Notes on mineralogy. No. VIII. On the felspar and mica of the granite of Canton / by Samuel Haughton.

Contributors

Haughton, Samuel. Royal College of Surgeons of England

Publication/Creation

[London]: [Taylor and Francis], [1859]

Persistent URL

https://wellcomecollection.org/works/rzw3betb

Provider

Royal College of Surgeons

License and attribution

This material has been provided by This material has been provided by The Royal College of Surgeons of England. The original may be consulted at The Royal College of Surgeons of England. Where the originals may be consulted. This work has been identified as being free of known restrictions under copyright law, including all related and neighbouring rights and is being made available under the Creative Commons, Public Domain Mark.

You can copy, modify, distribute and perform the work, even for commercial purposes, without asking permission.



Wellcome Collection 183 Euston Road London NW1 2BE UK T +44 (0)20 7611 8722 E library@wellcomecollection.org https://wellcomecollection.org

NOTES ON MINERALOGY .- No. VIII.

ON THE

FELSPAR AND MICA

OF THE

GRANITE OF CANTON.

BY

THE REV. SAMUEL HAUGHTON, M.A., F.R.S., FELLOW OF TRINITY COLLEGE, AND PROFESSOR OF GEOLOGY IN THE UNIVERSITY OF DUBLIN.

THE granite of the neighbourhood of Canton is composed of grey quartz, a light flesh-coloured or creamy-white felspar, in large crystals, and a black glossy mica (crystals \(\frac{3}{4}\) by \(\frac{1}{2}\) inch) imbedded in the felspar and accompanied by quartz.

The following analyses will show the chemical character of

these minerals:-

Felspar of Canton Granite.

		Per cent.	Atoms.
Silica		64.48	1.433
Alumina		19.12	0.367 0.274
Peroxide of iron		0.56	$0.367 \\ 0.007$ 0.374
Lime		0.45	0.016
Magnesia		trace	0.996
Potash		12.52	0.266
Soda		3.24	$0.266 \atop 0.104$ 0.386
Loss by ignition		0.16	
		100.53	Witnesself be one

From the preceding analysis may be deduced the following relation among the atoms of silica, peroxides, and protoxides:—

The Rev. S. Haughton's Notes on Mineralogy.

Silica .			1433	4.00
Peroxides			374	1.04
Protoxides			386	1.08

From which it is plain that this felspar is orthoclase.

The analysis of the black mica is as follows:—

Black Mica of Canton Granite.

1	er cent.	
Silica	35.50	0.789
Alumina	20.80	$0.400 \\ 0.246$ 0.646
Peroxide of iron	19.70	0.246
Lime	0.56	0.020
Magnesia	4.46	0.223
Protoxide of iron	7.74	0.215 0.699
Protoxide of manganese	1.70	0.047
Potash	9.00	0.191
Soda	0.10	0.003
Loss by ignition	0.25	
	99.81	

From the preceding we obtain, in atoms,-

ALL THE REAL PROPERTY.			Atoms.	Oxygen ratio.	
Silica .			789	789	
Peroxides			646	646 7 870	
Protoxides			699	$\binom{646}{233}$ 879	

from which may be deduced the following:-

$$\left\{ 26(3\,\mathrm{RO}) + 74\,\mathrm{R}^2\,\mathrm{O}^3 \right\} + 90\,\mathrm{SiO}^3,$$
$$\left\{ \frac{1}{4}(3\,\mathrm{RO}) + \frac{3}{4}\,\mathrm{R}^2\,\mathrm{O}^3 \right\} \mathrm{SiO}^{\frac{9}{10}}.$$

These formulæ represent the analysis, and are sufficiently near to the formulæ for the Lepidomelane of Soltmann, and of the black uniaxal micas of Donegal and Leinster, to render it probable that they all are varieties of the same mineral. To show their analogy and difference, the following comparison may be useful:—

	Atoms of				
Depui.	Silica.	Peroxide.	Protoxide.	Water	
Lepidomelane	831	569	551	66	
Black mica of Ballyellin*.	790	624	538	477	
Black mica of Ballygihen*	804	647	515	433	
Black mica of Canton	789	646	699	139	

^{*} Quart. Journ. Geol. Soc. London, vol. xv. p. 129.

The mineralogical formulæ of the four minerals are :-

I. Lepidomelane:

$$\left[\frac{25}{100}(3\,RO) + \frac{75}{100}\,(R^2\,O^3)\right] SiO^{3^{\frac{11}{10}}}.$$

II. Black Mica of Carlow:

$$\left[\frac{23}{100} \left(3 \, RO\right) + \frac{77}{100} \left(R^2 \, O^3\right)\right] \! SiO^{3^{\frac{10}{10}}} \! .$$

III. Black Mica of Donegal:

$$\[\frac{21}{100} (3 \,\mathrm{RO}) + \frac{79}{100} (\mathrm{R}^2 \,\mathrm{O}^3) \] \mathrm{SiO}^{\frac{10}{100}}.$$

IV. Black Mica of Canton:

$$\left[\frac{26}{100} \left(3 \, RO\right) + \frac{74}{100} \left(R^2 \, O^3\right)\right] SiO^{\frac{9}{10}}.$$

It appears to me that the preceding formulæ, representing black micas from Russia, Ireland, and China, balance around a mean or average formula, which may be regarded as the type species of this mineral; viz.—

$$\[\frac{25}{100} (3 RO) + \frac{75}{100} (R^2 O^3) \] SiO^3.$$

This abstract or theoretical black mica probably exists only as an idea or conception in our minds, and may not have a concrete development in any place; but it must be regarded as an essential constituent of the original granite formed in the astronomical epoch by the cooling of our globe. All our researches tend to prove that there is an original or type-granite, characteristic of the azoic epoch of the earth's history, marked mineralogically by the presence of four important minerals,—

- 1. Quartz;
- 2. Orthoclase felspar;
- 3. Black mica;
- 4. White mica;

and marked chemically by the abundance of potash and the absence of lime.

Trinity College, Dublin, March 10, 1859. and off as belonger of heartfailes directly thereto so and the tides or consequent by and mixed on a representation of sold and the second of the ad government of the course bearing the sounds