# Copy of a printed diagram referenced as "Representation of the wave function for the ground state of a hydrogen atom"

## **Contributors**

Fuller, Watson, 1935-

## **Publication/Creation**

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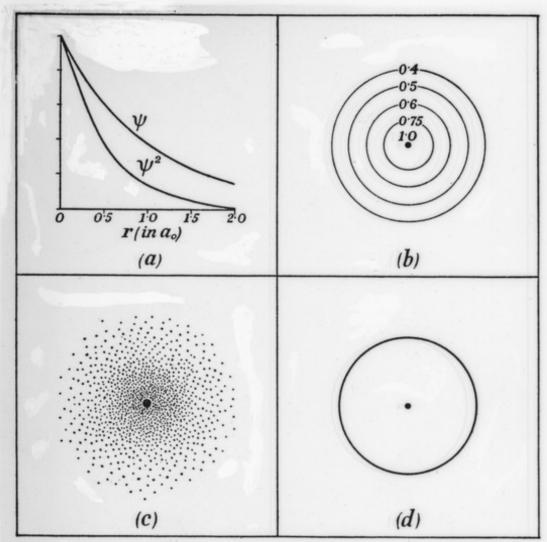
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quent use of it.

The senting the charge density which is often used for atoms. Instead of plotting the density



Representation of the wave function for the ground state of a hydrogen atom.

(a) graph of  $\psi$  and  $\psi^2$ .

(b) contours of  $\psi$ .

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winds are a

- (c) charge-cloud.
- (d) boundary surface.

re plot what is called the radial density  $4\pi r^2 \rho(r)$ . Since is the volume lying between the two spheres r, r+dr, it follows that  $4\pi r^2 \rho(r) dr$  is the total probability that the electron is at a distance between r and r+dr of the origin. Fig. 2 shows the graph of this r dial density. It is interesting, by way of a