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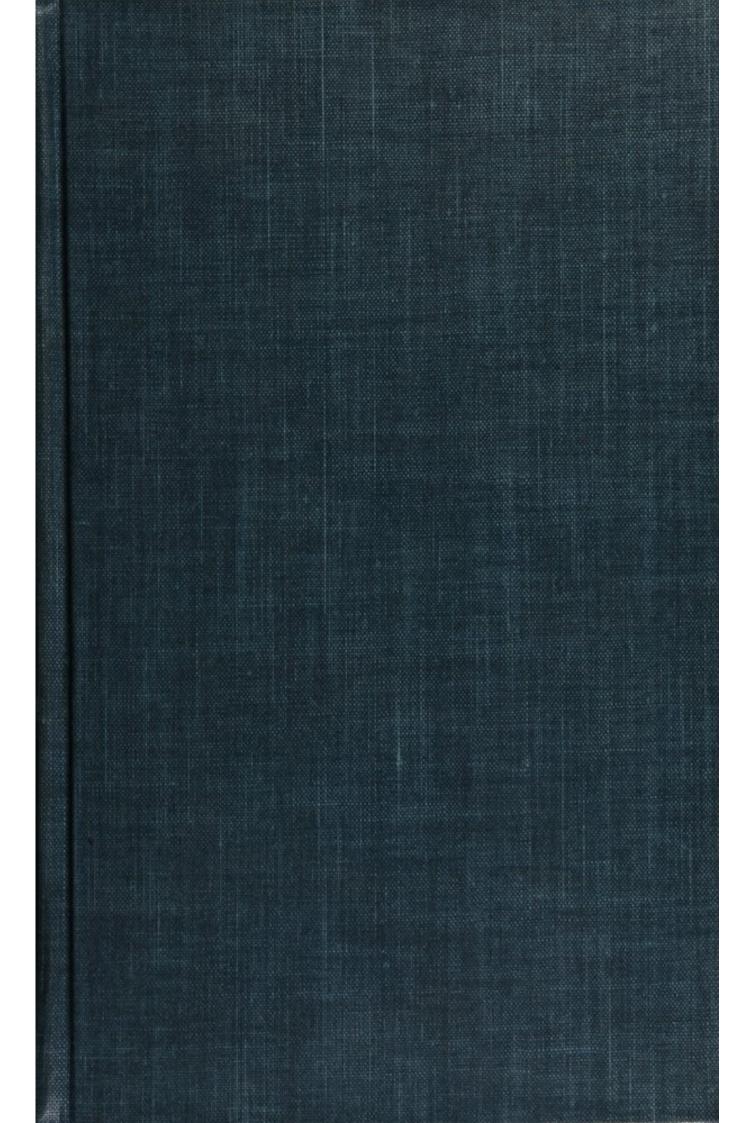
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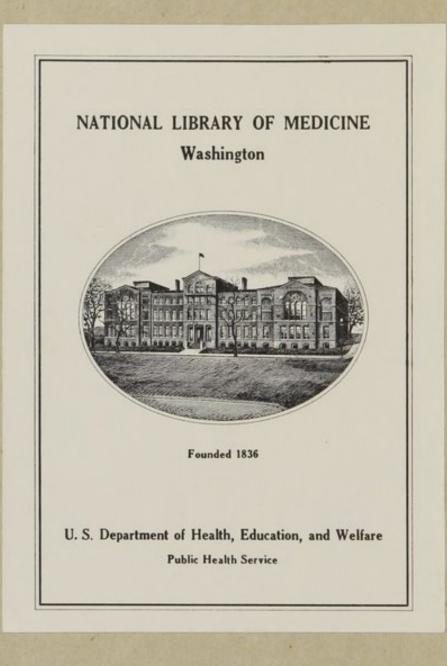
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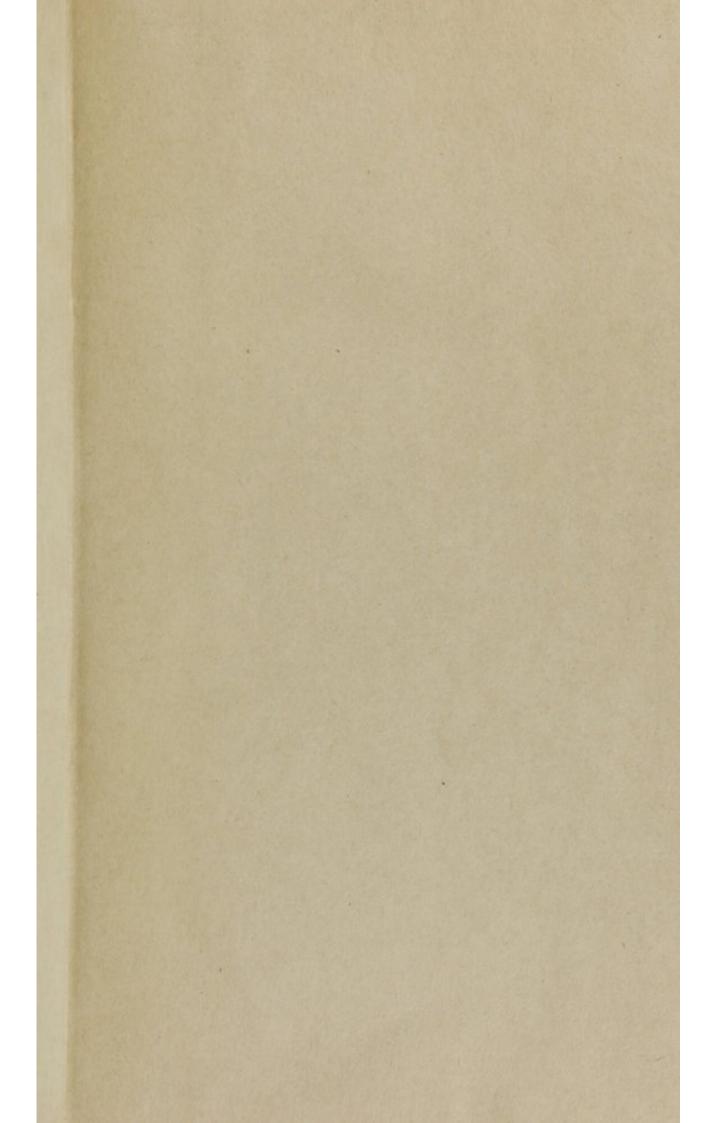
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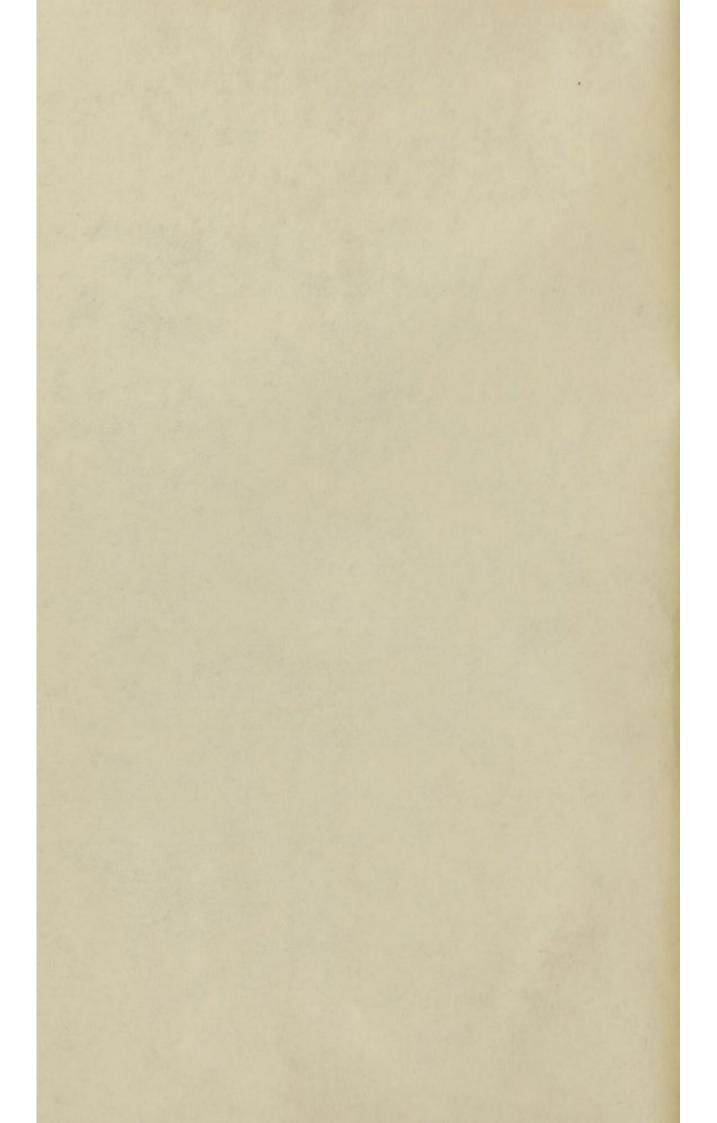


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A SHORT

# DISSERTATION

#### ONTHE

# Jamaica Bath Waters:

To which is prefixed,

AN

# INTRODUCTION

# CONCERNING

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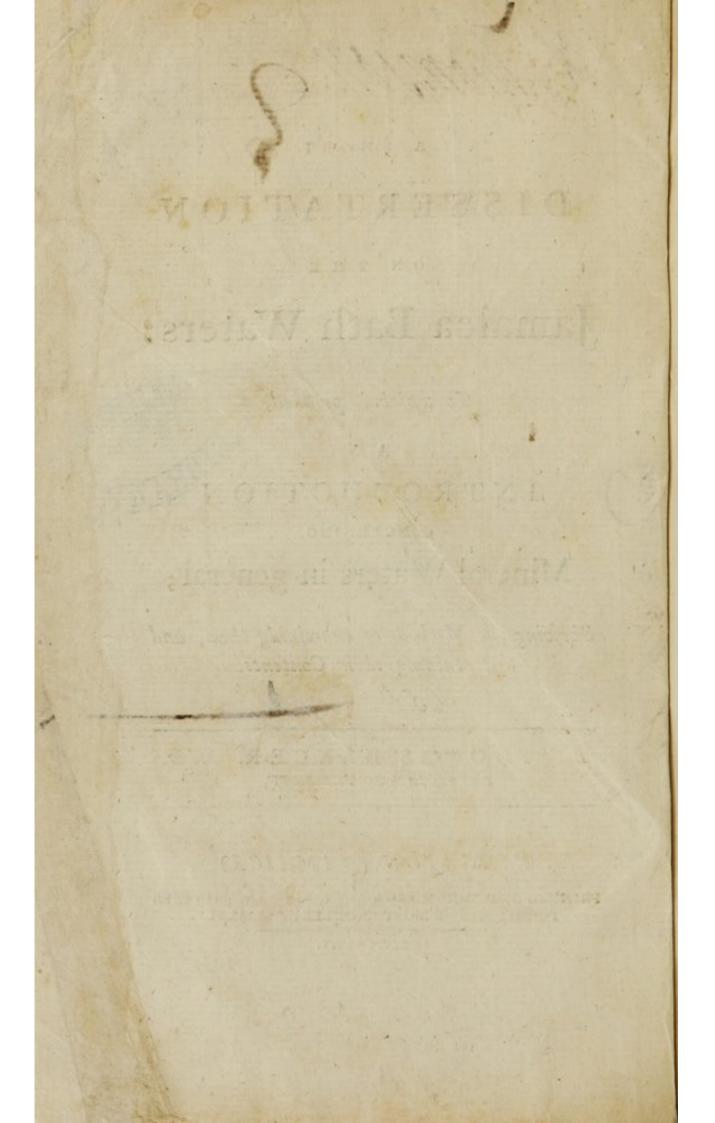
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BY THOMAS DANCER, M.D. PHYSICIAN TO THE BATH.

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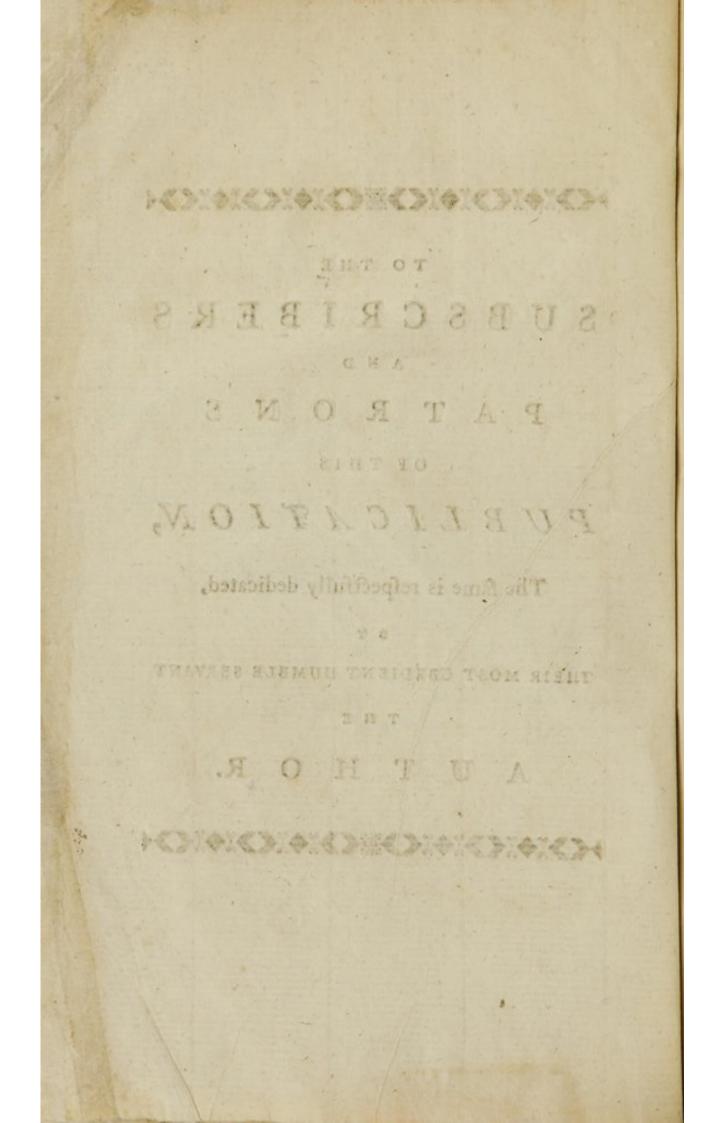
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# PREFACE.

nd enecles, yet it certainly tends to

Although the chemical analyfis of

THE virtues of the Bath Waters have been eftablished by the experience of almost a century, yet very little has been done to ascertain their particular nature. The account given by Dr. *Brown* in his history of Jamaica, as well as that in the history published in 1774 (which are the only ones that I have ever seen) are both not only defective, but extremely erroneous. What may have been done by my predecess, or others, in the investigation of these waters I am totally unacquainted with, as their experiments have never come to light:

# ( xviii )

I hope therefore that my humble attempts may not be unacceptable to the public.

Although the chemical analyfis of mineral waters may not be adequate to a full explanation of their virtues and effects, yet it certainly tends to throw confiderable light on their operation, and to direct to a more fafe, certain, and beneficial application of them. It is even useful to know what a mineral water does *not* contain, as will be feen from the miftaken notions that have been entertained concerning the waters here treated of,

To make the fubject univerfally intelligible, I have, in an introductory part, given a few chemical fketches relating to it; in which I can pretend to no novelty or merit, unlefs from the mode of arrangement, and from having avoided, as much as poffible, technical obfcurity.

( xix )

I might have enlarged much more than I have done upon the uses of the Bath waters, but it would have been impossible to particularife difeafes and fymptoms with fo much nicety, as to preclude the neceffity of advice, or make a book ferve the place of a phyfician; but I prefume I have faid enough to fhew their extensive utility, and to explain in fome degree their nature and operation. Could I have told how to make the publication more useful, and more deferving of the liberal patronage it has met with, no pains should have been wanting; but having done my beft, I have only to add my grateful acknowledgments, and to pray that the most favourable construction may be put on my endeavours.

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# I. Of the Introduction.

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# SECT. 1. On Chemical Bodies and their Affinities.

- 2. On the Impregnations, or Contents, of Mineral Waters in general.
- 3. On the Methods of examining and analyfing Mineral Waters.

# II. Of the Differtation.

SECT. 1. History and Analysis of the Bath Waters.

> 2. The Use and Administration of the Bath Waters in the Cure of Diseases.

# INTRODUCTION;

#### CONTAINING

A fhort Effay on Mineral Waters.

# SECTION I.

Medica, have from the earlieft ages engaged the attention of mankind: As they are various in their nature they afford a remedy for various diforders, but are particularly beneficial in complaints of the chronical kind, where all other medicines frequently fail. The improvements which have been

lately made in chemistry have enabled phyficians fo fully to explore and afcertain the nature of many of the mineral waters, as to imitate them with great exactitude; and, as fome of them owe their efficacy to a volatile principle that fuddenly escapes notwithstanding every means that can be made use of to detain it, the \* artificial impregnations of . waters may in a few inftances be preferable to those formed by nature; that is, when they cannot be used on the spot, or where they cannot be obtained in a perfect state, viz. without having previoufly fuffered fome change, if not lofs of their properties: But, in general, nature prepares this clafs of remedies much more perfectly in her laboratory than the chemist can do in his; for though we may by analyfis detect the feveral principles of a mineral water, yet we cannot always again

(2)

\* Should any of my readers be defirous of being inftructed in the method of preparing the mineral waters artificially, I muft beg leave to refer them to Dr. *Priefly*, or to a late pamphlet by Dr. *Elliet*, in which is explained the particular manner of imitating almoft all the moft ufeful mineral waters, whether faline, chalybeate, or fulphureous. — The apparatus for this purpofe, as invented by Dr. *Priefly*, and improved by Dr. *Naoth* and others, may be bought of the different druggifts in this ifland. combine those principles in the mode they before existed, and so as to give them the fame properties. Mineral waters will therefore ever retain their importance; and as they are, in some difeases of the human body, the most efficacious kind of remedy, so they are at the fame time the most agreeable one, and can with less inconveniency or difgust be persisted in, for a due length of time, than any other whatsoever.

(3)

Among the great number of mineral waters found in different parts of the world, I may with great fafety affert, that there cannot be a more valuable one than that which is the fubject of this publication, or that poffeffes greater virtues in the cure of fundry difeafes; to make the nature and effects of which better underftood, it will be neceffary to premife fomething concerning *mineral waters in general*, and of the methods made use of in examining them: I hope to explain the fubject fo as to render it intelligible to those who have never before ftudied it, and have no previous acquaintance with chemistry; in order to which, we must take a slight \* survey of chemical bodies, and of their several affinities or relations to each other.

(4)

# A General View of Chemical Bodies.

A LL the bodies in nature, confidered in relation to their chemical properties, are divifible into fix classes, viz.-1. Salts, -2. Inflammable Substances, -3. Earths and earthy Substances, -4. Metals and metallic Substances, -5. Water, -6. Air and aerial Substances.

## I. SALTS.

THESE are defined by Chemists to be fufible with beat, fapid in the mouth, and foluble in water; but the two last mentioned pro-

<sup>\*</sup> The feience of chemistry has such an extensive connection with the arts, and with the other branches of natural knowledge, that I flatter myscif the following outlines may be useful not merely in relation to mineral waters, but to many other subjects that require investigation—as, the process of sugar boiling, distillation of rum, manufacture of indigo, manufacture of pot-ash, &c. &c. all which depend directly on chemistry, and certainly require fome share of chemical knowledge.

perties, conftitute their most general character. It is not neceffary in such a sketch as this to study much precision; I hope therefore the chemists will not complain of my definitions being inaccurate and incomplete.

(5)

The following table exhibits a general view of falts, and requires only a fimple infpection to be perfectly underftood.

# TABLE OF SALTS.

## I. SIMPLE.

1. Acid.	Fossile.	Svitriolic, viz. Oil of Vitriol Nitrous, – Aqua Fortis Muriatic, – Spirit of Salt
		{Acetous, viz. Vinegar {Tartarous, — Tartar
2. Alkali.	Fixed.	{Vegetable, viz. Pearl Afkes {mild or Saltof Tartar cauftic Foffile, — Barilla
apinu or	Volatile.	{Volatile, viz. Sp. of Hartshorn {mild or Sal. Volatile caustic

#### H. COMPOUND.

1. Neutral, composed of Acid and Alkali.

2. Earthy, composed of Acids and Earths.

3. Metallic, composed of Acids and Metals.

The above view of the fimple falts, I think, can require no fort of explanation ; it is obvious from the arrangement, that they are generically diffinguished by being either acid or alkaline; the acid falts are either fosfile, or vegetable: Again, the foffile acids are of three kinds, vitriolic, nitrous, and muriatic ; the common names of which are fubjoined. The vitriolic acid is the fame thing as oil of vitriol; the nitrous acid is aqua fortis, or spirits of nitre; and the muriatic acid is spirit of falt. In the fame manner the alkaline falts are divided and fubdivided. By arranging fubftances in this method, the memory is much affisted, and descriptions rendered unneceffary. I shall proceed to enumerate the other compound falts in the fame manner.

# I. NEUTRAL SALTS.

ARE those which are formed by the union of an acid and alkali; having the properties of neither, but making (as chemists speak) a tertium quid; that is, a new substance fui generis, or of a distinct kind.

The following table exhibits a catalogue of most of the neutral falts, and shews the manner in which they are formed. It is to be thus understood ; viz. the vitriolic acid in the first column, combined with the vegetable alkali in the fecond, makes the neutral falt, in the third column, called vitriolated tartar :- Again, the vitriolic acid, with the fossile alkali in the next line, forms Glauber's falt, &c. &c.

# TABLE OF NEUTRAL SALTS.

ACID. ALKAL	I. NEUTRAL SALTS.
-------------	-------------------

I. Vitriolic.	{Foffile, -	Vitriolated Tartar Glauber's Salıs Vitriolic Ammoniac
2. Nitrons.	{Vegetable, - Foffile, - Volatile, -	Nitre, or Salt-petre Cubic Nitre Nitrous Ammeniae
3. Muriatic.	Vegetable, - Foffile, - Volatile, -	Digestive Salt Common Salt Crude Ammoniac
		Diuretic Salt ochelle, or Seignette do. Spirits of Mindererus

Table of fuch Earthy and Metallic falts as are found in Mineral Waters:

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#### I. WITH THE VITRIOLIC ACID, AND

SCalcareous earth, { Selenites, or Gyp-fum 1. Earths. Eofom Salts, or bit-Magnefia, ter pu ging do. Clay or particular earth, - Allum Zinc, White Vitriol 2. Metals SIron, Green Vitriol Copper, Blue Vitriol

II. WITH THE NITROUS ACID, AND Calcareous earth, - Calcareous nitre. (doub ful)

III. WITH THE MURIATIC ACID, AND Calcareous earth, Fixed ammoniac.

I have judged it needlefs in this place to enumerate any of the metallic falts, except fuch as are found in waters, and of which it was proper to know the composition. The next general class, is that of

# II. INFLAMMABLE SUBSTANCES.

THESE are fuch bodies as (without any definition) are known to take flame and confume with heat. Under this head are included all oils, refins, sulphur, ardent spirits, &c.

Q.

The inflammability of bodies depends on a principle they contain called by chemifts phlogifton; which principle exifts in all bodies, but in different quantities and in different flates of combination; whence their difference in refpect of inflammability and various other properties. So many phenomena in nature depend on this pervading principle, that the fludy of it conflitutes a main branch of chemical philofophy. We fhall fee hereafter what connection it has with the fubject of mineral waters, but at prefent we need take no further notice of it.

## III. METALLIC BODIES.

THE well known properties of these are their great specific gravity and fusibility with certain degrees of heat. All the metals contain a great portion of phlogiston; deprived of which they lose their metallic form, and are reduced to a fort of earth called calx, thus e, g. ceruste is the calx of lead. The metals, in their fossile state, are found in various forms :

( 10 )

1. Virgin or pure.

- 2. In ores combined with earths.
- 3. Mineralized with fulphur and arfenic
- 4. Diffolved by acids or by other means, and mixed with mineral waters.

## TABLE OF EARTHS.

#### I. ABSORBENTS.

The character of these is their effervescence with acids : they are of two kinds:

- Calcareous, viz. chalk, &c. which convert by heat into quick-lime, and become cauftic. This is effected by the expulsion of fixed air; which as foon as they imbibe again they lofe their caufficity and become mild.
- 2: Magnefia, which does not convert into quick-lime, or become cauftie as the preceding.

#### **II. CHRYSTALLINE OR FLINTY.**

These are hard and strike fire with steel: are either

- 1. Apyrous, *i. e.* indeftructible by heat as fome of the precious ftones, or
- 2. Vitrescent or fusible, running by heat into glass; such are common flints, fand, &c.

#### III. ARGILLACEOUS,

Or Clays, which do not effervesce with acids, nor melt into glass, but are unchangeable in the fire.

#### IV. MICACEOUS.

Stones of a fhining laminated or fibrous texture. They do not effervesce with acids, nor strike fire with steel, but are either

- 1. Talcky, which in a ftrong heat are vitrescent, or
- 2. Albertos, Amianthos, &c. indeftructible by heat.

N. B. Gypfum and gypfeous matters, which are fometimes confidered under the head of earths, are more properly earthy falts. (Vid. table of earthy falts)

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Spars are stones of very different natures, and arrange under abforbent earths, micaceous earths, &c.

Marles are absorbent earths.

# V. WATER.

THE properties of water are too well known to require any defcription. Pure elementary water is hardly to be obtained, but it may be confidered either as fuch, or elfe impregnated with foreign matters, as, particularly, in mineral waters.

## VI. AERIAL MATTERS.

AIR, as well as water, has by all philosophers been confidered as an elementary substance; but this seems to be brought into some doubt by the late celebrated experimenters in chemistry, who have discovered many species of air, and distinguish that by which animal life is supported, as atmosphe. ric or refpirable air : The other kinds more particularly deferving of note are,

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- 1. Fixed or mephitic air +.
- 2. Phlogifticated air, which is either
  - 1. Inflammable, taking fire, or
  - 2. Non-inflammable, extinguishing flame.

These kinds of air are variously produced, and exhibit wonderful phenomena; but this is not a place to enter into any further confideration of that curious subject. \*

Having now pointed out the general and diffinguishing characters of the feveral classes of chemical bodies, I shall proceed to explain the *affinities* of these, or the relations which they have to each other; upon which depend all the processes in chemistry, as well as many of the most important operations in nature.

> † Mephilic Air is demonstrated to be an Acid. (Vid. BEWLEY apud PRIESTLY.)
> \* Vid. PRIESTLY and others.

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## On Chemical Affinities.

HE affinities of bodies fignify their difpolition to unite, or the attraction they have for each other, which is owing to fome unknown relation between them : this attraction is greater between fome than others, and is therefore called elective : - For example, acids have an attraction both for alkalies and absorbent earths, but a much ftronger attraction for the former than the latter; in consequence of which, if an abforbent earth be first diffolved in an acid. and afterwards an alkali be added, the acid having a greater affinity to the latter will unite with it and deposit the earth. Again, acids in general have a greater attraction for earths than for metals, fo that if you add an earth to the folution of a metal in any acid. the metal will be precipitated, It is in this way that various combinations and decompofitions are effected, and that we are enabled to analyze bodies and detect their principles. I need not here give any further illustration of the fubject, as the application of it to the investigation of mineral waters,

which this is intended to explain, will exhibit a perfect view of it. All that is further required in this place, is, to point out the known laws of affinities, as they have been difcovered in the courfe of experiment and in the practice of chemistry: these have been reduced into tables: the following one is the most approved, as being corrected by some of the greatest philosophers of the age.

The table is thus to be underftood, viz. Under the head of acids is placed, firft, phlogifton, then alkali mild, &c. the meaning of which is, that acids have an attraction for those different substances according to the order in which they are placed, *i. e.* they have a greater attraction for phlogiston than alkalies, for alkalies than absorbents, earths, &c.

#### Table of Affinities or Elective Attractions.

#### I. ACIDS IN GENERAL.

Phlogiston
 Quick-lime, or caustic
 Fixed alkali,
 caustic mild
 Magnesia

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5. Calcareous earth mild
6. Volatile alkali
7. Earth of alum
7. Earth of alum
7. Pure or precious

7. Metallic fubstances metals. in general

#### 2. VITRIOLIC ACID.

Vid. acids in general.	6. Silver
1. Zinc	7. Tin
2. Nickell	8. Lead and Mercury
3. Cobalt	9. Volatile Alkali
4. Iron	10. Earth of Alum,
s. Copper	as a price prover pilled

#### 3. NITROUS ACID.

Vid. acids in gen.	5. Copper
1. Zinc	6. Arfenic
2. Lead or Tin	7. Mercury
3. Iron	8. Silver
4. Bifmuth and An-	9. Platina.
timony	Life tor as

#### 4. MURIATIC ACID.

Vid. acids in gen.	5. Copper
1. Zinc	6. Lead
2. Iron	7. Silver
3. Tin	8. Mercury
4. Regulus of Antim.	9. Gold.

#### 5. VEGETABLE ACID.

I. Iron

2. Copper.

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#### 6. MEPHITIC AIR, OF MEPHITIC ACID.

Metallic fubstances	Fixed alkali
Caustic calcar. earth	Earth of Magnefia
or quick-lime	Volatile Alkali

#### 7. ALKALIES IN GENERAL.

Vitriolic acid Nitrous acid Muriatic acid

Vegetable acid Oils. Phlogiston, or Sulphur

#### 8. CALCAREOUS EARTHS.

Vitriolic acid Nitrous acid Sulphur

Muriatic acid

Q. PHLOGISTON.

Nitrous acid Vitriolic acid Muriatic acid

Metallic bodies Alkalies

#### IO. SULPHUR.

Fixed alkali Quick-lime Iron Copper Lead Tin

Silver Antimony Mercury Arfenic Volatile alkali

### 11. METALLIC SUBSTANCES in general.

Muriatic acid Vitriolic acid Nitrous acid

Phlogiston Fixed air

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12. GOLD, ---- Aqua regia

#### 13. SILVER.

Muriatic acid Vitriolic acid Nitrous acid Lead, Copper

14. MERCURY.

Muriatic acid Vitriolic acid Nitrous acid Gold Silver Lead, &c.

#### 15. IRON.

Vitriolic acid Muriatic acid Nitrous acid Regulus of antimony

16. COPPER.

Vitriolic acid Muriatic acid Nitrous acid Vegetable acid

I flatter myfelf that the preceding sketches are sufficiently intelligible to every reader who may think it worth his pains to give the slightest attention to them; and that they will serve to render what follows on Mineral Waters easily comprehended.

#### CHAPTER II.

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#### On the CONTENTS of MINERAL WATERS.

A S this is not a work intended for the perufal of the learned, I do not think it requifite to enter largely into the inveftigation of the fubject, but shall content myfelf, according to the plan I fet off with, in giving a plain and easy introduction to it. Such of my readers as may be induced to study it more fully, and to extend their refearches into this pleasing and useful science, must have recours to other authors; amongs whom I would particularly recommend Dr. Faulkener, one that in this part of my work I am particularly indebted to.

The lift of matters contained in mineral waters was formerly much larger than at prefent, later experience having demonstrated the impossibility of many impregnations fupposed by the old chemists, particularly the nitrous and ammoniacal falts, of which we find fo frequent mention in former writers : The acids of these falts have never (like the vitriolic) been found separate, and if they were, they would still want a proper base, as neither the vegetable or volatile alkali have ever been discovered in a fossile state.

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TABLE exhibiting the feveral CONTENTS of MINERAL WATERS.

### I. SALTS.

1.	Acid, -	Vitriolic in a separate state.
2.	Alkali, -	Fossile Alkali.
3.	Neutral, -	Glauber's Salt, Common do.

#### II. INFLAMMABLES.

I. FOSSILE OIL, VIZ. PETROLEUM.

1. Separate, or float-	2. Combined with an
ing on the furface	Alkali in form of
of waters.	a Soap.

#### 2. SULPHUR.

1. Per se, diffused or	3. Combined with
fufpended.	Calcareous Earth
2. Combined with	in a cauftic ftate,
Alkali, forming	forming Calcareous
Hepar Sulphuris.	Hepar Sulphuris.

III. EARTHS and EARTHY SALTS.

I. CALCAREOUS EARTH.

1. In a caustic state,	ed Air.	
or deprived of fix-	2. Combined	with

forming Selenites. 3. Combined with

the Vitriolic Acid, Sulphur, forming Hepar Sulphuris.

#### 2. EARTH OF MAGNESIA diffolved.

1. By Vitriolic Acid, 2. By Muriatic Acid, forming Epfom conftituting a Salt that has no name. Salts.

#### 3. EARTH OF ALLUM.

Forming with Vitriolic Acid the Salt of Alum

#### IV. METALLIC MATTERS.

I. COPPER diffolved.

By the Vitriolic Acid Blue Vitriol. in the form of

#### 2. IRON diffolved.

1. By Fixed Air. form of Green Vit. 2. Vitriolic Acid in 3. Hepar Sulphuris.

3. ZINC diffolved. Vitriolic Acid, in form of White Vitriol.

### V. AERIAL MATTERS.

1. Common, or at- ed Air. mospheric Air, 3. Phlogisticated, ot 2. Mephitic, or fix- inflammable Air.

The above table comprehends, on the authority of the best Chemists, all the known impregnations of mineral waters. Many other matters, as I have faid, were formerly included; but I fuggested the reason why in particular the nitrous and ammoniacal falts cannot have an existence in mineral waters. For fimilar reafons none of the metals, except Iron, Copper, and Zinc, are ever found in waters, viz. because there are no foffile menftrua capable of diffolving them, the vitriolic acid and hepar fulphuris not acting on the other metals except by the affistance of heat, or under some other circumftances that are not supposed to take place in the bowels of the earth.

Arfenic has however, in particular, been conjectured to be prefent in fome waters; but in the opinion of the beft authors, this poifonous femi-metal is only foluble in water when deprived of its phlogifton or fulphur, in which ftate it cannot be found in ores. The existence of an \* actual Sulphur in waters has been long questioned, but it feems on very good evidence, as we shall see when we come to the manner of analysing waters, to be really found in some few of them. The aerial impregnations of waters are of late difcovery, and have thrown an entire new light upon the subject, as they account for several combinations among the fixed matters, as well as for the medicinal efficacy of many of the waters of principal note.

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I shall here subjoin a catalogue of the most important mineral springs in the various parts of the globe, with a summary of their contents, according to the latest and most approved analysis; which may serve not only to gratify curiosity, but to convey a general idea of their medicinal qualities and uses in the cure of diseases.

\* Some chemists have supposed the Sulphur found in mineral waters did not exist in the waters as a sulphur, but is formed afterwards by a new usion of the component principles.

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### CATALOGUE OF THE MOST CELEBRATED MINERAL WATERS.

Aix-la-Chapelle.—Those Waters contain an actual Sulphur, and are very hot.

Bath (English) Waters.—Contain fixed Air, phlogifticated Air, Hepar Sulphuris with Quick-lime, and a little Iron—

Vid. Faulkener, Nooth, Priestly. Bath (Jamaica) Vid. Analysis following. Barcges.—Hepar Sulphuris, Bitumen, and Sea Salt.

Buxton .- Similar to Briftol.

Briftol.—Calcareous Earth, marine Salt, and fome Sulphur. Vid. Elliott.
Caroline Baths (in Germany).—Pure alkaline Salt, calcareous Earth, and a little Iron.

Vid. Hoffman. Cheltenham. — Epfom, or bitter purging Salts. Hartfell.—Martial or green Vitriol in confiderable quantity. Lewis. Iflington.—A Chalybeate Water.

Moffat (in Scotland). - Sulphur and marine Salt. Plummer, Moffat (in Jamaica). — The fame as the above.

Vid. Analyfis by Dr. Mitchell. Pyrmont. — A great deal of fixed Air, by which Iron is diffolved in it. Brownrigg. Spa.—Fixed Air, Ocher, Sulphur.

Sedlitz (in Bohemia). - Epfom or purging Salts.

Teplitz (in Germany).—Is a Spring of pure hot Water. Hoffman.

Tilbury.—Alkaline Salt. Lewis.

- Wicklow (in Ireland). Copper, or blue Vitriol.
- Smith's (in Jamaica).—A martial Vitriol, like the Hartfell Spa. Dr. Clarke found that it is also impregnated with fixed Air, and sufpects the Iron to be kept diffolved by this as well as the Acid.

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Comparative View of the Temperature of the several Hot Springs in different Parts of the World.

Aix la Chapelle,	Farenhei	ŧ.	
Bain de	e l'Empereur, 12	27	
Bain de	es Pauvres, 11	12	
Briftol Hot-well	7	76	
Buxton, -	5	59	
Mallow, -	6	53	
Bath (in England.) Cr	rofs Bath - 11	2	
Ki	ing's do 11	6	
Ho	ot do. – 11	6	
Ho	ot Spring - 11	6	
Barcges, highest,	II	2	
(Jamaica)	12	6	
Dax (in Guienne), faid	l to be - 14	.0	
Japan Hot Waters, faid to be near the boil-			
ing point.	internation of the state		
* Musq. Sb. Waters, near the boiling point.			

\* I am informed by my friend *Major Lawrie*, Superintendent of the Mufquito-fhore, (a Gentleman well known for his active fervices in the defence of that valuable country, where he refided many years, and employed a very attentive obfervation on every nfeful and curious object,) that fo great is the heat of this foring that either animal or vegetable fubflances, placed in the refervoir into which it runs, are in a moderate fpace of time rendered fit for ufe, and that the Indians accordingly frequently drefs their food in it. It appears neverthelefs to contain no mineral impregnation.

## Of the Method of Analyfing Mineral Waters.

I T has been before remarked, that there is no water to be obtained perfectly pure. That which is most free from impurities is fnow water, next rain water \*, and then the waters of limpid rivers; but they all contain more or lefs of earthy and faline matter, even after repeated distillations, which has favoured the hypothesis of the ancient philosopher, that earth and all things were formed of water.

The first thing to be attended to in the examination of any water is not its pellucidity, or apparent purity, but its weight, the purest water being always the lightest. The gravity of water is ascertained either by weighing a certain measure of it in a well adjusted pair of scales, or by the water posse called the hydrometer. Where proper scales

 $<sup>\</sup>dagger$  A fufpicion has been entertained, that the rain water of this country contained vitriolic acid; to determine this I collected fome in a glafs veffel during a heavy flower, which I tried with the folutions of filver, lead, and mercury in the nitrous acid;— With the firft, there appeared a light purple tinge, which at length became reddift; with the lead it dropped a white precipitate, and the fame with mercury, which I allowed to remain for 24 hours, without obferving any change to yellow, as floud have happened had there been any vitriolic acid prefent.—Vid. page 16.)

and weights, or the above inflrument cannot be procured, another method has been propofed for comparing the relative weight of the water of any mineral fpring with the other waters in the neighbourhood, viz. by tincturing the water with faffron, or any other colouring matter that makes but little addition to its own weight, and then inferting the bottle filled with it in a veffel of the water you want to compare it with. If the coloured water is fpecifically heavier than the common water, it will of courfe run out and communicate a tincture to the water in which the bottle is immerfed.

### Table of the Weights of different Waters.

	and the loss of the second	Oz.	Dr.	. Gr.
Of distilled water,	I pint weighs	15	I	50
Rain water,		1		40
Spring water,	ditto	-		129
Sea water,	ditto		-	20

After having afcertained the apparent and fenfible qualities of any water with its relative weight to other water, or to pure water, we may proceed to try it by fome of the most ordinary and general methods, as follow:

- 1. If it lathers well with Soap.
- 2. If it curdles milk.
- 3. If it effervesces on the addition of acid or alkali.
- 4. If it changes the colours of blue flowers, infufed in it for fome time, either to red or green.
- If it turns milky on adding to it a few drops of any of the following folutions, viz.
  - 1. A folution of hepar fulphuris (i. e. liver of fulphur,) in water.
  - 2. A folution of faccharum faturni, (i.e. fugar of lead,) in diftilled water.
  - 3. A folution of lead in aqua fortis.
  - 4. A folution of filver in ditto.
  - Or, laftly, if it strikes a black or purple colour with the powder or tincture of Galls.

By these general trials we can discover not only the degree of purity in a water, but if it contains any mineral impregnation we can form some probable conjecture as to the nature thereof, to direct us in our further refearches. The explanation of the above modes of trial will be feen prefently as we go on to point out more particularly how the various fubftances contained in mineral waters may be feverally detected.

### I. To discover the Aerial Impregnations of Waters.

I. FIXED AIR,—first, gives to waters a sparkling appearance and poignant taste, fimilar to Champaigne or Perry, which liquors, as well as many others, contain a confiderable quantity of it \*.

Secondly, Waters containing fixed air added to lime water precipitates the calcareous earth which was only rendered foluble in water by being deprived of its fixed air.

Thirdly, Waters containing fixed air have the fingular property of diffolving iron.

<sup>\*</sup> Waters containing fixed air are difficultly kept, the bottlea burfting; Mr. SHACKLEFORD lately returned from the continent of North America, informs me, that, at the famed plains of Saratoga, he faw, in company with his fellow traveller Dr. JAMES, (a gentleman well known for his medical abilities) a mineral water that contained fo much elaftic air, that bottles only about half filled were broke by it.

II. INFLAMMABLE AIR in waters, as alfo fixed air, may be collected, by tying an oiled bladder over the neck of a bottle containing the water. Inflammable air is lighter than common air, has a peculiar fmell, and takes flame \* when it approaches any burning body.

A more accurate method of trial may be feen in the works of that learned philosopher Dr. *Priefly*, to whom I must refer my readers for more ample instruction respecting this and the several other species of phlogisticated air.

### To discover an acid in Waters.

THE only acid found in waters in a feparate ftate (vid. table,) is the vitriolic, and most commonly that species of it called the volatile fulphureous, from the smell, &c. which it has from the phlogiston with which it is combined.

In whatever form the vitriolic acid enters

<sup>\*</sup> The damps of mines are either fixed or inflammable air.

into waters, it may be detected in the following manner:

1. By the infusion of blue flowers in the water, which, if there be an acid, will become *red* \*.

This laft, though a common one, is by itfelf not decifive, becaufe fixed air and alum in waters will effect the fame change.

2. By adding to the water a little alkaline falt, or a few drops of oil of Tartar.

If the acid be prefent in any quantity, it will excite an effervescence with the alkali.

3. By dropping into the water a tincture of foap in ardent fpirits, provided the water contains an acid it will unite immediately with the alkali of the foap, and the oil, which is the other ingredient in the compofition of foap, is feparated and renders the water milky: This ferves to explain the effect of hard waters in general in curdling foap.

\* Alkalies turn blue vegetables of a green colour, (vid. p. 17.)

4. By dropping into the water, a folution of hepar fulphuris, which is a combination of fulphur and alkali: The fame attraction, where there is an acid, takes place here as above in No. 3; the acid unites with the alkali, and the fulphur precipitated makes the water turbid and milky.

I observed that an aluminous impregnation gives to waters tried in the above manner most of the same appearances as the vitriolic acid in a separate state; but to distinguish aluminous water from those containing an acid, we may,

1. Repeat the above experiment on the waters after they have flood for fome time; if the changes produced were caufed by an alum in them, the fame refult will follow on making the fame trials; but the acid of waters being, as was before faid, generally the volatile vitriolic, this upon the water flanding for fome time flies off.

2. Waters containing an acid will effervesce with and disfolve magnesia, remaining perfectly clear, (provided only the neceffary quantity of magnefia is made use of,) but when they contain an alum, the addition of magnefia will cause a precipitation, the vitriolic acid having a greater attraction for that than the aluminous earth.

There remains one more teft, which ferves to difcover the vitriolic acid either in a feparate or combined state; this is an unfaturated \* folution of lead in the nitrous acid, which being added to any water that contains the vitriolic acid, this acid (viz. the vitriolic) having a greater attraction for lead than the nitrous, (vid. table of affinities, ) unites with the lead in place of the latter, and the fubstance formed by this union, called a plumbum corneum, being not foluble in water, disturbs the transparency of it. It is requifite in the above experiment that there should be a redundancy of acid in the teft, to prevent the lead from being precipitated by any calcareous earth the water may at the fame time contain.

\* Unfaturated, fignifies, not fo much as the acid is capable of diffolving.

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## To discover an Alkali in Waters.

HE alkali found in mineral waters we have faid is always the foffile, the vegetable being generated only on the furface of the earth.

An alkali in waters is difcovered,

1. By their turning the blue flowers to a green colour.

2. By their effervescing on the addition of an acid.

3. By their precipitating chalk from its folution in aqua fortis.

N. B. The aqua fortis fhould be faturated with the chalk, (*i. e.* contain as much as it is capable of diffolving.) or elfe the part of the acid that remains unfaturated will diffolve the alkali in the water fo that it will not precipitate the chalk.

4. By their precipitating an ochre from folutions of iron in the vitriolic acid, or from a folution of green vitriol in water, the acid of the vitriol having a greater attraction for the alkali than the metal, (vid. table of affinities.)

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As the prefence of a volatile alkali in waters is not admitted, it is needlefs to adduce any of the criteria. I shall however, for form's fake, observe, that,

1. A folution of corrofive fublimate in diftilled water is precipitated by the volatile alkali in a white, but by the fixed alkali in 2 red or brown powder.

2. If a water contains any quantity of volatile alkali, it will acquire, from copper immerfed in it, a fine blue tincture.

#### To discover the Neutral Salts in Mineral Waters.

#### I. GLAUBER'S SALT.

THIS fait is known,

1. By its tafte and the form of its chryftals, which are hexagonal: these may be easily obtained from waters containing it by a flow evaporation till a pellicle forms on the furface, and then placing the liquor at rest for the chrystals to shoot: the addition of a little spirit of wine towards the end affists the chrystallization.

2. By the folution of the falt in water coagulating when spirit of wine is added to it; the fpirits attracting a part of the moifture from the falt, it begins to refume its chrystalline form.

3. By its fufibility in a gentle heat, which is a property peculiar to this neutral falt.

4. By no precipitation taking place on adding an alkali to it when diffolved in water, which always happens to falts with an earthy bafe, as Epfom falt, &c. (vid. table of affinities.)

5. By the precipitation of a yellow powder (*i. e.* a turbith mineral) on adding a folution of mercury in the nitrous acid, the mercury leaving the nitrous acid to unite with the vitriolic.

## II. COMMON SALT, is known,

1. By the figure of its chrystals, which are cubes, and which may be obtained by long boiling, and then placing the liquor to cool.

2. By its decrepitation (crackling) on being placed on a hot iron. 3. By its decomposition and fuffocating fumes, on adding either the vitriolic or the nitrous acid, which have a greater attraction for the alkaline base than the muriatic acid, of which it is formed, (vid. affinities.)

4. Waters containing fea falt, or the muriatic acid, precipitate lead and filver from aqua fortis, the muriatic acid having a greater attraction for those metals than the nitrous.

5. The refiduum of waters containing fea falt, added to aqua fortis, conflitute an aqua regia capable of diffolving gold: It may be tried with gold leaf, or a mark on the touch flone.

Although the presence of the other neutral falts in mineral waters be fo very doubtful, I shall nevertheless point out the characteristic properties by which they may be distinguished.

TREFCUENCIAS INC. LAG

1. Nitre is known, 1st, by the peculiar form of its chrystals, -2d, by its deflagration for manner of burning, -3d, by its turning the flesh of animals red.

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2. Calcareous nitre, or nitre composed of an earthy base, does not deflagrate as common nitre, but blifters, and being subjected to a strong heat, the acid is expelled and the earth converted into quick-lime.—A solution of calcareous nitre is also exposed by the addition of alkali.

3. Common ammoniac may be known by its giving out the volatile alkali on adding a fixed one, the acid having a greater affinity for the latter: the volatile alkali is immediately perceived by its pungent vapours when applied to the noftrils.

## To discover Sulphur in Waters.

FEW of the waters which, from their fmell and other properties, have been called fulphureous, contain a *real* or *attual* fulphur, but either a hepar fulphuris, or elfe a fulphureous Gas, (*i. e.* the phlogifton) in fome mode of combination hitherto not very well underftood.

Waters that contain an actual fulphur, as it is only fufpended, not diffolved, generally deposit it about the fides of the veffels or refervoirs in which they are placed; but when prefent only in a fmall quantity, fo as not to be perceptible in the foregoing manner, the fand or mud of the fprings may be dried and thrown on a plate of hot iron; if there be any fulphur contained therein the fumes will prefently discover it.

Waters containing a *hepar fulphuris*, or fulphur united with an alkali, (in which form it is rendered foluble in water,) are diftinguished,

1. By their fætid fmell, which refembles that of the fcourings of a foul gun.

By their tarnishing the white metals,
 as filver, of a purple or black colour, and gold, of a deep yellow or copper colour.

3. By their lactefcence on the addition of acids, which by attracting the alkali precipitate the fulphur.

4. By their turning a folution of fugar of lead of a brown colour, and by their vapours rendering visible characters wrote with the above folution, which is the common fym-. pathetic ink. The phlogiston which these vapours contain partly revive the metal in the faccharum faturni, and thus make the characters visible.

Sulphur diffolved in water by means of calcareous earth in a cauftic ftate, (or quicklime) exhibits none of the above figns of an alkaline hepar fulphuris. It is detected however, by the addition of alkali, which precipitates the calcareous earth, and by forming a true hepar fulphuris, the waters then begin to affume the qualities enumerated in the preceding paragraph.

### To discover Fossile Oil in Waters.

**THIS** is generally found floating on their furface, but may be blended with the water by means of an alkali, or quick-lime, in form of a foap.

Such waters will curdle on the addition of acids, just the fame as hard waters do with foap.

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### . To discover Calcareous Earth in Waters.

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## THIS is foluble in water;

1. By being deprived of fixed air, or in the ftate of quick-lime, which is very well known from the manner of making lime water; from which the calcareous earth is again precipitated on reftoring fixed air, as, e. g. in breathing into the water through a tube.

There is a quantity of fixed air generated in refpiration \*, and this mode of trial may fufficiently ferve; but befides this there is another method made ufe of—Fixed air produced from the effervefcence of an acid with an alkali, or with chalk, may be received into a bladder and afterwards made to pafs through the water to be examined in a gentle ftream: if the water contains a calcareous earth, it will be precipitated, by the fixed air rendering it unfoluble.

\* Dr. Prießly supposes that the fixed air produced in refpiration is not thrown off from the lungs, but is precipitated from the common air by means of the phlogiston extricated in refpiration. The argillaceous, ochrey, and flinty earths can only be fuspended or diffused, and cannot therefore constitute any impregnation in waters.

### To discover Selenites in Water.

SELENITES, or gypfum, (vid. table of earthy falts,) which is the matter of all hard waters, is discovered,

1. By the well known effect of fuch waters in curdling foap.

2. By the evaporation of these waters thin laminous chrystals, like the scales of fishes, may be obtained, of a rough astringent taste, which are difficultly soluble.

3. By adding a folution of mercury in the nitrous acid a turbith mineral is formed, as before explained (vid. page 7.)

## To discover Epsom Salts, and Magnefia, in Waters.

EPSOM SALTS, composed of the vitriolic acid and magnesia, (vid. table,) are obtained from waters by evaporation and chrystallization, in the fame manner as Glauber's falts, from which they are eafily diftinguished.

1. By the precipitation of magnefia on adding an alkali: Glauber's falts having no alkaline bafe fuffers no decomposition.

2. By their not being fufible as Glauber's falts.

They are diffinguished from common falt,

1. By giving out no fumes on the addition of oil of vitriol, (vid. page 7.)

To discover Alum in Waters.

1. ALUM in waters shew many of the figns of an acid, (vid. page 16.)

2. By evaporating waters that contain an alum, they acquire a rough auftere tafte.

3. An alkali added to au aluminous water, precipitates the earth in *floculi*, or little eurdled clouds, (not in a powder) which, if too much alkali be not made use of, will be rediffolved by the superabundant quantity of the vitriolic acid which alum contains.

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### To discover the Metallic Contents of Mineral Waters.

**I**RON:—The common tefts of the prefence of iron are powdered galls, or tincture of galls in brandy, and the phlogifticated alkali \*. Waters containing iron, with the tincture of galls, ftrike a purple or black colour: fometimes the addition of a minute portion of alkali is neceffary to make this experiment fucceed. The trial with the phlogifticated alkali is a much more accurate and pleafing one; the waters which contain iron ftrike with this a beautiful Pruffian blue.

Iron being diffolved in waters by feveral different means, the fame tefts do not univerfally apply.

I. Waters which contain iron, diffolved by fixed air, drop it on the escape of the air;

<sup>\*</sup> The *pblogificated alkali*, or Pruffian alkali, as it is called, (becaufe of its ufe in the manufacture of Pruffian blue,) is made by the calcination of tartar and bullocks blood. The alkali of the tartar is thus charged with a quantity of inflammable matter, and is faid to be *phlogificated*.—For the explanation of the procefs by which Pruffian blue is formed. *Vid. Macquer's Distinguate de Chemie*.

hence the difficulty of transporting many of the chalybeate waters.

II. Waters containing iron, diffolved by the volatile fulphureous acid, likewife quickly lofe their properties by flanding, but at first are to be tried as the following, viz.

III. Those that contain a vitriol or iron, diffolved by the common vitriolic acid.

Martial vitriol in waters is afcertained,

1. By tincture of galls and phlogifticated alkali.

2. By alkali or calcarcous earth, which precipitate an ochre.

3. By evaporation; fome waters, as Hartfell, affording chrystals.

IV. Waters that contain iron, by means of an hepar fulphuris, are known,

1. By their fmell.

2. By their not depositing any ochre on the addition of alkali.

3. By their not affording chrystals on evaporation. V. Waters that contain iron by means of a calcareous hepar fulphuris;

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1. Afford no chrystals on evaporation.

2. Drop an ochre with cauftic (though not with mild alkali) it will have no fmell.

### To discover Copper in Waters.

1. THE Pruffian alkali precipitates copper, from waters containing it, in a red powder.

2. Volatile alkali, added to waters containing copper, gives them a fine blue or fapphire colour.

3. An iron wire immerfed in a cupreous water is tranfmuted into copper; the vitriolic acid, which keeps the copper diffolved in the water, having a greater attraction for the iron, takes that up, and deposits the copper in its place \*.

\* It is faid that the waters of Wisklow in Ireland contain fo much copper, that they obtain in this way a quantity fufficient to become an article of fale.

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#### To discover Zinc in Waters.

ZINC is precipitated from waters by the Pruffian alkali, in the form of a *white* powder.

By proper evaporation chrystals of white vitriol may be obtained, for it is in the form of a vitriol, or diffolved by the vitriolic acid, that this femi-metal is found in waters.

#### SECTION II.

#### I. ON THE BATH WATERS.

THESE hot fprings, fituated in the eaft end of the ifland, were first discovered about the year 1695, and soon after being found to be a powerful remedy in the cure of the dry belly-ache and some other prevailing diseases of the climate, they were purchased by the country, with the adjacent lands, for the public use; an hospital was sounded for the reception of poor people, and it was the object of the legislature, in order to make the place of the utmost possible benefit to the Island, to establish a township; accordingly commissioners were appointed, who were formed into a body corporate, vested with authority for granting of lands and making the neceffary laws and regulations refpecting the town and the bath. The public defign was much promoted by the zeal and liberal donations of fome private gentlemen, particularly of Peter Valette, Efq. (whofe public and benevolent character is too well known in this Island to require any Eulogium) and the place began to be visited, not merely on account of the falubrity of the waters, but as a fashionable sejour.

Nothing could have obstructed the progrefs of this town, or hindered the completion of the public plan, but the unfortunate political factions that prevailed in the country during its infant state; which destroying the harmony of private life, prevented the principal families from reforting here as formerly, and the place has fince that time fallen inte confiderable decline. The House of Assem-

bly has, neverthelefs, not fuffered the public to be deprived of the benefit of the waters by withholding any neceffary grants for the keeping up of roads and buildings to make the place be conveniently vifited by invalids; and the liberal fum they have lately given for these uses, it is hoped, will, by rendering the baths more convenient, and the place more agreeable, caufe a greater conflux of company to it. There are belides many other circumftances that concur in inviting people to the place, and in removing the objections that were formerly made against it; amongst which I shall first mention the advantage of a better road, in confequence of the late turnpike act; next to that of the change of of climate, which was formerly, in this neighbourhood, fo exceedingly rainy as to be hardly habitable: the quantity of rain which has been known to fall in a given time, (to perfons not acquainted with the Weft-Indies, and have not feen what are called the feasons,) would seem incredible,above 40 perpendicular inches have fallen in about the space of 6 or 8 hours, which is nearly double the quantity that, on a medium, falls in Great-Britain through a whole year. The progrefs of cultivation having occasioned the falling of the adjacent woods, has produced a great alteration in the state of the seafons, particular years excepted, as the last, which was a very wet one over almost the whole Island.

Laftly, I must not omit to mention the Botanic Garden instituted here about five years ago, which is already in a flourishing state, being stocked with a great variety of the most rare and useful plants collected from every quarter of the globe, and cannot therefore fail of furnishing out a great deal of entertainment not only to the cultivators of natural science, but to every one visiting the place.

### Analyfis of the Bath Waters.

THE waters (for there are feveral fprings) iffue from fundry clefts and fiffures in the rocks on the fide of a fmall river, called from thence the Sulphur River, whose fource is in that stupendous pile of mountains which run east and west through the Island: the several springs are fituated very near to each other, no others of the fame nature having been difcovered in the neighbourhood; in the parifh of Portland indeed, (on the oppofite fide of the great ridge) there is a finall one of the fame kind, but of weaker impregnation, and lower temperature.

There appears to be no difference in the waters ifluing from the feveral fprings, except in their temperature, which is in fome of them ten or twelve degrees lefs than in the principal one, which forms a current of nearly four inches diameter, and fupplies the baths on the oppofite fide of the river.

The water when received into a glafs at the fountain fide is perfectly transparent, but has a foctid fulphureous finell and tafte, which it retains in fome degree for feveral hours; this however ultimately flies off without any change or precipitation; and the water afterwards appears to be a very pure common water \*.

\* This water is extremely well fitted for bottling, and proves not inferior to Briftol water: The late Dr. M<sup>c</sup>Kenzie, who refided for fome time at this place, had fome of it by him in bettles for feveral years, which remained perfectly fweet and good. The immediate effects which follow on drinking the bath waters are, frequent eructations of wind, fometimes a degree of Vertigo or head-ache, and ficknefs at the ftomach; a copious flow of perfpiration and urine; after which an exhilaration of mind generally enfues, with an encrease of appetite, and at night natural reft.

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# GRAVITY.

THE specific gravity of the water appears to vary a good deal from that of common water, as will be seen from the following experiments; when taken from the spring it is indeed lighter, on account of its heat and other causes, than common water, but cooled to the same point it becomes heavier, from the earthy and saline matter with which it is impregnated.

# EXPERIMENTS.

1. AN hydrometrical gauge plunged into the Bath waters, taken as hot as possible from the spring, sunk 4 inches 5-10ths. 2. The above mentioned inftrument funk in common water heated to the fame point, (viz. about 125 deg.) only 3 inches 6-10ths.

3. The fame gauge funk in the Bath water, when cold, 1 inch 3-10ths.

4. The fame gauge funk in the river water 1 inch 5-10ths.

The Bath water is therefore to common water, when hot, as 45 to 36, when cold, as 13 to 15.

By weighing it as accurately as I could with a common beam, I found the gravity of the Bath water to exceed that of the river I drahm and 20 grains in the pint.

It feems extraordinary that the Bath waters fhould, when hot, be lighter than common water of the fame temperature, and when cold, heavier; I can only explain it from the phlogiston which it contains, many waters become heavier after a decomposition takes place in their component principles: The Buxton waters, e. gr. placed in an exhausted receiver, gave out no air bubbles,

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but turned whitish, and weighed 16 grains in the pint heavier immediately after, (vid. Monro on Mineral Waters).

#### TEMPERATURE.

THE temperature of the feveral fprings, fo far as my obfervations have reached, is uniformly the fame under all the differences of weather and variations of the atmosphere; though, if we may credit the experiments made by gentlemen fome years ago, the water appears to have been hotter than I have ever found it, and to have varied a few degrees in its temperature, at different times.

#### FAHRENHEIT.

Heat of the main fpring by my ther-

· mometer,	-	-	-	r27
Ditto of the low	v fpring,		-	124
Ditto of the up	per do.	-	-	114
Ditto of the high	ghest do.	• -	-	112
Ditto of the wa	ter runn	ing from	the	
guttering int	o the bat	ths, -	-	122
Ditto of the wa	ater brou	ght in a f	four	
gallon jug t	to the t	own, dif	tant	
nearly two n		-	-	118

It might be expected, that I should here explain the caufes of heat in these fprings, but the common hypothesis concerning the generation of hot waters is very unfatisfactory, and I shall not prefume to offer any of my own: It may be fufficient to obferve, that the heat of fome waters is entirely adventitious, not owing to any principles which they contain, but acquired by paffing through strata in the vicinity of volcanos or heated matters. Others appear to owe their heat either to fome union or decomposition in the foffile matters through which the waters pafs, and by which they become impregnated; the nature of this union or decomposition may be various, but has never been clearly understood, unless perhaps in the cafe of pyrites, on which the heat of the thermæ, or hot springs, is supposed most commonly to depend.

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### Experiments to ascertain the Contents of the Bath Waters.

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EXPERIMENT I. THE Bath water, placed in the exhausted receiver of the air pump, gave out very few air bubbles.

Having no air-pump I could not make this experiment myfelf; but it was tried fome years ago by a medical gentleman \*, who anylyfed the Moffat Waters in this neighbourhood, and I believe with fufficient accuracy.

EXPERIMENT II. A bottle filled with the Bath water immediately from the fountain, had an oiled bladder tied over the mouth of it, after which it was placed in a veffel of hot water : The heat of the water being encreafed to the boiling point, a fmall quantity of air was feparated, which was fecured by tying a ligature round the neck of the bladder, and afterwards made to pafs in

<sup>\*</sup> The account of this analysis is anonymous, but I am informed ed it was wrote by Dr. Mitchell, formerly a practitioner in Blue Mountain Valley.

a gentle stream through lime-water; but no precipitation enfued, as happened by breathing through the fame lime-water (vid. p. 30.)

The Bath waters have been fuppofed to contain a confiderable quantity of fixed air \*, but from what circumstance I cannot conceive, as they are totally defitute of that fparkling appearance and poignant taste that distinguish such waters, and the above experiments plainly demonstrate that they have no such impregnation.

EXPERIMENT III. A bottle accurately filled with the Bath water was inverted in a veffel fall of the fame water, fo as to allow no air to enter the bottle; the veffel with the water in this position was then placed over the fire, and the heat of the water encreased to the boiling point,—the water in the bottle was observed to descend about half an inch; but on suffering it to cool, it afcended again, so as to leave no vacuum in the bottle.

" Vid. Hiftory of Jamaica.

EXPERIMENT IV. A bottle filled with the water was closely stopped by a cork, which was perforated by a crooked tube, the other end of which entered an inverted phial filled with water, and placed in a vessel of water after the manner of Dr. *Priestly*: Being allowed thus to remain for 24 hours, there was no defcent of the water in the phial.

The two laft experiments ferve to fhew that the Bath waters contain no air of any kind, unlefs a fmall quantity of common or atmospheric air, which I apprehend was what was extricated in experiment II. and III.—It is hence doubtful if the Bath waters contain any phlogisticated or inflammable air as has been supposed.

EXPERIMENT V. Some of the fine blue nowers of a convolvulus were infufed for an hour or two in the Bath water without fuffering any change in their colour; on adding a fingle drop of the oil of vitriol, the colour of the flowers was immediately changed from a blue to a bright red. EXPERIMENT VI. Some drops of oil of tartar being added to a glass of the Bath water, no effervescence or other change enfued.

EXPERIMENT VII. Some drops of a folution of hepar fulphuris being added to a glass of the Bath water, no lactescence was produced, till I dropped in a little oil of vitriol.

EXPERIMENT VIII. Some drops of an unfaturated folution of lead in the nitrous acid, were added to a glass of Bath water \*, which effected no alteration.

EXPERIMENT IX. Some drops of a folution of mercury in the nitrous acid were added to a glafs of the Bath water, no yellow precipitation enfued +.

From these experiments (N° 5, 6, 7, 8, 9,) it is very apparent that the Bath waters contain no vitriolic acid (vid. page 31) in a separate state.

> \* See this experiment explained, page 34. † Ditto do. do. page 37.

EXPERIMENT X. Some drops of the oil of vitriol were added to a glass of the Bath water, no ebullition or effervescence ensued.

In the account given of the Bath waters in the hiftory of Jamaica, it is faid, that acids dropped into the Bath water caufe an ebullition; but it is certainly a miftake, and I believe it has arifen from the obferver not rightly diftinguishing between an ebullition and the difficult mixture of the acid and the water; the acid, being fo very ponderous, on being dropped into the water falls through it, and caufes an appearance that might be miftaken for an effervescence, by perfons not properly acquainted with the fubject.

EXPERIMENT XI. A folution of fal martis being dropped into the Bath water, no ochrous precipitation followed, (vid. p. 35, fec. 4.)

EXPERIMENT XII. Some drops of a nicely faturated folution of chalk in the nitrous acid were added to a glafs of Bath water, but no precipitation enfued, (vid. p. 35, fec. 3.) These experiments, (N° 10, 11, 12,) clearly shew that there is no alkali in the Bath water, (vid. p. 35.) but less the alkali might be in too diffused a state to be discovered by the above tests, I repeated the experiments on the water after considerable evaporation, and the result was the same.

EXPERIMENT XIII. To the Bath waters I added a fmall portion of alkaline falt, and obferved no change, but the fame experiment being tried, after evaporating the water a little, a white precipitation followed, (vid. p. 14.)

EXPERIMENT XIV. A few drops of a faturated folution of lead in the nitrous acid, added to the Bath water, caufed a milkinefs, (vid. p. 14.)

The fame effect followed on adding a clear folution of faccharum faturni.

EXPERIMENT 15. The water during evaporation, with a very gentle heat, deposited a quantity of reddish brown earth, which I separated by decanting the liquor from time to time as it collected at the bottom of the vessel. EXPERIMENT XVI. To the earthy precipitate obtained in the preceding experiment, I added fome diffilled water, but on ly a part of it diffolved.

EXPERIMENT XVII. To the foregoing precipitate repeatedly washed in fresh quantities of distilled water, I added some drops of the oil of vitriol, which caused a flight effervescence; but great part of it remained undiffolved, which was also infoluble in the nitrous acid, and with difficulty soluble in water.

EXPERIMENT XVIII. After further evaporation of the water I placed it at reft for 12 hours, and then found it covered with thin laminous chryftals like the fcales of fifnes, which being taken off and put in diftilled water, great part of them continued to float therein, remaining undiffolved for fome days: the tafte of these chryftals was rough and earthy, not faline. These are the properties of the selenitic compound (vid. p. 8); which the following experiment more clearly demonstrate them to have been. EXPERIMENT XIX. To a folution of the chryftals (exp. 18) in diffilled water, I added fome drops of the folution of mercury in the nitrous acid, which caufed a yellow precipitation (vid. p. 37, fec. 5.)

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EXPERIMENT XX. That portion of the earth which was diffolved by the vitriolic acid in experiment 15, was precipitated again by adding a few drops of oil of tartar. I washed the precipitate in distilled water, and then re-diffolved it in fresh vitriolic acid; the liquor formed thereby was bitter and faline, refembling that of epfom falts.

I repeated this folution and precipitation feveral times, fo as to leave no doubt of the earth being a magnefia.

The foregoing experiments made on the earth precipitated during evaporation, were all repeated on the refiduum left after evaporation to drynefs, with nearly the fame refult, only that I found it difficult to obtain the earth entirely pure, fome of the faline matter adhering, notwithftanding repeated affusions of diftilled water. Magnefia in waters is generally combined with the vitriolic acid, conftituting epfom falts; but as there are no figns of vitriolic acid in this water, unlefs in the felenitic compound, it is not improbable that the magnefia is blended with the fea falt, which this water appears, from the fubfequent experiments, to contain, forming with the muriatic acid a fixed ammoniac (vid. p. 8. fec. 3.)

EXPERIMENT XXI. On dropping a folution of filver in the nitrous acid into the Bath water, a white curdled precipitation immediately took place, (vid. p. 38. fec. 4.)

EXPERIMENT XXII. The precipitate in the preceding experiment, with a little tartar, being rubbed on polifhed brafs, gave it, by the affiftance of heat, a filver colour \*.

Having these evident proofs, (experiment 20, 19,) of the presence of muriatic falt, I endeavoured to obtain it in its chrystallized state.

EXPERIMENT XXIII. A quantity of the refiduum was diffolved in diffilled water,

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<sup>\*</sup> This precipitate called a Luna Cornea is the fubitance made afe of by artifts for fome kinds of filvering.

and the liquor decanted clear from the earthy fediment, which was then flowly evaporated, first by the fire, and then by the fun. Regular cubical chrystals were thus obtained, having all the properties of common falt, (vid. p. 37.)

EXPERIMENT XXIII. A portion of the refiduum (experiment 15,) being thrown on a hot iron, gave out no fumes or fmell of fulphur.

EXPERIMENT XXIV. The incrustation adhering to the rocks about the fpring was also burnt, but yielded no proofs of its containing any fulphur.

EXPERIMENT XXV. A new coined bright dollar was placed in the water at the place of its iffue, which in a few minutes was tarnifhed of a deep purple colour,—a bright piece of gold coin was turned of a deep copper colour. The fame trials being made at the extremity of the gutturing, where the water is difcharged into the Baths, the metals were but flightly tarnifhed after lying a confiderable time, EXPERIMENT XXVI. Having made certain characters on paper with the fympathetic ink, (*i. e.* a folution of faccharum faturni in water) I tied the paper over a broad mouthed bottle filled with the Bath water, the vapours of which foon rendered the characters vifible.

EXPERIMENT XXVII. Saccharum faturni diffolved in the Bath water caufes no red or brown precipitation.

These experiments (N° 23 and 24) prove that the Bath waters contain no actual fulphur (vid. p. 39); but the following ones (N° 25 and 20) evidently shew the prefence of the phlogiston in some form (vid. p. 40): Why experiment 27 does not succeed I cannot tell, unless from the phlogistic principle being so very volatile as not to remain long enough to produce the effect.

The fætor of these waters indicate the presence of a hepar sulphuris (vid. p. 40) > but this is rendered doubtful by the experiments made with acids (vid. experiment 10) which cause no precipitation, and others such that the waters contain no alkali. From what has been faid (vid. p. 41) concerning hepar fulphuris with quick-lime, it is obvious that fulphur cannot be fufpected in the Bath waters, in this form.

In what flate of combination then, are we to fuppofe the phlogifton exiftent in thefe waters? According to the late difcoveries, refpecting the various fpecies of phlogifticated air, it feemed exceedingly probable that it was in fome fuch form : This opinion feems, however, overturned by experiment 2d and 3d, in which I found it impoffible to obtain any air that was permanent.

The mode of combination must still remain a defideratum, as chemistry has not, that I know of, found out any other methods of trial for afcertaining it.

Before I difinifs this fubject I must not omit taking notice of Dr. Brown's hypothefis concerning these waters, (vid. the Natural History of Jamaica) which is, that they contain the volatile vitriolic acid connected with a calcareous earth: they certainly contain Jelenites, but this affords no explanation refpecting the phlogifton. Befides, how the volatile acid can be fuppofed combined with the calcareous earth, I do not conceive; for felenites is one of the moft fixed falts we know of. As he has taken no notice of any of the other contents of thefe waters, which were very obvious on the flighteft trial, I fufpect that he never made any experiment on them, but fpoke merely from conjecture.

EXPERIMENT XXVIII. Into a glafs of the Bath water I dropped a fmall quantity of the tincture of galls in brandy; no apparent change took place after ftanding fome time, till I added a folution of iron, when it immediately affumed a deep purple colour.

EXPERIMENT XXIX. I repeated the foregoing experiment, ufing, inftead of the galls, the phlogifticated alkali; but no effect followed till I added the iron, when the ufual beautiful blue precipitation immediately took place (vid. p. 45.)

It is hence (viz. from N° 28 and 29) obvious, that these waters have no metallic impregnation, though they have been generally fuppofed to contain fome iron; and Mr. L—— (vid. Hift. Jamaica) fays, that there is an ochrey precipitation about the rocks over which the waters run. The yellow flimy incrustation about the rocks may have fomething of the appearance of an ochre, but from the following experiment it does not feem to contain any iron.

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EXPERIMENT XXX. I took a quantity of the above mentioned flimy matter, adhering about the rocks where the fpring is, which having first dried in the fun, I mixed with oil and charcoal dust, and placed it in the strong heat of a furnace for some hours; a blackish gritty powder was found in the crucible, which was not in the least affected by the magnet,

EXPERIMENT XXXI. I then fubjected it to the fire without any admixture, fufpecting that it was of a vegetable origin, and would yield a pot-afh; after keeping it for a confiderable time in a very violent heat, I obtained a powder refembling brick duft, reddifh and gritty; a part of this powder effervesced with acids and diffolved, but most of it retained its gritty form. Pot-ash being at first in a caustic state, this experiment must not be tried immediately after removing it from the furnace, unless you throw in fixed air to the water you mix it with.

I infer from the foregoing experiment, (N° 31) that the flimy matter of the rock is partly of a vegetable nature, produced by the putrefaction of leaves, &c. falling into the water, and partly ftoney, fome particles of fand and rock being entangled in it.

EXPERIMENT XXXII. On the Rock.-The rock being kept for fome time in a ftrong heat, was partly converted into quicklime.

EXPERIMENT XXXIII. The rock powdered and mixed with charcoal, &c. as in experiment 30,—fhewed no figns of iron; fome of the iron ore in the neighbourhood being tried in the fame manner, was afterwards attracted very ftrongly by the magnet. Having thus related and explained my feveral experiments made on the Bath waters, I shall next, after summing up the results in a general table, proceed to shew their effects on the human body, and their cure in difeases.

### Table of Experiments and Refults.

EXPERIMENTS.

RESULTS.

)	Sie	No fixed air.
	ew t	No inflammable a
	hat	No separate acid.
-	the	No separate alka.
6,	ВА	
	TH	A selenetic salt.

Earth of magnefia. Common marine falt. No actual fulphur. Agood deal of phlogiston No hepar fulphuris.

Nº 1, 2, 3, -	She
— I, 2, 3, 4, -	ew th
- 5, 6, 7, 8, 9,	hat t
- IO, II, I2, -	the ]
- 13, 14, 15, 16,	BATH
17, 18, -	
- 20,	WAT
- 21, 22,	m m
- 23, 24,	s co
- 25, 26,	ntai
10, 11, 12, -	1 - [

( 7.2 )

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#### CHAPTER IV.

#### On the Effects and Uses of the Bath Waters.

TAVING learned by analysis the mat-I ters contained in mineral waters, we can fometimes with certainty determine, a priori, what will be their effect on the human body; but not univerfally; for notwithstanding the feveral principles of a water may be clearly detected, yet their mode of combination being unknown, their uses can only be conjectured, till experience has afcertained them. The Somerfetshire Bath Waters, e.g. contain, according to the lateft and beft analyfis by Dr. Faulkener and others, a little common falt, some bepar sulphuris with quick-lime, a minute portion of iron, (viz. only one 37th of a grain in a pint of the water) some selenites and fixed air. However active these principles may be, I apprehend it will be impoffible, from the fmall proportion they contain of each, to account fully for their extensive influence in the cure of difeases. There will be as much difficulty in accounting fatisfactorily for the operation

of our waters; yet from comparing their effects together, their general action feems to be *ftimulant*; which power they owe chiefly to the fulphureous gas, or phlogistic principle, the fixed contents being in too fmall a quantity to be productive of any material operation.

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That the Bath waters act as a ftimulant is very obvious from the immediate effects which the drinking of them produce, viz. eructations of wind, vertigo, head-ache, and fever,  $\mathfrak{Cc}$ .  $\mathfrak{Cc}$ . (viz. p. 53) as well as from the nature of feveral difeafes in which they are found remedial.

Dr. Brown, in his Natural Hiftory of the Ifland, fays of thefe waters, " that they are " remarkably beneficial in all capillary ob-" ftructions or diforders proceeding from " weaknefs or the want of proper glandular " fecretions; in all lentors or vifcidities pro-" ceeding from the inaction of the folid fyf-" tem; in confumptions, nervous fpafms " and weakneffes: They reftore the appetite " and ufual action of the vifcera, invigorate " the circulation, warm the juices, open the " fkin and urinary paffages, ftrengthen the " nerves, and feldom fail of producing eafy " fleep at night.

Mr. L \* \* \* \*, in his well wrote Hiftory of the Island, gives nearly the fame account of these waters;—He fays, " they " excite appetite, promote urine, produce " fleep, cure ulcers, strengthen the nerves, " and cure palsey.

These effects, most of which I have seen verified in a great number of cases, evidently indicate a stimulant power. 1stly, The expulsion of flatulencies, increase of appetite,  $\mathfrak{S}_c$ , shew that the water excites the action of the stomach. 2dly, The vertigo, head-ache, and sever, which they sometimes induce, are consequent on an encreased circulation. 3dly, Their diuretic effects, in like manner, proceed from their stimulant action on the kidneys: their operation in this respect, is in part owing to the quantity taken in as a diluent, but they have certainly a further effect than as a mere diluent; for in some cafes \* the quantity of urine discharged has greatly exceeded the whole quantity of fluid made use of. 4thly, An increase of perspiration may be variously accounted for; 1stly, the waters may prove diaphoretic from their action in the stomach, between which and the extreme vessels of the body there is a particular consent: 2dly, They must be diaphorctic as being diluent; but besides, they may, 3dly, from their sulphureous contents, act as a stimulant in the extreme capillary vessels.

Whether the Bath waters may not alfo have fome *alterant* and *antifpafmodic* power, I fhall not determine, but it would feem probable that they are poffeffed of both.

Having faid thus much to explain the general nature and action of the Bath Waters, I shall now confider their uses in particular difeases.

I. The Dry Belly-ache feems to have been the complaint in which the good effects of the Bath waters were first most eminently ex-

<sup>\*</sup> I am obliged to my friend Dr. Clarke for a very particular cafe of this nature.

perienced, and they have ever fince been applied to as a most fovereign remedy against the paralytic torpor and relaxation that fupervene to the excruciating fpafms which by turns affect the bowels and limbs of people in this horrid difease. I must not in this place enter largely into the confideration of either the causes or cure of this complaint; but whatever it depends on, whether on the action of any poifon taken in (as of lead, acid, ardent spirit, &c.) on an acrimonious bile in the first passages, or on a constriction of the furface from cold, concurring with an irritable state of the bowels, it is obvious, that, after removing the conftipation, the principal indication of cure is to reftore the lost tone. As spasm, or a violent state of contraction in the muscular fibres, produces the alternate state of atonia or relaxation, fo vice versa, atonia gives occasion to the return of spafm: Accordingly we see that patients, once affected with this diforder, are fubject to frequent relapses, which nothing can fo effectually prevent, as exciting the action of the inteftinal canal. With this view various medicines are employed, but the

Bath waters have been found, of all others, the most useful remedy: Their effects in such cases seem almost miraculous.

The use of the Hot Baths jointly with the waters, 1st, by the grateful stimulus they impart to the nervous system, and, 2d, by causing a free determination to the surface, contribute greatly inremoving the torpor and palsey of the extremities.

II. Genuine Palfey, or Hæmiblegia, is a a diforder totally of a different nature from the preceding. There is in this cafe an affection of the brain, or of the nerves in their origin; but notwithstanding fuch a diffimilitude, the Bath waters may, in conjunction with other remedies, prove extremely beneficial. A remarkable inftance has lately occurred of the efficacy of the Bath waters in paralytic diforders, -- A gentleman under recovery from a long fever, by exposure to cold air, fuddenly loft the use of all his limbs ; Having tried the use of several remedies with little effect, he had recourse to the Bath waters, by which he was in a few weeks perfectly recovered.

III. The use of the Bath waters is highly conducive to convalescents of every class, efpecially after fevers: By exciting appetite and invigorating the bowels, it restores the tone of the whole softem, and obviates a tendency to relapse.

IV. The effects which the Bath waters have on the nervous fystem, and in affisting the functions of the stomach, shew them to be a very fit remedy in all *bysterical* and *fexual* complaints.

V. I shall next confider how far the Bath waters are useful in visceral obstructions, or in complaints of the liver and spleen; patients labouring under these complaints have frequently had recourse to these waters, and, I believe, have not always been disappointed of receiving benefit, but I nevertheless think that in such complaints they are rather an ambiguous remedy: So far as they strengthen the functions of the stomach and the bowels they may be useful, but from their stimulant or heating quality, they may sentens do hurt by exciting inflammation. Persons who have undergone a mercurial regimen for the removal of these complaints, may, however, derive great advantage from the waters: The

danger of topical inflammation being removed, the use of them will afterwards contribute much to their recovery. This leads me to speak of their effects after falivation in other diforders.

VI. Mercury, though the only certain remedy which fome difeafes admit of, is neverthelefs an hazardous one; under the most careful management it fometimes makes great depredations in the conftitution, particularly in warm climates, where the tone of the fystem being once impaired, is difficultly recovered. In this view, and for removing fome fymptoms, which mercury of itfelf is not a cure for, the Bath waters are highly beneficial in the lues venerea: They ferve to fupport the general health, and perhaps affift in washing out the venereal virus. With the Hot Baths they are of great use in relieving the nocturnal pains, cleaning and curing the ulcers, eruptions, Bc. Bc.

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They have in fome inftances difcovered a fingular efficacy in ftopping old gleets \*. They may operate in two ways in the cure of fuch complaints; 1ft, as a diluent in wafhing out fome latent virus; or, 2dly, by their tonic power in the ftomach, and in bracing up the whole fyftem. It is, however, immaterial how they act, if they produce the effect, which is unqueftionable: I have no doubt of their proving equally fo in the *fluor albus*.

VII. The Bath waters are not only ferviceable in the lues venerea, but alfo in that more loathfome diftemper of negroes, the yaws: By the drinking of the waters, and the ufe of the hot baths, the eruption is thrown out more plentifully on the furface of the body, and the matter eliminated; by which the internal fyftem feems to be relieved, and the cure expedited. Nothing can be more pernicious than the general practice of negroes in this diforder, I mean

\* Dr. Irvine, a gentleman univerfally known for his many virtues, as well as by his long refidence in this neighbourhood, has favoured me with a very fingular cafe of this kind, where a perfon was cured by the Bath waters of a gleet of eleven years ftanding.-I know of fome other cafes almost as fingular. that of washing in the cold rivers. The furface being by that means constringed, the yaws are repelled, which I believe is frequently the cause of subsequent bone-ache and distortion.

VIII. Every kind of cutaneous diftemper may receive advantage from the Bath waters. An inveterate *leprofy* was relieved by it; but I apprehend that in cutaneous defœdations there may be found waters ftill more ufeful than those of the Bath. Their use in absterging and drying up foul ulcers, is fo well known I shall fay nothing on the subject.

IX. Dr. Brown mentions confumption among the difeafes in which the Bath waters are ufeful; but if he means a confumption of the lungs, I fhould fear he is miftaken: Confidering the degree of phlogiftic diathefis, or inflammatory difpofition that prevails in this difeafe, I conceive the Bath waters may rather tend to aggravate than relieve the fymptoms; but I have no experience, and have not been able to collect any facts on the fubject. X. I fhould not either fuppofe the Bath waters indicated in *dropfy*, but a cure in that diforder has been obtained by fuch various and even oppofite means, that I fhould not think the trial of them improper where other remedies have failed. The Bath waters being poffeffed of an active principle, capable of producing material changes in the fyftem, may accidentally excite the action of the abforbent veffels, and caufe the waters to be carried off by the different outlets of the body:

XI. In the *jaundice* they are useful after the obstruction in the biliary ducts is removed, for carrying off the bile in the circulation, and for strengthening the impaired functions of digestion,  $\Im c$ .

XII. The diuretic effect of the Bath waters make them ufeful in *nephritic* complaints, in which they operate both as a diluent and by a fpecific quality without much ftimulus. In fuch *nephritic* affections as depend on a gouty diathefis, the Bath waters are calculated to anfwer extremely well; which will be obvious from what I have to remark on their effects in the gout.

XIII. The gout is a difease depending not on a morbid matter, as has been commonly supposed by physicians as well as others, but on a certain state of the moving fibres, which belongs to a particular temperament or constitution that is frequently hereditary, being transmitted from father to fon in the fame way as a likeness of features and other constitutional peculiarities : Hence there is no cure to be obtained in the gout but by obviating the effects of this particular temperament, which is only done in one way, viz. by exercise and abstinence from animal food. The difease, as an articular affection, seems falutary, tending to preferve the conftitution; but when it becomes irregular and mifplaced, it is then dangerous and deftructive. Every cafe of irregular and mifplaced gout probably depends on the want of a neceffary degree of tone in the ftomach, to determine the affection to the extremities. We from hence fee how the Bath waters are ferviceable in the gout. The waters and hot

baths, by exciting the action of the ftomach, and of the extreme veffels, call back the gout to the joints, after being repelled, or determine it there, when wandering about the body.

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When the conftitution is much debilitated, and the gout occasions affections of the head and stomach particularly, instead of producing inflammation in the joints, it is called *atonic*; that is, from the want of *tone* the difease which should be seated in the extremities seizes on other parts, or is not able to form in the joints. In cases of this nature the Bath waters are very beneficial.

The inhabitants of the West-Indies make frequent voyages to Great-Britain on account of health when perhaps they have a remedy nearer at hand. In the atonic gout Dr. *Cullen* recommends a warm climate, and I therefore embrace the opportunity of fuggesting to Jamaica gentlemen residing in Great-Britain, that under this discase they have in their own Island a climate more propitious, and in the Bath waters a remedy not less efficacious, and which deferved to be no lefs celebrated that those of England.

I might go on to illustrate the effects of the Bath waters in a variety of other difeases, but I shall take notice of only one more, viz.

XV. Chronical rheumatism, which fignifies, habitual pains affecting particularly the large joints of the body without much fever, and fometimes with little fwelling, but with debility and coldness of the affected part. Rheumatic affections are the complaints of people in cold climates; but the inhabitants of the West-Indies are not exempt from them. Exposition to rain, night dews, and the north winds that blow at a particular feafon of the year, occafion fuch diforders, that are apt, notwithftanding every care and attention to medicine, to return and become chronical. The Bath waters stand very much recommended in fuch cafes, though at first they feem to aggravate the difeafe, the pains being for fome time exafperated : By perfeverance, however, the patient is fune of obtaining relief.

Having, as I prefume, faid enough to fhew the general utility of these Bath waters, and to direct to the use and application of them; nothing remains but to make a few remarks on the method of using them.

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Rules for the drinking of mineral waters can only be general; they are nevertheless of use. The following ones are such as I think more particularly require attention.

I. To begin the use of the waters with fmall draughts, which may be repeated at short intervals, as the fudden ingurgitation of a large quantity may fometimes cause uneasines and bad effects, (vid. p. 53.)

II. To use now and then a gentle laxative as occasion may require, to prevent a conftipated state of the bowels.

III. To drink them as nearly to the fountain or place where they iffue, on account of the extreme volatility of their medicinal principle, which is fo remarkable that it is comparatively of little use to drink them at any distance; they lose even in their transportation across the river \*, (vid. p. 66, sec. 4.)

IV. To continue the use of them for a due length of time.

The character of any medicine being eftablifhed, people are commonly difpofed to expect too much from it. There are inftances where the effects of the Bath waters have been manifefted very fuddenly, but in fome chronical cafes it is requifite they fhould be perfifted in for fome length of time: Patients, after having defpaired of obtaining any relief, have, by perfeverance, obtained a perfect recovery from the most inveterate difeafes.

V. To use the hot baths in the afternoon rather than in the morning, as they tend to procure ease and fleep: Care should be nevertheless taken to avoid cold on returning from the Baths, and they ought not to be used too soon after meals.

The inconvenience which at prefent attends getting to the fpring will be foon removed by a bridge that is to be built across the river.

VI. Laftly, as to regimen.—I know of no particular one required by the Bath waters. It has been cuftomary to avoid wine and fruits, not as I apprehend, from any difagreement thefe have with the waters, but becaufe they have not been fuited to the complaints for which the waters were drunk : A little wine may be fometimes very neceffary, but in general it is better totally to relinquifh it, as the precife limits are difficult to be afcertained, and as it may interfere with the ufe of the waters.

That I may omit nothing that can be of use in directing to the application of these waters, I shall here subjoin a catalogue of all such diseases as they are likely to prove beneficial in.

" Quot et quam diversi morbi curentur, thermarum et aquarum medicatarum usu! Ad bas confugere toties coquntur ægri, decantissima alia remedia experti absque ullo fructu, &c. In Thermis, aqua nativo calore laxat omnia & emollit, venis bibulis cutaneis se insinuat, sanguini permiscetur,

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« obstructa loca alluit; et si potentur simuliz " salubres illæ aquæ, tutum et potens habetur. " remedium, &cc. &cc." Vid. V. Swieten Comm. Tom. III. page 345.

### CATALOGUE of DISEASES,

in which the Bath Waters are useful.

A BORTION, (viz. after) Anorexia, or want of Deafnefs. appetite. Aithma, 1. Nervous, 2. Gouty. Cancer. Chorea, Sp. Vit. and Epilepfy. Cholera morbus, (viz. after) Convultions in general to obviate a return Colic, viz. 1. Belly-ache, 2. Hy-Sterical, 3. Gouty. Confumption, (doubt- Fluor albus. ful.)

Crapula, or after furfeits. Dyfury, or pain and difficulty in paffing. urine. Elephantiafis. Fevers, (viz. when recovering from, and of, particularly, nervous fevers.) Flatulencies. Gleets.

#### Green ficknels. Gutta ferena. Gout, viz. the atonic and irregular. Herpes, impetigo, &c. Hypochondria, and Hysterics. laundice. Ifchias, or hip-gout Ifchury, ftoppage of urine. Lethargy. Leprofy. Lues venerea, vid. p. Lumbago, or rheumatifm in the loins. Menfes (ad movendas) Nephritis, 1. From Stone, 2. From Gout.

Palpitations. Palfey, vid. p. Poifons, (for recovery after) vid. Rheumatifm, р. Ricketts: Salivation, (after) Sterility. Stone in the kidneys or bladder. Strangury. Stupor. Suppreffion of urine. Tendons contracted. Tetanus, locked jaw. Torpor. Vertigo. Ulcers, mali moris.

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## CATALOGUE of PATIENTS who were either cured or greatly relieved by the drinking of the Bath Waters.

[Taken from a Memorandum-Book of P. Valette, Efq. kept by him for that purpose before any Surgeon or Physician was appointed to the faid Bath.]

MR. B. Harris, cured of —— Mr. G. Galbraith, of rheumatic pains. Mr. M'Queen, of a lingering fever. Mr. Rofs, of ditto.

Mr. James Wation, of a dry belly-ache .---

N. B. He was eased by the first draught of the water.

Mr. Alexander Grant, of the gravel.

John Gale, Efquire, of a lingering fever and lofs of appetite.

\* William Hicks, Efq. of a fwelling in the fpleen.

John Gofling, of an intermittent.

John Robertson, of ditto.

Charles Anfell, of a dry belly-ache.

Charles Webb, of a lingering fever.

Charles M'Farquhar, of a dry belly-ache.

Mr. Renny, of a lingering fever.

- Miss Barry, of a belly-ache with palfey of the limbs.
- Miss Gardiner, of a dry belly-ache and great weakness in the limbs.

Mrs. Puley, of a lingering fever and pain in the ftomach.

Mr. Joseph Morris, of a dry belly-ache. Mrs. Kilby, of loss of appetite.

T. B. Ruffell, of a dry belly-ache.

Mr. Clarke, of the gravel.

John Puley, Efq. of a lingering fever.

Mr. T. Fox, of the dry belly-ache and gravel. Thomas Allan, of the dry belly-ache and confumption.

\* Mr. Cargil, of the fciatica in four weeks. Peter Carr, of ditto.

Mrs. Nelfon, of ditto.

- Mrs. Forbes, of the hysterics, and loss of appetite and fleep.
- \* Capt. Jof. Lawrence, in one week greatly relieved of a fwelling of the fpleen and venereal ftrangury.

Mr. Maitland, of the rheumatifm.

Capt. J. Lawrence, of the gleet, strangury, &c. Mrs. Garrioch, of a swelling of the spleen-

Twice diffipated, and intermittent fever.

Mrs. M'Farquhar, of a belly-ache and weaknefs of the limbs.

\* Mr. Alex. M'Farlane, of the fciatica.

Colonel Price, of a lingering fever, lofs of appetite and fleep,

Mr. S. Woolery, of a complication of diftempers.

Mr. Lawrence Brodbelt, of the gravel, when given over by the faculty and his friends. Mr. Golding, (of Withywood) of the gravel. Captain Simon Booth, of a fwimming in his head, lofs of appetite, &c.—Was eight different times relieved of the above complaints.

\* Daniel Hanmer, Esq. of a violent rheumatic complaint in 23 days time.

Thomas Robertson, of an intermittent fever of 14 months standing, by 3 weeks use of the water.

- Mr. Drower, of a fever and belly-ache in one week.
- James Craddock, of the gravel when reduced extremely low. N. B. A great deal of gravel difcharged.

Christopher Terry, of the gravel, perfectly recovered in 5 weeks.

- William Gordon, (Clarendon) of lownefs of fpirits and depraved appetite.
- Mr. Yanckey, (Guanaboa) of ditto.
- \* Alex. Low, Bricklayer, of the gravel and violent pain in 4 days.

N. B. Those marked with an asterism \*, did not stay long enough to obtain a perfect cure, but were greatly relieved.

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# CATALOGUE of PLANTS

the most remarkable for Use or Beauty, which are in the Botanical Garden at Bath.

Aurus, Cinnamomum, The Cinnamon tree 1 -----, Saffafras, Saffafras tree \_\_\_\_, Campbora, Camphor tree Garcinia, Mangosteena, Mangosteen Mangifera, Indica, Mangoe tree Cycas, Circinnalis, Sagoe palm Gum Arabic tree Mimofa, Nilotica, Candleberry myrtle Myrica, Cerifera, E. Ind. Tallow tree Croton, Sebiferum, Peach tree Amygdalus, Perfica, Almond tree \_\_\_\_, Communis, Juniperus, Bermudiensis, B. Juniper or cedar Chinefe Arbor vitæ Orientalis, Thuya Manna Afh Fraxinus, Ornus, Cupresius, Sempervirens, Upright Cypress Cape Jeffamine Gardenia, Florida, Magnolia, Grandiflora, Laurel leaf tulip tree Scotch Lilac tree Lilac, Syringa, Sarsaparilla, Sarsaparilla Smilax, Vanilla EpidendrumVanilla, Hedyfarum, Movens, Moving plant Hibifcus, Rosa Sinensis, Chinese role.

N. B. There are a great many others no lefs ufeful or beautiful, but being more common it was not thought neceffary to enumerate them.

There are also feveral curious and unknown East-India plants, prefented by (the never to be forgotten friend of this Island) Lord Rodney, amongst which is a most elegant palm, supposed to be the walking cane.

FINIS.

Control States

Epid . dram Fanillas

Hibileus, Rola Sinenfu,

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