für Pharmacie, 1856-62.

UNITED STATES OF AMERICA.

CHEMICAL SOCIETY OF PHILADELPHIA.

Founded in 1792 at Philadelphia, under the presidency of James Woodhouse. Number of members unknown. The Society was in existence for more than ten years.

Publication: Memoir on the Supply and Application of the Blowpipe [etc.] by Robert Hare, 1802.

COLUMBIAN CHEMICAL SOCIETY OF PHILADELPHIA.

Founded August, 1811, at Philadelphia, under the presidency of James Cutbush. Honorary members, 69; junior members, 13.

Publication: Memoirs of the Columbian Chemical Society of Philadelphia. Vol. I., 1813.

CHEMICAL SECTION OF THE AMERICAN ASSOCIATION FOR THE AD-VANCEMENT OF SCIENCE.

A migratory organization, founded in 1875 as a Sub-Section; it became Section C of the A. A. A. S. in 1882 at the second meeting in Montreal. In 1900: Chairman of the Section, Jas. Lewis Howe; members, 89; fellows, 181.

Publications: The Proceedings of the A. A. A. S. has a division containing papers read before the Section of Chemistry.

AMERICAN CHEMICAL SOCIETY.

Founded April 20, 1876, in New York City. In 1900: President, William McMurtrie; honorary members, 10; members, 1546; associates, 123.

Publications: Proceedings of the American Chemical Society, 1877-78; Journal of the American Chemical Society, 1879-1900.

Note.—In 1900 the Society had 12 sections: Rhode Island, Cincinnati, New York, Washington, Lehigh Valley, Chicago, Nebraska, North Carolina, Columbus, North Eastern, Philadelphia, and Michigan.

Association of Official Agricultural Chemists.

Founded September 8, 1884, at Philadelphia. In 1900: President, B. W. Kilgore; members, 350.

Publications: Methods of Analyses, 1884-88; Proceedings, 1889-1900.

NOTE.—Conventions of the Official Agricultural Chemists had been held prior to 1884; in 1880 at Washington and Boston; in 1881 at Cincinnati; and in May, 1884, at Atlanta.

CHEMICAL SOCIETY OF WASHINGTON.

Founded at Washington in 1884. In 1893 became the Washington Section of the American Chemical Society, retaining also its name as above. In 1900: President, H. Carrington Bolton; members, 114.

Publication: Bulletin of the Chemical Society of Washington, 1884-92.

NEW ENGLAND ASSOCIATION OF CHEMISTRY TEACHERS.

Founded February 19, 1898. Meetings are held in New England. In 1900: President, Rufus P. Williams; honorary members, 8; active members, 50; associates, 22.

Publications: Circulars of Information and Reports, 1898-1900. Also Registers.

VICTORIA.

SOCIETY OF CHEMICAL INDUSTRY OF VICTORIA.

Founded in 1900 under the Presidency of Orme Masson; membership, about 100.

ADDENDUM.

Société d'Arcueil.

Founded in 1807 at Arcueil. Dissolved in 1822. Members (at any one time), 12.

Publication : Mémoires de physique et de chimie. Paris, 3 vols., 8vo. 1807-17.

NOTE.—This private organization was founded by C. L. Berthollet; the meetings were held at his country house in Arcueil, near Paris. The membership included: La Place, C. L. Berthollet and his son A. B. Berthollet, Biot, Gay Lussac, Humboldt, Thénard, Decandolle, Collet-Descotils, Berard, Chaptal, Dulong, Poisson, Malus.

The foregoing list does not include Academies of science nor Associations of general science (with a few exceptions); it does not embrace societies having for their object industries involving chemical processes in part, excepting the refining of sugar; nor does it include the numerous societies of brewers and of beer-making, among which may be named the following:

- BRAU-INDUSTRIE VEREIN IM KÖNIGREICH BÖHMEN, founded at Prague in 1874, and publishing the Böhmische Bierbrauer.
- DEUTSCHE BRAUERBUND, founded at Nürnberg in 1861, and publishing the Allgemeine Hopfen-Zeitung.
- WÜRTTEMBERGISCHE BRAUERBUND, founded at Waldsee in 1872, and publishing the Schwäbische Bierbrauer.
- BADISCHE BRAUERBUND, founded at Nürnberg in 1876, and publishing the Hopfenlaube.
- DEUTSCHE BRAUMEISTER VEREIN, founded at Berlin, 1887, and publishing the Deutsche Brau-Industrie.
- Association générale des brasseurs Belges, founded at Brussels in 1874, and publishing Revue des Bieres.

COUNTY BREWERS' SOCIETY, England, publishing since 1871 the Brewers' Guardian.



SMITHSONIAN MISCELLANEOUS COLLECTIONS

PART OF VOLUME XLVI

INDEX TO THE LITERATURE

OF

GALLIUM

1874-1903

PREPARED BY

PHILIP E. BROWNING, PH. D.



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LETTER OF TRANSMITTAL.

WASHINGTON AND LEE UNIVERSITY, DEPARTMENT OF CHEMISTRY, LEXINGTON, VA., October 18, 1904.

The Committee of the American Association for the Advancement of Science having charge of Indexing Chemical Literature has voted to recommend to the Smithsonian Institution for publication the following:

INDEX TO THE LITERATURE OF GALLIUM, 1875-1903;

INDEX TO THE LITERATURE OF GERMANIUM, 1886-1903;

both prepared by Philip E. Browning, Ph. D., of the Kent Chemical Laboratory of Yale University.

> JAMES LEWIS HOWE, Chairman.

Mr. S. P. LANGLEY,

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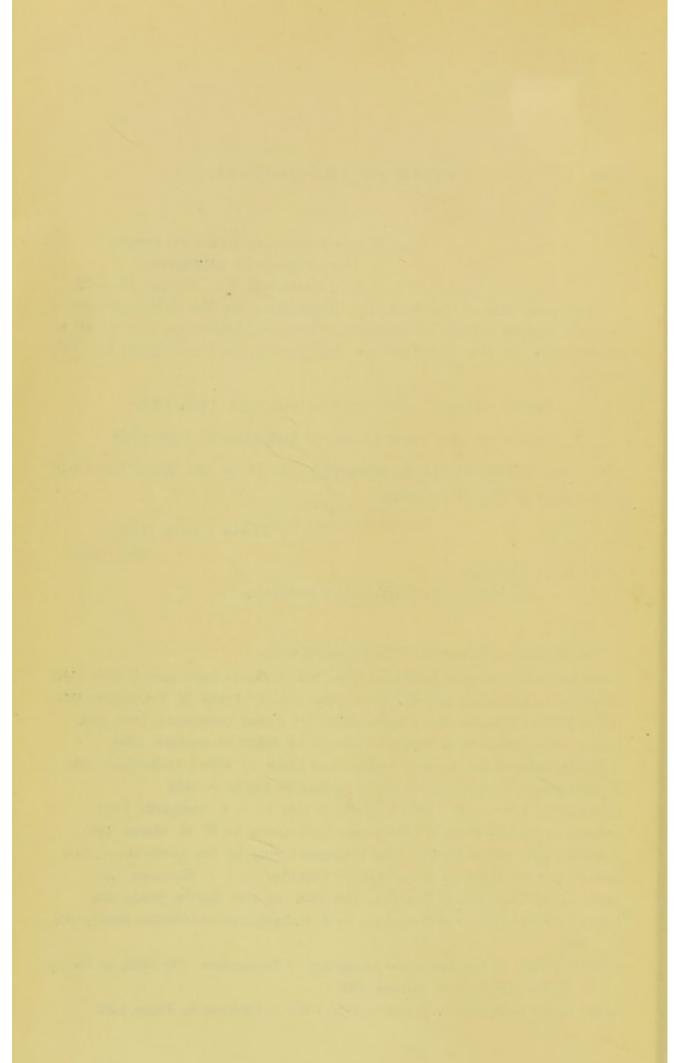
Secretary of the Smithsonian Institution.

This publication forms one of the following series :

Index to the Literature of Uranium, 1785-1885, by Henry Carrington Bolton, 1885. Index to the Literature of Columbium, 1801-1887, by Frank W. Traphagen, 1888. Index to the Literature of the Spectroscope, by Alfred Tuckerman, 1888, 1902. Index to the Literature of Thermodynamics, by Alfred Tuckerman, 1890. A Bibliography of the Chemical Influence of Light, by Alfred Tuckerman, 1891. A Bibliography of Aceto-Acetic Ester, by Paul H. Seymour, 1894. Index to the Literature of Didymium, 1842-1893, by A. C. Langmuir, 1895. Indexes to the Literature of Cerium and Lanthanum, by W. H. Magee, 1895. A Bibliography of the Metals of the Platinum Group, by Jas. Lewis Howe, 1897. Review and Bibliography of the Metallic Carbides, by J. A. Mathews, 1898. Index to the Literature of Thallium, 1861-1897, by Miss Martha Doan, 1898. Index to the Literature of Zirconium, by A. C. Langmuir and Charles Baskerville, 1899.

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Index to the Literature of Thorium, 1817-1902, by Cavalier H. Joüet, 1903.



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INDEX TO THE LITERATURE OF GALLIUM.

(1875 - 1903.)

PREPARED BY PHILIP E. BROWNING.

1875: (1). LECOQ DE BOISBAUDRAN. (Discovery.)

Compt. rend., LXXXI, 493; Ber. VIII, 1355, 1680; Ztschr. Anal. Chem., XVI, 239; Bull. Soc. Chim. (Paris), n. f., XXIV, 370; Amer. J. Sci., (3), XI, 320; Jsb. (1875), 205; Pogg. Ann., CLVIII, 494; Chem. News, XXXII, 159, 294; Amer. Chemist, VI, 146; Pharm. J. Trans., (3), VI, 282; N. Arch. Ph. Nat. liv., 283; Ann. Chim. Phys., (5), X, 100; J. Chem. Soc. (Lond.), XXX, 190; Chem. Centrbl. (1875), 658; Ding. Pol. J., CCXVIII, 376; Tidsskrift, (1), XIV, 349; Gazz. Chim. Ital., VIII, 24; Phil. Mag., L, 414; Monit. Scientif. (1876), 88; Berg. u. Hüttenmännische Ztg. (1876), 198, 207, 237; Arch. der Pharm., v, 352; Deutsche Industriezeit (1875), 731.

- 1875: (2). Hugo. (Objection to name.) Compt. rend., LXXXI, 530.
- 1875: (3). MENDELEEFF. (Prediction previous to discovery.)

Compt. rend., LXXXI, 969; J. Chem. Soc. (Lond.), XXX, 530; Chem.
News, XXXII, 293; Jsb. (1875), 207; Bull. Soc. Chim., n. f., XXV, 295; Chem. Centrbl. (1875), 817; Phil. Mag., (5), 1, 542.

1876: (1). LECOQ DE BOISBAUDRAN. (Spectrum.)

Compt. Rend., LXXXII, 168; Chem. News, XXXIII, 35; Phil. Mag., (5), 1, 176; Amer. Chemist, vi, 299; Chem. Centrbl. (1876), 194.

1876: (2). LECOQ DE BOISBAUDRAN. (Physical and chemical properties of the metal.)

Compt. rend., LXXXII, 1036, 1037; Bull. Soc. Chim. (Paris), (2), XXV, 400, 521; XXVI, 158, 433; Arch. Ph. Nat., LVI, 45; Chem. News, XXXIV, 150, 183; Phil. Mag., (5), II, 398, 479; Pogg. Ann., CLVIII, 494; Chem. Centrbl. (1876), 451, (1877), 19; Gazz. Chim. Ital., VII, 32; Ber., IX, 64, 1608, 1807.

1876: (3). LECOQ DE BOISBAUDRAN. (Extraction.)

Compt. rend., LXXXII, 1098; LXXXIII, 636; Bull. Soc. Chim. (Paris),
(2), XXVII, 49, 144; J. Chem. Soc. (Lond.), XXX, 275; XXXI, 48, 521;
Chem. Centrbl. (1876), 452, 705; Gazz. Chim. Ital., VII, 34; Chem.
News, XXXIII, 230; XXXIV, 173; Ber., 1X, 726, 731; Phil. Mag., (5),
II, 480.

1876: (4). LECOQ DE BOISBAUDRAN. (Physical properties.)

Compt. rend., LXXXIII, 611, 1100; Phil. Mag., (5), 1, 175; 11, 398;
 Wag. Jsb., XXIII, 7; Chem. News, XXXIII, 193; Bull. Soc. Chim. (Paris), XXVI, 458; Arch. d. Pharm., VII, 453.

- 1876: (5). DELACHANAL and MERMET. (Presence in zinc.)
 Bull. Soc. Chim. (Paris), n. f., xxv, 197; xxvii, 49; Chem. Centrbl. (1876), 339; Ber., x, 91; Wag. Jsb., xxii, 1; xxiii, 9.
- 1876: (6). LECOQ DE BOISBAUDRAN. (Reactions, behavior toward reagents.)

Compt. rend., LXXXII, 663, 824; Chem. Centrbl. (1876), 721; (1877), 51; Chem. News, XXXIV, 217.

- 1876: (7). LECOQ DE BOISBAUDRAN. (Gallium crystals.) Compt. rend., LXXXIII, 1044; J. Chem. Soc. (Lond.), XXXI, 440; Chem. News, XXXV, 11; Chem. Centrol. (1877), 65.
- 1877: (1). MUIR. (Comparison with Ekaaluminum.)
 Phil. Mag., (5), III, 281; Chem. Centrbl. (1877), 434; Wag. Jsb., xxIII, 8.
- 1877: (2). LECOQ DE BOISBAUDRAN. (Review of work.)
 Ann. Chim. Phys., (5), x, 100; Chem. Centrol. (1877), 178; Gazz. Chim. Ital., vii, 332; Chem. News, xxxv, 148, 157, 167.

1878: (1). LECOQ DE BOISBAUDRAN and JUNGFLEISCH. (Extraction.)

Compt. rend., LXXXVI, 475; Amer. J. Sci., (3), XV, 473; Phil. Mag., (5), v, 318; Jsb. (1878), 251; J. Chem. Soc. (Lond.), XXXIV, 374, 556, 837; Chem. Centrbl. (1878), 210; Chem. News, XXXVII, 121; Monit. Scientif. (1878), 290; Chem. Industrie (1878), 130; Wag. Jsb., XXIII, 9; XXIV, 5; Bull. Soc. Chim. (Paris), XXVII, 144; XXX, 501; Amer. Chemist, VII, 309.

1878: (2). LECOQ DE BOISBAUDRAN and JUNGFLEISCH. (Properties of the metal.)

Compt. rend., LXXXVI, 577; Jsb. (1878), 253; Chem. Centrbl. (1878), 276; Chem. News, XXXVII, 142.

- 1878: (3). LECOQ DE BOISBAUDRAN. (Halogens.) Comp. rend., LXXXVI, 756; Jsb. (1878), 254; Chem. Centrol. (1878), 322.
- 1878: (4). DUPRE. (Researches.)

Compt. rend., LXXXVI, 720; Amer. J. Sci., (3), XV, 474; Jsb. (1878), 254; Bull. Soc. Chim. (Paris), n. s., XXX, 503; J. Chem. Soc. (Lond.), XXXIV, 472; Chem. Centrbl. (1878), 322; Wag. Jsb., XXIV, 7; Chem. News, XXXVII, 184.

1878: (5). BERTHELOT. (Physical constants.)

Compt. rend., LXXXVI, 786; Amer. J. Sci., (3), XVII, 166; Phil. Mag.,
(5), VII, 75; Ann. Chim. Phys., (5), XV, 242; Jsb. (1878), 78; J.
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1878: (7). LECOQ DE BOISBAUDRAN. (Equivalent.)

- Comp. rend., LXXXVI, 756, 941; Bull Soc. Chim. (Paris), n. s., XXIX, 385; J. Amer. Chem. Soc., r, 320; J. Chem. Soc. (Lond.), XXXIV, 646; Wag. Jsb., XXIV, 8; Chem. Centrbl. (1878), 387; Chem. News, XXXVI, 216; Tidsskrift, (1), XVII, 144.
- 1878: (8). LECOQ DE BOISBAUDRAN. (Alloys with aluminum.) Compt. rend., LXXXVI, 1249; Chem. Centrbl. (1878), 483; Chem. News, XXXVII, 274; Wag. Jsb., XXIV, 9.
- 1878: (9). LECOQ DE BOISBAUDRAN. (Atomic weight.)
 Bull. Soc. Chim. (Paris), n. s., XXXII, 393; Amer. J. Sci., (3). XVI, 137; Jsb. (1878), 250; Chem. News, XXXVII, 138.
- 1878: (10). REGNAULD. (Electrochemistry.)

Compt. rend., LXXXVI, 1457; Jsb. (1878), 135; Chem. Centrol, (1878), 561; Wag. Jsb., XXV, 9.

- 1879: (1). LOCKYER. (Heating of metal in vacuo.) Chem. News, x1, 101; Jsb. (1879), 176; Compt. Rend., LXXXIX, 514.
- 1879: (2). JUNGFLEISCH. (Separation from blendes.)
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- 1880: (1). SCHUCHT. (Electrolysis of salts.)
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- 1880: (2). CORNWALL. (Occurrence in American blendes.)
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- 1881: (1). CLARKE. (Atomic weight.) Amer. Chem. J., 111, 263; Phil. Mag., (5), X11, 101; Jsb. (1881), 7.
- 1881: (2). LECOQ DE BOISBAUDRAN. (Anhydrous chlorides.)
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 xL, 1103; XLII, 364; Chem. Centrbl. (1881), 645; (1882), 5; Chem.
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- 1881: (3). CLARKE. (Atomic weight.) Amer. Chem. J., 111, 263; Phil. Mag., (5), X11, 101; Jsb. (1881), 7.
- 1882: (1). LECOQ DE BOISBAUDRAN. (Oxychloride.) Compt. rend., xciv, 695; Jsb. (1882), 287; J. Chem. Soc. (Lond.), xLII, 698; Chem. Centrbl. (1882), 284; Chem. Ztg. (1882), vr, 266.
- 1882: (2). LECOQ DE BOISBAUDRAN. (Decomposition of protochloride.) Compt. rend., xcv, 18; J. Chem. Soc. (Lond.), xLII, 1167.

- 1882: (3). LECOQ DE BOISBAUDRAN. (Precipitants.)
 Comp. rend., xciv, 1154, 1228; Jsb. (1882), 1295; J. Chem. Soc. (Lond.), xLII, 897; Chem. Centrbl. (1882), 418.
- 1882: (4). LECOQ DE BOISBAUDRAN. (Separations.) FROM NA., K., LI., Cs., RB., BA., SR., CA., MG., AL., CR.
 - Compt. rend., xciv, 1228; Jsb. (1882), 1295; Ann. Chim. Phys., (6), 11, 176; Chem. Ztg. (1882), vi, 493.

FROM BE., CE., Y., FE., TH.

Comp. rend., xciv, 1439; Jsb. (1882), 1295; Ann. Chim. Phys., (6), 11, 176; Chem. Centrbl. (1882), 519.

FROM ZR., MN., ZN.

FROM CO., NI., TL.

Compt. rend., xcv, 157; Jsb. (1882), 1295; Ann. Chim. Phys., (6), 11, 176; Bull. Soc. Chim. (Paris), xxxix, 547; Chem. Centrbl. (1882), 606.

FROM IN., CD.

Compt. rend., xcv, 410; Jsb. (1882), 1295; Ann. Chim. Phys., (6), 11, 176; Bull. Soc. Chim. (Paris), xxx1x, 547; Chem. Centrbl. (1882), 646.

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Compt. rend., xcv, 503; Jsb. (1882), 1295; Ann. Chim. Phys., (6), 11, 176; Bull. Soc. Chim. (Paris), xxx1x, 547; Chem. Centrbl. (1882), 727.

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- 1883: (2). DONATH and MAYRHOFER. (Atomic volume.) J. Chem. Soc. (Lond.), XLII, 1323; Ber., XVI, 1588; Jsb. (1883), 24.
- 1883: (3). RABUTEAU. (Physiological effect.) Compt. rend. de la Soc. de Rive (1883) 310, Chem. Centrbl. (1884), 64.

Compt. rend., xciv, 1625; xcix, 526; Jsb. (1882), 1295; Ann. Chim. Phys. (6), 11, 176; Chem. Centrbl. (1882), 519.

^{1883: (1).} LECOQ DE BOISBAUDRAN. (Separations.) FROM RH., IR., RU., Os., As., SE.

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- 1884: (1). LECOQ DE BOISBAUDRAN. (Separations.) FROM B. (Organic matter.)
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- 1884: (2). LECOQ DE BOISBAUDRAN. (Solubility of the ferrocyanide.) Compt. rend., xcix, 526; Jsb. (1884), 1602.
- 1884: (3). CARNELLY. (Relation of color to atomic weight.)
 Phil. Mag., (5), XVIII, 130; Ber. (1884), 2151; Chem. News, L, 193; Jsb. (1884), 43.
- 1884: (4). CLARKE. (Atomic weight.) Chem. News, XLIX, 260, 273; Chem. Ztg. (1884), 930.
- 1885: (1). EHRLICH. (Extraction.) Chem. News, LI, 115; Chem. Ztg. (1885), 78; Jsb. (1885), 496.
- 1885: (2). LECOQ DE BOISBAUDRAN. (Alloys with indium.)
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 Centrbl. (1885), 297; Chem. Ztg. (1885), 1, 470.
- 1885: (3). GLADSTONE. (Refraction equivalent.) Phil. Mag., (5), xx, 162; Jsb. (1885), 310.
- 1885 : (4). KUNERT. (Extraction.) Chem. Ztg. (1885), 1x, 1826 ; Ber., x1x, 74 ; Jsb. (1885), 496.
- 1886: (1). LECOQ DE BOISBAUDRAN. (Identity with austrium.) Compt. rend., CII, 647, 1436; Jsb. (1886), 407; Dingl. Pol. J., CCLXI, 96; Wag. Jsb., XXXII, 224.
- 1886: (2). LECOQ DE BOISBAUDRAN. (Estimation.) Ann. Chim. Phys., (6), x1, 429.
- 1886: (3). WILLGEROOT. (As halogen transferrer.)
 - J. Prakt. Chem., xxxv, 142, 391; Jsb. (1887), 618; Bull. Soc. Chim. (Paris), x1.VIII, 346; J. Chem. Soc. (Lond.), LII, 326; Chem. Ztg. Rep., 1887, 43; Chem. Centrbl. (1887), 507.
- 1887: (1). LECOQ DE BOISBAUDRAN. (Red fluorescence of the oxide with chromium.)

Compt. rend., crv, 330, 1584; Chem. News, Lv1, 12; Ber., xx, 456R; Jsb. (1887), 358.

- 1887: (2). LECOQ DE BOISBAUDRAN. (Volatility of the chloride.) Ann. Chim Phys. (1887), (6), xr, 420; Chem. Ztg. Rep. (1887), 186.
- 1888: (1). LECOQ DE BOISBAUDRAN. (Fluorescence of compounds.) Compt. rend., cv, 1228; Chem. Centrbl. (1888), 462.
- 1888: (2). FRIEDEL and CRAFTS. (Vapor density of the chloride.)
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- 1888: (3). NILSON and PETTERSSON. (Valence and the chloride.)
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- 1889: (1). BARTLETT. (Occurrence.) Chem. Soc. Ind. J., VIII, 896; Jsb. (1889), 341.
- 1889: (2). RAMSAY. (Molecular weight.) J. Chem. Soc. (Lond.), LV, 531.
- 1890: (1). WINKLER. (Reduction of the oxide by magnesium.) Ber., XXIII, 788; J. Chem. Soc. (Lond.), LVIII, 693.
- 1891: (1). CLARKE. (Atomic weight.) Chem. News, LXIII, 76; Jsb. (1891), 79.
- 1892: (1). LECOQ DE BOISBAUDRAN. (Spark spectrum.)
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- 1893: (1). WILDE. (Spectrum.) Proc. Roy. Soc., LIII, 369; Jsb. (1893), 151.
- 1893: (2). GLADSTONE. (Molecular refraction and dispersion.) Phil. Mag., xxxv, 365; Ber., xxv, 357r; Chem. News, LXVII, 94; Jsb. (1893), 42.
- 1893: (3). KIRTLAND. (Occurrence in Australian blendes.) Australian Assoc. Adv. Sci. (1893), 266; J. Chem. Soc. (Lond.), LXX, 183.
- 1895: (1). LECOQ DE BOISBAUDRAN. (Atomic weight.) Compt. rend., cxx, 361.
- 1896: (1). HARTLEY and RAMAGE. (Occurrence.) Lond. Roy. Soc. Proc., Lx, 35; Amer. J. Sci., (4), 11, 378; Jsb. (1896), 554; J. Soc. Chem. Indust., xvi, 367.
- 1897: (1). HARTLEY and RAMAGE. (Occurrence.)
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- 1897: (2). WINKLER. (History of the discovery.) Ber., xxx, 13.
- 1897: (3). WYRUBOFF. (Silico-tungstate.) Bull. Soc. Franc. Min., xix, 219; J. Chem. Soc. (Lond.), LXXII, 173.
- 1898: (1). LANDOLT, OSWALD, and SEUBERT. (Atomic weight.) Ber., XXXI, 2762.

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- 1898: (2). HARTLEY and RAMAGE. (Occurrence in iron ores, etc.) Lond. Roy. Soc. Proc., LX, 393; J. Chem. Soc. (Lond.), LXXIV, 236; Chem. Centrbl. (1897), 1, 455; Ztschr. anorg. chem., XVIII, 232; Dublin Roy. Soc. Proc., n. s., VIII, 703.
- 1899: (1). MEYER. (Magnetic properties.) Monatsh. f. Chem., xx, 380.
- 1899: (2). HARTLEY and RAMAGE. (Spectrum.) Astrophy. J., 1x, 214.
- 1901: (1). HARTLEY and RAMAGE. (Occurrence.) Lond. Roy. Soc. Proc., LXVIII, 99; Dublin Roy. Soc. Sci. Trans., VII; Amer. J. Sci., (4), XI, 323.
- 1904: (1). RIMATORI. (Occurrence in Sardinian blendes.)
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SMITHSONIAN MISCELLANEOUS COLLECTIONS

PART OF VOLUME XLVI

INDEX TO THE LITERATURE

OF

GERMANIUM

1886-1903

PREPARED BY PHILIP E. BROWNING



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LETTER OF TRANSMITTAL.

WASHINGTON AND LEE UNIVERSITY, DEPARTMENT OF CHEMISTRY, LEXINGTON, VA., October 18, 1904.

The Committee of the American Association for the Advancement of Science having charge of Indexing Chemical Literature has voted to recommend to the Smithsonian Institution for publication the following:

INDEX TO THE LITERATURE OF GALLIUM, 1875-1903;

INDEX TO THE LITERATURE OF GERMANIUM, 1886-1903;

both prepared by Philip E. Browning, Ph. D., of the Kent Chemical Laboratory of Yale University.

JAS. LEWIS HOWE, Chairman.

Mr. S. P. LANGLEY,

Secretary of the Smithsonian Institution.

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INDEX TO THE LITERATURE OF GERMANIUM. (1886–1903.)

PREPARED BY PHILIP E. BROWNING.

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PART OF VOLUME XLVI

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OF

INDIUM

1863-1903

PREPARED BY

PHILIP E. BROWNING, PH. D. of the Kent Chemical Laboratory of Yale University



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LETTER OF TRANSMITTAL.

WASHINGTON AND LEE UNIVERSITY, DEPARTMENT OF CHEMISTRY,

LEXINGTON, VA., March 25, 1905.

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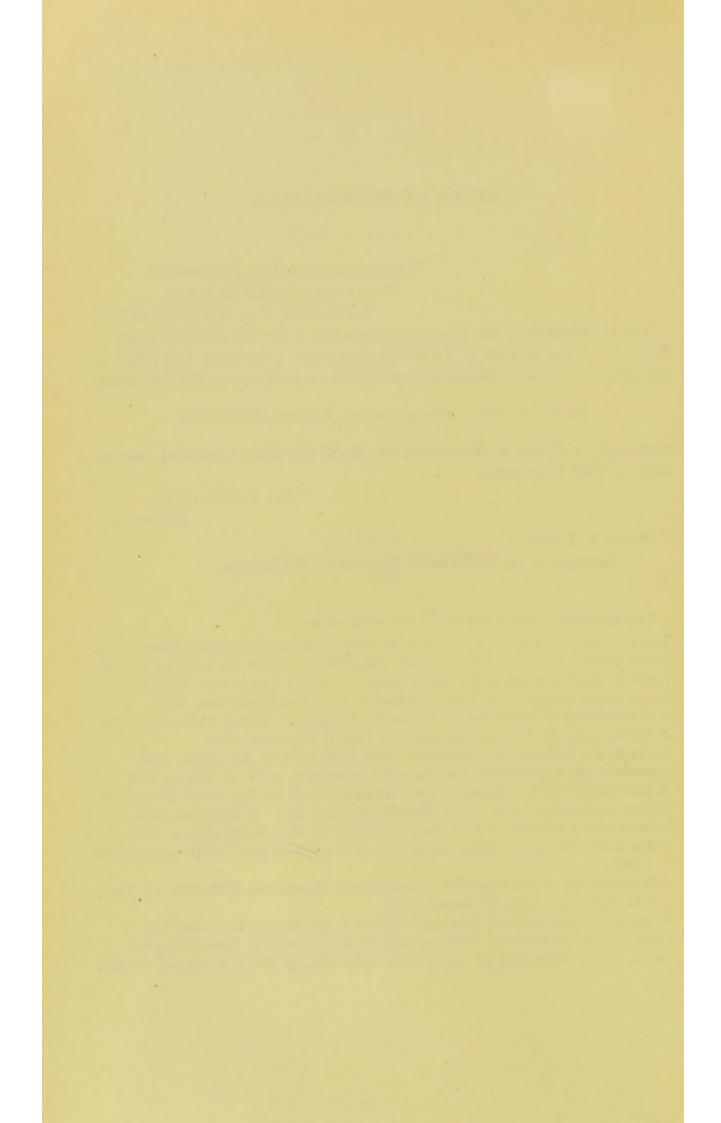
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(1863 - 1903.)

PREPARED BY PHILIP E. BROWNING.

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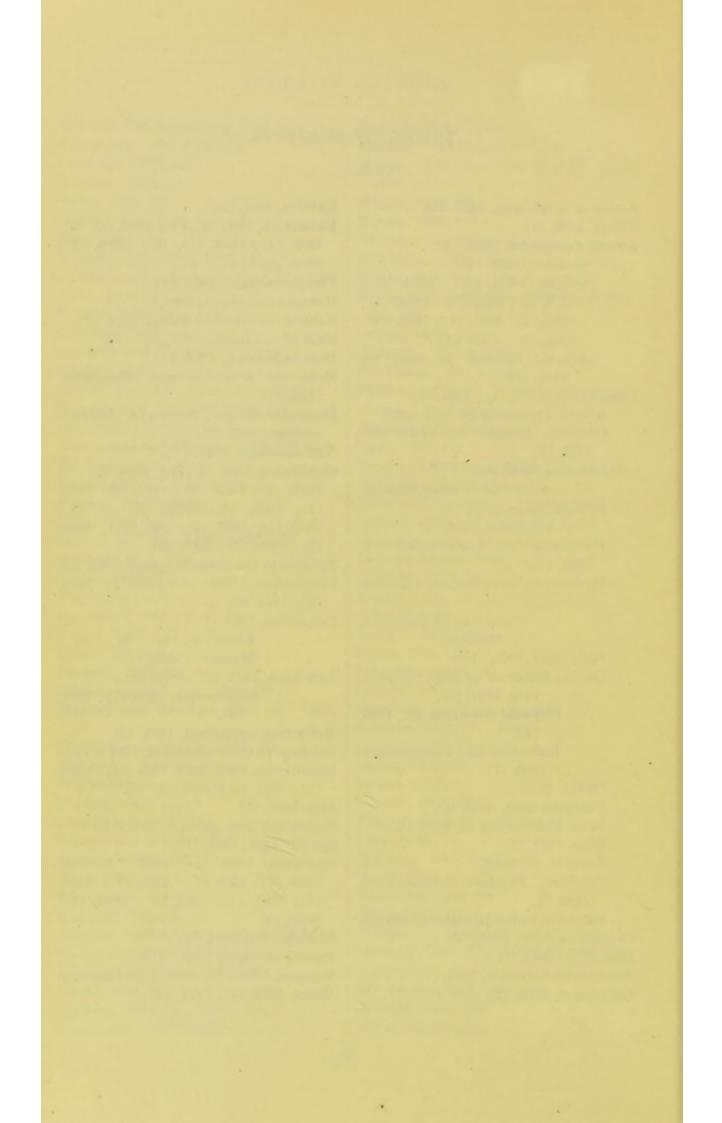
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SMITHSONIAN MISCELLANEOUS COLLECTIONS.

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INDEX TO THE LITERATURE OF THALLIUM,

1861-1896.

by MARTHA DOAN.



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LETTER OF TRANSMITTAL.

WASHINGTON, July 6th, 1898.

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INDEX TO THE LITERATURE OF THALLIUM, 1861-1896, by Miss Martha Doan.

H. CARRINGTON BOLTON, Chairman.

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Secretary of the Smithsonian Institution.



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BY MARTHA DOAN.

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SMITHSONIAN MISCELLANEOUS COLLECTIONS.

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OF

THORIUM.

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PREFACE.

This Index to the Literature of Thorium has been prepared after a very laborious and painstaking search through many scientific and technical journals.

Most of the references have been verified, and usually the original article heads the list, but in some few cases this was difficult to determine.

It is not offered as absolutely complete, and the compiler requests that any one using the index would send corrections and addenda to him.

Minerals now recognized as containing thorium have been mentioned only in such cases when the earth has been found. The patent literature relative to the use of thorium in the arts is not included.

C. H. J.

Columbia University, New York, 1902.



(1817 - 1902.)

BY CAVALIER H. JOUET, PH. D.

1817: 1. GAHN, WALLMANN, EGGERTZ, BERZELIUS. Undersökning af några i trakten kring Fahlun funna Fossilier, och af deras Lagerställen.

> Afh. Fys. Kemi, 1818, 5, 1–93; Oken, Isis, 1819, col 391–409; J. für Chem. (Schweigger), 1817, 21, 25–43; Ann. Phil. (Thomson), 1817, 9, 160–161, 452–460; Ann. chim. phys., 1817, 5, 5–21; Quart. Jour. Sci. Arts, 1817, 2, 443; Ann. Mines, 1818, [1], 3, 151–160; Roy. Soc. C. Sci. Papers, 1867, 1, 340, and 1868, 2, 457, 754.

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- 1817: 3. GAHN. Thorine, eine neue Erde. Oken, Isis, 1817, 1, col 1317-1320; Roy. Soc. C. Sci. Papers, 1868, 2, 754.
- 1817: 4. BERNHARDI. Das allgemeine Krystallizations system der chemischen Elemente. "Thorinium."

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> Ann. der Phys. Pogg., 1818, 59, 247-254; Roy. Soc. C. Sci. Papers, 1867, 1, 333.

- 1821: 6. BERZELIUS. Nya metalliska Kroppar. "Thorium." Årsb. Phys. Kemi, 1821, 66; Berzelius' Jsb., 1822, 1, 50; Archiv. Bergbau, 1823, 8, 376.
- 1821: 7. BERZELIUS. Thorjord funnen på Bornholm (now problematical).

Årsb. Phys. Kemi, 1821, 57; Berzelius' Jsb., 1822, 1, 40.

1823: 8. BERZELIUS. Undersökning af flusspats-syran och dess märkvärdigaste föreningar. "Tillagg om Thorjorden" (proves to be yttrium phosphate).

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Ann. chim. phys., 1824, 26, 39–43; 1824, 27, 53–67, 167–177, 287–308, 337–359; 1825, 29, 295–314, 337–372;
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1825, 44, 348-350; Årsb. Phys Kemi, 1825, 118; Berzelius' Jsb., 1826, 5, 112, 113; Ann. Mines, 1826 [I], 12, 190; Quart. Jour. Sci. Arts, 1825, 18, 156, 157; Annals Phil. (Thomson), 1824, 8, 330-343, 450-457; 1824, 9, 124-131; 1824, 10, 116-130; Roy. Soc. C. Sci. Papers, 1867, 1, 335.

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> Ann. des sci. naturelles, 1825, No. 5, 430–432; Ann. Phil. (Thomson), 1826, 11, 23–24; Edin. J. Sci., 1825, 3, 332–334; Roy. Soc. C. Sci. Papers, 1867, 1, 335.

- 1826: 10. Wöhler. Ueber den Pyrochlor, eine neue Mineral species "Ceroxyd" (unrein).
 - Ann. der Phys. Pogg., 1826, 7, 417-428; Ztschr. Kryst, 1826, 2, 385-389; Ber., 1882, 15, 3150*a*; Berzelius' Jsb., 1828, 7, 175-176; Årsb. Phys. Kemi, 1827, 172-173; Beudant. Min., 1832, vol. 2, 649, 756; Rammelsberg's Min. Chem., 1875, 2d ed., 371-375; Roy. Soc. C. Sci. Papers, 1872, 6, 411.
- 1827: 11. Rose. Pyrochlore, a new mineral species. Edin. J. Sci., 1827, 6, 358-361.
- 1828: 12. BERZELIUS. Ueber den Thorit, ein neues mineral und eine darin enthaltene neue Erde, die Thorerde.
 - Ann. der Phys. Pogg., 1829, 15, 633-634; Berzelius' Traité de Chimie, French ed., 1846, 2, 179-184: Rammelsberg's Min. Chem., 1860, 544-546; Edin. J. Sci., 1829, 1, 207-209; 1829, 2, 223-225; Quart. Jour. Sci. Arts, 1829, 2, 412-413; 1830, 1, 58-104; 1830, 1, 417-419; Gmelin-Kraut, Handb. anorg. Chemie, 1874-1886, II¹, 881; Hensmans, Repertoire, 1829, June; Phil. Mag., 1829, 6, 392-393; Roy. Soc. C. Sci. Papers, 1867, 1, 336.
- 1829: 13. BERZELIUS. Undersökning af ett nytt mineral som innehåller en förut obekant jord.
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 [7], 58, 463, July 22; J. de pharm., 1829, 15, 488–489; Am. J. Sci.,
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- 1830: 20. BERZELIUS. Atomengewichte der einfachen Körper. Pharm. Centrbl., 1830, 8-10.
- 1830: 21. BERZELIUS. Untersuchung einer minerals von Brevig, Norwegen. "Thorium," "Thorit."

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- 1831: 23. BERZELIUS. Om Vanadin och dess egenskaper. "Vanadinsyrad Thorjord."
 - Kongl. Sv. Vet. Acad. Handl., 1831, 1-67; Ann. der Phys. Pogg., 1831, 22, 1-67; Ann. chim. phys., 1831, 47, 337-409; J. für Chem. (Schweigger), 1831, 62, 121-124; 323-374; Berzelius, Traité de Chimie, 1831, t. 4, 642-686; J. für Chem. (Schweigger), 1831, 63, 26-54; Årsb. Phys. Kemi, 1831, 99-110; Berzelius' Jsb., 1832, 11, 97-108; J. tech. chem., 1831, 1, 141-142; Ztschr. Physik u. Mathematik, 1831, 9, 391-392; Phil. Mag., 1831, 10, 321-337; 1831, 11, 7-20; Magazin für Pharm., 1831, 33, 249-253; Roy. Soc. C. Sci. Papers, 1868, 2, 336; 1872, 6.
- 1832: 24. BERZELIUS. Recherches sur la thorine, nouvel oxyde. Ann. der Phys. Pogg., 1829, 16, 385-415; Ann. Mines, 1832 [3], 1, 98-106.
- 1832 : 25. BERZELIUS. Analyse du thorite minéral contenant une nouvelle terre.

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- 1832: 27. BERZELIUS. Mention of false discovery of Thorium (xenotime).

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- 1833: 29. WÖHLER. Thorerde im Pyrochlor.
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- 1833: 31. BERZELIUS. Undersökning af tellurens egenskaper. "Tellursyrlig thorjord."
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- 1837 : 36. BERZELIUS. Atomgewichte der einfachen Körpers. Pharm. Centrbl., 1837, 1–2.
- 1838: 37. BERZELIUS. Atomgewichte der einfachen Körpers. Pharm. Centrbl., 1838, 1–2.

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- 1839: 40. Rose, G. Beschreibung einiger neuen Mineralien des Urals. "Tschewkinit."

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- 1839: 41. KERSTEN. Untersuchung des Monazits, eines Thorerde und Lantanoxyd enthaltenden Minerals vom Urals.
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- 1839: 42. WÖHLER. Analyse des Pyrochlors. Miask and Brevig.
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- 1840: 45. Rose, Gustav. Ueber die Identität des Edwardsit und Monazit.

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- 1842: 49. BERZELIUS. Atomgewichte der einfachen Körper. Pharm. Centrbl., 1842, 1-2.
- 1842: 50. NORDENSKIÖLD. Utkast till ett examinations-system för mineralierne.

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- 1842: 51. Rose. "Pyrochlor" and "Monazit." Reise nach dem Ural, 1842, 2, 64-66, 87-92, 447.
- 1843: 52. BERZELIUS. Atomgewichte und Aequivalente der einfachen Körper.

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1844: 53. BERZELIUS. Atomgewichte und Aequivalente der einfachen Körper.

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- 1844: 55. Rose. Ueber die Titansäure. "Ueber die in Natur vorkommenden Mineralien, Tschewkinit." (Rose finds no thoria, but later Hermann does find thoria.)

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INDEX

TO THE

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LITERATURE OF DIDYMIUM

1842-1893

BY

A. C. LANGMUIR, PH. D.



CITY OF WASHINGTON PUBLISHED BY THE SMITHSONIAN INSTITUTION



LETTER OF TRANSMITTAL.

NEW YORK, JULY 24, 1894.

The Committee of the American Association for the Advancement of Science having charge of Indexing Chemical Literature has voted to recommend to the Smithsonian Institution for publication the three following Indexes: —

AN INDEX TO THE LITERATURE OF CERIUM.¹

AN INDEX TO THE LITERATURE OF LANTHANUM.¹ Both by W. H. Magee, Ph. D.

AN INDEX TO THE LITERATURE OF DIDYMIUM. By A. C. Langmuir, Ph. D.

The latter has already appeared in the School of Mines Quarterly, No. 1, Vol. XV.

H. CARRINGTON BOLTON,

Chairman.

To the SECRETARY of the SMITHSONIAN INSTITUTION.

¹ These Indexes are printed as Smithsonian Publication No. 971.



INDEX TO THE LITERATURE OF DIDYMIUM - 1842-1893.

By A. C. LANGMUIR, PH. D.

THE following paper is offered to chemists with the hope that it may be of some value to them in their researches on an element of great theoretical and scientific interest, particularly as an example of the wonderful results accomplished by the use of the spectroscope in modern chemistry. The voluminous literature of didymium affords a striking illustration of the pursuit of science for its own sake, and with no reward beyond the satisfaction of having advanced the cause of truth.

Original work, at the present time, must always be preceded by a long and painstaking search through the literature, which consumes no inconsiderable amount of time. Anything which can lighten the labors of the investigator in this direction is sure to be a welcome addition to the literature.

In 1882 Dr. H. Carrington Bolton originated the idea of indexing the literature of each of the chemical elements, and a Committee on Indexing Chemical Literature was appointed by the American Association for the Advancement of Science. The committee annually reports the progress made during the year, the reports being published in the *Chemical News* and in American journals.

The following elements have been indexed : ---

Columbium. — Index to the literature of, 1801–1887, by Frank W. Traphagen, Smithsonian Miscellaneous Collections, No. 663, Washington, 1888.

- Iridium. Bibliography of the metal, 1803-1885, by N. W. Perry, in Mineral Resources of the United States, 1883-1884, p. 588; School of Mines Quarterly, 1885, p. 114; Chem. News, 1885, 51, p. 32.
- Manganese. Index to the literature of, 1596-1874, by H. C. Bolton, Annals of the Lyceum of Natural History, New York, Vol. II., Nov., 1875.
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- Vanadium. Index to the literature of, 1801–1876, by G. Jewett Rockwell, Annals of the New York Academy of Sciences, Vol. I., No. 5, 1877.

The general plan of the following index corresponds with that of the others published. The indexes at the end of every volume of each journal were consulted, unless an index covering a series of years was available. The French journals proved to be very troublesome in this respect, as indexes at the end of the volume are often omitted, and the general indexes are seldom detailed enough to be of much value. This was especially true of the *Bull. Soc. Chim.* and the *Ann. Chim. Phys.*

The abbreviations used are those given by H. Carrington Bolton in his "Select Bibliography of Chemistry, 1492-1892," Smithsonian Miscellaneous Collections, No. 840, Washington, 1893.

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1878	Claes	Absorption spectrum.	Ann. der Phys., Pogg. [2], 3, 404.
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1880	E. F. SMITH	Electrolytic estimation.	Ber., 1880, 754.
1880	MARIGNAC		Arch. ph. nat. [3], 3, 413.
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-00	Margaret		Jsb., 1880, 295.
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1880	PERONI and		Gazz. chim. ital., 10, 390.
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1880	Cossa		Gazz. chim. ital., 10, 465.
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1880	Cossa	Tungstate.	Gazz. chim. ital., 10, 467.
			Ber., 1881, 107. J. Chem. Soc., 40 , 225.
1880	Soret	Spectrum.	Jsb., 1880, 293. Compt. rend., 91 , 378.
			Chem. Centibl., 1880, 662. Jsb., 1880, 210.
1880	Soret	Ultra violet	Jsb. rein. Chem., 1880, 74. Arch. ph. nat. [3], 4, 261.
00	6	absorption spectrum.	Jsb., 1880, 214.
1881	CLARKE	At mic weight.	Am. Chem. J., 3, 263. Phil. Mag. [5], 12, 101. Jsb., 1881, 7.
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			J. Chem. Soc., 41, 68. Chem. News, 46, 16.
1882	Clarke	At'mic weight.	Jsb., 1882, 283, 285. Am. Chem. J., 4 , 76.
1882	HARTLEY	Separation from cerium.	Chem. News, 45, 40. Chem. Centrbl., 1882, 151.
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1882	BOISBAUDRAN	Separation from gallium.	Compt. rend., 94 , 1439. Jsb., 1882, 1296.
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			Chem News, 47, 203. Jsb., 1883, 37.
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			Chem. News, 48 , 39, 74. Jsb., 1883, 361.
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1883	STOLBA		Chem. Centrbl., 1883, 313.
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			Ber., 1883, 1860. Jsb., 1883, 354.
1883	J. L. SMITH	Occurrence in samarskite.	Am. Chem. J., 5, 80.
		Estimation by	Chem. News, 48, 13, 29. J. Chem. Soc., 46, 111.
		spectroscope.	Ber., 1883, 1886. Jsb., 1883, 1562.
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SMITHSONIAN MISCELLANEOUS COLLECTIONS.

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INDEX TO THE LITERATURE

ZIRCONIUM.

BY

A. C. LANGMUIR, PH.D., and CHARLES BASKERVILLE, PH.D.



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LETTER OF TRANSMITTAL.

WASHINGTON, MAY 10th, 1899.

The Committee on Indexing Chemical Literature, appointed in 1882 by the American Association for the Advancement of Science, unanimously recommends for publication by the Smithsonian Institution the following :

INDEX TO THE LITERATURE OF ZIRCONIUM,

by A. C. Langmuir, Ph.D., and Charles Baskerville, Ph.D.

H. CARRINGTON BOLTON, Chairman.

MR. S. P. LANGLEY,

Secretary of the Smithsonian Institution.



PREFACE.

An "Index to the Literature of Zirconium" was begun independently by the two authors. Learning of this, and in order to avoid further unnecessary duplication, it was decided to combine the references then on hand and to divide the remaining labor. The resulting product is thought to contain most of the important references to the element, and is offered to the Committee of the American Association for the Advancement of Science on Indexing Chemical Literature in hopes of its proving of value. The references are brought up to January 1st, 1899.

Besides our private libraries, we have sought references to the subject in the libraries of Columbia, Johns Hopkins, and the North Carolina Universities and the North Carolina Geological Survey. To the gentlemen in charge of these libraries we wish to extend our thanks for their courtesy and co-operation.

> A. C. LANGMUIR, C. BASKERVILLE.

APRIL, 1899.



1789-1899

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BY A. C. LANGMUIR AND CHARLES BASKERVILLE.

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- 1857: 67. WARREN. (Double sulphate with potassium.) Ann. der Phys. (Pogg.), 102, 449-53; Jsb., 1857, 158.
- 1857: 68. DEVILLE and TROOST. (Comparison with titanium.) Compt. rend., 45, 821.
- 1857: 69. DEVILLE and TROOST. (Chloride vapor-density.)
 Compt. rend., 45, 821; Ann. chim. phys. [4], 5, 117; Ann. Chem. (Liebig), 105, 217; Phil. Mag. [4], 15, 459; Instit., 1857, 280; Arch. ph. nat. [2], 1, 191; Jsb., 1857, 11, 12.
- 1858: 70. DEVILLE and CARON. (Artificial preparation of Zircon.)
 Compt. [rend., 46, 764; Ann. Chem. (Liebig), 108, 56; Ann. chim. phys. [4], 5, 109; Instit., 1858, 133; Rep. chim. pure, 1, 16; J. prakt. Chem., 74, 157; Jsb., 1858, 2.
- 1859: 71. MALLET. (Metal. Nitride.)
 Am. J. Sci. [2], 28, 349-54; A. A. A. S. Proc., 1859, 217-20; Ann. Chem. (Liebig), 113, 362; J. de Pharm. [3], 37, 233; Rep. chim. pure, 2, 160; Chem. Centrbl., 1860, 94; Jsb., 1859, 145.
- 1859: 72. Ротука. (Opening up Zircon.) Jsb., 1859, 277.
- 1859: 73. DEVILLE. (Vapor density of chloride.)
 Compt. rend., 45, 821; Ann. chim. phys. [3], 58, 281, 282; Ann. der Phys. (Pogg.), 108, 639; Jsb., 1859, 150.
- 1859: 74. STROHMEYER. (Separation from iron.) Ann. Chem. (Liebig), 113, 127; Chem. Centrbl., 1860, 285; Jsb., 1859, 678.
- 1859: 75. G. Rose. (Isomorphism with silica.)
 Ann. der Phys. (Pogg.), 107, 602–4; Chem. News, 1, 131; Jsb., 1859, 151.
- 1860: 76. MARIGNAC. (Fluoride. Researches. Norium.)
 Compt. rend., 50, 952-5; Ann. chim. phys. [3], 60, 257-99; Ann. Chem. (Liebig), 116, 359; Arch. ph. nat. [2], 8, 121-5; Rep. chim. pure, 3, 39; Phil. Mag., [4], 20, 87; J. prakt. Chem., 80, 426; Chem. Centrbl., 1860, 603-5; Jsb., 1860, 134-40.
- 1861: 77. NORDENSKJÖLD. (Crystalline form of oxide.)
 Ann. der Phys. (Pogg.), 114, 625, 626; J. prakt. Chem., 85, 431; Jsb., 1861, 201.

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- 1861: 79. DEVILLE. (Artificial production of Zircon.)
 - Compt. rend., 52, 780; Rep. chim. pure, 3, 219; Instit., 1861, 141;
 J. prakt. Chem., 86, 35; Ann. Chem. (Liebig), 120, 176; Chem. News, 5, 57; Phil. Mag. [4], 21, 496, 497; Chem. Centrbl., 1862, 660; Jsb., 1861, 2, 3.
- 1864: 80. PISANI. (Separation from titanium. Determination.)
 Compt. rend., 57, 298; Ztschr. anal. Chem., 4, 416, 417; Chem.
 News, 10, 91, 218; Bull. Soc. chim. (Paris), 2, 353, 354; Chem.
 Centrbl., 1865, 289-91.
- 1864 : 81. NYLANDER. (Existence of two earths in Zirconia.) Acta Universitatis Lundensis.
- 1865: 82. TROOST. (Metal.)
 - Compt. rend., 61, 109; Instit., 1865, 226; Bull. Soc. chim. (Paris), 5, 212; Arch. ph. nat., 24, 376; Ann. Chem. (Liebig), 136, 349; J. prakt. Chem., 97, 171; Chem. News, 12, 45; Monit. sci. (Quesneville), 7, 752; Ztschr. Chem., 1865, 561; Phil. Mag. [4], 30, 257; Am. J. Sci. [2], 50, 261; Jsb., 1865, 182-4.
- 1865: 83. HERMANN. (Composition of minerals.)

J. prakt. Chem., 95, 123; Chem. Centrbl., 1865, 735.

1865: 84. DEVILLE and CARON. (Artificial production crystallized Zirconia.)

Ann. chim. phys. [4], 5, 109, 117.

1865: 85. Phipson. (Preparation of metal.)

Compt. rend., 61, 745; J. prakt. Chem., 96, 447; Bull. Soc. chim. (Paris), 5, 353; Monit. sci. (Quesneville), 7, 1007; Chem. News, 12, 171; Ztschr. Chem., 9, 30; Chem. Centrbl., 1866, 63; Jsb., 1865, 184.

1865: 86. HERMANN. (Ferrocyanide.)

J. prakt. Chem., 95, 127; Jsb., 1865, 709.

1865: 87. HJORTDAHL. (Sodium salts.)

Compt. rend., 61, 175, 213; Ann. Chem. (Liebig), 137, 34, 236; Instit., 1865, 251; Chem. News, 12, 58, 69; J. de Pharm. [4], 3, 148; Bull. Soc. chim. (Paris), 5, 213; Monit. sci. (Quesneville), 7, 805; Ztschr. Chem., 1865, 619; Jsb., 1865, 184–6.

- 1866: 88. HERMANN. (Preparation from Zircon.) J. prakt. Chem., 97, 330; Ztschr. Chem., 1866, 717; Jsb., 1866, 189.
- 1866: 89. HERMANN. (Identity of zirconium and norium.)
 - J. prakt. Chem., 97, 321; Bull. Soc. chim. (Paris), 6, 383–5; Chem. News, 14, 33; Ztschr. Chem., 1866, 443; Jsb., 1866, 191.

- 1866: 90. HERMANN. (Separation from earths.)
 J. prakt. Chem., 97, 337; Ztschr. anal. Chem., 5, 381-4; Bull. Soc. chim. (Paris), 6, 385-7; Chem. News, 14, 33; Jsb., 1866, 797-8.
- 1868 : 91. WEBSKY. (Occurrence in kochelite.) Jahrb. Miner., 1868, 607 ; Jsb., 1868, 1013.
- 1868: 92. FIZEAN. (Expansion of Zircon by heat.) Compt. rend., 66, 1005; Ann. der Phys. (Pogg.), 135, 380; Phil. Mag. [4], 36, 31; Jsb., 1868, 52.

1868: 93. CARON. (Zirconia light.)

Compt. rend., 66, 850, 1040; Ann. chim. phys. [4], 14, 311-5; Monit.
 sci. (Quesneville), 1868, 496, 899; Ztschr. Chem., 11, 536; Chem.
 News, 17, 276; Wagner's Jsb., 14, 754, 756; Jsb., 1868, 979.

- 1869: 94. STREIT and FRANZ. (Separation from titanium.)
 J. prakt. Chem., 108, 75; Ztschr. anal. Chem., 9, 388-90; Ztschr. Chem., 13, 256; Chem. Centrbl., 1870, 98, 319; Jsb., 1869, 915.
- 1869: 95. TESSIÉ DU MOTAY. (Zirconia light.) Wagner's Jsb., 15, 730; Chem. News, 19, 107, 213, 310; Dingl. Polyt. J., 191, 252.
- 1869: 96. DARKER. (Zirconia light.)
 Chem. News, 19, 499; Dingl. Polyt. J., 194, 519; Wagner's Jsb., 15, 730.
- 1869: 97. HARRISON. (Zirconia light.) Mechan. Mag., 1869, 458; Polyt. Centr., 1869, 1461; Wagner's Jsb., 15, 731.
- 1869: 98. PAYEN. (Zirconia light.)
 Genie industriel, 1869, 161; Polyt. Centr., 1869, 966; Dingl. Polyt.
 J., 193, 433; Wagner's Jsb., 15, 752.

1869 : 99. Vogt. (Zirconia light.) Köln. Zeit., 1869, No. 159 ; Wagner's Jsb., 15, 750-1.

1869 : 100. ——. (Zirconia light.)

Pharm. J. [2], 11, 81-2.

- 1869 : 101. Тнаце́м. (Spectrum.) Ann. chim. phys. [4], 18, 228.
- 1869: 102. Sorby. (A new element, jargonium, in Zircon.)

Lond. Roy. Soc. Proc., 17, 511; Ann. der Phys. (Pogg.), 138, 58-65;
Chem. News, 19, 121-3, 142, 181; Ber., 2, 126, 193, 337, 382; Bull.
Soc. chim. (Paris), 12, 36; Ztschr. Chem., 1869, 221, 403; Ann.
chim. phys. [4], 18, 487; Am. J. Sci. [2], 48, 405; Wagner's Jsb., 15, 1; Jsb., 1869, 261.

1869: 103. Forbes. (Jargonia.)

Intellectual Observer, 9, 291; Chem. News, 19, 277.

- 1869: 104. SORBY. (Spectrum of mixtures of Zirconium and uranium.)
 Ber., 3, 146; Bull. Soc. chim. (Paris), 14, 40; Chem. Centrbl., 1870, 369; Lond. Roy. Soc. Proc., 18, 197; Phil. Mag. [4], 39, 65.
- 1869 : 105. Сниксн. (Idem.) Chem. News, **20**, 9.
- 1870: 106. MELLISS. (Metal. Salts.)

Bull. Soc. chim. (Paris), 14, 204; Ber., 4, 57; Chem. News, 22, 23;
 Ann. Chem. (Liebig), 153, 238; Ztschr. Chem., 1870, 296; Jsb., 1870, 328; Chem. Centrbl., 1870, 392.

- 1870: 107. HUGGINS. (Spectrum.) Lond. Roy. Soc. Proc., 18, 548.
- 1870: 108. FRANZ. (Metal. Preparation of pure salts.)
 - Ber., 3, 58-60; Bull. Soc. chim. (Paris), 13, 507; Ztschr. Chem., 1870, 468; Chem. Centrbl., 1870, 131; Wagner's Jsb., 16, 1, 2; Jsb., 1870, 329.
- 1870: 109. WUNDER. (Isomorphism with titanic and stannic oxides.)
 J. prakt. Chem. [2], 2, 206-12; Chem. News, 22, 215; Ztschr. Chem., 1870, 286; Jsb. Min., 1870, 1000; Gazz. chim. ital., 1, 527; Chem. Centrol., 1870, 663.
- 1871: 110. MENDELÉJEFF. (Fluoride.)

Ber., 4, 933.

- 1871 : 111. RATH. (Occurrence in diorite.) Ann. der Phys. (Pogg.), 144, 250.
- 1871: 112. KNOP. (Artificial production of crystals.)
 - Ann. Chem. (Liebig), 157, 363; 159, 36; Bull. Soc. chim. (Paris), 15, 190; Ztschr. Chem., 1871, 397; J. Chem. Soc. (Lond.), 24, 805; Jsb., 1871, 322.
- 1871: 113. TROOST and HAUTEFEUILLE. (Oxychloride.)
 - J. prakt. Chem. [2], 4, 298; Compt. rend., 73, 563; Instit., 1871, 49;
 Bull. Soc. chim. (Paris), 16, 240; J. Chem. Soc. (Lond.), 24, 1000;
 Gazz. chim. ital., 1, 628; Jsb., 1871, 293.
- 1871: 114. TROOST and HAUTEFEUILLE. (Spectrum.)
 - Compt. rend., 73. 620; Instit., 1871, 77; Arch. ph. nat., 42, 178;
 Bull. Soc. chim. (Paris), 16, 229; Ztschr. Chem., 1871, 465; J.
 Chem. Soc. (Lond.), 24, 1147; Jsb., 1871, 169.
- 1871: 115. RAMMELSBERG. (Separation from niobic and tantalic acids.) J. Chem. Soc. (Lond.), 25, 195; Ber., 4, 875.

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- 1872: 116. RAMMELSBERG. (Action of heat on amorphous oxide.) Ber., 5, 1006.
- 1872 : 117. RAMMELSBERG. (Sulphate.) Ber., 5, 1005.
- 1872: 118. TROOST and HAUTEFEUILLE. (Action of silicon chloride on oxide.)

Compt. rend., 75, 1819; Ber., 6, 34; Gazz. chim. ital., 2, 27; Jsb., 1872, 226.

- 1872 : 119. NORDENSKJÖLD. (Occurrence in nohlite.) Bull. Soc. chim. (Paris), 18, 178.
- 1873: 120. MALLARD. (Behavior of oxide with sodium carbonate.) Compt. rend., 75, 472; Gazz. chim. ital., 3, 84.
- 1873: 121. HANNAY. (Zirconia.)

J. Chem. Soc. (Lond.), 26, 703–10; Chem. News, 27, 232; Ber., 6, 571; Am. Chemist, 4, 194; Jsb. rein. Chem., 1, 73; Gazz. chim. ital., 3, 468.

1873: 122. MIXTER and DANA. (Specific heat.)

Ann. Chem. (Liebig), 169, 388; Chem. Centrbl., 1873, 721; Bull. Soc. chim. (Paris), 21, 68; Jsb. rein. Chem., 1, 73; J. Chem. Soc. (Lond.), 27, 118; Am. J. Sci. [3], 7, 506-7; Gazz. chim. ital., 3, 577; 5, 107; Jsb., 1873, 58.

1873: 123. PAYKULL. (Compounds: Hydroxide, oxychloride, double chlorides, sulphate, arsenate, and ortho-phosphate.)

Oefvers. Sv. Vetensk. Akad. Förh., 1873, 22; Ber., 1873, 1467; Jsb. rein. Chem., 1, 73; Bull. Soc. chim. (Paris), 20, 65; Chem. News, 28, 45; J. Chem. Soc. (Lond.), 26, 1105; Chem. Centrbl., 1873, 594; Gazz. chim. ital., 3, 484; Am. Chemist, 4, 393; Jsb., 1873, 263.

- 1873: 124. JANNETTAZ. (Propagation of heat by Zirconia.) Ann. chim. phys. [4], 29, 33.
- 1874: 125. TROOST and HAUTEFEUILLE. (Spectrum.) Compt. rend., 73, 620; Ztschr. anal. Chem., 13, 313-4.
- 1875: 126. ENDEMANN. (Soluble basic salts.)

Am. Chemist, 5, 326-7; Bull. Soc. chim. (Paris), 24, 466; J. prakt. Chem. [2], 11, 219; Jsb. rein. Chem., 3, 84; J. Chem. Soc. (Lond.), 28, 1162; Gazz. chim. ital., 6, 428; Chem. Centrbl., 1875, 339; Jsb., 1875, 219.

1875: 127. Nilson. (Selenites.)

"Researches on Salts of Selenious Acid," Upsala, 1875.

1876 : 128. NILSON. (Platinate.) Ber., 9, 1143 ; Bull. Soc. chim., 27, 209 ; Jsb., 1876, 294.

- 1876: 129. HORNBERGER. (Comparison with silicon. Preparation pure chloride. Alkaline tartrates. Cyanogen compounds.)
 - Am. Chem. (Liebig), 181, 232; Bull. Soc. chim. (Paris), 26, 493; Jsb. rein. Chem., 4, 90; J. de Pharm. [4], 25, 154; J. Chem. Soc. (Lond.), 30, 275; Chem. News, 33, 261; Gazz. chim. ital., 7, 166, 232; Chem. Centrbl., 1876, 435; Jsb., 1876, 240.
- 1877 : 130. Рнплер. (Technology.) Jsb., 1877, 1121.
- 1877 : 131. DRAPER. (Zirconia cylinders for oxyhydrogen light.) Am. J. Sci. [3], 14, 208 ; Chem. Centrol., 1877, 673.
- 1877 : 132. VINCENT. (Behavior towards trimethylaniline.). Bull. Soc. chim. (Paris), 27, 194; Chem. Centrol., 1878, 263.
- 1878 : 133. Рнплер. (Zirconia light.) Monit. sci. (Quesneville), 20, 481.
- 1878: 134. MALLET. (Occurrence in sipylite.) Am. J. Sci. [3], 14, 397; Chem. Centrol., 1878, 7.
- 1879: 135. FRIEDEL and CRAFTS. (Action of chloride on organic compounds.)

Bull. Soc. chim. (Paris), 31, 531; Ber., 12, 373.

- 1879: 136. PAYKULL. (Compounds.) Ber. 12, 1719.
- 1880 : 137. Ängström. (Oxide, diamagnetic.) Jsb., 1880, 338 ; Dammer II., 1, 619.
- 1880: 138. SORET. (Spectrum.) Arch. ph. nat. [3], 4, 261; Jsb., 1880, 214.
- 1880 : 139. NILSON and PETTERSSON. (Specific heat.) Compt. rend., 91, 232 ; Ber., 13, 1461 ; Jsb., 1880, 237.
- 1880: 140. VINCENT. (Behavior towards dimethylaniline.) Bull. Soc. chim., Paris, 33, 156-8; Ztschr. anal. Chem., 19, 479; Chem. Centrbl., 1880, 279.
- 1881: 141. CLARKE. (Atomic weight.)

Phil. Mag. [5], 12, 101; Am. Chem. J., 3, 263; Jsb., 1881, 7.

- 1881 : 142. CROOKES. (Phosphorescent spectrum.)
 Bakerian Lecture, May 31, 1883, Lond. Roy. Soc. Proc., 32, 206;
 Ann. chim. phys. [5], 23, 555; Compt. rend., 92, 1281; Chem.
 News, 43, 237; Jsb., 1881, 131.
- 1882 : 143. WELLER. (Action of hydrogen peroxide on.) Ber., 14, 2592 ; Jsb., 1882, 1292.

- 1882 : 144. CLASSEN. (Electrolytic separation.) Ber., 14, 2783 ; Zeit. anal. Chem., 22, 421 ; Chem. Centrol., 1882, 233.
- 1882 : 145. LEVY and BOURGEOIS. (Microchemical reaction.) Compt. rend., 94, 812 ; Chem. News, 45, 240 ; Jsb., 1882, 1527.
- 1882 : 146. BOISBAUDRAN. (Separation from gallium.) Compt. rend., 94, 1154 ; Chem. News, 45, 207 ; Jsb., 1882, 1296.
- 1883: 147. DONATH and MAYRHOFER. (Affinity.) Ber., 16, 1588; Jsb., 1883, 26.
- 1884 : 148. CLARKE. (Atomic weight.) Chem. Ztg., 8, 930.
- 1884: 149. STOLBA. (Opening up Zircons.) Chem. News, 49, 174; J. Chem. Soc., 46, 821; Jsb., 1884, 1594.
- 1885 : 150. GROSHAUS. (Density.) Rev. Trav. chim. pays bas., 4, 236 ; Jsb., 1885, 53.
- 1885 : 151. Скоокез. (Spectrum when samarium is present.) Compt. rend., 100, 1380; Lond. Roy. Soc. Proc., 38, 414; Chem. News, 51, 301; Jsb., 1885, 332.
- 1885 : 152. DEMARÇAY. (Separation from titanium.) Compt. rend., 100, 740-742 ; Rep. anal. Chem., 1885, 186 ; Chem. Centrbl., 56, 283 ; Jsb., 1885, 1929.
- 1885 : 153. CLÉVE. (Peroxide.) Bull. Soc. chim. (Paris), 43, 57 ; Ztschr. anal. Chem., 28, 699 ; Jsb., 1885, 492.
- 1885: 154. LINNEMANN. (Qualitative composition. Opening up Zircons.)

Monatsh. Chem., 6, 335-47; Chem. Ztg., 9, 1244; Ber., 18, c, 459, 460; J. Chem. Soc., 48, 1042; Chem. News, 52, 233, 240; Chem. Centrbl., 56, 666, 667.

1885: 155. LINNEMANN. (Absorption spectra of Zircons.)

Monatsh. Chem., 6, 531, 536; Ber., 18, c, 605; J. Chem. Soc., 48, 1173; Chem. News, 52, 220; Chem. Centrbl., 56, 907; Jsb., 1885, 2271.

1886: 156. BAILEY. (Separation and estimation by hydrogen peroxide.)

J. Chem. Soc., 49, 149–152; Chem. News, 53, 55, 260; Ann. Chem.
 (Liebig), 232, 352; Ber., 19, c, 319; Chem. Ztg., 10, 1, 148, 677;
 Pharm. J. [3], 16, 1022; Chem. Centrbl., 57, 172, 451.

- 1886: 157. BAILEY. (Separation and estimation by hydrogen peroxide.)
 - J. Chem. Soc. (Lond.), 49, 481–6; Chem. News, 53, 160; Am. J. Sci., 26, 470; Ztschr. anal. Chem., 28, 699; Ann. Chem. (Liebig), 232, 352; Ber., 19, c, 881; Chem. Centrbl., 57, 682; Jsb., 1886, 1942.
- 1886: 158. LINNEMANN. (Zirconia light.)

Monatsh. Chem., 6, 899–908; J. Chem. Soc. (Lond.), 50, 417; Chem. Centrbl., 57, 263, 264; Wagner's Jsb., 32, 381–4; Jsb., 1885, 2167.

1886: 159. HAUTEFEUILLE and MARGOTTET. (Phosphate.)

Compt. rend., 102, 1017–1019; Ber., 19, c, 387; J. Chem. Soc. (Lond.), 50, 670; Chem. News, 53, 252; Chem. Centrbl., 57, 468; Jsb., 1886, 447.

1886: 160. VAN DER PLAATS. (Atomic weight.)

Ann. chim. phys. [6], 7, 501; Zeitschr. anal. Chem., 26, 276.

- 1886 : 161. TROOST and OUVRARD. (Double potassium phosphate.) Compt. rend., 102, 1422-7 ; Ber., 19, c, 659 ; J. Chem. Soc., 50, 853 ; Chem. Centrol., 57, 594 ; Jsb., 1886, 453, 454.
- 1887: 162. MEVER and WILKINS. (Action of carbon tetrachloride on oxide.)

Ber., 20, 683; Jsb., 1887, 379.

- 1887 : 163. DITTE. (Behavior towards sulphuric acid. Selenate.)
 Compt. rend., 104, 172 ; Jsb., 1887 [1], 547, 549 ; Dammer. II, 1, 622.
- 1887 : 164. RAMMELSBERG. (Separation from thorium, cerium, etc.) Sitzber. Akad. Wissin., Berlin, 1886, 441 ; Ber., 20, c, 413.
- 1887 : 165. PICCINI. (Action of hydrogen peroxide.) Gazz. chim. ital., 17, 486 ; Jsb., 1887, 551.
- 1887 : 166. DEMARÇAY. (Action of carbon tetrachloride on Zirconia.) Compt. rend., 104, 113 ; Ber., 20, с. 96 ; Chem. Centrbl., 58, 214 ; Jsb., 1887, 380.
- 1887: 167. WILLGERODT. (Action as a chloridizing agent.)

J. prakt. Chem. [2], 35, 391; Ber., 20, c, 312; Chem. Centrbl., 58, 720; Jsb., 1887, 618.

1887: 168. WEIBULL. (Crystalline forms of Zirconyl chloride and bromide; also sulphate.)

Ber., 20, a, 1394–6; J. Chem. Soc. (Lond.), 52, 778; Chem. Centrol., 58, 778; Jsb., 1887, 553. 1887: 169. TROOST and OUVRARD. (Double sodium phosphate. Comparison with thorium.)

Compt. rend., 105, 30-4; Ber., 20, c, 534; J. Chem. Soc. (Lond.), 52, 1017; Chem. News, 56, 57; Chem. Centrbl., 58, 1015; Jsb., 1887, 554-6.

- 1887: 170. HINSBERG. (Attempt to prepare Zirconium ethyl. Iodide.)
 Ann. Chem. (Liebig), 239, 253-6; Ber., 20, c, 413; J. Chem. Soc. (Lond.), 52, 896; Chem. News, 56, 219; Chem. Centrbl., 58, 1016; Jsb., 1887, 553.
- 1887: 171. TROOST and OUVRARD. (Zircon not isomorphous with thorium silicate.)

Compt. rend., 105, 255; Chem. Centrbl., 58, 1098; Jsb., 1887, 556; Ber., 20, c, 534.

- 1887: 172. WELSBACH. (Welsbach burners.)
 Ber., 20, c, 406; Chem. News, 55, 192; Chem. Centrbl., 1887, 1125; Jsb., 1887, 2670; German Patent, 39, 162.
- 1888 : 173. HAUTEFEUILLE and PERREY. (Artificial preparation of Zircon.)

Compt. rend., 107, 1000, 1001; Ber., 22, c, 94; J. Chem. Soc. (Lond.),
56, 355; Chem. News, 59, 11; Monit. sci. (Quesneville), 33, 199;
Chem. Centrbl., 1889 [1], 127; Jsb., 1888, 638.

1888 : 174. BLÖMSTRAND. (Constitution of silicates containing Zirconium.)

Ztschr. Kryst.; 15, 83, 84; Chem. Centrbl., 1889 [1], 821; Jsb., 1888, 637.

- 1888 : 175. KEEPORT. (Application in gold metallurgy.) Ber., 21, c, 458 ; Wagner's Jsb., 34, 369 ; Jsb., 1888, 2650 ; German Patent, 43, 231.
- 1888 : 176. CARNELLEY and WALKER. (Action of heat on hydrate.) J. Chem. Soc. (Lond.), 53, 68, 82 ; Ber., 21, 131.
- 1888 : 177. SCHMIDT and HAENSCH. (Emissive power of Linnemann's light.)

Ann. der Phys. (Pogg.), Berbl., 12, 244; Jsb., 1888, 2838.

- 1889 : 178. RIÖRDAN. (Preparation from eudialyte.) Chem. Centrbl., 1889, 533.
- 1889 : 179. WELSBACH. (Preparation of pure nitrate.) Chem. Ztg., 13 [2], 1192 ; American Patent, 409, 653.
- 1889 : 180. STOLBA. (Opening up Zircon.) Listy chemické, 13, 117, 118 ; Chem. Centrbl., 1889, 1, 297.

1889: 181. DAVIS. (Separation from aluminium.)

- Amer. Chem. J., 11, 26-9; Ztschr. anal. Chem., 29, 454, 455; Ber.,
 22, c, 300; J. Chem. Soc. (Lond.), 56, 551; Chem. News, 59, 100,
 101; Chem. Centrbl., 60, 1, 454; Jsb., 1889, 2388.
- 1889: 182. DAY. (Production in United States.)

Mineral Resources, U. S. 6th report; J. Soc. Chem., Ind., 8, 591.

- 1889: 183. BAILEY. (Atomic weight. Oxychloride. Peroxide.)
 - Lond. Roy. Soc. Proc., 46, 74–87; Chem. News, 60, 6–8, 17, 18, 32;
 J. Chem. Soc. (Lond.), 58, 705; Ztschr. anal. Chem., 29, 743–7;
 Nature, 36, 568; Ber., 22, c, 655, 666; Ztschr. physikal Chem., 4, 494; Chem. Centrbl., 1889 [2], 311, 312; Jsb., 1889, 113–6; Brit. Assn. Trans., 1887, 636.
- 1890: 184. HAUTEFEUILLE and PERREY. (Action of hydrochloric acid gas on oxide.)

Compt. rend., 110, 1038; Ber., 23, c, 428; J. Chem. Soc. (Lond.), 58 1071.

- 1890 : 185. WARREN. (Precipitation by magnesium.) Chem. News, 61, 183 ; Ber., 23, c, 560 ; Jsb., 1890, 42.
- 1890: 186. HIRSCHWALD. (Solubility in microcosmic salt bead.) J. prakt. Chem. [2], 41, 360; Jsb., 1890, 2421.
- 1890: 187. WINKLER. (Reduction of oxide by magnesium.) Ber., 23, b, 2664-8; J. Chem. Soc. (Lond.), 58, 1375: Chem. Centrol., 1890 [2], 644, 645; Jsb., 1890, 432.
- 1890: 188. Імкау. (Preparation from ores.) J. Soc. Chem. Ind., 9, 941; English Patent, 16, 555.
- 1890: 189. Kochs. (Zirconia light.)

Dingl. Polyt. J., 278, 235–40; J. Soc. Chem. Ind., 10, 37; Wagner's Jsb., 37, 62; Eng. and Mining J., 51, 466; Jsb., 1890, 2850.

1891: 190. VENABLE. (Preparation of pure chloride.)

J. anal. Chem., 5, 551; J. El. Mitchell Sc. Soc., 8, 20; Chem. Ztg., 15, 328; J. Chem. Soc. (Lond.), 62, 412; Chem. News, 64, 315, 316; Chem. Centrbl., 1891 [1], 149; Jsb., 1891, 575.

- 1891 : 191. DROSSBACH. (Zirconia pencils.) Chem. Ztg., 15 [1], 328 ; Chem. Centrbl., 1891 [1], 772, 773.
- 1891: 192. BEHRENS. (Microchemical reaction.) Ztschr. anal. Chem., 30, 156; Chem. News, 64, 124.
- 1891: 193. MOREHEAD. (Analysis of Zircon.) J. El. Mitchell Sc. Soc., 8, 24.

- 1891: 194. WINKLER. (Action of magnesium on oxide. Hydride.)
 Ber., 24, a, 888; J. Chem. Soc. (Lond.), 60, 802; Bull. Soc. chim.
 (Paris), [3], 6, 173; Chem. Centrol., 1891 [1], 912; Jsb., 1891, 499.
- 1891 : 195. VENABLE. (Occurrence.) J. El. Mitchell Sc. Soc., 8, 74.
- 1891 : 196. OUVRARD. (Alkaline Zirconates.)
 Compt. rend., 112, 1444-6 ; Ber., 24, c, 694 ; J. Chem. Soc. (Lond.),
 60, 1431 ; Monit. sci. (Quesneville), 37, 868 ; Chem. News, 64, 26 ;
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INDEXES

TO THE

LITERATURES OF CERIUM AND LANTHANUM

BY

W. H. MAGEE, Ph. D.



CITY OF WASHINGTON PUBLISHED BY THE SMITHSONIAN INSTITUTION



LETTER OF TRANSMITTAL.

NEW YORK, JULY 24, 1894.

The Committee of the American Association for the Advancement of Science having charge of Indexing Chemical Literature has voted to recommend to the Smithsonian Institution for publication the three following Indexes: —

AN INDEX TO THE LITERATURE OF CERIUM.

AN INDEX TO THE LITERATURE OF LANTHANUM. Both by W. H. Magee, Ph. D.

AN INDEX TO THE LITERATURE OF DIDYMIUM.¹ By A. C. Langmuir, Ph. D.

The latter has already appeared in the School of Mines Quarterly, No. 1, Vol. XV.

H. CARRINGTON BOLTON, Chairman.

To the SECRETARY of the SMITHSONIAN INSTITUTION.

¹ This Index is printed as Smithsonian Publication No. 972.



INDEXES TO THE LITERATURES OF CERIUM AND LANTHANUM.

By W. H. MAGEE, PH. D.

INTRODUCTION.

THE following indexes to the literatures of cerium and lanthanum were prepared during the course of some work on the former element. They are not offered as being absolutely correct, but all the more important articles bearing upon the elements are certainly indexed, and usually the original article heads the list. In some few cases, however, it was difficult to determine the original. Whenever the journal was to be found on the library shelves the references were verified. No single library, however, contains all the journals to which references will be found.

That the indexing of chemical literature is of great and growing importance is evident; that the work should be as nearly perfect as possible is equally true. Yet few except those who have attempted the task realize the difficulty and labor involved. I would ask, therefore, as regards these indexes, that any one using them, and all chemists interested in the study of cerium and lanthanum, should send corrections and addenda to W. H. Magee, care of Professor L. M. Dennis, Cornell University, Ithaca, N. Y., so that after a few years perfectly correct indexes may be prepared.

The Indexes are arranged on the same plan as that of the Index to Uranium, published by Dr. H. Carrington Bolton in 1870, and followed by several other chemists. The abbreviations used are in the main those of the standard list printed in Bolton's Bibliography of Chemistry.

CORNELL UNIVERSITY, ITHACA, N. Y., July 21, 1894.



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SMITHSONIAN MISCELLANEOUS COLLECTIONS.

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A BIBLIOGRAPHY

OF THE

ANALYTICAL CHEMISTRY

OF

M A N G A N E S E. 1785-1900.

BY

HENRY P. TALBOT AND JOHN W. BROWN.



CITY OF WASHINGTON: PUBLISHED BY THE SMITHSONIAN INSTITUTION.

1902.

The Anickerbocker press. Rew york

LETTER OF TRANSMITTAL.

WASHINGTON, May 11th, 1901.

The Committee on Indexing Chemical Literature, appointed in 1882 by the American Association for the Advancement of Science, has voted to recommend to the Smithsonian Institution for publication the following:

"A Bibliography of the Analytical Chemistry of Manganese, 1785-1900," by Henry P. Talbot and John W. Brown.

This forms one of the following series:

Index to the Literature of Uranium, 1785–1885, by Henry Carrington Bolton, 1885.

Index to the Literature of Columbium, 1801–1887, by Frank W. Traphagen, 1888.

- Index to the Literature of the Spectroscope, by Alfred Tuckerman, 1888.
- Index to the Literature of Thermodynamics, by Alfred Tuckerman, 1890.
- A Bibliography of the Chemical Influence of Light, by Alfred Tuckerman, 1891.

A Bibliography of Aceto-Acetic Ester, by Paul H. Seymour, 1894. Index to the Literature of Didymium, 1842-1893, by A. C. Langmuir, 1895.

Indexes to the Literature of Cerium and Lanthanum, by W. H. Magee, 1895.

- A Bibliography of the Metals of the Platinum Group, by Jas. Lewis Howe, 1897.
- Review and Bibliography of the Metallic Carbides, by J. A. Mathews, 1898.

Index to the Literature of Thallium, 1861–1897, by Miss Martha Doan, 1898.

Index to the Literature of Zirconium, by A. C. Langmuir and Charles Baskerville, 1899.

> HENRY CARRINGTON BOLTON, Chairman.

MR. S. P. LANGLEY, Secretary Smithsonian Institution.



PREFACE.

In 1875 Dr. H. Carrington Bolton published an "Index to the Literature of Manganese, 1596-1874," comprising the references to the entire available journal literature relating to manganese and its compounds which had appeared before the latter date. The general plan of his index did not, however, provide for the separate classification of the articles containing analytical data, and as the journal literature since 1874 has increased so enormously in volume as to render the search for such data very laborious, it was deemed worth while to compile a separate bibliography bearing upon the qualitative detection and quantitative separation and determination of manganese, for the use of analytical chemists. For this purpose Dr. Bolton's work has been reviewed almost from its beginning, and while we are in part indebted to his "Index" for the references from 1785 to about 1830 (having received his permission to make use of them), yet a number of these early journals were independently examined by us as indicated in our list. From 1830 the compilation is practically independent, although we desire to express our obligation to the Bolton "Index" for the opportunity afforded for the verification of a portion of our work. All the originals of the references have been examined except those marked with an asterisk.

The compilation of material for this Bibliography was practically completed when the "Bibliography of Steel-Works Analysis," published by Brearley in the *Chemical News*, 1899, came to our attention. We are indebted to this Bibliography for an opportunity to verify a portion of our later references and for some four or five references which we had omitted.

The abbreviations used in this Bibliography are those recommended by the Committee on Indexing Chemical Literature of the American Association for the Advancement of Science. The original reference is placed first, and, in general, articles corresponding to references which follow those to the *Chemisches Centralblatt* or the *Jahresbericht der Chemie* are decidedly briefer than the original papers. It has not been practicable, however, to rigidly maintain this distinction.

PREFACE

The Subject Index has been based upon such an examination of the original articles as was possible with a reasonable expenditure of time, and is based therefore upon the salient points rather than upon minute details, although an attempt has been made to carry the subdivision of subjects as far as possible. Under the heading "Applications of Quantitative Methods" (page 111 *et seq.*), only those references are, in general, included in which the title of the article specifies the material analyzed. This is particularly true of irons, steels, ferromanganese, and spiegeleisen.

We desire to express our obligation to Dr. H. C. Bolton for the permission to make use of his "Index," as well as for assistance in the examination of a file of journals, and to Mr. A. C. Davis for his valuable assistance in the examination of proof-sheets.

This compilation was made possible by a ready access to the valuable libraries of the Massachusetts Institute of Technology, notably the William Ripley Nichols Chemical Library, but we also wish to express our appreciation of the courtesies extended to us by the Boston Public Library, the Library of the American Academy of Arts and Sciences, the Boston Society of Natural History, the Surgeon-General's Office at Washington, the Library of Congress, the libraries of Yale, Columbia, Lehigh, and Harvard Universities, the Massachusetts College of Pharmacy, and the John Crerar and Astor Libraries. Professor James Lewis Howe's excellent "Bibliography of the Platinum Metals" has served as a model for the arrangement of our data.

> Henry P. Talbot. John W. Brown.

Massachusetts Institute of Technology, Boston

JANUARY, 1902.

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Mém. de l'Inst., 10, 286; Ann. chim. phys. (2), 43, 380. Separation from iron and zinc by electrolysis.

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J. für Chem. (Schweigger), 60, 346.

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J. für Chem. (Schweigger), 62, 192; Pharm. Centrbl., 1831, 461; Berzelius' Jsb. (1833), 12, 164.

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Ann. chim. phys. (2), 60, 252; Pol. Centrbl., 1836, 286; Ann. Chem. (Liebig), 18, 47; Arch. Pharm., 58, 128.

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Ann. Chem. (Liebig), 107, 101.

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- 1858: 4. SPILLER, J. On Some Remarkable Circumstances Tending to Disguise the Presence of Various Acids and Bases in Chemical Analysis.

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Chem. News, 1, 4.

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C. R., 50, 610.

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J. prakt. Chem., 82, 257; Chem. Centrbl., 1861, 525; Chem. News, 4, 150; Rep. chim. pure, 3, 66.
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Ann. Chem. (Liebig), 119, 129; Dingl. pol. J., 161, 373.

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 Ann. Chem. (Liebig), 120, 243; Ztschr. anal. Chem., 1, 217.
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Chem. News, 6, 123; Rep. chim. appl., 5, 26.

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J. Chem. Soc. (Lond.), 3, 206.

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Ztschr. anal. Chem., 7, 340; Jsb. Chem., 1868, 227; Ztschr. Chem., 1869, 306.

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Ztschr. anal. Chem., 8, 370; Chem. Centrbl., 1870, 530; Jsb. Chem., 1869, 887; Ztschr. Chem., 1870, 285; Bull. soc. chim. (2), 14, 44; Chem.-techn. Rep., 9, b, 122.

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C. R., 69, 168; Jsb. Chem., 1869, 891; L'Inst., 1869, 243.

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Ztschr. anal. Chem., 8, 137.

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Am. J. Sci. (2), 44, 213; Chem. Centrbl., 1870, 62.

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Am. J. Sci. (2), 50, 240; Chem. News, 22, 210; Chem. Centrbl., 1870, 706.

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Monit. scientif., 12, 277.

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> Chem. News, 23, 290; Jsb. Chem., 1871, 930; Bull. soc. chim. (2), 16, 93; J. Chem. Soc. (Lond.), 24, 757; Ztschr. anal. Chem., 11, 79; Ztschr. Chem., 1871, 413.

Detection by means of potassium ferricyanide.

1871: 2. CHATARD, T. M. Contribution to Chemistry from the Laboratory of the Lawrence Scientific School.

> Am. J. Sci. (3), 1, 419; Jsb. Chem., 1871, 928; Chem. Centrbl., 1871, 426; Chem. News, 24, 196; Ztschr. anal. Chem., 11, 308; J. Chem. Soc. (Lond.), 26, 531.

> Quantitative application of Crum's test. Ammonium oxalate employed to titrate for the permanganic acid. See 1845 : 2.

1871: 3. KÄMMERER, H. Ueber die Anwendung des Broms statt des Chlors zu analytischen Zwecken.

Ber., 4, 218; Jsb. Chem., 1871, 866; Ztschr. Chem., 1871, 444;
Ztschr. anal. Chem., 10, 464.
Recommends use of bromine for manganese precipitations.

1871: 4. LUCK, E. Beiträge zur Braunsteinanalyse nach den Fresenius- Will'schen Verfahren.

> Ztschr. anal. Chem., 10, 310; Jsb. Chem., 1871, 929; J. Chem. Soc. (Lond.), 25, 264; Dingl. pol. J., 202, 305; Pol. Centrbl., 1871, 1568.

Determination of the accuracy of the method.

1871: 5. ROWAN, T. On the Estimation of Manganese in Spiegeleisen and Ferro-manganese.

> Chem. News, 23, 279; J. Chem. Soc. (Lond.), 24, 756; Am. Chemist, 2, 75. Precipitation of manganese as carbonate.

1871: 6. TAMM, H. On a New Method of Estimating Zinc.

Chem. News, 24, 150; Jsb. Chem., 1871, 932; Ztschr. Chem., 14, 467; Bull. soc. chim. (2), 16, 261.
Determination of manganese as phosphate.

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Chem. News, **26**, 81. Comments on Tamm's article. See 1871: 6.

- 1872: 2. BÖTTGER, R. Nachweisung von Spuren von Mangan.
 - *Jsb. phys. Ver. Frankfurt, 10, 388; Vierteljsb. prakt. Pharm., 21, 418; Jsb. Chem., 1872, 911; Ztschr. anal. Chem., 1872, 433; Chem. News, 24, 192; Chem.-techn. Rep., 10, b, 150; J. Frank. Inst., 93, 87.

Detection of small amounts of manganese by contact of substance with fused potassium chlorate.

1872: 3. FRESENIUS, R. Ueber die Bestimmung des Mangans auf gewichtsanalytischem Wege.

Ztschr. anal. Chem., 11, 290 and 413; Jsb. Chem., 1872, 908; Am. Chemist, 3, 472; J. Chem. Soc. (Lond.), 26, 409.

Determination as protosesquioxide after precipitation as manganous carbonate, hydroxide, hydrated peroxide (Guyard), or oxalate; as pyrophosphate (Gibbs) or as sulphide (Fresenius, Classen).

1872: 4. HORNER. The Spectra of Manganese in Blowpipe Beads.

Chem. News, 25, 139; J. Chem. Soc. (Lond.), 25, 524. Detection by means of absorption spectra of blowpipe beads. 1872: 5. KESSLER, F. Beiträge zur Analyse des Roheisens und Stahls.

Ztschr. anal. Chem., 11, 255; Monit. scientif., 15, 826; Chem. News, 28, 158.

Separation from iron by acetate and sulphate methods. Precipitation of manganese by bromine, solution in antimonious chloride, and titration with potassium permanganate.

1872: 6. KESSLER, F. Ueber die Bestimmung des Mangans in Roheisen, Stahl, und Stabeisen.

Ber., 5, 60; Oester. Ztschr. Berg-u. Hüttenw., 20, 405; Dingl. pol. J., 205, 332 and 439; Pol. Centrbl., 1872, 1608; Technologiste, Dec. 1872; Chem. Centrbl., 1872, 617; Chem. News, 27, 14; J. Chem. Soc. (Lond.), 25, 925; Chem.-techn. Rep., 11, b, 199; Wagner's Jsb., 19, 11; Am. Chemist, 4, 76; Iron, 2, 326.
See 1872: 5.

1872: 7. LECLERC, A. Dosage du manganèse dans les sols et dans les végétaux.

C. R., 75, 1209; Chem. News, 26, 296; Jsb. Chem., 1872, 910; Chem. Centrbl., 1872, 88; Bull. soc. chim. (2), 19, 177; Ztschr. anal. Chem., 12, 308; Dingl. pol. J., 206, 366; J. Chem. Soc. (Lond.), 26, 193; Chem.-techn. Rep., 12, a, 193; Am. Chemist (1875), 5, 267; Arch. Pharm., 202, 268.

Oxidation to permanganic acid by means of lead peroxide, or red lead, and titration with mercurous nitrate.

- 1872: 8. PICHARD. Dosage du manganèse dans les minerais de fer, les fontes, les aciers, par un procédé colorimetrique.
 - C. R., 75, 1821; Dingl. pol. J., 207, 136; Jsb. Chem., 1872, 909; Chem. News, 27, 85; Bull. soc. chim. (2), 19, 253; Ztschr. anal. Chem., 12, 308; J. Chem. Soc. (Lond.), 26, 407; Chem.-techn. Rep., 12, a, 195; Berg- u. hüttenm. Ztg., 32, 91.

Oxidation to permanganic acid by means of lead peroxide in nitric acid solution.

1872: 9. DE REZENDE. Note sur un procédé de séparation du fer et du manganèse.

Ann. des Mines (7), 1, 418. Separation from iron by means of cupric oxide.

1872: 10. TAMM, H. On an Improved Mode of Estimating Manganese.

> Chem. News, 26, 37; Am. Chemist, 3, 145; Jsb. Chem., 1872, 910; Monit. scientif., 14, 973; Bull. soc. chim. (2), 19, 121.

> Precipitation with ammonium carbonate from solutions containing ammonium chloride. Separation from iron by the succinate method, and from zinc and nickel by ammonium carbonate.

- 1872: 11. TAMM, H. On the Metallurgy of Manganese, and the Docimastic Assaying of Manganese Ores. Chem. News, 26, 111; Am. Chemist, 3, 177; Pol. Centrbl., 1872, 1348. Dry assay with various fluxes.
- 1873: I. BRÜNNER, A. Schnell durchführbare colorimetrische Probe auf Mangangehalt des Roheisens, Stahls, Eisens, und Erze.
 - Oester. Ztschr. Berg- u. Hüttenw., 21, 341; Chem. Centrbl., 1873, 757; Bull. soc. chim. (2), 21, 278; J. Chem. Soc. (Lond.), 27, 604 and 816; Chem.-techn. Rep., 12, b, 196; Wagner's Jsb., 20, 10; Pol. Centrbl., 1873, 1367; Dingl. pol. J., 210, 278.
 Conversion to sodium manganate and comparison of solutions.
- 1873: 2. GIBBS, W. On the Estimation of Manganese as Pyrophosphate.

Chem. News, 28, 51; Jsb. Chem., 1873, 934; J. Chem. Soc. (Lond.), 27, 92.

Precipitation by means of salt of phosphorus. See 1867: 3.

- 1874: I. KOPPMAYER, M. Ueber A. Brünner's colorimetrische Probe auf Mangangehalt des Stahls, Eisens, und der Erze.
 - Dingl. pol. J., 211, 133; Jsb. Chem., 1874, 988; Chem. Centrbl., 1874, 138; J. Chem. Soc. (Lond.), 27, 1009; Pol. Centrbl., 1874, 395; Berg- u. hüttenm., Ztg., 33, 109.
 Regards Brünner's method as valueless. See 1873: 1.
- 1874: 2. MORRELL, T. T. Estimation of Manganese.
 Am. Chemist, 5, 213; Jsb. Chem. 1874, 988.
 Colorimetric method depending upon the liberation of iodine in solution, after precipitation by bromine.
- 1874: 3. PIESSE, C. H. The Estimation of Silicon, Graphite, Manganese, Aluminium, and Calcium in Pig Irons.

Chem. News, 29, 110; Jsb. Chem., 1874, 986; Bull. soc. chim. (2), 22, 67; J. Chem. Soc. (Lond.), 27, 711.
Separation from iron by the basic acetate method and precipitation

1874: 4. PARRY, J. Estimation of Manganese in Spiegeleisen.
 Chem. News, 29, 86; Jsb. Chem., 1874, 987; Am. Chemist, 4, 434;
 J. Chem. Soc. (Lond.), 27, 712; Bull. soc. chim. (2), 22, 68.
 Adaptation of Fresenius-Will method, after solution in nitric acid and ignition of the residue left on evaporation.

by bromine.

- 1874: 5. POUCHET, A. G. Revue des méthodes d'analyse des produits industriels: Titrage et assai des manganèses. Monit. scientif., 16, 1130.
 - Comments of Mohr (1855: 1), Fresenius-Will (1843: 3), and Bunsen (1853: 1) methods for the determination of peroxide; also of method involving oxidation of sulphurous acid by chlorine evolved, and precipitation of barium sulphate. Outlines of

methods for the determination of total acid consumed, and of manganese in "Weldon Mud."

1874: 6. WILLIS, A. Estimation of Manganese in Spiegeleisen.

Chem. News, 29, 150; Jsb. Chem., 1874, 987. Comments on Piesse, 1874: 3.

1875: 1. BOUSSINGAULT. Études sur la transformation du fer en acier par la cémentation.

Ann. chim. phys. (5), 5, 184; Dingl. pol. J., 224, 80; Jsb. Chem., 1877, 1064; Chem. Centrbl., 1877, 376.

Separation from iron by basic acetate method, and precipitation by hypochlorite. Determination of small quantities by means of lead peroxide and titration with mercurous nitrate; also electrolytic deposition.

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* Annals Lyceum Nat. Hist. N. Y., Nov., 1875. Compilation of journal literature on manganese and its compounds.

1875: 2. KERN, S. Estimation of Manganese in Spiegeleisen, Iron, and Steel.

Chem. News, 32, 100; Jsb. Chem., 1875, 955; Ztschr. anal. Chem., 16, 505; J. Chem. Soc. (Lond.), 29, 110; Chem.-techn. Rep., 14, a, 278; Wagner's Jsb., 22, 19; Am. Chemist, 6, 192.

Precipitation as manganous hydroxide, reduction in hydrogen, separation of iron by magnet, and ignition to protosesquioxide.

1875: 3. LUNGE, G. Ueber die neusten Fortschritte in der Sodaund Chlorkalk-Industrie in England.

> Dingl. pol. J., 215, 157; Pol. Centrol., 1875, 853.
> Determination in "Weldon Mud" by ferrous sulphate and permanganate.

1875: 4. MORRELL, T. T. Note on the Estimation of Manganese in Spiegeleisen.

Am. Chemist, 6, 45; Jsb. Chem., 1875, 954.

Separation from iron by the basic acetate method, precipitation by bromine, and colorimetric determination by the liberation of iodine. (See 1874: 2.)

1875: 5. VOGEL, H. W. Ueber die Absorptionsspectren einiger Salze der Metalle der Eisengruppe und Anwendung in der Analyse.

> Ber., 8, 1533; Dingl. pol. J., 219, 533. Detection by means of absorption spectra of permanganic acid.

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Gazz. chim. ital., 1876, 464; Jsb. Chem., 1876, 1000; Chem. News, 35, 75; J. Chem. Soc. (Lond.), 32, 223.

Detection in plant ashes by the color of the residue after evaporation with nitric acid.

1876: 2. FRESENIUS, R. Methode zur Analyse alkalischer Mineralwasser.

Ztschr. anal. Chem., 15, 222 and 225.

Determination in mineral water by precipitation as sulphide and ignition in hydrogen.

- 1876: 3. GALBRAITH, W. The Determination of Manganese in Spiegeleisen.
 - Chem. News, 33, 47; Am. Chemist, 6, 462; Oester. Ztschr. Berg- u. Hüttenw., 25, 31; Jsb. Chem., 1876, 999; Dingl. pol. J., 221, 448; Ztschr. anal. Chem., 16, 506; J. Chem. Soc. (Lond.), 28, 750; Wagner's Jsb., 22, 18; Berg- u. hüttenm. Ztg., 35, 355; Chem.-techn. Rep., 15, 479.
 - Solution in nitric acid, evaporation and ignition; solution of residue with ferrous ammonium sulphate and titration for the excess.
- 1876: 4. KERN, S. Estimation of Manganese in Cast Iron.
 - Chem. News, 33, 90; J. Chem. Soc. (Lond.), 29, 962; Dingl. pol. J., 221, 188; Am. Chemist, 7, 76; Rev. univers. des Mines, 39, 199; Bull. soc. chim. (2), 26, 474.
 - Precipitation of iron and manganese by potassium hydroxide, solution of manganese by addition of ammonium chloride, filtration, precipitation as sulphide, strong heating with sulphuric acid, and weighing as mangano-manganic oxide.
- 1876: 5. PETERS, S. On the Estimation of Manganese in Iron and Steel.

Chem. News, 33, 35; Jsb. Chem., 1876, 999; Dingl. pol. J., 221, 486; J. Chem. Soc. (Lond.), 29, 750; Wagner's Jsb., 22, 19; Chem.-techn. Rep., 15, 480.

Colorimetric method. Oxidation to permanganic acid by lead peroxide.

- 1876: 6. PHIPSON, T. L. Determination of Manganic Oxide. Chem. News, 34, 19 and 39. Determination of manganese peroxide in the presence of sesquioxide.
- 1877: I. BOLTON, H. C. Schemes of Analysis Executed in the School of Mines, Columbia College. Am. Chemist, 7, 307. Determination as pyrophosphate (Gibbs' method, 1867: 3).
- 1877: 2. CHAPMAN, E. J. On Some Blow-pipe Reactions. Chem. News, 35, 13 and 26; J. Chem. Soc. (Lond.), 31, 489. Detection by means of sodium carbonate.

- 1877: 3. CLASSEN, A. Quantitative Bestimmung des Mangans durch Fällung als Manganoxalat.
 - Ztschr. anal. Chem., 16, 315; J. Chem. Soc. (Lond.), 32, 804; Dingl. pol. J., 225, 515; Chem. Centrbl., 1877, 503; Chem.-techn. Rep., 16, 636.

Precipitation as oxalate and weighing as protosesquioxide.

1877: 4. CLASSEN, A. Zur Trennung des Mangans von Kalk.

 Ztschr. anal. Chem., 16, 318; Jsb. Chem., 1877, 1055; Chem. Centrbl., 1877, 470; J. Chem. Soc. (Lond.), 32, 805; Dingl. pol. J., 225, 515; Chem.-techn. Rep., 16, 636.

Separation by means of oxalic acid not practicable.

1877: 5. CLASSEN, A. Ueber die Abscheidung des Mangans als wasserfreies Sulfür.

Ztschr. anal. Chem., 16, 319; Jsb. Chem., 1877, 1062; Chem. Centrbl., 1877, 470; J. Chem. Soc. (Lond.), 32, 514.
Precipitation as sulphide in presence of potassium oxalate.

1877: 6. CLASSEN, A. Ueber eine neue Methode zur Trennung des Eisens von Mangan, Kobalt, Nickel und Zink.

> Ber., 10, 1316; Jsb. Chem., 1877, 1064 and 1066; Chem. Centrbl., 1877, 602; Ztschr. anal. Chem., 16, 471; Bull. soc. chim. (2), 30, 409.

> Separation from iron by means of neutral potassium oxalate and acetic acid.

- 1877: 7. CLASSEN, A. Quantitative Bestimmung von Mangan, Kobalt, Nickel und Zink durch Fällung als Oxalate. Ber., 10, 1315; J. Chem. Soc. (Lond.), 32, 924. See 1877: 3.
- 1877: 8. DEBY. Determination of Manganese in Spiegel, Iron, and Steel at Terrenoire.

* Report, 1877, II; Berg- u. hüttenm. Ztg., 37, 391.

Oxidation to permanganic acid by lead peroxide and titration with arsenious acid.

- 1877: 9. FUNARO, A. Della seperazione quantitativa del ferro e del manganese nei minerali ferro-manganici.
 - Gazz. chim. ital., 7, 286; Jsb. Chem., 1877, 1064; Chem. Centrbl., 1877, 661; Ber., 10, 1383; J. Chem. Soc. (Lond.), 32, 805; Dingl. pol. J., 225, 610; Chem.-techn. Rep., 16, 635.

Separation from iron by means of ammonium benzoate or succinate.

1877: 10. HANNAY, J. B. Note on a New Manganese Reaction.

J. Chem. Soc. (Lond.), 33, 269; Jsb. Chem., 1877, 1063; Chem. Centrbl., 1878, 41; Chem. News, 36, 212; Bull. soc. chim. (2), 30, 412; Ber., 10, 2052; Chem.-techn. Rep. 16, 637.

Precipitation by potassium chlorate in nitric acid solution, and determination by gravimetric and volumetric methods.

1877: 11. KERN, S. Quantitative Analysis of Certain Metals in Iron and Steel.

Chem. News, 35, 67; Jsb. Chem., 1877, 1057; J. Chem. Soc. (Lond.), 32, 647; Eng. Min. J., 24, 127.

Determination in chrome-iron alloys. Use of sodium hypochlorite for precipitation advised.

- 1877: 12. KERN, S. On the Estimation of Manganese in Spiegeleisen and Ferro-manganese.
 - Chem. News, 35, 247 and 270; Ber., 10, 975; Jsb. Chem., 1877, 1062; Chem. Centrbl., 1877, 457; Dingl. pol. J., 225, 392; Chem. -techn. Rep., 16, 635.

Determination in alloys by ignition of mixed oxides of iron and manganese in hydrogen, and then in chlorine, leaving a residue of protosesquioxide. Also direct ignition of the alloy with ammonium chloride.

1877: 13. KRÄMER, C. Zur Trennung des Mangans von Eisen.

Ztschr. anal. Chem., 16, 334; Jsb. Chem., 1877, 1063; J. Chem. Soc. (Lond.), 32, 805.

Separation from iron by the basic acetate method. See Stöckmann, 1877: 20.

1877: 14. MUNROE, C. E. The Estimation of Manganese as Pyrophosphate.

> Am. Chemist, 7, 287; Jsb. Chem., 1877, 1061; Iron, 9, 555. The influence of ammonia upon the determination as phosphate.

- 1877: 15. PARREÑO, A. G. Determination du manganèse métallique par la voie volumétrique.
 - Ann. chim. phys. (5), 11, 571; Jsb. Chem., 1877, 1062; Chem. Centrbl., 1877, 615; Am. J. Sci. (3), 14, 418; J. Chem. Soc. (Lond.), 32, 924; Pharm. Centrh., 18, 396; Chem.-techn. Rep., 16, 636.

Ignition of mineral to mangano-manganic oxide, treatment with hydrochloric acid and determination of the iodine liberated from potassium iodide, by the chlorine evolved.

1877: 16. PERREY. (Title unknown.)

* Bull. de Rouen, 1877, 104; Jsb. Chem., 1877, 1063; Chem. Centrbl., 1878, 15; Dingl. pol. J., 226, 194; Chem.-techn. Rep., 16, 640.

Comparison of Fresenius-Will (1843: 3), Hempel (1858: 1), Gay-Lussac (1829: 1), Mohr (1855: 1), and Bunsen (1853: 1) methods for analysis of pyrolusite.

- 1877: 17. RICHE, M. A. Note sur la dosage du manganèse, du nickel, du zinc et du plomb.
 - C. R., 85, 226; Jsb. Chem., 1877, 1066; Chem. News, 36, 96; Ztschr. anal. Chem., 17, 216; Bull. soc. chim. (2), 29, 378; Dingl. pol. J., 239, 380; J. Chem. Soc. (Lond.), 32, 924; Eng. Min. J., 24, 222; Chem.-techn. Rep., 16, 637; Rev. univers. des mines, 2, 297; Chem. Ind., 7, 27.

Electrolytic determination.

- 1877: 18. RILEY, E. On the Estimation of Manganese in Spiegeleisen and Iron in Manganiferous Ores.
 - J. Iron Steel Inst., 1877, a, 52; Chem. News, 35, 175; J. Chem. Soc. (Lond.), 32, 1; Chem. Centrbl., 1877, 376; Oester. Ztschr. Bergu. Hüttenw., 25, 424; Dingl. pol. J., 227, 493; Iron, 9, 617, 711, and 746; Jsb. Chem., 1877, 1061; Ber., 10, 911; Bull. soc. chim. (2), 29, 282; Berg- u. hüttenm. Ztg., 36, 223; Wagner's Jsb., 23, 21; Chem.-techn. Rep., 16, 633.

Separation from iron by the acetate method, precipitation by bromine, and ignition; also determination by difference.

- 1877: 19. ROSENTHAL, G. Ueber die Fällung des Mangans mit Wasserstoffsuperoxyd.
 - Dingl. pol. J., 225, 154; Jsb. Chem., 1877, 1037; Chem. Centrbl., 1877, 651; Chem. News, 36, 147; Ztschr. anal. Chem., 17, 364; J. Chem. Soc. (Lond.), 32, 923; Bull. soc. chim. (2), 32, 364; Berg- u. hüttenm. Ztg., 36, 324; Chem.-techn. Rep., 16, 635.

Separation from iron by the acetate method, and precipitation by means of hydrogen peroxide.

1877: 20. STÖCKMANN, C. Ueber die Bestimmung von Mangan und Phosphor im Spiegeleisen.

> Ztschr. anal. Chem., 16, 172; Jsb. Chem., 1877, 1063; Dingl. pol. J., 225, 108; J. Chem. Soc. (Lond.), 32, 648; Chem. News, 36, 275; Monit. scientif., 19, 1274.

Separation from iron by the acetate method. See 1877: 13.

1878: 1. Bong, G. Sur un bleu au manganèse.

Bull. soc. chim. (2), 29, 199; Jsb. Chem., 1878, 1129.Detection with the aid of a flux of silica, metallic sodium, and calcium carbonate.

1878: 2. DESHAYES, V. Dosage du manganèse dans les fers, fontes et aciers; dans les spiegels, ferro-manganèses et minerais.

Bull. soc. chim. (2), 29, 541; Jsb. Chem., 1878, 1062; Chem. News, 38, 70; J. Chem. Soc. (Lond.), 34, 808; Bull. soc. ind. minerale (2), 7, 163.

Oxidation to permanganic acid by lead peroxide, and titration with arsenious acid.

1878: 3. MATZURKE, G. Zur Trennung von Eisen und Mangan. Ztschr. anal. Chem., 17, 78; Jsb. Chem., 1878, 1061. Separation from iron by acetate method. See Krämer, 1877: 13.

- 1878: 4. MORAWSKI, T., and STINGL, J. Ueber eine maassanalytische Bestimmung des Mangans.
 - J. prakt. Chem. (2), 18, 96; Jsb. Chem., 1878, 275; Chem. Centrol., 1878, 758; Chem. News, 38, 297; J. Chem. Soc. (Lond.), 36, 277; Ztschr. anal. Chem., 18, 471; Ber., 11, 1933; Bull. soc. chim. (2), 32, 603; Chem.-techn. Rep., 17, b, 241.

- 1878: 5. MORAWSKI, T., and STINGL, J. Zur Bunsen'schen Braunsteinbestimmungsmethode.
 - J. prakt. Chem. (2), 18, 101; Chem. Centrbl., 1878, 759; Jsb. Chem., 1878, 275; Ztschr. anal. Chem., 18, 471; Ber., 11, 1933; J. Chem. Soc. (Lond.), 36, 278.

Modification of an apparatus for the Bunsen method for the analysis of pyrolusite.

1878: 6. MÜLLER, F. C. G. Untersuchungen über den Bessemerprocess.

Ber., 11, 552; Ztschr. Ver. d. Ing., 22, 467.

Separation from iron by the acetate method and precipitation by chlorine.

1878: 7. PROCHASKA, J. Der Siemens-Martinprocess im Südbahnwalzwerke im Graz.

Oester. Ztschr. Berg- u. Hüttenw., 26, 116; Berg- u. hüttenm. Ztg., 37, 147.

Oxidation to permanganate acid by red lead and titration by ferrous sulphate.

1878: 8. RICHE, A. Mémoire sur le dosage du manganèse, du plomb du cuivre, du zinc et du nickel; et sur l'analyse des alliages de ces métaux.

Ann. chim. phys. (5), 13, 508; Berg- u. hüttenm. Ztg., 37, 26; Jsb. Chem., 1878, 1062; J. Chem. Soc. (Lond.), 34, 750.
Electrolytic determination.

1878: 9. WRIGHT, C. R. A., and LUFF, A. P. Researches on Some Points in Chemical Dynamics.

J. Chem. Soc. (Lond.), 33, 526.

A study of the oxides of manganese with reference to the bearing of the results reached upon current analytical methods.

1879: I. BEILSTEIN, F., and JAWEIN, L. Ueber eine directe Trennung des Mangans von Eisen.

Ber., 12, 1528; Iron, 14, 587; Jsb. Chem., 1879, 1045; Chem. News, 40, 300; Ztschr. anal. Chem., 19, 77; Chem. Ztg., 3, 630; Bull. soc. chim. (2), 32, 604; J. Chem. Soc. (Lond.), 38, 61; Dingl. pol. J., 234, 254; Wagner's Jsb., 26, 320; Monit. scientif., 22, 811;

Determination by potassium permanganate in slightly acid solution.

J. Am. Chem. Soc., 1, 533; Berg- u. hüttenm. Ztg., 38, 360; Am. Chem. J., 2, 73; Arch. Pharm., 215, 449; J. Iron Steel Inst., 1880, 354-

Separation from iron by iodine in cyanide solution; also by precipitation by potassium chlorate in nitric acid; precipitation as sulphide.

1879: 2. CARNOT, A. Sur l'emploi de l'hydrogène sulfuré par voie sèche dans les analyses.

Bull. soc. chim. (2), **32**, 161; C. R., **89**, 167; Jsb. Chem., **1879**, 1024. Determination as sulphide, with use of Rose crucible. See 1860: 5.

1879: 3. CLASSEN, A. Zur Trennung des Mangans von Zink.

Ztschr. anal. Chem., 18, 194; Chem. Centrbl., 1879, 366; J. Chem.
Soc. (Lond.), 36, 1055; Chem. Ztg., 3, 253; Chem. News, 40, 33;
J. Am. Chem. Soc., 1, 327.

Criticism of Tamm procedure. See 1872: 10.

1879: 4. CLASSEN, A. Ueber eine neue quantitative analytische Methode von vielfacher Anwendbarkeit.

Ztschr. anal. Chem., 18, 379, 380, and 396; Chem. News, 40, 33; Chem. Ztg., 3, 676; Bull. soc. chim. (2), 35, 91; J. Chem. Soc. (Lond.), 36, 969.

Separation from iron and aluminum with the aid of potassium oxalate.

1879: 5. CLASSEN, A. Ueber eine neue Methode zur Trennung des Eisenoxyds und der Thonerde von Mangan.

Ztschr. anal. Chem., 18, 175; Chem. Centrbl., 1879, 365; Jsb. Chem., 1879, 1045; Bull. soc. chim. (2), 33, 446; J. Chem. Soc. (Lond.), 36, 1055; J. Am. Chem. Soc., 1, 325.
See 1879: 4.

- 1879: 6. KESSLER, F. Ueber die Bestimmung des Mangans, besonders in Eisen-Manganlegirungen.
 - Ztschr. anal. Chem., 18, 1; Iron, 13, 643, 675, and 706; Jsb. Chem., 1879, 1050; Chem. Centrbl., 1879, 90; J. Chem. Soc. (Lond.), 36, 341; Chem. Ztg., 3, 30; J. Am. Chem. Soc., 1, 83; Am. Chem. J., 1, 363; J. Frank. Inst., 107, 411; Dingl. pol. J., 232, 91; J. Iron Steel Inst., 1880, 353.

Separation from iron by means of sodium sulphate, precipitation by bromine, solution of the peroxide by antimonious chloride, and titration with potassium permanganate.

1879: 7. LEDEBUR, A. Zur chemischen Untersuchung des Eisens und seiner Erze.

Berg- u. hüttenm. Ztg., 38, 47.

Comparison of Kessler (1879: 6), Riley (1877: 18), Müller (1851: 2), the sulphide and indirect methods of determination.

1879: 8. MACKINTOSH, J. B. (Correspondence.)

School Mines Quart., 1, 127.

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1879: 9. PATTINSON, J. On a New Volumetric Method of Determining Manganese in Manganiferous Iron Ores, Spiegeleisen, Steel, etc.

J. Iron Steel Inst., 1879, a, 209; b, 335; Rev. univers. des mines, 10, 135.

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 - J. Chem. Soc. (Lond.), 37, 128; Chem. News, 40, 261; Jsb. Chem., 1880, 1182; Chem. Centrbl., 1880, 103; Bull. soc. chim. (2), 36, 261; Dingl. pol. J., 236, 350.

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- 1879: 13. RÖSSLER, C. Ueber eine neue Bestimmung des Mangans mit Anwendung des Volhard'schen Silbertitrirverfahrens.
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 chim. (2), 33, 281; Wagner's Jsb., 25, 16; J. Am. Chem. Soc., 1, 329; Dingl. pol. J., 233, 86.
 - Precipitation with ammoniacal silver nitrate, filtration of the mangano-silver compound, and titration for the excess of the silver by the sulphocyanide procedure.

See 1879: 9.

- 1879: 14. VOLHARD, J. Zur Scheidung und Bestimmung des Mangans.
 - Ann. Chem. (Liebig), 198, 318; Jsb. Chem., 1879, 1048; Chem. News, 40, 207; J. Chem. Soc. (Lond.), 38, 141; Ztschr. anal. Chem., 20, 271 and 285; Ber., 12, 2175; Bull. soc. chim. (2), 34, 715; Berg- u. hüttenm. Ztg., 39, 150; J. Iron Steel Inst., 1880, 355.

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- 1880: 4. DUNSTON, W. R. The Analysis of Steel.
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- 1880: 5. HASWELL, A. E. Volhard's Titrirung des Mangans mit übermangansaures Kali.

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Confirmation of Volhard's procedure. (1879: 14.)

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> Chem. News, 40, 273; J. Chem. Soc. (Lond.), 38, 289. See title.

1880: 7. v. JÜPTNER, H. Volhard's Methode der Trennung und Bestimmung des Mangans.

Oester. Ztschr. Berg- u. Hüttenw., 28, 168.

Favorable comment upon the Volhard method. (1879: 14.)

- 1880: 8. JURISCH, K. Zur Prüfung des Weldonschlamms. Chem. Ind., 1880, 193; Dingl. pol. J., 237, 312; Wagner's Jsb., 26, 324. Determination of peroxide in "Weldon Mud" and of total acid consumed on solution.
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Ztschr. anal. Chem., 19, 17; Chem. News, 41, 213; Jsb. Chem., 1880, 1140. Electrolytic precipitation.

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Chem. News, 41, 179; Jsb. Chem., 1880, 1183. Precipitation is incomplete by means of chloride of lime except in the presence of ferric chloride. See 1880: 16.

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Criticism of method used by Lunge, 1880: 11. See also 1881: 12.

- 1880: 15. Rössler, C. Ueber eine neue maassanalytische Bestimmungsmethode des Mangans und des Kobalts.
 - Ann. Chem. (Liebig), 200, 323; Jsb. Chem., 1880, 1182; Chem. Centrbl., 1880, 250; Ztschr. anal. Chem., 19, 75; Chem. News, 41, 184; Chem. Ztg., 4, 86; J. Chem. Soc. (Lond.), 38, 347; Dingl. pol. J., 235, 391.
 - Determination with the aid of silver and the Volhard silver titration. See 1879: 12.

1880: 16. WELDON. Volumetric Estimation of Manganese.

Chem. News, 41, 207; Jsb. Chem., 1880, 1183.

Precipitation by means of chloride of lime found to be complete. See 1880: 13.

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J. Chem. Soc. (Lond.), 37, 22; Chem. News, 40, 261; Jsb. Chem., 1880, 316; Chem. Centrbl., 1880, 66; Chem. Ztg., 4, 86; Ber., 13, 427.
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1880: 18. WRIGHT, C. R. A., and MENKE, A. E. Volumetric Determination of Manganese.

> J. Chem. Soc. (Lond.), 37, 42. Comment on Pattinson's method. See 1879: 9.

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Separation from zinc with the use of ammonium sulphocyanate.

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> Separation from iron by precipitation with iodine from cyanide solution, and also by potassium chlorate in nitric acid solution.

1881: 2. CLASSEN, A. Electrolytische Bestimmungen und Trennungen.

Ber., 14, 2772; Dingl. pol. J., 242, 440; Jsb. Chem., 1881, 1151;
Bull. soc. chim. (2), 37, 525; J. Chem. Soc. (Lond.), 42, 896; Wagner's Jsb., 28, 448; Ztschr. anal. Chem., 22, 417; School Mines Quart., 3, 302.

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1881: 3. CLASSEN, A., and v. REIS, M. Electrolytische Bestimmungen und Trennungen.

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Electrolytic separation from iron, and determination by electrolysis.

1881: 4. DELVAUX, G. Séparation de l'oxyde de nickel et l'oxyde de cobalt.

C. R., 92, 723; Jsb. Chem., 1881, 1188; Ztschr. anal. Chem., 21, 111. Separation from cobalt by sulphuretted hydrogen in acetic acid solution, and from nickel by oxidation on standing in the air.

1881: 5. DESHAYES, V. Revue métallurgique.

Bull. soc. chim. (2), 36, 121.
Comments on Ford's and colorimetric methods of determination in steels. See 1881:8.

- 1881: 6. DONATH, E. Ueber eine volumetrische Bestimmung von Chrom und Mangan neben Eisenoxyd und Thonerde.
 - Ber., 14, 982; Chem. News, 43, 253; Jsb. Chem., 1881, 1184; Chem. Centrbl., 1881, 469; Ztschr. anal. Chem., 22, 245; Chem. Ztg., 5, 304; J. Chem. Soc. (Lond.), 40, 760; Bull. soc. chim. (2), 37, 92; Wagner's Jsb., 27, 355; Dingl. pol. J., 242, 391.

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1881: 7. EMMERTON, F. A. Chemical Methods for Analysing Rail Steel.

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Determination by the Volhard method. (1879: 14.)

1881: 8. FORD, S. A. Method for the Estimation of Manganese in Spiegels, Irons, and Steels.

Trans. Am. Inst. Min. Eng., 9, 397; Eng. Min. J., 32, 6.

Precipitation by means of potassium chlorate, re-solution in acid and final precipitation as manganese ammonium phosphate.

1881: 9. FORGUIGNON. Recherches sur la fonte malleable et sur le recuit des aciers.

Ann. chim. phys. (5), 23, 447.

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1881: 10. ILES, M. W. Decomposition of Slags and Silicates.

Chem. News, 43, 78; J. Chem. Soc. (Lond.), 40, 645.
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Trans. Am. Inst. Min. Eng., 10, 101; J. Iron Steel Inst., 1881, b, 655; School Mines Quart., 3, 51.
General discussion of methods. Results of twelve chemists upon the same sample.

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> Dingl. pol. J., **242**, 371. Reply to Post, 1880: 14.

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- Comments on Fresenius-Will (1843:3), Bunsen (1853:1), and iron methods for the determination of the peroxide.
- 1881: 14. v. REIS, M. A. Ueber die Benutzung einiger oxalsauren Salze in der Analyse.

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> Precipitation of iron and manganese by sodium bicarbonate, and titration with potassium permanganate in the presence of the precipitate.

1881: 16. TERREIL, A. Liqueur volumétrique pour le dosage des composés suroxygénés ou agissant comme corps oxydants.

Bull. soc. chim. (2), 35, 551; Chem. Centrbl., 1881, 569; Jsb. Chem., 1881, 1155.

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1881: 17. TROILIUS, M. Chemical Methods for Analysing Rail Steel. Trans. Am. Inst. Min. Eng., 10, 173; Eng. Min. J., 32, 300; J. Iron Steel Inst., 1881, b, 654; School Mines Quart., 3, 52.

Separation from iron by the acetate method, precipitation by bromine in ammoniacal solution, and ignition to mangano-manganic oxide.

1881: 18. WILLIAMS, F. A. A Volumetric Estimation of Manganese in Pig Iron and Steel.

> Trans. Am. Inst. Min. Eng., 10, 100; Jsb. Chem., 1882, 1288; Dingl. pol. J., 246, 241; J. Iron Steel Inst., 1881, b, 656; Iron, 18, 540; Wagner's Jsb., 28, 15.

> Precipitation by means of potassium chlorate, solution in oxalic acid, and titration for the excess of the latter.

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C. R., 94, 1625; Jsb. Chem., 1882, 1296; Chem. News, 46, 3. Nine methods for the separation from gallium.

- 1882: 2. CABOT, J. W. Chemical Methods for Analysing Rail Steel. Trans. Am. Inst. Min. Eng., 10, 191. Separation from iron by the acetate method, and precipitation by bromine.
- 1882: 3. DARTON, N. H. On the Estimation of and Separation of Manganese.

Scientif. Am. Supplem. 1882, 5168; Chem. Ind., 5, 201; Wagner's Jsb., 28, 448; Rep. anal. Chem., 2, 216; J. Soc. Chem. Ind., 1, 468. Determination of peroxide in pyrolusite by solution with potassium oxalate, absorption of carbon dioxide by barium hydroxide, and titration for the excess of the latter.

- 1882: 4. DEWEY, F. P. Chemical Methods for Analysing Rail Steel. Trans. Am. Inst. Min. Eng., 10, 194. Determination by Kent's method. See 1881; 11.
- 1882: 5. DIEHL, W. Zur maassanalytische Bestimmung der Hyperoxyde.

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Chem. News, 45, 137; Jsb. Chem., 1882, 302.

A study of the oxides of manganese and their relation to the Guyard method. See 1863: 2.

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> Monit. scientif. (3), 12, 778; Jsb. Chem., 1882, 1293. Separation from zinc by ammonium sulphocarbonate.

1882: 8. HASWELL, A. E. Maassanalytische Bestimmung und Trennung der Metalle.

> Rep. anal. Chem., 2, 243; J. Iron. Steel Inst., 1882, 743. Determination by Volhard method. No details.

1882: 9. JEWETT, J. Influence of Acetic Acid on the Separation of Iron as a Basic Acetate from Manganese, Zinc, Cobalt, and Nickel.

> Am. Chem. J., 1, 251; Ztschr. anal. Chem., 21, 262. See title.

1882: 10. KEISER, E. H. The Electrolytic Separation of Manganese in Pig Iron and Steel.

> Eng. Min. J., **33**, 131. See title.

1882: 11. LEDEBUR, A. Eine colorimetrische Manganbestimmung.

Berg- u. hüttenm. Ztg., 41, 417; Jsb. Chem., 1882, 1288; Chem. Centrbl., 1882, 733; Ztschr. anal. Chem., 22, 607; Ber., 15, 2926; Wagner's Jsb., 29, 15; Stahl u. Eisen, 2, 626; Rep. anal. Chem., 2, 346; J. Chem. Soc. (Lond.), 44, 242; J. Soc. Chem. Ind., 2, 249; Techn.-chem. Jahrb., 5, 11; Dingl. pol. J., 248, 215; Chem. -techn. Rep., 21, b, 211.

Oxidation by lead peroxide to permanganic acid and comparison with solutions of potassium permanganate.

1882: 12. LEFORT, J., and THIEBAULT, P. De l'influence de la gomme arabique dans certaines réactions chimiques.

J. de pharm. (5), 6, 169; Pharm. J. Trans. (3), 13, 301; Jsb. Chem., 1882, 1259.

Influence of gum arabic on the precipitation as sulphide.

1882: 13. Löwe, J. Ueber den qualitativen Nachweis und quantitativen Bestimmung des Arsens—sowie einiger in geringer Menge in gediegenen Kupfer des Handels vorkommenden Metalle.

Ztschr. anal. Chem., 21, 516.

Separation in the analysis of commercial copper.

1882: 14. MILLS, E. J., and BECKET, J. H. Researches on Chemical Equivalence.

Phil. Mag. (5), 13, 170. Separation from nickel.

1882: 15. TAMM, A. Die üblichsten Eisenanalysen. (Title from Chem. Centrbl.)

* Jern.-Kont. Ann., 1882, 123; Chem. Centrbl., 1882, 766; Jsb. Chem., 1882, 1288; Berg- u. hüttenm. Ztg., 41, 448. Comparison of methods of determination.

1882: 16. TROILIUS, M. Bestämning af mangan i jern.

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- Precipitation by means of potassium chlorate, solution and separation of iron, re-precipitation by bromine, and weighing as protosesquioxide.
- 1882: 17. WAGNER, A. Ueber das Verhalten von Braunstein und Chlorkalk beim Glühen mit Chromoxyd und kohlensaurem Natron unter Ausschluss der Luft.

Ztschr. anal. Chem., 21, 493; Chem. News, 45, 80.

Determination of peroxide from the amount of chromate formed during the heating.

1883: 1. GOETZ. Die Bestimmung des Mangans in Eisen.

Dingl. pol. J., 248, 215.

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1883: 2. GUYARD, A. Recherche qualitative du manganèse dans le zinc de commerce, les cendrése de zinc et les calamines,—et recherches du bismuth dans le plomb commercial, au moyen de l'électrolyse.

C. R., 97, 673; Bull. soc. chim. (2), 40, 420; Jsb. Chem., 1883, 1514; Ber., 16, 2691; Chem. News, 48, 193; J. Chem. Soc. (Lond.) 46, 368 and 640; Chem. Ztg., 7, 1611; Rep. anal. Chem., 3, 379; Berg- u. hüttenm. Ztg., 42, 587.

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 - Chem. News. 47, 2; Jsb. Chem., 1883, 1566; Chem. Centrbl., 1883, 199; J. Chem. Soc. (Lond.), 44, 513; Ztschr. anal. Chem., 23, 60; Ber., 16, 262; Chem. Ztg., 7, 141; Chem. Ind., 6, 82; Chem.-techn. Rep., 22, 236; Wagner's Jsb., 29, 437; Dingl. pol. J., 248, 303.

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- 1883: 4. HAMPE, W. Zwei neue maassanalytische Manganbestimmungsmethoden.
 - Chem. Ztg., 7, 1103; Jsb. Chem., 1883, 1565; Ber., 16, 2531; Stahl.
 u. Eisen, 3, 638; Wagner's Jsb., 30, 141; Berg- u. hüttenm. Ztg.,
 42, 536; Ztschr. anal Chem. (1885), 24, 422; Techn.-chem.
 Jahrb., 6, 27.

Precipitation by means of potassium chlorate, solution by ferrousammonium sulphate, and titration for the excess of the latter. Evaporation of nitric-acid solution, heating of residue with phosphoric acid, and titration of the phosphate formed with a solution of ferrous sulphate.

1883: 5. HEMPEL, C. W. Zur maassanalytischen Bestimmung der Hyperoxyde.

> Dingl. pol. J., 247, 144. Criticism of Diehl's article. See 1882: 5.

1883: 6. v. JÜPTNER, H. Das übermangansaure Kali als Titrirflüssigkeit.

> Oester. Ztschr. Berg- u. Hüttenw., 31, 502. Comments on Volhard method. (1879: 14.)

1883: 7. KERL, B. (Title unknown.) Dingl. pol. J., 250, 416. Confirmation of Särnström method. See 1881: 15.

1883: 8. KNOP, W. Zur Analyse der Silicate.

Ber. königl. Sach. Ges. Wiss. (math.-phys. Classe), 1882, 35; Ztschr. anal. Chem., 22, 558.
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1883: 9. MACKINTOSH, B. The Volumetric Determination of Manganese.

> Trans. Am. Inst. Min. Eng., 12, 79; Chem. News, 48, 176; Am, Chem. J., 5, 290; Iron, 22, 464; Jsb. Chem., 1883, 1569; J. Iron Steel Inst., 1883, b, 761; J. Chem. Soc. (Lond.), 46, 220; Ber., 16, 2939; Berg- u. hüttenm. Ztg., 43, 302.

- 1883: 10. MEINEKE, C. Titrirung des Mangans durch übermangansaures Kali.
 - Rep. anal. Chem., 3, 337; Jsb. Chem., 1883, 1567; Ztschr. anal.
 Chem. (1885), 24, 430; Ber., 16, 3074; Wagner's Jsb., 29, 437;
 Berg- u. hüttenm. Ztg., 43, 23; Chem. Ztg., 7, 1609.

Separation from iron by zinc oxide, addition of the solution to an excess of permanganate, and titration for the excess with antimonious chloride.

- 1883: 11. MEINEKE, C. See 1883: 10.
- 1883: 12. ORLOWSKI, A. Ersetzung des Schwefelwasserstoffs bei dem systematischen Gang der qualitativen chemischen Analyse durch unterschwefligsaures Ammon.

Ztschr. anal. Chem., 22, 364. Detection of manganese.

Precipitation by potassium chlorate. Williams method. See 1881: 18.

- 1883: 13. RAIMOND, E. Nouvelle méthode volumétrique de dosage du manganèse dans les aciers, fontes, ferro-manganèses, etc.
 - Rev. univers. des mines, 13, 460; * Armengaud. Publ. Ind., 1883, 189; Jsb. Chem., 1883, 1673; Chem. Centrbl., 1884, 156; Chem. News, 48, 23; Chem. Ztg., 8, 344; Wagner's Jsb., 29, 18; J. Iron Steel Inst., 1883, b, 783; Chem.-techn. Rep., 23, a, 236; Rep. anal. Chem., 4, 137; Berg- u. hüttenm. Ztg., 43, 166; J. Chem. Soc. (Lond.), 48, 840; Dingl. pol. J., 250, 416.
 - Precipitation by means of potassium chlorate, and use of ferrous sulphate.
- 1883: 14. Särnström, C. G. Ueber volumetrische Manganbestimmung im Eisen.

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Comments on Schoeffel and Donath method of determination. See 1883: 17.

1883: 15. SÄRNSTRÖM, C. G. Tillägg till uppsat sen om manganprof medelst titrering.

> Jern.-Kont. Ann., 1883, 400; Stahl u. Eisen, 4, 127; Wagner's Jsb., 30, 23. See 1881: 15.

- 1883: 16. SCHMITT. Einführung einheitlicher Untersuchungsmethoden bei Manganbestimmungen in Eisen.
 - Stahl u. Eisen, 3, 487; Wagner's Jsb., 29, 18; Oester. Ztschr. Bergu. Hüttenw., 32, 164.

Request for the appointment of a commission to propose uniform methods for the determination of manganese.

- 1883: 17. SCHOEFFEL, R., and DONATH, E. Ueber eine neue Methode der volumetrischen Bestimmung des Mangans, insbesondere in Eisen und Stahl.
 - Oester. Ztschr. Berg- u. Hüttenw., 31, 229; Jsb. Chem., 1883, 1567;
 Chem. Centrbl., 1883, 332; Dingl. pol. J., 248, 421; Chem. Ztg.,
 7, 587; Ztschr. anal. Chem. (1885), 24, 427; Ber., 16, 1690; Wagner's Jsb., 29, 15; Stahl. u. Eisen, 3, 374; J. Iron Steel Inst., 1883,
 a, 381; Rep. anal. Chem., 3, 207; Berg- u. hüttenm. Ztg., 42, 231;
 Pharm. Runds, 1883, 352; Chem.-techn. Rep., 22, b, 235.

Volumetric determination by the addition of the manganese solution to alkaline permanganate solution.

1883: 18. SCHUCHT. Zur Elektrolyse.

Ztschr. anal. Chem., 22, 493; Jsb. Chem., 1883, 1512; Dingl. pol. J., 254, 298.

Electrolytic precipitation.

1883: 19. STONE, G. C. The Determination of Manganese in Spiegel.

Trans. Am. Inst. Min. Eng., 11, 323; Eng. Min. J., 35, 318; Iron, 22, 378; J. Iron Steel Inst., 1883, a, 366; Dingl. pol. J., 250, 416; Techn.-chem. Jahrb., 6, 27.

Precipitation by means of potassium chlorate, solution in oxalic acid, and titration with permanganate. Doubt as to the composition of the precipitate produced by the chlorate.

1883: 20. STONE, G. C. The Volumetric Determination of Manganese.

> Chem. News, 48, 273; Jsb. Chem., 1883, 1569; J. Chem. Soc. (Lond.), 46, 499; Wagner's Jsb., 29, 19. Reply to Mackintosh. 1883: 9.

1883: 21. TROILIUS, M. The Determination of Manganese in Spiegel, Ferro-manganese, Steel, etc.

Trans. Am. Inst. Min. Eng., 12, 73; J. Iron Steel Inst., 1883, b, 761;
Iron, 22, 397; Jern.-Kont. Ann., 1883, 466; Berg- u. hüttenm.
Ztg., 43, 284; Jsb. Chem., 1884, 1599; Chem. Centrbl., 1884, 716;
Ber., 17, 386, Ref.; J. Chem. Soc. (Lond.), 48, 597; Stahl. u. Eisen,
4, 126; Wagner's Jsb., 30, 18; J. Soc. Chem. Ind., 3, 523; 4, 137.
Use of potassium bichromate in connection with the Williams method of determination. See 1881: 18.

1883: 22. WOLFF, N. Ueber die Anwendung eines mit Bromdämpfen geschwängerten Luftstromes zur Fällung des Mangans.

Ztschr. anal. Chem., 22, 520; Jsb. Chem., 1883, 1566; Chem. Centrbl., 1884, 156; J. Chem. Soc. (Lond.), 46, 640; Ber., 16, 3075; Chem. News, 49, 201; Wagner's Jsb., 29, 438; Rep. anal. Chem., 3, 364; J. Am. Chem. Soc., 5, 244.

Precipitation by bromine in ammoniacal solution.

- 1883: 23. ZULKOWSKY, K. Zur Bestimmung des Mangans in Eisenerzen. (*Title from Dingl. pol. J.*)
 - * Ber. oester. chem. Ges., 1883, 3; Jsb. Chem., 1883, 1569; J. Chem. Soc., 46, 116; Dingl. pol. J., 248, 259; Wagner's Jsb., 28, 1;
 * Chemiker u. Droguist, 1883, 62; Rep. anal. Chem., 3, 124; Chem.-techn. Rep., 22, b, 236.

Precipitation as sulphide, ignition, solution of the residue in sulphurous and nitric acids, and titration with permanganate.

1884: I. ANGER, C. Prüfung der Mangantitrirung mit Kaliumpermanganat in alkalischer Lösung.

Stahl. u. Eisen, 4, 156; Wagner's Jsb., 30, 397; Dingl. pol. J., 254, 139.

Titration with permanganate in the presence of sodium carbonate in excess.

- 1884: 2. ATKINSON, A. J. Volumetric Estimation of Manganese. Chem. News, 49, 25; Jsb. Chem., 1884, 1599. Determination in spiegeleisen by difference, after determination of the iron volumetrically. See Holdich, 1884: 8.
- 1884: 3. BLOXAM, C. L. Estimation of Manganese in Cast Iron and Spiegeleisen.

Chem. News, 50, 112; Jsb. Chem., 1884, 1599; Chem. Centrbl., 1884, 849; Ber., 17, 508, Ref.; Chem. Ztg., 8, 1436; J. Iron Steel Inst., 1884, b, 584; J. Chem. Soc., 48, 84; Wagner's Jsb., 31, 18; J. Am. Chem. Soc., 6, 242; Berg- u. hüttenm. Ztg., 43, 520; Iron, 24, 271; Chem. Ind., 7, 362; Chem.-techn. Rep., 23, b, 186; 24, 249.
Separation from iron by double acetate precipitation with the addition of phosphates, and final precipitation of manganese as phosphate.

- 1884: 4. CLASSEN. A. Entgegnung. (Wieland, 1884: 17.)
 Ber., 17, 2351; Jsb. Chem., 1884, 1540; Ztschr. anal. Chem., 24, 247.
 Electrolytic determination. See also 1885: 4.
- 1884: 5. CLASSEN, A. Quantitative Analyse durch Elektrolyse.

Ber., 17, 2472 and 2484; Jsb. Chem., 1884, 1543; Ztschr. anal. Chem., 24, 255.

Separation from copper and chromium by electrolysis.

1884: 6. GMELIN, O. Chemische Notizen für der Giesserei-Techniker.

Oester. Ztschr. Berg- u. Hüttenw., 1884, No. 49; Berg- u. hüttenm. Ztg., 44, 23, Ref.

A modification of the Volhard method of determination.

1884: 7. Намоwsку. Ueber eine neue Anwendung des Wasserstoffsuperoxyds in der chemischen Analyse. (*Title from Chem.* Ztg.)

* Ber. oester. Ges. z. Förd. Chem. Ind., 1884, 8; Jsb. Chem., 1884, 1562; Rep. anal. Chem., 1884, 220; Chem. Ztg., 8, 789.

Precipitation by hydrogen peroxide and ignition to manganomanganic oxide.

1884: 8. HOLDICH. Volumetric Estimation of Manganese.

Chem. News, 49, 9 and 57; Jsb. Chem., 1884, 1598. Determination in spiegeleisen by difference, after volumetric determination of the iron.

1884: 9. HOLTHOF, C. Ueber Fällung des Mangans mit Brom.

Ztschr. anal. Chem., 23, 491; Jsb. Chem., 1884, 1598; Chem. Centrbl., 1885, 67; Ber., 18, 34, Ref.; Wagner's Jsb., 30, 397; Bergu. hüttenm. Ztg., 44, 55; J. Soc. Chem. Ind., 4, 367.

Precipitation by bromine in ammoniacal solution. Comments on Beilstein and Jawein chlorate method. See 1879: 1.

1884: 10. ILES, M. W. Manganese in Slags Formed by Argentiferous Lead Smelting.

School Mines Quart., 1884, 223; Chem. News, 50, 194; Berg- u. hüttenm. Ztg., 44, 16.

Comments on Haswell's method of determination. See 1880: 5.

- 1884: 11. LEDEBUR, A. Ueber Manganbestimmung im Eisenbetriebe.
 - Chem. Ztg., 8, 910, 927, and 963; Iron, 24, 558; Jsb. Chem., 1884, 1597; J. Chem. Soc. (Lond.), 44, 242; J. Iron Steel Inst., 1884, a, 269; Berg- u. hüttenm. Ztg., 43, 452; J. Soc. Chem. Ind., 3, 522.
 Comments on the Pattinson (1879: 9), Hampe (1883: 4), and Volhard (1879: 14) methods.
- 1884: 12. MACKINTOSH, J. B. The Influence of Organic Matter and Iron on the Volumetric Determination of Manganese.
 - Trans. Am. Inst. Min. Eng., 13, 39; Iron, 24, 224; Jsb. Chem., 1884, 1599; Chem. News, 50, 75; J. Chem. Soc. (Lond.), 48, 85; Ber., 18, 126; Chem. Ztg., 8, 1144; Berg- u. hüttenm. Ztg., 43, 302; Bull. soc. chim. (2), 41, 354; Eng. Min. J., 37, 440.

Influence of organic matter on the Williams method of determination (1881:18).

1884: 13. MACKINTOSH, J. B. Manganese Methods.

School Mines Quart., 6, 35. Comparison of methods.

- 1884: 14. MAUMENÉ, E. J. Sur l'existence du manganèse dans les animaux et les plantes et sur son rôle dans la vie animale.
 - Bull. soc. chim. (2), 41, 451; (2), 42, 305; C. R., 98, 1416; Jsb. Chem., 1884, 1436.

Detection in plants, wines, and cereals by oxidation to permanganic acids. Various procedures briefly mentioned.

- 1884: 15. MEINEKE, C. Bestimmung des Mangans durch Permanganat. (*Title from Rep. anal. Chem.*)
 - * Chem. Verst. Mitt., 1884, 63; Rep. anal. Chem., 5, 1; Jsb. Chem. 1884, 1596; Chem. Ztg., 9, 432; Ztschr. anal. Chem., 24, 423; Chem. Ind., 8, 86; Ber., 18, 125, Ref.; J. Am. Chem. Soc., 7, 91; Dingl. pol. J., 257, 202.

Comments on Guyard (1863: 2), Volhard (1879: 14), Morawski and Stingl (1878: 4), and Ledebur (1884: 11) methods of determination.

- 1884: 16. STONE, G. C. The Determination of Manganese in Spiegel.
 - Trans. Am. Inst. Min. Eng., 12, 295 and 514; School Mines Quart.,
 6, 24; J. Iron Steel Inst., 1884, a, 335; Eng. Min. J., 36, 228; 37, 138; Berg- u. hüttenm. Ztg., 42, 442.
 - Shows variation in results obtained by different chemists when working upon the same sample.

1884: 17. WIELAND, J. Ueber elektrolytische Bestimmungen.
 Ber., 17, 1611 and 2931; Jsb. Chem., 1884, 1542; Chem. News, 50, 211.

Electrolytic determination. See Classen, 1884: 4.

1884: 18. WOLFF, N. Eine maassanalytische Bestimmung des Mangans.

Stahl u. Eisen, 4, 702; J. Iron Steel Inst., 1885, a, 301; Wagner's Jsb., 31, 12; Berg- u. hüttenm. Ztg., 44, 20; Dingl. pol. J., 257, 199; Chem.-techn. Rep., 23, b, 186.

1885: I. BLOXAM, C. L. On the Detection of Iron, Aluminium, Chromium, Manganese, Cobalt, Nickel, Calcium, and Magnesiums (as Phosphates) in the Precipitate Produced by Ammonia.

Chem. News, 52, 109; Chem. Centrbl., 1885, 942; J. Chem. Soc., 48, 1264; Chem. Ind., 8, 324.

Detection by means of phosphates.

- 1885: 2. CHARPENTIER, P. Sur une nouvelle méthode d'analyse volumétrique, applicable aux essais des bioxydes de manganèse.
 - C. R., 101, 316; Chem. Centrbl., 1885, 715; Jsb. Chem., 1885, 1937; Chem. News, 52, 87; J. Iron Steel Inst., 1886, a, 401; Ztschr. anal. Chem., 28, 733; J. Chem. Soc. (Lond.), 48, 1162; Wagner's Jsb., 32, 345; Rep. anal. Chem., 5, 352; Dingl. pol. J., 259, 103; Chem.-techn. Rep., 24, b, 248; 25, a, 207; Chem. Ind., 9, 59.

Determination of peroxide by passage of chlorine evolved on solution in hydrochloric acid into ferrous sulphate solution, and titration for ferric iron with potassium hydroxide in presence of potassium sulphocyanide.

1885: 3. CHEEVER, B. W. Estimation of Manganese, Carbon, and Phosphorus in Iron and Steel.

> Trans. Am. Inst. Min. Eng., 14, 372; J. Iron Steel Inst., 1885, 736. Comments on the Williams method. (1881: 18)

1885: 4. CLASSEN, A. Bemerkungen zu der Antwort des Herrn Wieland.

Ber., 18, 168.

Determination by electrolysis. See 1884: 4 and 1884: 17.

1885: 5. CLASSEN, A. Quantitative Analyse durch Elektrolyse.
Ber., 18, 1793; Jsb. Chem., 1885, 1883; Bull. soc. chim. (2), 45, 893; Dingl. pol. J., 259, 93; Ztschr. anal. Chem., 25, 110.
Electrolytic separation from iron.

1885: 6. DEANE, L. M. Ferricyanide of Manganese. Chem. News, 51, 164 and 248; Jsb. Chem., 1885, 1937. Solubility of manganese ferricyanide. Note on an error in Fresenius' Qualitative Analysis. See 1885: 8.

Separation from iron by means of zinc oxide, and titration with permanganate in the presence of the iron precipitate.

1885: 7. DIEHL, W. Zur Bestimmung des Mangans.

Chem. Ind., 8, 206; Chem. Centrbl., 1885, 713; Jsb. Chem., 1885, 1934; J. Chem. Soc. (Lond.), 50, 101; Wagner's Jsb., 31, 89; Rep. anal. Chem., 5, 300; Dingl. pol. J., 258, 95; Chem. Ztg., 8, 1502; Chem.-techn. Rep., 24, b, 248.

Precipitation by bromine and ammonia, solution in hydrochloric acid and estimation of the iron by the amount of iodine liberated from potassium iodide.

1885: 8. DRAPER, C. N. Solubility of Manganese Ferricyanide in Hydrochloric Acid.

> Chem. News, 51, 226; Jsb. Chem., 1885, 1937. See 1885: 6.

1885: 9. HAMPE, W. Ein einfaches Verfahren zur Trennung des Zinks von allen Metallen seiner Gruppe.

Chem. Ztg., 9, 543; Chem. Centrbl., 1885, 603; Jsb. Chem., 1885, 1938; Ztschr. anal. Chem., 24, 588.

- Separation from zinc by precipitation with hydrogen sulphide in presence of formates.
- 1885: 10. HAMPE, W. Die maassanalytische Bestimmung des Mangans in Legirungen, Mineralien u. s. w. mittelst Kaliumchlorats.
 - Chem. Ztg., 9, 1083 and 1515; Berg- u. hüttenm. Ztg., 44, 328; Monit. scientif., 27, 1046; Chem. Centrbl., 1885, 714; Jsb. Chem., 1885, 1936; Ber., 18, 580, Ref.; J. Iron Steel Inst., 1885, b, 652; Wagner's Jsb., 31, 311; Rep. anal. Chem., 5, 299; Techn.-chem. Jahrb., 8, 17 and 83; J. Soc. Chem. Ind., 4, 690; Analyst, 10, 191.
 - Precipitation with potassium chlorate, solution of the precipitate, addition of ferrous sulphate, and titration with permanganate. See 1883: 4.
- 1885: 11. v. JÜPTNER, H. Eine neue Manganbestimmungsmethode auf gewichtsanalytischem Wege.
 - Chem. Ztg., 9, 692; Jsb. Chem., 1885, 1935; Ztschr. anal. Chem.,
 25, 217; J. Iron Steel Inst., 1885, a, 247; Wagner's Jsb., 31, 14;
 J. Soc. Chem. Ind., 4, 510; Dingl. pol. J., 257, 201; Analyst, 10,
 149; Chem.-techn. Rep., 24, 249.

Precipitation as manganous sulphide, after removal of the iron with barium carbonate, solution of the precipitate in acetic acid, evaporation and ignition to mangano-manganic oxide.

1885: 12. KALMANN, W., and SMOLKA, A. Ueber eine neue Methode zur Bestimmung des Mangans in Spiegeleisen, Ferromangan und den wichtigsten Erzen.

> Wien, Acad. Ber. (2 Abt.), 91, 49; Monatsh., 6, 65; Chem. Centrbl., 1885, 235; Jsb. Chem., 1885, 1936; Ztschr. anal. Chem., 24, 590;

Chem. News, 51, 230; J. Chem. Soc. (Lond.), 48, 690; Ber., 18, 198, Ref.; Bull. soc. chim. (2), 45, 356; Wagner's Jsb., 31, 17; Monit. scientif., 27, 1205; Oester. Ztschr. Berg- u. Hüttenw., 34, 130; J. Soc. Chem. Ind., 4, 419; Dingl. pol. J., 257, 204; Arch. Pharm., 223, 314; Iron, 25, 516; J. Iron Steel Inst., 1885, 248.

Fusion with sodium carbonate and borax, solution of the fused mass, addition of an excess of ferrous sulphate, and titration for this excess with permanganate.

1885: 13. LUNGE, G. Ueber die Analyse von übermangansauren Kali und Braunstein durch Wasserstoffsuperoxyd.

Ber., 18, 1872; J. Chem. Soc. (Lond.), 48, 1162; Chem. Ind., 8, 255;
J. Soc. Chem. Ind., 4, 613; Chem.-techn. Centr. Anz., 1885, 775;
Chem.-techn. Rep., 24, b, 248.

Addition of the permanganate to an excess of hydrogen peroxide, in the presence of sulphuric acid, and measurement of the evolved oxygen.

- 1885: 14. MATHESIUS, W. Eine neue gewichtsanalytische Manganbestimmung.
 - Chem. Ztg., 8, 1777; Ber., 1885, 34, Ref.; Ztschr., anal. Chem., 25, 116; Chem. News, 52, 224; 53, 74; Wagner's Jsb., 31, 18; Monit. scientif., 27, 982.

Separation from iron by means of zinc oxide and precipitation by the Wolff method. See 1883: 22.

1885: 15. MEINEKE, C. Zur Maassanalyse des Mangans.

Chem. Ztg., 9, 1478 and 1787; Dingl. pol. J., 257, 202; Rep. anal. Chem., 5, 389; Berg- u. hüttenm. Ztg., 45, 35.

Determination by weighing as sulphate, by titration with permanganate and antimony chloride, and by the chlorate method. See 1883: 10.

1885: 16. OSMOND. Méthode colorimétrique de dosage du manganèse.

Bull. soc. chim. (2), 43, 66; Chem. Centrbl., 1885, 234; Jsb. Chem., 1885, 1935; J. Iron Steel Inst., 1885, a, 275; J. Chem. Soc. (Lond.), 48, 690; Ztschr. anal. Chem., 25, 552; Ber., 18, 344, Ref.; Chem. Ind., 8, 119; Wagner's Jsb., 31, 15; Dingl. pol. J., 257, 201; Arch. Pharm., 223, 285; Chem.-techn. Rep., 24, 249.

- Oxidation by means of lead peroxide in the presence of metaphosphates and nitric acid.
- 1885: 17. REINHARDT, C. Ueber N. Wolff's gewichtsanalytische Manganbestimmung.

Stahl und Eisen, 5, 81; Ztschr. anal. Chem., 25, 114; Wagner's Jsb., 31, 14; Dingl. pol. J., 257, 201.

Separation from calcium in the presence of ammonium acetate. See 1883: 22.

1885: 18. REINHARDT, C. Ueber Manganbestimmung.

Stahl und Eisen, 5, 782; Wagner's Jsb., 32, 9; Berg.- u. hüttenm. Ztg., 45, 47; Chem. Ztg., 8, 15, Ref.; Chem.-techn. Rep., 24, b, 247.

Discussion of Wolff's gravimetric and volumetric methods. See 1883: 22; 1884: 18, and 1885: 20.

1885: 19. SCHLAGDENHAUFFEN. Ueber die maassanalytische Bestimmung des Mangans. (*Title from Chem. Centrbl.*)

> * J. de pharm. (5), 10, 337; Chem. News, 50, 249; Chem. Centrbl., 1885, 146; Jsb. Chem., 1885, 1936; Chem. Ztg., 8, 1767; J. Chem. Soc. (Lond.), 48, 442; J. Iron Steel Inst., 1886, 1020.

> Comments on the Leclerc method and historical discussion of the Lenssen, Guyard, Morawski and Stingl and Rössler methods.

1885: 20. WOLFF, N. Die maassanalytische Bestimmung des Mangans.

Stahl. u. Eisen, 5, 529; Jsb. Chem., 1885, 1935; J. Chem. Soc. (Lond.), 48, 1264; Wagner's Jsb., 31, 12; Dingl. pol. J., 259, 199; J. Iron Steel Inst., 1885, 301.

Titration with permanganate in the presence of ferric oxide.

1886: I. ATKINSON, R. W. Estimation of Manganese.

J. Soc. Chem. Ind., 5, 365 and 467; Dingl. pol. J., 262, 136; Monit. scientif., 28, 1043; Jsb. Chem., 1886, 1934; J. Chem. Soc. (Lond.), 52, 399; Wagner's Jsb., 32, 4; J. Anal. Chem., 1, 72.

Gravimetric determination in ores by use of ammonium acetate and bromine; also criticisms of Pattinson's Method. See 1879: 9, and 1886: 20.

- 1886: 2. BARLOW, J. J. A New Method of Precipitating and Estimating Manganese, also Iron indirectly, by means of Hydrogen Peroxide.
 - Chem. News, 53, 41; J. Iron Steel Inst., 1886, a, 392; Jsb. Chem., 1886, 1935; J. Chem. Soc. (Lond.), 50, 393; Ber., 19, 219, Ref.; Wagner's Jsb., 32, 4; Chem. Ztg., 10, 32, Ref.; Berg- u. hüttenm. Ztg., 45, 350; Dingl. pol. J., 261, 260; Techn.-chem. Jahrb., 8, 17; Chem.-techn. Rep., 25, a, 206.

Separation from zinc, cobalt, nickel, and alkaline-earths, also the simultaneous determination of iron and manganese. The mixed precipitates are ignited, weighed, treated with hydrochloric acid, and the chlorine generated absorbed in potassium iodide.

1886: 3. BEHRENS, T. H. On the Micro-chemical Analysis of Minerals.

> Chem. News, 54, 289. Detection by micro-chemical tests.

1886: 4. BEIN, S. Ueber die quantitative Abscheidung und Bestimmung des Zinks.

> Rep. anal. Chem., 1886, 275; Jsb. Chem., 1886, 1939. Separation from iron by the succinate method, and from zinc by treating the ignited oxides with acetic acid.

1886: 5. BLUM, L. Ueber die directe Trennung des Mangans von Eisen.

Ztschr. anal. Chem., 25, 519; Chem. Centrbl., 1887, 97; Jsb. Chem., 1886, 1934; Chem. News, 55, 236; Chem. Ztg., 11, 251; Ber., 19, 850, Ref.; J. Chem. Soc. (Lond.), 52, 183; Chem. Ind., 9, 383; Wagner's Jsb., 32, 344; Rep. anal. Chem., 6, 662; Analyst, 11, 234; J. Am. Chem. Soc., 9, 10; Dingl. pol. J., 262, 335.

Separation from iron by precipitation as ferrocyanide from ammoniacal tartrate solution.

1886: 6. CARNOT, A. Séparation successive du cuivre, du cadmium, du zinc et du nickel ou du cobalt (fer et manganèse).

Bull. soc. chim. (2), 46, 812; C. R., 102, 621 and 678; Eng. Min.
J., 41, 340; Jsb. Chem., 1886, 1948; Chem. News, 53, 196; Ztschr.
anal. Chem. (1889), 28, 344; Chem. Ztg., 11, 4, Ref.; Ber., 19, 364, Ref.; J. Chem. Soc. (Lond.), 50, 650; J. Iron Steel Inst., 1887, a, 470.

Separation from nickel and iron by means of hydrogen sulphide in acetic acid solution.

1886: 7. CHRISTENSEN, O. T. Beiträge zur Chemie des Mangans und des Fluors.

> J. prakt. Chem., 34, 41; 35, 161; Chem. News, 54, 96; 55, 153. Detection by the formation of permanganic acid on the electrolysis of the manganese salt in hydrofluoric acid solution.

1886: 8. CHEEVER, B. W. Colorimetric Estimation of Manganese in Steel.

Trans. Am. Inst. Min. Eng., 15, 102; J. Iron Steel Inst., 1885, b, 736; J. Anal. Chem., 1, 88.

Criticisms of the method of determination by means of lead peroxide and nitric acid.

- 1886: 9. CLASSEN, A. See 1884: 4.
- 1886: 10. CLASSEN, A. See 1884: 5.
- 1886: 11. CLASSEN, A., and LUDWIG, R. Quantitative Analyse durch Elektrolyse.

Ber., 19, 323; Jsb. Chem., 1886, 1894. Separation from mercury by electrolysis.

1886: 12. DEANE, L. M. On the Separation of Silica in the Estimation of Manganese in Pig Iron, and On the Estimation of Phosphorus in Pig Iron and Steel.

Chem. News, 54, 174; Jsb. Chem., 1886, 1932; Ber., 19, 851, Ref.;
J. Chem. Soc. (Lond.), 52, 183; Chem.-techn. Centr. Anz., 1887, 187; Chem.-techn. Rep., 26, a, 295.

Separation from iron by the basic acetate method, and final separation from silica after igniting and weighing the manganese oxide.

1886: 13. HUNT, A. E. The Estimation of Manganese in Iron and Steel by the Color Method.

Trans. Am. Inst. Min. Eng., 15, 104; J. Iron Steel Inst., 1886, b, 1020; J. Anal. Chem., 1, 89.

Oxidation with lead peroxide in the presence of nitric acid.

1886: 14. LANGBEIN. E. Zur Nickelanalyse.

Rep. anal. Chem., 1886, 423; Jsb. Chem., 1886, 1937; Dingl. pol. J., 261, 495.

Separation from nickel by electrolysis.

1886: 14a. LÖSEKANN, G., and MEYER, T. Eine neue Methode der Zinkbestimmung.

Chem. Ztg., 10, 729.

Separation from zinc by means of phosphate impracticable.

- 1886: 15. MEINEKE, C. Eine Methode schneller Bestimmung des Mangans in Eisensorten mittels Permanganat.
 - Rep. anal. Chem., 6, 252; Jsb. Chem., 1886, 1933; Ber., 19, 464, Ref.; Chem. Ind., 9, 194; Stahl u. Eisen, 6, 444; Wagner's Jsb., 32, 5; Berg- u. hüttenm. Ztg., 46, 43; J. Soc. Chem. Ind., 5, 508; Techn.-chem. Jahrb., 9, 18; Chem.-techn. Rep., 25, a, 205.

Stahl u. Eisen, 6, 98; Wagner's Jsb., 32, 5; Berg- u. hüttenm. Ztg., 45, 349; Techn.-chem. Jahrb., 8, 17; J. Iron Steel Inst., 1886, 392.
Separation from iron by the basic acetate method, precipitation by bromine, and subsequently as manganous carbonate; or, separation by the acetate method, and re-precipitation with hydrogen peroxide and determination as mangano-manganic oxide.

Stahl. u. Eisen, 6, 590; Chem. Ztg., 11, 216, Rep.; Wagner's Jsb., 32, 5; Rep. anal, Chem., 6, 595; Berg- u. hüttenm. Ztg., 45, 452; J. Iron Steel Inst., 1886, 1022; Chem.-techn. Rep., 25, a, 206.
Comments on the Meineke method. (1883:10.)

Use of permanganate method (Meineke) with antimonious chloride. See 1883: 10.

^{1886: 16.} MÜLLER, C. G. Ueber eine schnelle und scharfe Methode zur gewichtsanalytischen Bestimmung des Mangans im Spiegeleisen und Ferromangan.

^{1886: 17.} MÜLLER, C. G. Ueber die neue Meinekesche Manganbestimmung.

1886: 18. MOORE, T. Quantitative Chemical Analysis by Electrolysis.

Chem. News, 53, 209; Jsb. Chem., 1886, 1895; J. Chem. Soc. (Lond.), 50, 921; Eng. Min. J., 41, 371.

Electrolytic deposition and estimation as oxide.

1886: 19. MOORE, T. On the Estimation of Nickel in Mattes, Ores, Slags, etc.

Chem. News, 54, 300; Jsb. Chem., 1886, 1938; Ztschr. anal. Chem., 26, 732.

Separation from nickel by electrolysis.

- 1886: 20. PATTINSON, J. The Volumetric Test for Manganese.
 J. Soc. Chem. Ind., 5, 422; Monit. scientif., 28, 1048; Jsb. Chem., 1886, 1934.
 Reply to the criticisms of Atkinson. See 1886; 1.
- 1886: 21. PERILLOU. Dosage rapide du carbon, du phosphore, etc.
 Bull. soc. ind. mineral. (2), 13, 108; Berg- u. hüttenm. Ztg., 45, 6, and 32; Jsb. Chem., 1886, 1933; Ber., 1886, 181, Ref.; Wagner's Jsb., 32, 11.

Volumetric method; oxidation by lead peroxide, and titration with ferrous sulphate in nitric acid solution.

1886: 22. REINHARDT, C. Gewichtsanalytische Manganbestimmung. Chem. Ztg., 10, 323, 357, and 372; Berg- u. hüttenm. Ztg., 45, 163; J. Soc. Chem. Ind., 5, 391.

> Separation from iron by the basic acetate method, and precipitation by (1) a current of bromine-ammonia, or (2) bromine in hydrochloric acid, and a current of gaseous ammonia.

1886: 23. REINHARDT, C. Ueber Mangantitrationsmethoden nach N. Wolff, E. Belani, Hampe und Meineke.

Stahl. u. Eisen, 6, 150; Wagner's Jsb., 32, 9; Berg- u. hüttenm.
Ztg., 45, 192; J. Iron Steel Inst., 1886, 393.
See 1884: 18; 1883: 4; 1883: 10; 1887: 17.

- 1886: 24. SCHÖFFEL, R., and DONATH, E. Ueber die volumetrische Bestimmung des Mangans.
 - Monatsh. Chem., 7, 639; Wein. Acad. Ber. (2 Abth.), 94, 844; Stahl u. Eisen, 7, 30; Dingl. pol. J., 264, 34; Berg- u. hüttenm. Ztg., 1887, 60; Oester. Ztschr. Berg- u. Hüttenw., 35, 70; Chem. Centrbl., 1887, 152 and 285; Jsb. Chem., 1887, 2429; J. Chem. Soc. (Lond.), 52, 399; Ztg. angew. Chem., 1887, a, 159; Chem. Ztg., 11, 111, Rep.; Ber., 20, 115, Ref.; Bull. soc. chim. (2), 49, 48; Chem. Ind., 10, 230 and 279; Wagner's Jsb., 33, 273; J. Iron Steel Inst., 1887, a, 468; Techn.-chem. Jahrb., 9, 18 and 98; Chem.-techn. Rep., 26, a, 296.
 - Volumetric method; addition of an excess of permanganate and titration with arsenious acid.

1886: 25. SETTERWALL, A. Om bestämmande i jern och stöl m. m. af dels mangan förmedelst titrering.

Jern.-Kont. Ann., 1886, 427; *Z. O. S. Berg. u. H. V., 25, 410; Techn.-chem. Jahrb., 9, 19.

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Separation from cobalt by precipitation as sulphide after the formation of cobaltic cyanide by the addition of a solution of hydrocyanic acid.

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1887: 12. v. KNORRE, G. Ueber die Verwendbarkeit des Nitroso- β napthols in der quantitativen Analyse.

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J. Iron Steel Inst., 1887, a, 470.
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- 1887: 13. LAX, E. Beiträge zur Maassanalyse des Mangans. (Title from Berg- u. hüttenm. Ztg.)
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 - Chem. Ztg., 11, 137; Chem. Centrbl., 1887, 230; J. Chem. Soc. (Lond.), 52, 531; Berg- u. hüttenm. Ztg., 46, 135; J. Soc. Chem. Ind., 6, 456; Techn.-chem. Jahrb., 10, 16.
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Separation from iron by zinc oxide, precipitation by bromine in the presence of zinc oxide, solution in ferrous sulphate, and titration with potassium permanganate.

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J. Anal. Chem., 2, 291; Ztschr. anal. Chem., 30, 616; Chem. Ztg., 12, 250, Rep.; J. Chem. Soc. (Lond.), (1892) 62, 103.
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Precipitation by means of hydrogen peroxide, solution in oxalic acid, and titration with potassium permanganate.

- 1888: 3. FRIEDMANN, A. Zur Bestimmung des Mangans in Eisen.
 - Stahl u. Eisen, 8, 315; Jsb. Chem., 1888, 2553; Ztschr. angew. Chem., 1888, 415; Wagner's Jsb., 34, 196; Techn.-chem. Jahrb., 11, 19; J. Iron and Steel Inst., 1888, b, 328.

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Precipitation with chlorate, solution without filtration with the aid of ferrous sulphate or oxalic acid, and titration with permanganate.

1888: 8. KLEIN, J. Ueber einige neue Reactionen.

Chem. Ztg., 12, 1321; Berg- u. hüttenm. Ztg., 47, 425; Arch. Pharm., 227, 77.

Test for manganese by means of hydrogen peroxide in alkaline solution.

1888: 9. DE KONINCK, L. L., and LECRENIER, A. Bestimmung des verfügbaren Sauerstoffs in den Hyperoxyden mittels gasförmigen Salzsäure.

Ztschr. angew. Chem., 1888, 353; Wagner's Jsb., 34, 515; Eng. Min. J., 47, 460; Berg- u. hüttenm. Ztg., 1888, 295; Chem.-techn. Rep., 27, a, 235.

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- 1888: 12. OETTEL, F. Ueber die Analyse des Neusilbers.

Ztschr. anal. Chem., 27, 16; Jsb. Chem., 1888, 2554. Separation from iron and cobalt by electrolysis; precipitation by bromine, and determination as mangano-manganic oxide.

- 1888: 13. REINHARDT, C. Zur Bestimmung des Mangans in siliciumreichen aber manganarmen Roheisensorten.
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- 1888: 14. v. REIS, M. A. Vorschläge zur Einführung von einheitlichen analytischen Methoden für Eisenhüttenlaboratorien.

Stahl. u. Eisen, 8, 97; Techn.-chem. Jahrb., 10, 16.

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1888: 16. SCHÜRMANN. Ueber die Verwandtschaft der Schwermetalle zum Schwefel.

Ann. Chem. (Liebig), 249, 329.

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Chem. Ztg., 12, 446; Chem. Centrbl., 1888, 645; J. Chem. Soc. (Lond.), 56, 188; Chem.-techn. Rep., 27, a, 257.

Determination in the ashes of food-stuffs by treatment with sodium nitrate, sulphuric acid, and lead peroxide, and titration for the permanganate formed with a standard ferrous salt solution. See 1888: 20.

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 - J. Chem. Soc. (Lond.), 53, 182; Jsb. Chem., 1888, 2552; Ztschr. anal. Chem., 32, 367; Chem. Ztg., 12, 92, Rep.; Ber., 21, 374, Ref.; J. Iron Steel Inst., 1888, b, 329; J. Anal. Chem., 2, 197; Techn.-chem. Jahrb., 10, 16; Chem. News, 57, 48.

Oxidation with lead peroxide, and titration with ammonium oxalate. See 1871: 2.

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J. Chem. Soc. (Lond.), 53, 179; Chem. Ztg., 12, 155.

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- 1888: 20. WEISSMANN, G. Kurze Methode der Manganbestimmung im Roheisen, Stahl, etc.
 - Chem. Ztg., 12, 205; Chem. Centrbl., 1888, 423; Ber., 1888, 311, Ref.; Ztschr. anal. Chem., 32, 366; Chem. Ind., 11, 212; Wagner's Jsb., 34, 198; J. Chem. Soc. (Lond.), 54, 992; J. Iron and Steel Inst., 1888, a, 377; 1893, b, 531; Berg- u. hüttenm. Ztg., 47, 113; J. Soc. Chem. Ind., 7, 235; Dingl. pol. J., 267, 528; Techn.-chem. Jahrb., 10, 17; Chem.-techn. Rep., 27, a, 234.

Modification of the Chatard method. Oxidation to permanganic acid by lead peroxide and titration with ferrous ammonium sulphate. See 1871: 2 and 1888: 17.

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> Iron, 30, 360 and 504; Stahl und Eisen, 8, 607; Wagner's Jsb., 35, 173. Separation from iron by the acetate method, precipitation by bromine, and determination as mangano-manganic oxide, as manganous sulphate, or by difference after determination of the iron.

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Chem. Ztg., 13, 1339; Chem. Centrbl. 1889, b, 859; Jsb. Chem., 1889, 2399; J. Chem. Soc. (Lond.), 58, 419; Stahl u. Eisen, 9, 961; Berg- u. hüttenm. Ztg. 48, 429; J. Anal. Chem., 3, 425; Techn.-chem. Jahrb., 12, 13.

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1889: 2. BLUM, L. Eine Fehlerquelle bei der Trennung geringer Manganmengen von viel Kalk durch Schwefelammonium.

Ztschr. anal. Chem., 28, 454; Chem. Centrbl., 1889, b, 513; J. Chem. Soc. (Lond.), 56, 1087; Ber., 22, 706, Ref.; Chem. Ind., 13, 89; Stahl u. Eisen, 9, 960; School Mines Quart., 11, 69; J. Iron Steel Inst., 1890, a, 372; J. Soc. Chem. Ind., 8, 922; Analyst, 14, 192.

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- 1889: 5a. GOOCH, F. A. and WHITFIELD, J. E. Analyses of Waters of the Yellowstone National Park.

Bull. U. S. Geol. Surv., No. 47, 27. Determination in mineral waters.

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Arch. Pharm. (3), 27, 77; Chem. Centrbl., 1889, a, 391; Jsb. Chem., 1889, 2398; Chem. Ztg., 13, 83, Rep.; Ber., 22, 171, Ref.; J. Chem. Soc., 56, 653; Berg- u. hüttenm. Ztg., 48, 164; Chem. -techn. Rep., 28, a, 240.

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1889: 7. KOHN, C. J., and WOODGATE, J. The Application of Electrolysis to Quantitative Analysis.

J. Soc. Chem. Ind., 8, 256; Chem. Centrbl., 1889, b, 54; Jsb. Chem., 1889, 2304.
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1889: 8. DE KONINCK, L. L. Zur Prüfung der Reagentien.

Ztschr. angew. Chem., 1889, 4; Jsb. Chem., 1889, 2299; Rev. univers. des mines, 1889, 308; Berg- u. hüttenm. Ztg., 48, 183; Chem. News, 59, 230.

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Determination of peroxides by use of standard solutions of bichromate and ferrous sulphate.

- 1889: 11. McCulloch, N. The Volumetric Estimation of Cobalt in Presence of Nickel, Manganese, and other Metals.
 - Chem. News, 59, 51; Chem.-techn. Rep., 29, 240; Chem. Ztg., 13, 38, Rep.

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> Criticism of Carnot's article on the action of hydrogen peroxide on the salts of manganese.

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 - Chem. Ztg., 13, 1220; Chem. Centrbl., 1889, 811; Jsb. Chem., 1889, 2407; Ztschr. anal. Chem., 30, 340; Ber., 22, 711, Ref.; J. Soc. Chem. Ind., 9, 108; J. Anal. Chem., 3, 429.

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1889: 14. NEUMANN, G. Quantitative Bestimmung von Zink neben Mangan.

Ztschr. anal. Chem., 28, 57; Jsb. Chem., 1889, 2408; J. Chem. Soc. (Lond.), 56, 549; J. Soc. Chem. Ind., 8, 62; J. Anal. Chem., 4, 69; Chem. Ztg., 23, 38, Rep.

Separation of zinc as sulphide from a formic acid solution.

1889: 15. RADAU, C. Zur Kenntniss vanadinsaurer Salze.

Ann. Chem. (Liebig), 251, 154.

Separation from vanadium by fusion with sodium carbonate, solution of the fused mass in water, precipitation of hydrated manganese peroxide with alcohol, re-solution in hydrochloric acid, and re-precipitation with hydrogen peroxide and ammonia.

1889: 16. REITMAIR, O. Kalkbestimmung bei Gegenwart von Phosphorsäure, Eisen, Thonerde und Mangan.

> Ztschr. angew. Chem., 1889, 358; Ztschr. anal. Chem., 31, 314. Separation from calcium by means of oxalates.

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Oester. Ztschr. Berg- u. Hüttenw., 36, 608; Chem. Centrbl., 1889, a, 64; Berg- u. hüttenm. Ztg., 48, 153; Chem.-techn. Centrl. Anz., 1889, 64; Pharm. Centrh., 30, 189; Chem.-techn. Rep. 28, a, 240.

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School Mines Quart., 11, 355; Jsb. Chem., 1890, 2442; J. Soc. Chem. Ind., 9, 973.
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1890: 4. CARNOT, A. (Discussion.)

Bull. soc. chim. (3), 3, 594; Chem. Ztg., 14, 637; Oester. Ztschr. Berg- u. Hüttenw., 38, 450.

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Ztschr. anal. Chem., 29, 28; Jsb. Chem., 1890, 2440. Determination in chromite.

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 61, 1138; J. Iron Steel Inst., 1891, b, 326; School Mines Quart., 12, 262; Berg- u. hüttenm. Ztg., 50, 186; Eng. Min. J., 51, 635; J. Soc. Chem. Ind., 10, 387; Chem.-techn. Rep., 31, c, 268.
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 J. Anal. Chem., 5, 368 and 374.
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 Ztschr. anal. chem., 31, 69; Chem. Ztg., 15, 142, Rep.; Ber. 24, 675, Ref.; School Mines Quart., 12, 337; 13, 75; Eng. Min. J., 52, 386; J. Soc. Chem. Ind., 10, 659; J. Anal. Chem., 5, 659.
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Wagner's Jsb., 37, 209; Berg- u. hüttenm. Ztg., 50, 185; J. Anal. chem., 5, 237; Chem.-techn. Rep., 31, a, 267.

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Use of Volhard method of determination (1879: 14).

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Ztschr. angew. Chem., 1891, 476. Determination in commercial aluminum.

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 - Iron Age, 47, 528; J. Iron Steel Inst., 1891, a, 443; 1892, a, 491; Stahl u. Eisen, 11, 927; Wagner's Jsb., 37, 147.

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 - Chem. Z^{*}g., 15, 882; Jsb. Chem., 1891, 2481; Chem. Centrbl., 1891,
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 Chem.-techn. Rep., 31, b, 297; Berg- u. hüttenm. Ztg., 50,
 390; Analyst, 16, 180.

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Chem. Ztg., 15, 149 and 186; Jsb. Chem., 1891, 2481; Chem. Centrbl., 1891, a, 470; J. Chem. Soc. (Lond.), 62, 916; Ztschr. anal. Chem., '30, 242; Wagner's Jsb., 37, 140; J. Iron Steel Inst., 1891, b, 326. Modification of Volhard procedure. Addition of sodium sulphate, and titration without filtration. See 1891: 25.

1891: 27. SMITH, E. F. The Electrolysis of Metallic Phosphates in Acid Solution.

Am. Chem. J., 13, 206; Jsb. Chem., 1891, 2400. Electrolytic separation from cadmium.

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Monit. scientif. (4), 6, 639; Chem. Centrbl., 1892, b, 632; Ztschr. angew. Chem., 1892, 704; Wagner's Jsb., 38, 118; J. Iron Steel Inst., 1893, a, 403; Berg- u. hüttenm. Ztg., 52, 59; J. Soc. Chem. Ind., 11, 1037.

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Modification of the Volhard method. Comments on Rürup procedure. See 1879: 14 and 1891: 26.

- 1892: 3. BLUM, L. Zur volumetrischen Bestimmung des Zinks.
 Ztschr. anal. Chem., 31, 60; Berg- u. hüttenm. Ztg., 51, 164.
 Separation from zinc by means of bromine in alkaline solution.
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Rev. chim. indust., 3, 298.

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Detection by means of sodium hypobromite.

1892: 7. DONATH, E. Bemerkungen zur Vereinbarung einheitlicher Untersuchungsmethoden für Eisen und Stahl.

> Chem. Ztg., 16, 141. Criticism of report of Chemiker-Commission. See 1891: 5 and 1892: 5.

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Separation from iron by the acetate method.

- 1892: 12. v. REIS, M. A. Zur Bestimmung von Mangan im Eisen.
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Oxidation of organic matter by the use of barium peroxide. Volumetric determination by the Volhard procedure. See 1887: 4 and 1879: 14.

- 1892: 13. v. REIS, M. A. Ueber Bestimmung des Mangans nach der Chloratmethode.
 - Chem. Ztg., 15, 1791; Stahl u. Eisen, 12, 28; J. Chem. Soc. (Lond.), 62, 1132.

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- 1892: 15. ROTHE, J. W. Trennung des Eisens von anderen Elementen nach einem neuen Verfahren.

Mitth. Kgl. Verst. zu Berlin, 10, 132; Iron, 40, 404; J. Iron Steel Inst., 1892, b, 510.

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- 1892: 16. RUBRICIUS, H. Nachtrag zur modificierten Volhard-'schen Manganprobe.
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- 1892: 20. VAN GRUNDY, C. P. Note on Textor's Rapid Method for the Determination of Manganese.

Proc. Eng. Soc. Western Penna., 8, 158; J. Iron Steel Inst., 1892, b, 512.

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 Bull. soc. chim. (3), 9, 214; Chem. Centrbl., 1893, b, 156; Chem. News, 68, 15 and 301.
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Bull. soc. chim. (3), 9, 646; C. R., 116, 1295; Chem. Centrbl., 1893,
b, 191; J. Chem. Soc. (Lond)., 64, b, 497; J. Soc. Chem. Ind.,
12, 897; Ztschr. anorg., Chem., 6, 81.

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Determination by measurement of the gas evolved when manganese dioxide is titrated with hydrogen peroxide. See 1885: 13, 1890: 1, and 1893: 3.

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Comments on the procedures of Guyard and Donath. See also 1893: 1, 1863: 2, and 1881: 6.

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> Ztschr. anorg. Chem., **3**, 193. Detection by means of sodium peroxide.

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Bull. soc. chim. (3), 9, 248; J. Chem. Soc. (Lond.), 64, b, 498; School Mines Quart., 14, 362; Ztschr. anorg. Chem., 4, 479.

Precipitation by means of sodium carbonate, solution in nitric acid, precipitation by potassium chlorate, and determination as phosphate, sulphide, or oxide; also volumetrically by methods of Pattinson, Guyard, Gay-Lussac, and Campredon. Separation from iron by the acetate method. See also 1893: 8.

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J. Am. Chem. Soc., 15, 113; Chem. Centrbl., 1893, b, 393.

Precipitation by means of potassium chlorate, and titration with hydrogen peroxide

1893: 11. KOSMAN, B. Zur Trennung von Eisen und Aluminium, Mangan, Zink und Calcium.

Stahl u. Eisen, 13, 431; Chem. Centrbl., 1893, b, 155; J. Chem. Soc. (Lond.), 64, b, 600; Ztschr. anorg. Chem., 4, 397.

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> J. Am. Chem. Soc., **19**, 547; Chem. Centrbl., **1897**, b, 538. Volumetric determination as ferrocyanide.

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Gasometric determination, with the aid of hydrazine-sulphate.

1897: 15. SCHNEIDER, L. Ein Beitrag zu den vergleichenden Manganbestimmungen in Stahl und Eisen von L. Rürup.

Chem. Ztg., 21, 41; Chem. Centrbl., 1897, a, 436; J. Iron Steel Inst., 1898, a, 534; J. Chem. Soc. (Lond.), 74, b, 94; Analyst, 22, 110.
Comments on Rūrup's comparative determinations. See 1896: 14.
Reference to colorimetric methods by oxidation to permanganic acid.

1897: 16. STONE, G. C., and VAN INGEN, D. A. The Ferrocyanides of Zinc and Manganese.

J. Am. Chem. Soc., 19, 542; Chem. Centrbl., 1897, b, 538. Volumetric determination as ferrocyanide.

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> J. Am. Chem. Soc., 19, 434; J. Chem. Soc. (Lond.), 74, b, 54. Separation from iron by means of sodium nitrite.

1898: 1. AUSTIN, M. On the Estimation of Manganese Separated as the Carbonate.

> Am. J. Sci. (4), 5, 382; Ztschr. anorg. Chem., 17, 272; Chem. Centrbl., 1898, b, 65; Chem. News, 77, 243; 78, 239; Ztschr. angew. Chem., 1898, 581 and 1131; Chem. Ztg., 22, 212, Rep.; Wagner's Jsb., 44, 121; Chem.-techn. Rep., 37, 610; J. Chem. Soc. (Lond.), 74, b, 646; School Mines Quart., 20, 303.

1898: 2. BIALOBSZCSKI, M. Die Anwendung saurer Lösungen von arseniger Säure in der Maassanalyse. (*Title from Ztschr. anal. Chem.*).

> * Pharm. Ztschr. f. Russland, **35**, 785; Ztschr. anal. Chem., **37**, 445. Determination of manganese peroxide by the use of arsenious acid.

1898: 3. BREARLEY, H. Separations from Chromic Acid. II. The Separation of Manganese.

Chem. News, 77, 131; Chem. Centrbl., 1898, a, 961; J. Chem. Soc., 74, b, 409.

Separation from chromic acid by the use of sodium carbonate in cold solution and by the use of sodium hydrogen phosphate.

1898: 4. CAMPREDON, L. Sur le dosage rapide des principaux éléments des produits sidérurgiques.

Rev. chim. indust., 9, 306.

Comparison of the Schneider method with the Volhard method. See 1885: 15, and 1879: 14.

1898: 4a. DENIGES. Réactions de quelques métaux de groupe du fer en milieu glycériné.

* Bull. soc. pharm. Bordeaux, 1898, 97; Ann. chim. anal. appl., 3, 230.

Detection in the presence of cobalt and nickel by means of reactions occurring in alkaline solutions containing glycerine.

Precipitation as carbonate, and determination as the oxide or the sulphate.

1898: 5. ENGEL, C. Analyse électrolytique, dosage des métaux precipitables par le sulphure ammonique.

L'Éclairage electrique, 14, 106; Chem. Centrbl. 1898, b, 557; Ztschr. Elektrochem., 5, 37; School Mines Quart., 20, 302; J. Soc. Chem. Ind., 17, 796.

Electrolytic separation from iron, cobalt, and nickel.

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 - J. Am. Chem. Soc., 20, 504; Chem. Centrbl., 1898, b, 508; Chem. Ztg., 22, 198, Rep.; Bull. soc. chim. (3), 22, 9; Chem.-techn. Rep. 37, 610; Wagner's Jsb., 44, 121; J. Chem. Soc. (Lond.), 74, b, 540; School Mines Quart., 20, 303; J. Soc. Chem. Ind., 17, 796; Analyst, 23, 303.

Use of hydrofluoric acid to hold the silica in solution when precipitating manganese peroxide by the Williams method (1881: 18).

- 1898: 7. GOOCH, F. A., and AUSTIN, M. The Estimation of Manganese as the Sulphate and as the Oxide.
 - Am. J. Sci. (4), 5, 209; Ztschr. anorg. Chem., 17, 264; Chem. Ztg.,
 22, 212, Rep.; Chem. Centrbl., 1898, b, 1150; J. Chem. Soc. (Lond.), 74, b, 646; J. Iron Steel Inst., 1898, b, 558; Chem. News,
 77, 255; Bull. soc. chim. (3), 20, 694; Wagner's Jsb., 44, 121; Eng. Min. J., 65, 585; Chem.-techn. Rep., 37, 610.
 Determination as sulphate and as oxide.
- 1898: 8. GOOCH, F. A., and AUSTIN, M. On the Determination of Manganese as the Pyrophosphate.
 - Am. J. Sci. (4), 6, 233; Ztschr. anorg. Chem., 18, 339; Chem. Centrbl., 1899, a, 378; Chem. News, 78, 239 and 246; Chem. Ztg., 22, 319, Rep.; Bull. soc. chim. (3), 22, 197; Wagner's Jsb., 44, 122; J. Chem. Soc. (Lond.), 76, 128; School Mines Quart., 20, 303 and 400; Analyst, 24, 52.

Study of the proper conditions for precipitation as phosphate.

- 1898: 9. GOOCH, F. A., and AUSTIN, M. On the Condition of Oxidation of Manganese Precipitated by the Chlorate Process.
 - Am. J. Sci. (4), 5, 260; Ztschr. anorg. Chem., 17, 253; Chem. Centrbl., 1898, a, 1203; Ztschr. angew. Chem., 1898, 664; Chem. News, 77, 269 and 279; Chem. Ztg., 22, 212, Rep.; Wagner's Jsb., 44, 121; J. Chem. Soc. (Lond.), 74, b, 645; School Mines Quart., 20, 303; J. Soc. Chem. Ind., 17, 796.
 - Precipitation with sodium chlorate and solution of the oxide in a sulphuric acid and potassium iodide solution, and titration for the liberated iodine with thiosulphate; or reduction of the oxide with arsenious acid and titration of the excess of that reagent.

- 1898: 10. DE GRAMONT, A. Analyse spectrale de quelques minéraux non conducteurs par les sels fondus et réactions des éléments.
 - C. R., 126, 1513; J. Chem. Soc. (Lond.), 74, b, 636. Detection by spectrum analysis.
- 1898: 11. HILLEBRAND, W. F. See 1897: 8a.
- 1898: 12. JANNASCH, P., and ALFFERS, F. Ueber quantitative Metalltrennungen in ammoniakalischer und saurer Lösung durch Hydroxylamin und durch Hydrazin. (II) Die Trennung des Quecksilbers von Molybdän und Wolfram, sowie von den Metallen der Schwefelammoniumgruppe.

Ber., 31, 2383; J. Chem. Soc. (Lond.), 76, b, 60. Separation from mercury.

1898: 13. KAEPPEL, F. Zur quantitative Bestimmung des Mangans und Trennung des Eisens von Mangan durch Elektrolyse.

Ztschr. anorg. Chem., 16, 268; Chem. Centrbl., 1898, a, 962; Ztschr. angew. Chem., 1898, 435; Chem. Ztg., 22, 118, Rep.; Chem. News, 77, 201; 79, 195; Bull. soc. chim. (3), 22, 811; Chem.-techn. Rep., 37, 287 and 610; Oester. chem. Ztg., 1, 13; Ztschr. Elektrochem., 1898, 41; Wagner's Jsb., 44, 287; J. Chem. Soc. (Lond.), 74, b, 354; School Mines Quart., 19, 430; 20, 400; J. Soc. Chem. Ind., 17, 605; Analyst, 23, 221; Dingl. pol. J., 310, 16.

Deposition from faintly acid solution partly as metal and partly as peroxide.

1898: 14. LEHNKERING, P. Untersuchung von Eisenerzen. (Title from Wagner's Isb.).

* Ztschr. öffenlich. Chem., 1898, 459; Wagner's Jsb. 44, 120; J. Soc. Chem. Ind., 17, 951; J. Chem. Soc. (Lond.), 76, 251.

Recommends the Volhard-Wolff method for determination in ores. Comments on the Hampe method. See 1884: 18, and 1883: 4.

- 1898: 15. MURMANN, E. Bemerkungen zur Bestimmungen des Zinks und Mangans als Sulfid.
 - Wien. Akad. Ber. (2b), 107, 434; Monatsh. Chem., 19, 404; Chem. Centrbl., 1898, b, 1035; Analyst, 24, 51; Ann. chim. anal. appl., 4, 203.; J. Soc. Chem. Ind., 17, 1186; J. Chem. Soc. (Lond.), 76, 126; Chem. News, 81, 60.

Addition of mercuric chloride, precipitation of mercuric sulphide and manganous sulphide together, and expulsion of the mercuric sulphide by the ignition of the precipitate, in presence of hydrogen, in a special form of crucible.

Oester. chem. Ztg., 1, 383; Chem. Centrbl. 1898, b, 1282; School Mines Quart., 20, 303.

Gravimetric determination as sulphide.

^{1898: 16.} MURMANN, E. Bemerkungen zur Analyse von Schmiedeeisen.

- 1898: 17. NOTHOMB, M. Apparat zur Werthbestimmung des Braunsteins.
 Chem. Ztg., 22, 80; Chem. Centrbl., 1898, a, 631; Analyst, 23, 111.
 Determination by loss of weight on treatment with oxalic acid.
- 1898: 18. PICHARD, P. Recherche et dosage rapide du manganèse dans les plantes et les terres végétates par une méthode colorimétrique.
 - C. R., 126, 550; Chem. Centrbl., 1898, a, 753; Chem. News, 77, 108; Chem.-techn. Rep., 37, 286; J. Soc. Chem. Ind., 17, 273; Ann. chim. anal. appl., 3, 123.

Oxidation in a nitric acid solution by means of lead peroxide.

1898: 19. PICHARD, P. Contribution à la recherche du manganèse dans les minéraux, les végétaux et les animaux.

C. R., 126, 1882; Chem. Centrbl., 1898, b, 381; J. Soc. Chem. Ind., 17, 807; School Mines Quart., 19, 429; J. Chem. Soc. (Lond.), 76, 40.

Detection by colorimetric test. See also 1898: 18.

1898: 20. VITALI, D. Ueber den Nachweis des Mangans. (Title from Chem. Centrbl.).

* Boll. chim. Farm., 37, 545; Chem. Centrbl., 1898, b, 942; J. Chem. Soc. (Lond.), 76, 251; Ann. chim. anal. appl., 3, 408.
Detection by the use of bromates in a sulphuric acid solution.

1898: 21. WOLMAN, L. Beitrag zur quantitativen Elektrolyse von Schwermetallen.

Ztg. Elektrochem., 3, 537; J. Chem. Soc. (Lond.), 74, b, 50.
Influence of an oxalate, pyrophosphate, or acetic acid on the results obtained by electrolytic deposition from nitric-acid solution.

1899: I. BREARLEY, H. The Estimation of Manganese by means of Potassium Permanganate.

> Chem. News, **79**, 47 and 83. A query on Mr. Daw's article on the Volhard process (1899; 4).

- 1899: 2. BREARLEY, H. Iron Separations with Alkaline Salts. Chem. News, 79, 193; J. Chem. Soc. (Lond.), 76, 815. Separation from iron by the acetate method.
- 1899: 3. BREARLEY, H. A Bibliography of Steel Works Analysis. Chem. News, 80, 233, 245, 257, 271.

A compilation of references from the Chemical News, 1860-1899, Journal of the Chemical Society (London), 1885-1898, and the Journal of the Iron and Steel Institute, 1880-1899, bearing on manganese in its relations to iron and steel analysis.

1899: 4. DAW, F. W. The Estimation of Manganese by Means of Potassium Permanganate.

Chem. News, 79, 25, 58, and 104; Chem. Centrbl., 1899, a, 504;
Ztschr. angew. Chem. 1899, 279; Chem. Ztg., 23, 44, Rep.; Bull. soc. chim. (3), 22, 443; J. Iron Steel Inst. 1899, a, 465; Chem. -techn. Rep., 38, 272; Wagner's Jsb., 45, 130; School Mines Quart., 20, 302; Analyst, 24, 110; J. Chem. Soc. (Lond.), 76, 334.
Criticism of the Volhard method (1879: 14).

1899: 5. DUNNINGTON, F. P. Composition of Manganese Pyrophosphate.

Chem. News, 79, 275.

A notice of an error in "Fresenius Quantitative Analysis" (1876, Vol. 1).

1899: 6. FERNBERGER, H. M., and SMITH, E. F. The Electrolysis of Metallic Phosphate Solutions.

> J. Am. Chem. Soc., 21, 1001. Electrolytic separation from copper.

1899: 7. FRIEDHEIM, C., and BRÜHL, E. Kritische Studien ueber die Anwendung des Wasserstoffsuperoxyds in der quantitativen Analyse.

Ztschr. anal. Chem., 38, 686; J. Soc. Chem. Ind., 19, 170.

Criticism of the work done by Jannasch regarding the use of hydrogen peroxide for the separation from copper, zinc, nickel, and chromium. See 1895: 10, 11, and 14, and 1896: 8 and 9.

1899: 8. HERTING, O. Beitrag zur Bestimmung des Kohlenstoffs, des Kupfers und Mangans im Eisen.

Ztschr. angew. Chem. 1899, 1193; ,Chem. Centrbl., 1900, a, 226; Wagner's Jsb., 45, 129.

Preference given to the Gooch-Austin and the Volhard-Wolff methods. See 1898: 8 and 1884: 18.

1899: 9. HESS, W. H., and CAMPBELL, E. D. A New Method for the Direct Determination of Alumina in the Presence of Iron, Manganese, Calcium, and Magnesium.

J. Am. Chem. Soc., 21, 776; J. Chem. Soc. (Lond.), 78, 50; Ann. chim. anal. appl., 5, 230.

Separation from aluminum by means of phenylhydrazine.

1899: 10. NAMIAS, R. Volumetric Estimation of Manganese. (Title from J. Chem. Soc.)

* Ann. Soc. chim. Melano, 1899, 54; Chem. Centrbl., 1899, a, 1224;
 J. Iron Steel. Inst., 1900, a, 433; J. Chem. Soc. (Lond.), 78, 50.
 Use of the Volhard method (1879: 14).

1899: 11. NATTERER, K. Chemische Untersuchungen im Rothen Meere.

Monatsh. Chem., 20, 12.

Colorimetric determination in the water of the Red Sea.

1899: 12. POZZI-ESCOT, M-. E. Analyse microchimique. Ann. chim. anal. appl., 4, 398. Detection by microchemical tests.

1899: 13. REICHARD, C. Ueber die maassanalytische Bestimmung des Mangans in den mangansauren Salzen durch alkalische Lösungen von arseniger Säure.

> Chem. Ztg., 23, 801; Chem. Centrbl., 1899, b, 886; J. Soc. Chem. Ind., 18, 1156; J. Chem. Soc. (Lond.), 76, b, 813; Chem. News, 82, 308.
> See title.

1899: 14. REICHARD, C. Ueber die quantitative Bestimmung der Mangansäure in Gegenwart von Mangansalzen, bezw. die Analyse der beiden Manganverbindungen neben einander mittels arseniger Säure in alkalischer Lösung.

> Chem. Ztg., 23, 867; Chem. Centrbl., 1900, a, 66; J. Chem. Soc. (Lond.), 78, 109; Ann. chim. anal. appl., 5, 394; Analyst., 25, 23. See title.

1899: 15. RIEDERER, E. J. Electrolytic Determination of Zinc in the Presence of Manganese.

J. Am. Chem. Soc., 21, 789; J. Chem. Soc. (Lond.), 78, 49; Ann. chim. anal. appl., 5, 266; Analyst, 25, 79. Separation from zinc by electrolysis.

- 1899: 16. J. T. Manganese in Chrome Steels. Chem. News, 79, 157. Note on the Ford-Williams method (1881: 18).
- 1900: I. BÖTTGER, W. Ueber die Bestimmung des Mangans als Pvrophosphat.

Ber., 33, 1019; Chem. Centrbl. 1900, a, 1140; J. Soc. Chem. Ind., 19, 564; Chem. News, 82, 247; J. Chem. Soc. (Lond.), 78, 443; Analyst, 25, 304.

Full discussion of the conditions which give the best results in the determination as pyrophosphate.

1900: 2. DAKIN, H. D. Zur Bestimmung von Mangan und Kobalt als Phosphat.

Ztschr. anal. Chem., 39, 784.

Precipitation in the presence of a moderate excess of ammonium chloride and determination by weighing as ammonium manganese phosphate or the pyrophosphate. Comments on the article by Gooch and Austin. See 1898: 8.

1900: 3. HILLEBRAND, W. F. Some Principles and Methods of Rock Analysis.

Bull. U. S. Geol. Surv., No. 176, 60.

- Separation from nickel and cobalt by means of the solubility of their sulphides. Precipitation as carbonate. Comments on the Jannasch and Cloedt method of separation from zinc by means of hydrogen peroxide. See 1895: 11 and 1899: 7.
- 1900: 4. HIORNS, A. The Electrolytic Estimation of Manganese in Manganese Ores.
 - Chem. News, 81, 15; Chem. Centrbl., 1900, a, 489; J. Chem. Soc. (Lond.), 78, 444; Ann. chim. anal. appl., 5, 230; School Mines Quart., 22, 94.
 - Precipitation by electrolysis and ignition to mangano-manganic oxide.
- 1900: 5. IBBOTSON, F., and BREARLEY, H. The Estimation of Manganese and Chromium in Tungsten Alloys.
 - Chem. News, 82, 209; Chem. Centrbl., 1900, b. 1188; Chem. Ztg., 24, 347, Rep.

Volumetric determination by oxidation to permanganate by means of lead peroxide in nitric acid solution, and titration with a reducing agent. Hydrofluoric acid used to aid in the solution of the alloy. (See also Norris, 1891: 19, and Ford and Bregowsky, 1898: 6).

1900: 6. IBBOTSON, F., and BREARLEY, H. The Estimation of Molybdenum in Steel and Steel-making Alloys.

> Chem. News, 81, 269; School Mines Quart., 22, 97. Influence of molybdenum on the determination of manganese by means of bromine and by the Williams-Ford method (1881: 18).

1900: 7. JERVIS, H. Note on the Estimation of Manganese in Steel.

Chem. News, 81, 171; Chem. Centrbl., 1900, a, 1038; J. Chem. Soc. (Lond.), 78, 444; Stahl u. Eisen, 20, 747; School Mines Quart., 22, 94.

Determination in molybdenum powders and in tungsten steels by oxidation to permanganate by means of lead peroxide in nitric acid solution, and titration with oxalic acid or ferrous sulphate.

1900: 8. JOUET, C. H. The Analysis of Slags and Cinders.

School Mines Quart., 22, 71.

Determination by the Volhard method (1879: 14). Also precipitation by means of chlorate, separation from iron by means of acetates, and precipitation as phosphate.

1900: 9. MCKENNA, A. G. The Analysis of Chrome and Tungsten Steels.

Eng. Min. J., 70, 124; Chem. Ztg., 24, 243, Rep; Analyst, 25, 301. Precipitation by means of potassium chlorate, solution of the precipitate in hydrochloric acid and potassium nitrite, separation from iron by means of acetates, re-precipitation by means of bromine, and ignition to mangano-manganic oxide.

- 1900: 10. MIGNOT, A. Dosage volumétrique du manganèse dans les fers, fontes et aciers.
 - Ann. chim. anal. appl., 5, 172; Chem. Centrbl., 1900, b, 65; J. Soc. Chem. Ind., 19, 854; J. Chem. Soc. (Lond.), 78, 690; School Mines Quart., 22, 94.

Determination by oxidation to permanganic acid by means of bismuth tetroxide and titration with hydrogen peroxide.

1900: 11. PATTINSON, J., and H. S. Note on the Determination of Manganese as Sulphide.

Chem. News, 81, 193; J. Chem. Soc. (Lond.), 78, 443; Chem. Centrbl., 1900, a, 1244.

Discussion of the incomplete precipitation as sulphide.

1900: 12. TRUCHOT, P. Analyse des cuivres et des mattes industrielles.

Ann. chim. anal. appl., 5, 442.

Separation from cobalt and nickel, and determination by precipitation as manganese ammonium phosphate. Determination in commercial copper.

QUANTITATIVE DETERMINATION OF MANGANESE.

(A) BY GRAVIMETRIC METHODS.

I. By precipitation as	1882: 4 Dewey
(a) carbonate.	1882: 16 Troilius
1819: 1 Brandes	1883: 22 Wolff
1830: 2 Fuss	1884: 9 Holthof
1836: 2 Thomson	1885: 17 Reinhardt
1851: 1 Laming	1885: 18 Reinhardt
1853: 8 Morfit and Booth	1886: 1 Atkinson
1867: 2 Forbes	1886: 16 Müller
1867: 4 Tosh	1886: 22 Reinhardt
1869: 8 Prior	1886: 26 Sprenger
1870: 8 Rowan	1886: 27 Wolff
1871: 5 Rowan	1880: 27 Wolff
1872: 3 Fresenius	1888: 12 Oettel
1872: 10 Tamm	1888: 14 v. Reis
1886: 16 Müller	1888: 21
1888: 10 Meineke	1889: 1 Alt
1893: 9 Jean	1800: 5 Fresenius and Hintz
1893: 9 Jean 1897: 8a Hillebrand	1893: 11 Kosmann
1897. sa filleorald 1898: 1 Austin	
1898: 3 Brearley	1893: 13 Parry and Morgan
	1894: 13 Saniter
1900: 3 Hillebrand	1896: 11 Mignot
(b) di-oxide, hydrated, by means of	1900: 6 Ibbotson and Brear-
(1) bromine. 1862: 1 Abel	ley
	1900: 9 McKenna
1871: 3 Kammerer	(2) chlorine.
1874: 3 Piesse	1865: 6 Warington
1874: 6 Willis	1878: 6 Müller
1877: 18 Riley	(3) electrolysis.
1879: 8 Mackintosh	1865: 4 Luckow
1879: 14 Volhard	1875: 1 Boussingault
1880: 4 Dunston	1877: 17 Riche
1880: 9 de Koninck	1878: 8 Riche
1881: 11 Kent	1880: 10 Luckow
1881: 17 Troilius	1881: 2 Classen
1882: 2 Cabot	1881: 3 Classen and v. Reis

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	ysis—Continued.
1882: 10	Keiser
1883: 18	Schucht
1884:4	
1884: 17	Wieland
1885:4	
1886: 18	Moore
1889:3	
	Kohn and Woodgate
	Smith and Fränkel
1891: 13	
1892: 18	
	Warwick
1893: 14	Contraction of the second s
1894: 2	
	Thomälen
1895: 5	
1895: 6	
1895: 16	Neumann
1896: 6	
1897: 7	Engels
1898: 13	Kaeppel
1898: 21	Wolman
1900:4	Hiorns
(4) hydrog	en peroxide.
1877: 19	Rosenthal
1884: 7	Hanowsky
1886: 16	
1887: 7	Donath and Zeller
1888: 11	Moore
1889: 15	Radau
1890:4	Carnot
1893: 2	Carnot
1894:4	Jones
1899: 7	Friedheim and Brühl
(5) hypoch	lorites.
	Reichardt
1875: 1	Boussingault
1877: 11	Kern
(6) lead pe	roxide in neutral solu-
tion.	
1852: 2	Gibbs
1853: 9	Parkinson
1860: 7	Rose
1879: 14	Volhard
	um chlorate.
	Hannay
	Beilstein and Jawein
	Beilstein and Jawein

1881: 8	Ford
1882: 16	Troilius
1884: 9	Holthof
1887: 18	
1893: 9	Jean
	Forestier
	Mignot
1896: 17	Viard
1900: 8	Joüet
1900: 9	McKenna
(c) manganes	se ammonium phos-
phate.	
1867: 3	Gibbs
1870: 11	Talbott
1871: 6	
1872: 1	
1872: 3	Fresenius
1873: 2	Gibbs
1877: 1	Bolton
1877: 14	Munroe
1881: 8	
1884: 3	
1887: 2	
1887: 3	Blair
1888: 10	
1890: 3	
1890: 12	
1893: 9	
1894: 13	
1896: 5	
1896: 11	
1898: 8	Gooch and Austin
1899: 5	Dunnington
1900: 1	Böttger
1900: 2	Dakin
1900: 8	Joüet
1900: 12	Truchot
	us hydroxide. Gurlt
1856: 1	Kern
1875: 2 1876: 4	Kern
(e) oxalate.	Kern
(e) Oxalate. 1870: 2	Gibbs
1870: 2	Leison
1872: 3	Fresenius
1877: 3	
1877: 7	Classen
(f) sulphide.	
1821: 2	Pfaff

(d)

(f) sulphide—Continued.	1877: 3 Classen
1857: 4 Terreil	1877: 12 Kern
1860: 2 Gorgeu	1878: 6 Müller
1860: 5 Rose	1879: 14 Volhard
1860: 6 Rose	1880: 4 Dunston
1861: 1 Fresenius	1881: 17 Troilius
1863: 3 Lippert	1882: 2 Cabot
1867: 2 Forbes	1882: 16 Troilius
1867: 4 Tosh	1883: 22 Wolff
1869: 1 Classen	1884: 7 Hanowsky
1869: 4 How	1885: 11 v. Jüptner
1870: 11 Talbott	1886: 16 Müller
1872: 3 Fresenius	1887: 7 Donath and Zeller
1876: 2 Fresenius	1887: 18
1876: 4 Kern	1888: 12 Oettel
1877: 5 Classen	1888: 14 v. Reis
1879: 1 Beilstein and Jawein	1888: 21
1879: 2 Carnot	1889: 3 Brand
1879: 7 Ledebur	1889: 7 Kohn and Woodgate
1880: 2 Delffs	1891: 21 Pattinson
1883: 23 Zulkowsky	1893: 9 Jean
1885: 11 v. Jüptner	1893: 13 Parry and Morgan
1888: 3 Friedmann	1894: 2 Classen
1888: 10 Meineke	1894: 13 Saniter
1888: 16 Schürmann	1895: 7 Forestier
1890: 5 Fresenius and Hintz	1896: 14 Rürup
1893: 9 Jean	1897: 8a Hillebrand
1894: 13 Saniter	1898: 1 Austin
1897: 8 Granger	1898: 7 Gooch and Austin
1898: 15 Murmann	1900: 4 Hiorns
1898: 16 Murmann	1900: 9 McKenna
1900: 11 Pattinson	(b) manganous oxide.
(g) vanadate.	1843: 2 Ebelmen
1887: 6 Carnot	1875: 2 Kern
I. By ignition to	(c) pyrophosphate. See "precipi
(a) mangano-manganic oxide.	tation as manganese-ammo
1836: 2 Thomson	nium phosphate."
1856: 1 Gurlt	(d) sulphate.
1860: 2 Gorgeu	1879: 14 Volhard
1865: 6 Warington	1885: 15 Meineke
1866: 4 Reichardt	1888: 21
1867: 2 Forbes	1898: 1 Austin
1867: 4 Tosh	1898: 7 Gooch and Austin
1870: 8 Rowan	(e) sulphide.
1872: 3 Fresenius	1860: 5 Rose
1874: 3 Piesse	1860: 6 Rose
1874: 6 Willis	1863: 3 Lippert
1875: 2 Kern	1876: 2 Fresenius
1876: 4 Kern	1879: 2 Carnot

I

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III. H	By the met	hod of
	Classen (d	
	1877: 3	
	1877: 7	
	1894: 9	Nass
(b)	Ford.	
		Deshayes
	1881: 8	Ford
	1896: 5	
	1896: 14	
(c)	Gibbs (py	rophosphate).
	1867: 3	Gibbs
	1871: 6	
	1872: 1	Allen
	1877: I	Bolton
		McKenna
(d)	Rüdorff.	
	1892: 18	
	1895: 8	Gröger
(e)	Wolff.	
	1883: 22	
		Mathesius
		Reinhardt
		Reinhardt
	1886: 27	
		Chemiker - Commis-
	sion	17
TV T	1893: 11	
	and the second	neous methods.
(a)	by differe: 1877: 18	
		Ledebur
		Atkinson
	1884: 8	
	1885: 7	
	1888: 21	
(b)	by dry as	
(~)	1872: 11	
		Büttgenbach
(c)		gen absorbed by alka
	line sol	
	1864: 4	Mittenzwey
(B) B	Y VOLUM	METRIC METHODS
I. By	titration	with potassium per
	nanganate	
	Direct titr	
	1862: 2	

1864:	6	Winkler
1865:	3	Habich
1872:		Fresenius
1878:		Morawski and Stingl
1879:		and the second sec
1880:	5	Haswell
1880:	7	v. Jüptner
1881:	6	Donath
1881:	7	Emmerton
1881:	15	
1883:	-	v. Jüptner
1883:		Meineke
1883:		
1883:		Särnström
1883:		
1883:	23	
1884:	1	Anger
1884:	6	Gmelin
		Meineke
		Wolff
1885:	18	Reinhardt
1885:		Wolff
		Reinhardt
1886:		Wolff
	27 28	
		Zimmermann Brand
1887:		
1887:		Jolles
1888:		Ghilian
1891:		Blum
	4	Brown Chamilton Commis
1891:	5	Chemiker - Commis-
		SION
1891:	9	Hampe
1891:	16	Moldenhauer
	25	
1891:	26	
1892:	I	Aller
1892:	4	Campredon
1892:		Donath
1892:		v. Reis
1892:		
1892:	17	Rubricius
1893:	I	Carnot
1893:	6	Gorgeu
1893:	9	Jean
1894:		Saniter
1894:	14	
1895:	2	Auchy
1895:	18	Thomas

(a) Direct titration—Continued.	1889: 12 McCulloch
1896: 1 Auchy	1890: 2 van Bemmeln
1896: 12 Mixer and Dubois	1893: 2 Carnot
1896: 13 Murkewitsch	1894: 4 Jones
1896: 14 Rürup	1895: 4 Carnot
1896: 15 Stone	1895: 7 Forestier
1897: 1 Auchy	1895: 19 Ulzer and Brüll
1897: 3 Brearley	(d) hypochlorite.
1897: 6 Devisse	1853: 4 Hempel
1897: 10 Longi and Camilla	1854: 2 Streng
1898: 4 Campredon	1855: 1 Mohr
1898: 14 Lehnkering	1855: 2 Müller
1890: 1 Brearley	1879: 9 Pattinson
1899: 4 Daw	1879: 10 Pattinson
1899: 8 Herting	1880: 13 Pattinson
1899: 10 Namias	1880: 16 Weldon
1900: 8 Jouet	1880: 18 Wright and Menke
(b) Indirect titration.	1884: 11 Ledebur
1883: 10 Meineke	1886: 1 Atkinson
1883: 11 Meineke	1887: 13 Lax
1885: 15 Meineke	1893: 9 Jean
1886: 15 Meineke	(e) potassium chlorate.
1886: 17 Müller	1877: 10 Hannay
1886: 24 Schöffel and Donath	1881: 18 Williams
1887: 13 Lax	1883: 4 Hampe
1891: 5 Chemiker - Commis-	1883: 9 Mackintosh
sion	1883: 13 Raimond
1891: 9 Hampe	1883: 19 Stone
1896: 7 Giorgis	1883: 20 Stone
1899: 13 Reichard	1883: 21 Troilius
II. ¹ By precipitation as di-oxide, solu-	1884: 11 Ledebur
tion with the aid of a reducing	1884: 12 Mackintosh
agent, and titration for the ex-	1885: 3 Cheever
cess of the latter.	1885: 10 Hampe
Precipitation by means of	1885: 15 Meineke
(a) bromine.	1887: 13 Lax
1872: 5 Kessler	1887: 15 Meineke
1872: 6 Kessler	1888: 7 Julian
1879: 6 Kessler	1888: 13 Reinhardt
1879: 7 Ledebur	1891: 4 Brown
1887: 14 Meineke	1891: 5 Chemiker - Commis-
1887: 17 Reinhardt	sion
1888: 13 Reinhardt	1891: 9 Hampe
1893: 12 Low	1891: 19 Norris
(b) chlorine.	1891: 23 v. Reis
1861: 4 Möller	1891: 28 Ukena
(c) hydrogen peroxide.	1892: 2 Bastin
1886: 2 Barlow	1892: 5 Chemiker - Commis-
1888: 2 Carnot	sion

IOI

1892: 9 Hampe
1892: 13 v. Reis
1893: 9 Jean
1893: 10 Julian
1893: 13 Parry and Morgan
1894: 5 Jones
1895: 2 Auchy
1895: 7 Forestier
1895: 19 Ulzer and Brüll
1896: 5 Dudley
1896: 14 Rürup
1897: 9 Julian
1898: 6 Ford and Bregowsky
1898: 14 Lehnkering
1899: 16 J. T.
1900: 6 Ibbotson and Brear-
ley
(f) reduction of manganate by al-
cohol.
1890: 13 Myhlertz
(g) sodium chlorate.
1898: 9 Gooch and Austin
1899: 8 Herting
Solution of the peroxide with the
aid of
(a) antimonious chloride.
Phase Konglas
1872: 5 Kessler
1872: 6 Kessler
1872: 6 Kessler 1879: 6 Kessler
1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur
1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide.
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid.
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay 1879: 9 Pattinson
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay 1879: 9 Pattinson 1879: 10 Pattinson
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay 1879: 9 Pattinson 1879: 10 Pattinson 1880: 13 Pattinson
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay 1879: 9 Pattinson 1879: 10 Pattinson 1880: 13 Pattinson 1880: 16 Weldon
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay 1879: 9 Pattinson 1879: 10 Pattinson 1880: 13 Pattinson 1880: 16 Weldon 1880: 18 Wright and Menke
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay 1879: 9 Pattinson 1879: 10 Pattinson 1880: 13 Pattinson 1880: 16 Weldon 1880: 18 Wright and Menke 1881: 18 Williams
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay 1879: 9 Pattinson 1879: 10 Pattinson 1880: 13 Pattinson 1880: 16 Weldon 1880: 18 Wright and Menke 1881: 18 Williams 1883: 4 Hampe
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay 1879: 9 Pattinson 1879: 10 Pattinson 1880: 13 Pattinson 1880: 16 Weldon 1880: 18 Wright and Menke 1881: 18 Williams 1883: 4 Hampe 1883: 9 Mackintosh
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay 1879: 9 Pattinson 1879: 10 Pattinson 1880: 13 Pattinson 1880: 16 Weldon 1880: 18 Wright and Menke 1881: 18 Williams 1883: 4 Hampe 1883: 4 Hampe 1883: 9 Mackintosh 1883: 13 Raimond
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay 1879: 9 Pattinson 1879: 10 Pattinson 1880: 13 Pattinson 1880: 16 Weldon 1880: 18 Wright and Menke 1881: 18 Williams 1883: 4 Hampe 1883: 9 Mackintosh 1883: 13 Raimond 1883: 19 Stone
 1872: 6 Kessler 1879: 6 Kessler 1879: 7 Ledebur (b) arsenious oxide. 1898: 9 Gooch and Austin (c) ferrous salts or oxalic acid. 1853: 4 Hempel 1877: 10 Hannay 1879: 9 Pattinson 1879: 10 Pattinson 1880: 13 Pattinson 1880: 16 Weldon 1880: 18 Wright and Menke 1881: 18 Williams 1883: 4 Hampe 1883: 4 Hampe 1883: 9 Mackintosh 1883: 13 Raimond

1884: 12 Mackintosh 1885: 3 Cheever 1885: 10 Hampe 1885: 15 Meineke 1886: 1 Atkinson 1887: 13 Lax 1887: 14 Meineke 1887: 15 Meineke 1887: 17 Reinhardt 1888: 2 Carnot 1888: 7 Julian 1888: 13 Reinhardt 1889: 12 McCulloch 1890: 2 van Bemmeln 1890: 13 Myhlertz 1891: 4 Brown 1891: 5 Chemiker - Commission 1891: 9 Hampe 1891: 19 Norris 1891: 23 v. Reis 1891: 28 Ukena 1802: 2 Bastin 1892: 5 Chemiker - Commission 1802: 9 Hampe 1892: 13 v. Reis 1893: 2 Carnot 1893: 9 Jean 1893: 12 Low 1893: 13 Parry and Morgan 1894: 4 Jones, H. C. 1894: 5 Jones, J. 1895: 2 Auchy 1895: 4 Carnot 1895: 7 Forestier 1805: 18 Thomas 1895: 19 Ulzer and Brüll 1896: 5 Dudley 1896: 14 Rürup 1898: 6 Ford and Bregowsky 1898: 14 Lehnkering 1899: 16 J. T. (d) hydrochloric acid (Bunsen). 1861: 4 Möller 1886: 2 Barlow (e) hydrogen peroxide. 1893: 10 Julian

(e) hydrogen peroxide—Continued.	1887: 19
1897: 9 Julian	1892: 20 Van Grundy
(f) potassium iodide.	(c) ferrous salts.
1898: 9 Gooch and Austin	1878: 7 Prochaska
(g) stannous chloride.	1886: 21 Perillou
1854: 2 Streng	1888: 17 Stein
1855: 1 Mohr	1900: 7 Jervis
1855: 2 Müller	(d) hydrogen peroxide.
III. By titration of permanganic acid,	1888: 15 Schneider
after oxidation by means of	1889: 17 Schneider
(a) bismuth tetroxide.	1892: 19 Schneider
1888: 15 Schneider	1895: 17 Reddrop and Ram
1889: 17 Schneider	age
1898: 4 Campredon	1898: 4 Campredon
1000: 10 Mignot	1900: 10 Mignot
(b) lead peroxide.	(e) mercurous nitrate.
1871: 2 Chatard	1872: 7 Leclerc
1872: 7 Leclerc	1881: 9 Forguignon
1877: 8 Deby	1885: 19 Schlagdenhauffen
1878: 2 Deshayes	IV. By the method of
1878: 7 Prochaska	(a) Chatard.
1881: 9 Forguignon	1871: 2 Chatard
1885: 19 Schlagdenhauffen	1888: 18 Thorpe and Hambly
1886: 21 Perillou	1888: 19 Thorpe and Hambly
1886: 25 Setterwall	1888: 20 Weissmann
1887: 1 Babbitt	(b) Carnot.
1887: 5 Cheever	1888: 2 Carnot
1887: 19	1889: 12 McCulloch
1888: 17 Stein	1890: 2 van Bemmeln
1888: 18 Thorpe and Hambly	1890: 4 Carnot
1888: 19 Thorpe and Hambly	1897: 2 van Bemmeln
1802: 19 Schneider	(c) Deshayes.
1892: 20 Van Grundy	1878: 2 Deshayes
1900: 5 Ibbotson and Brear-	1887: 19
ley	(d) Donath.
1900: 7 Jervis	1881: 6 Donath
(c) sodium bismuthate.	1893: 6 Gorgeu
1895: 17 Reddrop and Ram-	(e) Guyard.
age	1863: 2 Guyard
Titration of the permanganic acid	1865: 3 Habich
by means of	1882: 6 Dunn
(a) ammonium oxalate.	1884: 15 Meineke
1871: 2 Chatard	1893: 1 Carnot
1888: 18 Thorpe and Hambly	1893: 6 Gorgeu
1888: 19 Thorpe and Hambly	1893: 9 Jean
(b) arsenious oxide.	(f) Hampe.
1877: 8 Deby	1883: 4 Hampe
1878: 2 Deshayes	1884: 11 Ledebur
1886: 25 Setterwall	1885: 10 Hampe

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Ram-

(f) Hampe—Continued. 1886: 23 Reinhardt 1886: 24 Schöffel and Donath 1887: 13 Lax 1887: 15 Meineke 1888: 13 Reinhardt 1891: 5 Chemiker - Commission 1891: 23 v. Reis 1892: 5 Chemiker - Commission 1892: 9 Hampe 1892: 13 v. Reis 1894: 5 Jones, J. 1895: 19 Ulzer and Brüll 1896: 14 Rürup (g) Kessler. 1872: 5 Kessler 1872: 6 Kessler 1879: 6 Kessler 1887: 13 Lax (h) Leclerc. 1872: 7 Leclerc 1881: 9 Forguignon 1885: 19 Schlagdenhauffen (i) Lenssen. 1860: 3 Lenssen. 1864: 2 Fresenius (j) Meineke. 1883: 10 Meineke 1883: 11 Meineke 1885: 15 Meineke 1886: 15 Meineke 1886: 17 Müller 1886: 23 Reinhardt 1887: 13 Lax 1891: 5 Chemiker - Commission (k) Morawski and Stingl. 1878: 4 Morawski and Stingl 1884: 15 Meineke (1) Pattinson. 1879: 9 Pattinson 1880: 18 Wright and Menke 1884: 11 Ledebur 1886: 1 Atkinson 1886: 20 Pattinson 1887: 13 Lax 1891: 21 Pattinson

1803: 0 Jean 1894: 13 Saniter (m) Reinhardt. 1888: 13 Reinhardt 1891: 5 Chemiker - Commission (n) Rössler. 1879: 13 Rössler 1880: 15 Rössler 1894: 14 Seeliger (o) Rürup. 1891: 2 Blum 1891: 25 Rubricius 1891: 26 Rürup (p) Särnström. 1881: 15 Särnström 1883: 7 Kerl 1883: 15 Särnström 1890: 6 Hellman 1896: 12 Mixer and Dubois 1897: 1 Auchy (q) Schneider. 1889: 17 Schneider 1895: 17 Reddrop and Ramage 1898: 4 Campredon (r) Schöffel and Donath. 1883: 14 Särnström 1883: 17 Schöffel and Donath 1886: 24 Schöffel and Donath 1887: 13 Lax 1891: 5 Chemiker - Commission (s) Volhard. 1879: 14 Volhard 1880: 5 Haswell 1880: 7 v. Jüptner 1881: 7 Emmerton 1882: 8 Haswell 1883: 6 v. Jüptner 1883: 11 Meineke 1884: 6 Gmelin 1884: 11 Ledebur 1884: 15 Meineke 1887: 9 Jolles 1888: 6 Iles 1801: 2 Blum 1891: 4 Brown 1891: 16 Moldenhauer

(s) Volhard—Continued		
1801:	20	Namias
		Rubricius
		Rürup
1892:		
		v. Reis
		Rubricius
		Rubricius
		Saniter
		Auchy
1895:	18	Thomas
1896:	I	Auchy
1896:	13	Murkewitsch
1896:	14	Rürup
1896:	15	Stone
1897:	6	Devisse
1897:	10	Longi and Camilla
1898:	4	Campredon
		Lehnkering
1899:	I	Brearley
1899:		
		Herting
1899:	10	Namias
1900:		
(t) Weissm		
1888:		
		Weissmann
		Ulzer and Brüll
(u) William		
		Williams
	-	Mackintosh
		Troilius
		Mackintosh
		Cheever
1891:	4	Brown
		Bastin
1893:	13	Parry and Morgan
1895:	2	Auchy
		Dudley
1898:	0	Ford and Bregowsky
1899:		
1900:	0	Ibbotson and Brear-
ley . By miscellaneous methods.		
(a) by means of alkali sulphides.		
(a) by mea	Ins	Neumann
(b) by reduction of potassium ferri- cyanide.		
		Lenssen
1000.	0	LICHOSCH .

1864: 2 Fresenius (c) by means of potassium ferrocyanide. 1889: 13 Moldenhauer 1891: 3 Blum 1891: 14 Luckow 1801: 17 Moldenhauer 1897: 12 Miller 1897: 13 Miller and Mathews 1897: 16 Stone and van Ingen (d) by means of tartaric or malic acids. 1868: 2 Juette (e) by means of silver nitrate (indirect). 1879: 13 Rössler 1880: 15 Rössler 1894: 14 Seeliger (f) by the titration of manganate. 1881: 10 Iles 1885: 12 Kalmann and Smolka 1899: 14 Reichard (g) by the titration of manganic phosphate. 1883: 4 Hampe. 1801: 18 Moore (h) by means of iodine (indirect). 1890: 16 Vortmann (i) by solution of ignited oxide in reducing agents. 1876: 3 Galbraith 1886: 2 Barlow (C) BY COLORIMETRIC METH-ODS. I. By oxidation to permanganic acid by means of (a) bismuth tetroxide. 1895: 7 Forestier 1896: 11 Mignot (b) lead peroxide. 1872: 8 Pichard 1876: 5 Peters 1881: 5 Deshayes 1883: 1 Goetz 1882: 11 Ledebur 1886: 8 Cheever

1886: 13 Hunt

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(b) lead peroxide—Continued.	(E) S
1887: 5 Cheever	
1887: 16 Morgan	
1893: 13 Parry and Morgan	
1895: 7 Forestier	(a)
1896: I Auchy	
1896: 11 Mignot	
1897: 11 Lemaire	
1897: 15 Schneider	
1897: 15 Schieder 1898: 18 Pichard	(b)
II. By the formation of metaphos- phate.	
1885: 16 Osmond	
1801: 24 Rossi	
III. By the formation of manganate.	
1873: 1 Brünner	
1874: 1 Koppmayer	(0)
IV. By the liberation of iodine.	(c) 1
1874: 2 Morrell	
1875: 4 Morrell	
(D) GENERAL DISCUSSION OF	(d) ±
METHODS.	
1875: 1a Bolton	
1881: 11 Kent	
1882: 6 Dunn	
1882: 15 Tamm	5 (e) f
1883: 16 Schmitt	
1884: 13 Mackintosh	
1884: 16 Stone	
1885: 19 Schlagdenhauffen	
1887: 13 Lax	
1888: 10 Meineke	(f) f
1889: 4 Finkener	(1) 1
1891: 21 Pattinson	(g) f
1894: 13 Saniter	(g) 1
1895: 15 v. Jüptner	
1895: 16 Neumann	
1896: 1 Auchy	
1896: 4 Dewey	
1896: 10 v. Jüptner	
1896: 14 Rürup	
1897: 15 Schneider	
1898: 13 Kaeppel (electro-	(h) f
lytic)	(1)
1898: 21 Wolman (electro-	
lytic)	(2)
1899: 3 Brearley	

(E) SEPARATION FROM OTHER ELEMENTS.

(a) from alk	aline earths.	
1852: 2	Gibbs	
1853: 9	Parkinson	
1861: 4	Möller	
1861: 4 1886: 2	Barlow	
(b) from alu		
1860: 6	Rose	
1865: 2	Gibbs	
1865: 5		
1879:4	Classen	
1879: 14	Volhard	
1881: 3	Classen	
	Hess and Campbell	
(c) from arse		
1837:4		
1895: 13	Jannasch and Kam-	
	merer	
(d) from cad		
1889: 3		
1891: 27		
	Warwick	
1895: 14	Jannasch and Rött-	
(e) from calc	gen	
	Stromeyer	
1860: 6	Rose	
1877: 4	Classen	
1889: 2	Blum	
1889: 16		
1802: 14		
(f) from ceriu	ım.	
1864: 3	Gibbs	
(g) from chro	mium.	
1865: 2	Gibbs	
1884: 5		
1894: 6	Kassner	
1894: 11	Poleck	
1895: 10	Jannasch and Cloedt	
1898: 3	Brearley	
	Friedheim and Brühl	
(h) from cobalt by means of		
(1) chlorine		
1866: 5		
(2) citrates		
1892: 10	Moore	

(3) cyanides. 1841: 3 Liebig 1853: 2 Flajolot 1887: 10 Klobb 1889: 11 McCulloch (4) electrolysis. 1888: 12 Oettel 1889: 3 Brand 1891: 15 Le Roy 1898: 5 Engels (5) hydrogen peroxide. 1886: 2 Barlow 1887: 7 Donath and Zeller 1891: 10 Jannasch and Franzek 1896: 9 Jannasch and Lehnert (6) hypochlorite, hydrofluoric acid, and ammonia. 1841: 4 Ullgren (7) magnesium. 1832: 2 Döbereiner (8) Mercuric oxide. 1835: 2 Persoz (9) nitroso-β-napthol. 1806: 2 Burgass (10) oxalates. 1827: I Du Menil (11) phosphates. 1858: 2 Henry 1900: 12 Truchot (12) potassium permanganate. 1866: 5 Terreil (13) potassium polysulphide. 1845: 1 Cloez (14) silver nitrate and ammonia. 1839: 3 W. (15) sodium peroxide. 1893: 5 Clark (16) solubility of chlorides in ether. 1837: 1 Döbereiner (17) the solubilities of the sulphides. 1838: 2 Wackenroder 1846: 1 Barreswil 1847: 3 Rose 1847: 4 Strecker 1849: 1 Ebelmen

1865: 2 Gibbs 1866: 3 Frohde 1860: 7 Muck 1881: 4 Delvaux 1886: 26 Sprenger 1890: 5 Fresenius and Hintz 1807: 8a Hillebrand 1000: 3 Hillebrand (18) volatility of chloride. 1846: 4 Völker (i) from copper. 1860: 5 Luckow 1884: 5 Classen 1887: 11 v. Knorre 1887: 12 v. Knorre 1889: 3 Brand 1803: 14 Rüdorff 1895: 14 Jannasch and Röttgen 1806: 2 Burgass 1806: 8 Jannasch 1807: Sa Hillebrand 1800: 6 Fernberger and Smith 1899: 7 Friedheim and Brühl (j) from gallium. 1882: 1 de Boisbaudran (k) from iron by means of (1) acetates. 1841: 2 Henry 1862: 1 Abel 1865: 2 Gibbs 1866: 2 Eggertz 1866: 4 Reichardt 1867: 4 Tosh 1869: 2 Damour 1870: 8 Rowan 1872: 5 Kessler 1872: 6 Kessler 1874: 3 Piesse 1874: 6 Willis 1875: 1 Boussingault 1875: 4 Morrell 1877: 13 Krämer 1877: 18 Riley 1877: 19 Rosenthal 1877: 20 Stöckman 1878: 3 Matzurka 1878: 6 Müller

1879: 8 Mackintosh

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(I) acetates-Continued. 1880: 4 Dunston 1880: 6 Jewett 1881: 11 Kent 1881: 17 Troilius 1882: 2 Cabot 1882: 4 Dewey 1882: 9 Jewett 1884: 3 Bloxam 1886: 1 Atkinson 1886: 12 Deane 1886: 16 Müller 1886: 22 Reinhardt 1886: 26 Sprenger 1887: 18 -----1888: 4 Ghilian 1888: 10 Meineke 1888: 14 'v. Reis 1888: 21 -----1889: 9 Mayer (qualitative) 1892: 4 Campredon 1892: 11 Priwoznik 1892: 14 Riggs 1803: 0 Jean 1893: 11 Kosman 1893: 13 Parry and Morgan 1895: 7 Forestier 1897: 3 Brearley 1897: 4 Brearley 1899: 2 Brearley 1900: 8 Jouet 1900: 9 McKenna (2) ammonia in the presence of ammonium chloride. 1813: 1 Hatchett 1830: 2 Fuss 1876: 4 Kern (3) arsenates. 1827: I Du Menil 1827: 2 Quesneville 1829: 3 Martini (4) benzoic acid. 1806: 1 Berzelius 1812: 1 Pfaff 1829: 3 Martini 1836: 2 Thomson 1877: 9 Funaro (5) camphoric acid. 1832: 5 Kastner

(6) chlorate. See "precipitation by means of chlorate." See pp. 08 and IOI (7) chlorides. 1797: 1 Kirwan 1837: 4 Scheerer 1863: 3 Lippert (8) chlorine. 1853: 12 Schiel 1861: 4 Möller (9) electrolysis. 1830: 1 Becquerel 1881: 2 Classen 1881: 3 Classen and v. Reis 1882: 10 Keiser 1885: 5 Classen 1886: 10 Classen 1886: 11 Classen 1888: 12 Oettel 1889: 7 Kohn and Woodgate 1891: 15 Le Roy 1896: 6 Engels 1808: 5 Engel (10) ether. 1892: 15 Rothe (11) fusion with alkali and nitrate. 1833: 2 Planiawa 1894: 14 Seeliger (12) ferrocyanide. 1886: 5 Blum (13) hydrogen peroxide. 1888: 11 Moore (14) iodine. 1879: 1 Beilstein and Jawein 1881: 1 Beilstein and Jawein (15) a magnet after ignition in hydrogen. 1875: 2 Kern (16) neutralization with carbonates. 1700: 1 Vauquelin Pfaff 1812: I 1821: 1 Herschell 1831: I Fuchs 1831: 2 Liebig 1832: 2 Döbereiner 1832: 6 Liebig 1834: 1 Demarçay

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1890: 17 Warren

(g) Filtration aided by addition of a concentrated solution of sodium acetate.

1888: 16 Schürmann

- (h) Destruction of organic matter by means of barium peroxide before titration with permanganate.
 - 1887: 4 Brand

1892: 12 v. Reis

- (i) Determination of moisture in the analysis of pyrolusite.
 1855: 3 Fresenius
- (j) Use of hydrofluoric acid to hold silicic acid in solution.
 - 1891: 19 Norris
 - 1898: 6 Ford and Bregowsky
 - 1900: 5 Ibbotson and Brearley.

QUANTITATIVE DETERMINA-TION OF MANGANESE PEROXIDE.

I. By evolution of chlorine, and absorption in solutions of

- (a) alkaline hydroxides and determination of the hypochlorite formed.
 - 1829: I Gay-Lussac
 - 1835: 1 Gay-Lussac
 - 1844: 1 Ettling
 - 1869: 10 Sherer and Rumpf
 - 1870: 10 Sherer and Rumpf
 - 1870: 12 Tissandier
 - 1877: 16 Perrey
- (b) arsenious acid. 1853: 10 Price

1860: 4 Machnea

(c) ferrous sulphate.
1831: 3 Turner
1842: 4 Otto
1867: 1 Braun
1868: 3 Lunge
1869: 8 Prior
1885: 2 Charpentier
(d) potassium iodide, and titration

of iodine.

1853: 1 Bunsen

1853: 5 Krieger

1861: 4 Möller

1869: 10 Sherer and Rumpf

- 1870: 1 Fresenius
- 1870: 5 Pattinson
- 1870: 10 Sherer and Rumpf
- 1877: 15 Parreño
- 1877: 16 Perrey
- 1879: 11 Pickering
- 1881: 13 Lunge
- 1888: 9 de Koninck and Lecrenier
- (e) silver nitrate.

1843: 1 Baumann

- (f) stannous chloride. 1851: 2 Müller
- (g) sulphurous acid (precipitation of barium sulphate).
 - 1832: 3 Duflos
 - 1832: 4 Duflos
 - 1837: 2 Ebelmen
 - 1838: 1 Gieseler
 - 1874: 5 Pouchet
- II. By solution in presence of a reducing agent.
 - (a) antimonious chloride.
 1872: 5 Kessler
 - (b) arsenious acid. 1808: 2 Bialobszcski
 - (c) ferrous salts.
 - 1842: 2 Levol
 - 1847: 2 Levol
 - 1851: 4 Schabus
 - 1856: 2 Schreiner
 - 1869: 11 Teschenmacher and Smith
 - 1870: 5 Pattinson
 - 1877: 10 Hannay
 - 1880: 11 Lunge
 - 1881: 16 Terreil
 - 1889: 4 Finkener
 - 1889: 10 M'Kellar
 - (d) formic acid (with absorption of carbon dioxide).

1833: 1 Göbel

- (e) oxalates (with absorption of carbon dioxide).
 - 1843: 3 Fresenius and Will
 - 1847: 6 De Vry
 - 1861: 2 Kolbe

(e) oxalates—Continued. 1863: 1 Fresenius 1869: 6 Mohr 1869: 10 Sherer and Rumpf 1869: 11 Teschenmacher and Smith 1870: 5 Pattinson 1870: 10 Sherer and Rumpf 1871: 4 Luck 1877: 16 Perry 1881: 13 Lunge 1882: 3 Darton 1890: 1 Baumann (f) oxalic acid (volumetric). 1870: 6 Paul 1889: 4 Finkener (g) potassium iodide and acid. 1858: 1 Hempel 1882: 5 Diehl 1883: 5 Hempel (h) stannous chloride. 1865: 1 Alfraise 1883: 3 Harvey III. By gasometric methods. Measurement of (a) carbon dioxide. 1832: 1 Berthier 1833: 3 Zenneck (b) nitrogen. 1832: 1 Berthier 1833: 3 Zenneck 1897: 14 Purgotti (c) oxygen evolved from hydrogen peroxide. 1885: 13 Lunge 1890: 1 Baumann 1890: 9 Lunge 1800: 10 Lunge 1800: 11 Lunge 1893: 3 Carnot 1893: 4 Carnot 1894: 7 Kippenberger 1894: 8 Lunge 1895: 3 Bodländer (d) oxygen expelled on ignition. 1833: 3 Zenneck IV. By loss of weight of metallic copper. 1839: 1 Fikentscher

1839: 2 Fuchs 1851: 3 Personne and Lhermite Fikentscher 1859: I 1859: 2 Nolté 1861: 5 Quadrat 1864: 5 -----V. By fusion with chromic oxide and alkali. 1882: 17 Wagner VI. By the method of (a) Bunsen. 1853: 1 Bunsen 1853: 5 Krieger 1861: 4 Möller 1869: 10 Sherer and Rumpf 1870: 1 Fresenius 1870: 5 Pattinson 1870: 9 Sherer 1870: 10 Sherer and Rumpf 1874: 5 Pouchet 1877: 16 Perrey 1880: 13 Pattinson 1881: 13 Lunge 1889: 4 Finkener (b) Fresenius and Will. 1843: 8 Fresenius and Will 1847: 6 De Vry 1862: 2 Röhr 1863: 1 Fresenius 1860: 6 Mohr 1869: 10 Sherer and Rumpf 1869: 11 Teschenmacher and Smith 1870: 5 Pattinson 1870: 9 Sherer 1870: 10 Sherer and Rumpf 1871: 4 Luck 1874: 5 Pouchet 1877: 16 Perry 1881: 13 Lunge 1890: 1 Baumann (c) Gay-Lussac. 1829: 1 Gay-Lussac 1836: 3 Wittstein 1844: 1 Ettling 1860: 4 Machnea 1877: 16 Perry 1803: 9 Jean

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- (p) sodium hypobromite. 1892: 6 Deniges (q) sodium peroxide.
- 1893: 7 Hempel
- (r) spectrum analysis. 1862: 3 Simmler

1872: 4 Horner

1875: 5 Vogel 1880: 12 Parry and Tucker

1898: 10 de Gramont

(s) lead peroxide.

1889: 8 de Koninck

(t) separation from iron by means of nitrites.

1897: 17 Wynkoop

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